

ASSAM DOWN TOWN UNIVERSITY

Curriculum and Syllabus

Bachelor of Pharmacy

OUTCOME BASED EDUCATION FRAMEWORK

CHOICE BASED CREDIT SYSTEM

Version: 1.01

FACULTY OF PHARMACEUTICAL SCIENCE

July, 2023

PREAMBLE

Assam down town University is a premier higher educational institution which offers Bachelor, Master, and Ph.D. degree programs across various faculties. These programs, collectively embodies the vision and mission of the university. All the programs offered by the Faculty of Pharmaceutical Science of Assam down town University strictly follow the curriculum approved by the Pharmacy Council of India (PCI), the statutory body responsible for regulating the profession of pharmacy in India. This document contains outline of teaching and learning framework and complete detailing of the courses. This document is a guidebook for the students to choose desired courses for completing the program and to be eligible for the degree. This volume also includes the prescribed literature, study materials, texts, and reference books under different courses as guidance for the students to follow.

Recommended by the Board of Studies (BOS) meeting of the Faculty of Pharmaceutical Science held on dated 08/07/2023 and approved by the Emergent Academic Council(AC) meeting held on dated 28/07/2023

Chairperson, Board of Studies

Member Secretary, Academic Council

Vision

To become a Globally Recognized University from North Eastern Region of India, Dedicated to the Holistic Development of Students and Making Society Better

Missions

- 1. Creation of curricula that address the local, regional, national, and international needs of graduates, providing them with diverse and well-rounded education.
- 2. Build a diverse student body from various socio-economic backgrounds, provide exceptional value-based education, and foster holistic personal development, strong academic careers, and confidence.
- 3. Achieve high placement success by offering students skill-based, innovative education and strong industry connections.
- 4. Become the premier destination of young people, desirous of becoming future professional leaders through multidisciplinary learning and serving society better.
- 5. Create a highly inspiring intellectual environment for exceptional learners, empowering them to aspire to join internationally acclaimed institutions and contribute to global efforts in addressing critical issues, such as sustainable development, Climate mitigation and fostering a conflict-free global society.
- 6. To be renowned for creating new knowledge through high quality interdisciplinary research for betterment of society.
- 7. Become a key hub for the growth and excellence of AdtU's stakeholders including educators, researchers and innovators
- 8. Adapt to the evolving needs and changing realities of our students and community by incorporating national and global perspectives, while ensuring our actions are in harmony with our foundational values and objectives of serving the community.

Programme Overview

B. Pharm programme designed to enrich students' basic and advanced knowledge in the Pharmaceutical Science domain, the programme follows the courses mandated by Pharmacy Council of India (PCI) education regulations. The semester-wise course sequence and the entire B. Pharm curricula have been arranged to provide hands-on training and real-world exposure to traditional and modern practices, making graduates industry-ready. As pharmacists are true drug experts, B. Pharm students are exposed to allied science courses and core pharmaceutical courses, fostering their aptitude for research and advancements in new drug development technologies. Rules & Syllabus for the Bachelor of Pharmacy (B. Pharm) Course framed under Regulation 6, 7 & 8 of the Bachelor of Pharmacy (B. Pharm) course regulations 2014 as per by Pharmacy Council of India (PCI).

Duration of the course-

The course of study for B. Pharm shall extend over a period of eight semesters (four academic years) and six semesters (three academic ears) for lateral entry students.

I. Specific Features of the Curriculum

The BPharm curriculum is designed to align with the evolving needs of the pharmacy field and society at large. It offers a comprehensive blend of theoretical knowledge and practical applications essential for a profound understanding of pharmaceuticals, fostering the development of a wide array of skills. Encompassing vital subjects such as Pharmaceutical Chemistry, Pharmacology, Pharmaceutics, Pharmaceutical Analysis, Human Anatomy, Clinical Pharmacy, Pharmacognosy, and Pharmaceutical Microbiology, the B. Pharm program ensures a robust coverage of core topics pivotal to pharmacy education. This curriculum is thoughtfully designed to equip students with both theoretical acumen and hands-on proficiency, catering precisely to the requirements of the dynamic industry and the broader societal demands.

II. ELIGIBILITY Criteria:

First year B.Pharm: 10+2

Candidate shall have passed 10+2 examination conducted by the respective state/central government authorities recognized as equivalent to examination by the Association of Indian Universities (AIU) with English as one of the subjects and Physics, Chemistry, Mathematics (P.C.M) and or Biology (P.C.B /P.C.M.B.) as optional subjects individually. Any other qualification approved by the Pharmacy Council of India as equivalent to any of the above examinations.

B.Pharm lateral entry (to third semester):

A pass in D. Pharm. course from an institution approved by the Pharmacy Council of India under section 12 of the Pharmacy Act.

III. Program Educational Objectives (PEOs):

- **PEO-1:** AdtU Pharmacy graduates will be well prepared for successful careers as Pharmaceutical Professionals across diverse sectors including the pharmaceutical industry, healthcare, corporate institutions and government organizations.
- PEO-2: Pharmacy graduates will be academically prepared to become Registered

- Pharmacists, poised to make significant contributions to the advancement of the healthcare sector.
- **PEO-3:** The graduates will engage in professional practices to elevate their stature with a sense of responsibility and be successful in higher education, if pursued.

IV. Programme Specific Outcomes (PSOs):

- **PSO-1: Research Competency** Apply pharmaceutical knowledge in research, and collaborative projects thereby contributing to the continuous advancement of pharmaceutical science.
- **PSO-2: Entrepreneurial Proficiency** Exhibit entrepreneurial competency in capitalizing business opportunity.
- **PSO-3: Global Competency** Excel in the profession with global competency that attained through global certifications from international learning platforms.

V. Program Outcome (PO):

- **PO.1:** Pharmacy Knowledge: Possess knowledge and comprehension of the core and basic knowledge associated with the profession of pharmacy, including biomedical sciences; pharmaceutical sciences; behavioral, social, and administrative pharmacy sciences; and manufacturing practices.
- **PO.2:** Planning abilities: Demonstrate effective planning abilities including time management, resource management, delegation skills and organizational skills. Develop and implement plans and organize work to meet deadlines.
- **PO.3:** Problem analysis: Utilize the principles of scientific enquiry, thinking analytically, clearly and critically, while solving problems and making decisions during daily practice. Find, analyze, evaluate and apply information systematically and shall make defensible decisions.
- **PO.4:** Modern tool usage: Learn, select, and apply appropriate methods and procedures, resources, and modern pharmacy-related computing tools with an understanding of the limitations.
- **PO.5:** Leadership skills: Understand and consider the human reaction to change, motivation issues, leadership and team-building when planning changes required for fulfillment of practice, professional and societal responsibilities. Assume participatory roles as responsible citizens or leadership roles when appropriate to facilitate improvement in health and well-being.
- **PO.6:** Professional identity: Understand, analyze and communicate the value of their professional roles in society (e.g. health care professionals, promoters of health, educators, managers, employers, employees).
- **PO.7:** Pharmaceutical ethics: Honour personal values and apply ethical principles in professional and social contexts. Demonstrate behavior that recognizes cultural and personal variability in values, communication and lifestyles. Use ethical frameworks; apply ethical principles while making decisions and take responsibility for the outcomes associated with the decisions.
- **PO.8:** Communication: Communicate effectively with the pharmacy community and with society at large, such as, being able to comprehend and write effective reports, make effective presentations and documentation, and give and receive clear instructions.
- PO.9: The Pharmacist and society: Apply reasoning informed by the contextual knowledge

- to assess societal, health, safety and legal issues and the consequent responsibilities relevant to the professional pharmacy practice.
- **PO.10:** Environment and sustainability: Understand the impact of the professional pharmacy solutions in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.
- **PO.11:** Life-long learning: Recognize the need for, and have the preparation and ability to engage in independent and life-long learning in the broadest context of technological change. Self- access and use feedback effectively from others to identify learning needs and to satisfy these needs on an ongoing basis.

VI. Career Prospects:

B. Pharm graduates are equipped to assume diverse roles, such as Industrial Pharmacist (in the field of Production and Manufacturing, Formulation Development, Quality Assurance, Quality Control, Packaging, R & D etc.), Hospital and Community Pharmacist, Medical Representative, Sales Executive, Bulk Medicine Distributor, Lecturer (for D.Pharm Students), Entrepreneurship, Drug Inspector, Drug Analyst etc. After completion of B.Pharmacy the students may go for higher studies in different M. Pharmacy specializations or in other fields

CHAPTER-I: REGULATIONS

1. Short Title and Commencement

These regulations shall be called as "The Revised Regulations for the B. Pharm. Degree Program (CBCS) of the Pharmacy Council of India, New Delhi". They shall come into effect from the Academic Year 2016-17. The regulations framed are Subject to modifications from time to time by Pharmacy Council of India.

1. Minimum qualification for admission.

1.1. First year B. Pharm:

Candidate shall have passed 10+2 examination conducted by the respective state/central government authorities recognized as equivalent to 10+2 examination by the Association of Indian Universities (AIU) with English as one of the subjects and Physics, Chemistry, Mathematics (P.C.M) and or Biology (P.C.B /P.C.M.B.) as optional subjects individually. Any other qualification approved by the Pharmacy Council of India as equivalent to any of the above examinations.

1.2. B. Pharm lateral entry (to third semester):

A pass in D. Pharm. course from an institution approved by the Pharmacy Council of India under section 12 of the Pharmacy Act.

2. Duration of the program

The course of study for B. Pharm shall extend over a period of eight semesters (four academic years) and six semesters (three academic years) for lateral entry students. The curricula and syllabi for the program shall be prescribed from time to time by Pharmacy Council of India, New Delhi.

3. Medium of instruction and examinations

Medium of instruction and examination shall be in English.

4. Working days in each semester

Each semester shall consist of not less than 100 working days. The odd semesters shall be conducted from the month of June/July to November/December and the even semesters shall be conducted from December/January to May/June in every calendar year.

5. Attendance and progress

A candidate is required to putting at least 80% attendance in individual course considering theory and practical separately. The candidate shall complete the prescribed course satisfactorily to be eligible to appear for the respective examinations.

6. Program/Course credit structure

As per the philosophy of Credit Based Semester System, certain quantum of academic work viz. theory classes, tutorial hours, practical classes, etc. are measured in terms of credits. On satisfactory completion of the courses, a candidate earns credits. The amount of credit associated with a course is dependent upon the number of hours of instruction per week in that course. Similarly, the credit associated with any of the other academic, co/extracurricular activities is dependent upon the quantum of work expected to be putting for each of these activities per week.

Credit assignment

Theory and Laboratory courses

Courses are broadly classified as Theory and Practical. Theory courses consist of lecture (L) and /or tutorial (T) hours, and Practical (P) courses consist of hours spent in the laboratory. Credits (C) for a course is dependent on the number of hours of instruction per week in that course, and is obtained by using a multiplier of one (1) for lecture and tutorial hours, and a

multiplier of half (1/2) for practical (laboratory) hours. Thus, for example, a theory course having three lectures and one tutorial per week throughout the semester carries a credit of 4. Similarly, a practical having four laboratory hours per week throughout semester carries a credit of 2.

Minimum credit requirements

The minimum credit points required for award of a B. Pharm. degree is 208. These credits are divided into Theory courses, Tutorials, Practical, Practice School and Project over the duration of Eight semesters. The credits are distributed semester-wise as shown in Table IX. Courses generally progress in sequences, building competencies and their positioning indicates certain academic maturity on the part of the learners. Learners are expected to follow the semester-wise schedule of courses given in the syllabus. The lateral entry students shall get 52 credit points transferred from their D. Pharm program. Such students shall take up additional remedial courses of 'Communication Skills' (Theory and Practical) and 'Computer Applications in Pharmacy' (Theory and Practical) equivalent to 3 and 4 credit points respectively, a total of 7 credit points to attain 59 credit points, the maximum of I and II semesters.

7. Academic work

A regular record of attendance both in Theory and Practical shall be maintained by the teaching staff of respective courses

8. Course of study

The course of study for B. Pharm shall include Semester Wise Theory & Practical as given in Table – I to VIII. The number of hours to be devoted to each theory, tutorial and practical course in any semester shall not be less than that shown in Table – I to VIII.

Table-I: Course of study for semester I

Course code	Name of the course	No. of hours	Tutorial	Credit points
BP101T	Human Anatomy and Physiology I – Theory	3	1	4
BP102T	Pharmaceutical Analysis I–Theory	3	1	4
BP103T	Pharmaceutics I— Theory	3	1	4
BP104T	Pharmaceutical Inorganic Chemistry–Theory	3	1	4
BP105T	Communication skills—Theory*	2	-	2
BP106RBT BP106RMT	Remedial Biology/ Remedial Mathematics—Theory*	2	-	2
BP107P	Human Anatomy and Physiology–Practical	4	-	2
BP108P	Pharmaceutical Analysis I–Practical	4	-	2
BP109P	Pharmaceutics I–Practical	4	-	2
BP110P	Pharmaceutical Inorganic Chemistry – Practical	4	-	2
BP111P	Communication skills–Practical*	2	-	1
BP112RBP	Remedial Biology–Practical*	2	-	1
	Total	32/34\$/6#	4	27/29 ^{\$} /30 [#]

#Applicable ONLY for the students who have studied Mathematics / Physics / Chemistry at HSC and appearing for Remedial Biology (RB) course.

\$Applicable ONLY for the students who have studied Physics / Chemistry / Botany / Zoology at HSC and appearing for Remedial Mathematics (RM) course.

^{*} Non University Examination (NUE)

Table-II: Course of study for semester II

Course Code	Name of the course	No. of hours	Tutorial	Credit points
BP201T	Human Anatomy and Physiology II-Theory	3	1	4
BP202T	Pharmaceutical Organic Chemistry I–Theory	3	1	4
BP203T	Biochemistry-Theory	3	1	4
BP204T	Pathophysiology-Theory	3	1	4
BP205T	Computer Applications in Pharmacy–Theory*		-	3
BP206T	Environmental sciences-Theory*	3	-	3
BP207P	Human Anatomy and Physiology II-Practical	4	-	2
BP208P	Pharmaceutical Organic Chemistry I–Practical	4	-	2
BP209P	Biochemistry-Practical	4	-	2
BP210P	Computer Applications in Pharmacy–Practical*	2	-	1
	Total	32	4	29

^{*}Non University Examination (NUE)

Table-III: Course of study for semester III

Course code	Name of the course	No. of hours	Tutorial	Credit points
BP301T	Pharmaceutical Organic Chemistry II– Theory	3	1	4
BP302T	Physical Pharmaceutics I-Theory	3	1	4
BP303T	Pharmaceutical Microbiology–Theory	3	1	4
BP304T	Pharmaceutical Engineering-Theory	3	1	4
BP305P	Pharmaceutical Organic Chemistry II–Practical	4	-	2
BP306P	Physical Pharmaceutics I-Practical	4	-	2
BP307P	Pharmaceutical Microbiology–Practical	4	-	2
BP308P	Pharmaceutical Engineering—Practical	4	-	2
	Total	28	4	24

Table-IV: Course of study for semester IV

Course code	Name of the course	Name of the course No. of hours		
BP401T	Pharmaceutical Organic Chemistry III-Theory	3	1	4
BP402T	Medicinal Chemistry I–Theory	3	1	4
BP403T	Physical Pharmaceutics II—Theory	3	1	4
BP404T	Pharmacology I-Theory	3	1	4
BP405T	Pharmacognosy and Phytochemistry I-Theory	3	1	4
BP406P	Medicinal Chemistry I–Practical	4	-	2
BP407P	Physical Pharmaceutics II–Practical	4		2
BP408P	Pharmacology I-Practical	4	-	2
BP409P	Pharmacognosy and Phytochemistry I-Practical	4	-	2
	Total	31	5	28

Table-V: Course of study for semester V

Course code	Name of the course	No. of hours	Tutorial	Credit points
BP501T	Medicinal Chemistry II–Theory	3	1	4
BP502T	Industrial Pharmacy I—Theory	3	1	4
BP503T	Pharmacology II-Theory	3	1	4
BP504T	Pharmacognosy and Phytochemistry II–Theory	3	1	4
BP505T	Pharmaceutical Jurisprudence—Theory	3	1	4
BP506P	Industrial Pharmacy I–Practical	4	-	2
BP507P	Pharmacology II–Practical	4	-	2
BP508P	Pharmacognosy and Phytochemistry II— Practical	4	-	2
	Total	27	5	26

Table-VI: Course of study for semester VI

Course code	Name of the course	No. of hours	Tutorial	Credit points
BP601T	Medicinal Chemistry III–Theory	3	1	4
BP602T	Pharmacology III – Theory	3	1	4
BP603T	Herbal Drug Technology-Theory	3	1	4
BP604T	Biopharmaceutics and Pharmacokinetics-Theory	3	1	4
BP605T	Pharmaceutical Biotechnology–Theory	3	1	4
BP606T	Quality Assurance–Theory	3	1	4
BP607P	Medicinal chemistry III–Practical	4	-	2
BP608P	Pharmacology III- Practical		-	2
BP609P	Herbal Drug Technology– Practical	4	-	2
	Total	30	6	30

Table-VII: Course of study for semester VII

Course code	Name of the course	No.of hours	Tutorial	Credit points
BP701T	Instrumental Methods of Analysis –Theory	3	1	4
BP702T	Industrial Pharmacy II–Theory	3	1	4
BP703T	Pharmacy Practice—Theory	3	1	4
BP704T	Novel Drug Delivery System–Theory	3	1	4
BP705P	Instrumental Methods of Analysis–Practical	4	-	2
BP706PS	Practice School*	12	-	6
	Total	28	5	24

^{*}Non University Examination (NUE)

Table-VIII: Course of study for semester VIII

Sl No	Course	Name of the course	No. of	Tutorial	Credit
	Code		hours		points
1	BP801T	Biostatistics and Research Methodology	3	1	4
2	BP802T	Social and Preventive Pharmacy	3	1	4
	BP803ET	Pharma Marketing Management			
	BP804ET	Pharmaceutical Regulatory Science	1		
3	BP805ET	Pharma covigilance	1		
4	BP806ET	Quality Control and Standardization of			
		Herbals	3+3=	1+1=2	4+4=
	BP807ET	Computer Aided Drug Design	6		8
	BP808ET	Cell and Molecular Biology	1		
	BP809ET	Cosmetic Science	1		
	BP810ET	Experimental Pharmacology			
	BP811ET	Advanced Instrumentation Techniques	1		
	BP812ET	Dietary Supplements and Nutraceuticals			
5	BP813PW	Project Work	12	-	6
6	BP814EA	Extracurricular/ Co-curricular activities**			1
		Total	24	4	23

Table-IX: Semester wise credits distribution

Semester	Credit Points
I	27/29 ⁸ /30 [#]
II	29
III	24
IV	28
V	26
VI	30
VII	24
VIII	23
Total credit points for the program	211/213 ^{\$} /214 [#]

The credit points assigned for extracurricular and or co-curricular activities shall be given by the Principals of the colleges and the same shall be submitted to the University. The criteria to acquire this credit point shall be defined by the colleges from time to time.

\$Applicable ONLY for the students studied Physics / Chemistry / Botany / Zoology at HSC and appearing for Remedial Mathematics course.

#Applicable ONLY for the students studied Mathematics / Physics / Chemistry at HSC and appearing for remedial biology course.

9. Program Academic Committee Program Committee

- 1. The B. Pharm program shall have a Program Committee constituted by the Head of the institution in consultation with all the Heads of the departments.
- 2. The composition of the Program Committee shall be as follows:

 A senior teacher shall be the Chairperson; One Teacher from each department handling B.Pharm courses; and four student representatives of the program (one from each

academic year), nominated by the Head of the institution.

3. Duties of the Program Committee:

- i. Periodically reviewing the progress of the classes.
- ii. Discussing the problems concerning curriculum, syllabus and the conduct of classes.
- iii. Discussing with the course teachers on the nature and scope of assessment for the course and the same shall be announced to the students at the beginning of respective semesters.
- iv. Communicating its recommendation to the Head of the institution on academic matters.
- v. The Program Committee shall meet at least thrice in a semester preferably at the end of each Sessional exam (Internal Assessment) and before the end semester exam.

10. Examinations/Assessments

The scheme for internal assessment and end semester examinations is given in Table–X.

End semester examinations

The End Semester Examinations for each theory and practical course through semesters I to VIII shall be conducted by the university except for the subjects with asterix symbol (*) in table I and II for which examinations shall be conducted by the subject experts at college level and the marks/grades shall be submitted to the university.

Tables-X: Scheme for internal assessments and end semester examinations semester wise Semester I

Caursa anda		Internal Assessment				End Se Ex	Total	
Course code	Name of the course	Continuous Mode	Session: Marks	al Exams Duration	Total	Marks	Duration	Marks
BP101T	Human Anatomy and Physiology I– Theory	10	15	1 Hr	25	75	3 Hrs	100
BP102T	Pharmaceutical Analysis I— Theory	10	15	1 Hr	25	75	3 Hrs	100
BP103T	Pharmaceutics I-Theory	10	15	1 Hr	25	75	3 Hrs	100
BP104T	Pharmaceutical Inorganic Chemistry–Theory	10	15	1 Hr	25	75	3 Hrs	100
BP105T	Communication skills— Theory*	5	10	1 Hr	15	35	1.5Hrs	50
BP106RBT BP106RMT	Remedial Biology/ Mathematics—Theory*	5	10	1 Hr	15	35	1.5Hrs	50
BP107P	Human Anatomy and Physiology–Practical	5	10	4 Hrs	15	35	4 Hrs	50
BP108P	Pharmaceutical Analysis I— Practical	5	10	4 Hrs	15	35	4 Hrs	50
BP109P	Pharmaceutics I—Practical	5	10	4 Hrs	15	35	4 Hrs	50
BP110P	Pharmaceutical Inorganic Chemistry—Practical	5	10	4 Hrs	15	35	4 Hrs	50
BP111P	Communication skills— Practical*	5	5	2 Hrs	10	15	2 Hrs	25
BP112R BP	Remedial Biology– Practical*	5	5	2 Hrs	10	15	2Hrs	25
Total		70/75 ^{\$} /8 0 [#]	115/125 ^{\$} / 13 0 [#]	23/24 ^s /26 [#] Hrs	0#	490/525\$ _/ 540 [#]	31.5/33 \$/35 [#] Hrs	675/72 5 ^{\$} / 750 [#]

[#]Applicable ONLY for the students studied Mathematics/Physics/Chemistry at HSC and appearing for Remedial Biology (RB) course.

Semester II

Course code	Name of the course	In	ternal A	ssessment			emester ams	Total Marks
		Continuous	Session	al Exams	Total	Marks	Duratio	
		Mode	Marks	Duration			n	
BP201T	Human Anatomy and	10	15	1Hr	25	75	3Hrs	100
	Physiology II-Theory							
BP202T	Pharmaceutical Organic	10	15	1Hr	25	75	3Hrs	100
	Chemistry I–Theory							
BP203T	Biochemistry-Theory	10	15	1Hr	25	75	3Hrs	100
BP204T	Pathophysiology-Theory	10	15	1Hr	25	75	3Hrs	100
BP205T	Computer Applications in	10	15	1Hr	25	50	2Hrs	75
	Pharmacy-Theory*							
BP206T	Environmental sciences-	10	15	1Hr	25	50	2Hrs	75
	Theory*							
BP207P	Human Anatomy and	5	10	4Hrs	15	35	4Hrs	50
	Physiology II–Practical							
BP208P	Pharmaceutical Organic	5	10	4Hrs	15	35	4Hrs	50
	Chemistry I–Practical							
BP209P	Biochemistry- Practical	5	10	4Hrs	15	35	4Hrs	50
BP210P	Computer Applications in	5	5	2Hrs	10	15	2Hrs	25
	Pharmacy-Practical*							
	Total	80	125	20 Hrs	205	520	30 Hrs	725

^{*} The subject experts at college level shall conduct examinations

Semester III

Course		Internal Assessment					Semester	Total
code	Name of the course						xams	Marks
		Continuous	Session	al Exams	Total	Marks	Duration	
		Mode	Marks	Duration				
BP301T	Pharmaceutical Organic	10	15	1Hr	25	75	3Hrs	100
	Chemistry II–Theory							
BP302T	Physical Pharmaceutics I-	10	15	1Hr	25	75	3Hrs	100
	Theory							
BP303T	Pharmaceutical	10	15	1Hr	25	75	3Hrs	100
	Microbiology – Theory							
BP304T	Pharmaceutical	10	15	1Hr	25	75	3Hrs	100
	Engineering- Theory							
BP305P	Pharmaceutical Organic	5	10	4Hr	15	35	4Hrs	50
	Chemistry II–Practical							
BP306P	Physical Pharmaceutics I-	5	10	4Hr	15	35	4Hrs	50
	Practical							
BP307P	Pharmaceutical	5	10	4Hr	15	35	4Hrs	50
	Microbiology –Practical							
BP308P	Pharmaceutical	5	10	4Hr	15	35	4Hrs	50
	Engineering- Practical							
	Total	60	100	20	160	440	28Hrs	600

^{\$}Applicable ONLY for the Students studied Physics/Chemistry/Botany/Zoology at HSC and appearing for Remedial Mathematics (RM)course.

^{*} Non University Examination (NUE)

Semester IV

Course		Internal Assessment			End Semester Exams		Total	
code	Name of the course	Continuous	Session	al Exams	Total	Marks	Duration	Marks
		Mode	Marks	Duration	Total	Maiks	Duration	
BP401T	Pharmaceutical Organic Chemistry III–Theory	10	15	1Hr	25	75	3Hrs	100
BP402T	Medicinal Chemistry I— Theory	10	15	1Hr	25	75	3Hrs	100
BP403T	Physical Pharmaceutics II– Theory	10	15	1Hr	25	75	3Hrs	100
BP404T	Pharmacology I-Theory	10	15	1Hr	25	75	3Hrs	100
BP405T	Pharmacognosy I-Theory	10	15	1Hr	25	75	3Hrs	100
BP406P	Medicinal Chemistry I— Practical	5	10	4Hr	15	35	4Hrs	50
BP407P	Physical Pharmaceutics II– Practical	5	10	4Hrs	15	35	4Hrs	50
BP408P	Pharmacology I–Practical	5	10	4Hrs	15	35	4Hrs	50
BP409P	Pharmacognosy I— Practical	5	10	4Hrs	15	35	4Hrs	50
	Total	70	115	21 Hrs	185	515	31 Hrs	700

Semester V

		,	iicstei v					
Course		Internal Assessment				Total		
code	Name of the course	Continuous Mode		al Exams Duration	Total	Marks	3Hrs 3Hrs 3Hrs 3Hrs 4Hrs 4Hrs	Marks
BP501T	Medicinal Chemistry II– Theory	10	15	1Hr	25	75	3Hrs	100
BP502T	Industrial Pharmacy I— Theory	10	15	1Hr	25	75	3Hrs	100
BP503T	Pharmacology II-Theory	10	15	1Hr	25	75	3Hrs	100
BP504T	Pharmacognosy II— Theory	10	15	1Hr	25	75	3Hrs	100
BP505T	Pharmaceutical Jurisprudence – Theory	10	15	1Hr	25	75	3Hrs	100
BP506P	Industrial Pharmacy I— Practical	5	10	4Hr	15	35	4Hrs	50
BP507P	Pharmacology II– Practical	5	10	4Hr	15	35	4Hrs	50
BP508P	Pharmacognosy II– Practical	5	10	4Hr	15	35	4Hrs	50
	Total	65	105	17 Hr	170	480	27 Hrs	650

Semester VI

Course		In	Internal Assessment			End So Ex	Total	
code	Name of the course	Continuous		al Exams	Total	Marks	Duration	Marks
		Mode	Marks	Duration	1000	1/141113	Duracion	
BP601T	Medicinal Chemistry III– Theory	10	15	1Hr	25	75	3Hrs	100
BP602T	Pharmacology III-Theory	10	15	1Hr	25	75	3Hrs	100
BP603T	Herbal Drug Technology— Theory	10	15	1Hr	25	75	3Hrs	100
BP604T	Biopharmaceutics and Pharmacokinetics—Theory	10	15	1Hr	25	75	3Hrs	100
BP605T	Pharmaceutical Biotechnology– Theory	10	15	1Hr	25	75	3Hrs	100
BP606T	Quality Assurance– Theory	10	15	1Hr	25	75	3Hrs	100
BP607P	Medicinal chemistry III– Practical	5	10	4Hrs	15	35	4Hrs	50
BP608P	Pharmacology III– Practical	5	10	4Hrs	15	35	4Hrs	50
BP609P	Herbal Drug Technology– Practical	5	10	4Hrs	15	35	4Hrs	50
_	Total	75	120	18 Hrs	195	555	30 Hrs	750

Semester VII

Course	Name of the course	In	ternal A	assessment		Semesto	Total	
code	rame of the course	Continuous	Session	al Exams	Total	Marks	Duration	Marks
		Mode	Marks	Duration	Total	Maiks	Duration	
BP701T	Instrumental Methods of	10	15	1Hr	25	75	3Hrs	100
	Analysis – Theory					75 3Hrs		
BP702T	Industrial Pharmacy—	10	15	1Hr	25	75	3Hrs	100
B1 7021	Theory	10	13	1111	23	, ,	75 3Hrs	100
BP703T	Pharmacy Practice—Theory	10	15	1Hr	25	75	3Hrs	100
BP704T	Novel Drug Delivery	10	15	1Hr	1Hr 25	75	3Hrs	100
DI /041	System— Theory	10	13	1111	23	13	31118	100
BP705P	Instrumental Methods of	5	10	10 411	1.5	35	4Hrs	50
Dr /03r	Analysis –Practical	3	10	4Hrs	15	33	41118	30
BP706PS	Practice School*	25	-	-	25	125	5Hrs	150
	Total	70	70	8Hrs	140	460	21 Hrs	600

^{*} The subject experts at college level shall conduct examinations

Semester VIII

Course		In	ternal A	ssessment		End S Ex	Total	
code	Name of the course	Continuo	Session	al Exams	Total	Marks	Duration	Marks
		us Mode	Marks	Duration	Total	IVIAI KS	Duration	
BP801T	Biostatistics and Research Methodology –Theory	10	15	1Hr	25	75	3Hrs	100
BP802T	Social and Preventive Pharmacy – Theory	10	15	1Hr	25	75	3Hrs	100
BP803ET	Pharmaceutical Marketing— Theory							
BP804ET	Pharmaceutical Regulatory Science – Theory							
BP805ET	Pharma co vigilance-Theory							
	Quality Control and							
BP806ET	Standardization of Herbals-							100
Dr 800E 1	Theory	10+10	15 +15	1 +1	25 +25	75 +75	3 +3	+
BP807ET	Computer Aided Drug	=	=	=	=	=	3Hrs 3Hrs	100
DI OUTEI	Design– Theory	20	30	2Hrs	50	150	6 Hrs	=
BP808ET	Cell and Molecular Biology– Theory							200
BP809ET	Cosmetic Science– Theory							
BP810ET	Experimental Pharmacology— Theory							
	Advanced Instrumentation							
BP811ET	Techniques—Theory							
BP812PW	Project Work	-	-	-	-	150	4Hrs	150
	Total	40	60	4Hrs	100	450	16 Hrs	550

Internal assessment: Continuous mode

The marks allocated for Continuous mode of Internal Assessment shall be awarded as per the scheme given below.

Table-XI: Scheme for awarding internal assessment: Continuous mode

Theory					
Criteria	Criteria Maximum Ma				
Attendance (Refer Table–XII)	4	2			
Academic activities (Average of any 3 Activities eg. quiz, assignment, open book test, fieldwork, group discussion and seminar)	3	1.5			
Student–Teacher interaction	3	1.5			
Total	10	5			
Practical					
Attendance(Refer Table–XII) 2		2			
Based on Practical Records, Regular viva voce, etc. 3		3			
Total		5			

Table-XII: Guidelines for the allotment of marks for attendance

Percentage of Attendance	Theory	Practical
95–100	4	2
90– 94	3	1.5
85–89	2	1
80–84	1	0.5
Less than 80	0	0

11.2.1. Sessional Exams

Two Sessional exams shall be conducted for each theory / practical course as per the schedule fixed by the college(s). The scheme of question paper for theory and practical Sessional examinations is given below. The average marks of two Sessional exams shall be computed for internal assessment as per the requirements given in tables—X.

Sessional exam shall be conducted for 30 marks for theory and shall be computed for 15 marks. Similarly Sessional exam for practical shall be conducted for 40marks and shall be computed for 10 marks.

Question paper pattern for theory Sessional examinations for subjects having University examination

I.	Multiple Choice Questions (MCQs)	=	10x1 =	10
	OR			
TT	O(1) C C O C C O		05.2	10

II. Objective Type Questions (5x2) = 05x2 = 10

(Answer all the questions)

I. Long Answers (Answer 1 out of 2) = 1x10 = 10II. Short Answers (Answer 2 out of 3) = 2x5 = 10

Total = 30 marks

For subject shaving Non University Examination

I. Long Answers (Answer 1 out of 2) = 1x10 = 10II. Short Answers (Answer 4 out of 6) = 4x5 = 20

Total = 30

marks

Question paper pattern for practical sessional examinations

I. Synopsis = 10
 II Experiment = 25
 III Viva Voice = 05
 Total = 40 Marks

11. Promotion and award of grades

A student shall be declared PASS and eligible for getting grade in a course of B.Pharm. program if he/she secures at least 50% marks in that particular course including internal assessment. For example, to be declared as PASS and to get grade, the student has to secure a minimum of 50 marks for the total of 100 including continuous mode of assessment and end semester theory examination and has to secure a minimum of 25marks for the total 50 including internal assessment and end semester practical examination.

12. Carry forward of marks

In case a student fails to secure the minimum 50% in any Theory or Practical course as

specified in 12, then he/she shall reappear for the end semester examination of that course. However his/her marks of the Internal Assessment shall be carried over and he/she shall been titled for grade obtained by him/her on passing.

13. Improvement of internal assessment

A student shall have the opportunity to improve his/her performance only once in the Sessional exam component of the internal assessment. The re-conduct of the Sessional exam shall be completed before the commencement of next end semester theory examinations.

14. Re-examination of end semester examinations

Re-examination of end semester examination shall be conducted as per the schedule given in table XIII. The exact dates of examinations shall be notified from time to time.

Table-XIII: Tentative schedule of end semester examinations

Semester	For Regular Candidates	For Failed Candidates
I, III, V and VII	November/December	May/June
II, IV,VI and VIII	May/June	November/December

Question paper pattern for end semester theory examinations for 75 marks paper

I.	Multiple Choice Question (MCQs)	=	20x1 =	20
	OR			
	Objective Type Question (10x2)	=	10x2 =	20
	(Answer all the Question (10x2)			
II.	Long Answer (Answer 2 out of 3)	=	2x10 =	20
III.	Short Answer (Answer 7 of 9)	=	7x5 =	35
	Total	=	75 Mark	

For 50 marks paper

Total	=	50 marks	
II. Short Answers(Answer 6 out of 8)	=	6x5 =	30
I. Long Answers(Answer 2 out of 3)	=	2x10 =	20

For 35 marks paper

I. Long Answers (Answer 1 out of 2)	=	1x10	=	10
II. Short Answers (Answer 5 out of 7)	=	5x5	=	25
Total	=	35 mar	ks	

Question paper pattern for end semester practical examinations

I.	Synopsis	=	5
II.	Experiments	=	25
III.	Viva voce	=	5
	Total	=	35 marks

15. Academic Progression:

No student shall be admitted to any examination unless he/she fulfils the norms given in 6 Academic progression rules are applicable as follows:

A student shall be eligible to carry forward all the courses of I, II and III semesters till the IV semester examinations. However, he/she shall not be eligible to attend the courses of V semester until all the courses of I and II semesters are successfully completed.

A student shall be eligible to carry forward all the courses of III, IV and V semesters till the VI semester examinations. However, he/she shall not be eligible to attend the courses of VII semester until all the courses of I, II, III and IV semesters are successfully completed.

A student shall be eligible to carry forward all the courses of V, VI and VII semesters till the VIII semester examinations. However, he/she shall not be eligible to get the course completion certificate until all the courses of I,II,III,IV,V and VI semesters are successfully completed.

A student shall be eligible to get his/her CGPA upon successful completion of the courses of I to VIII semesters within the stipulated time period as per the norms specified in 26.

A lateral entry student shall be eligible to carry forward all the courses of III, IV and V semesters till the VI semester examinations. However, he/she shall not be eligible to attend the courses of VII semester until all the courses of III and IV semesters are successfully completed.

A lateral entry student shall be eligible to carry forward all the courses of V, VI and VII semesters till the VIII semester examinations. However, he/she shall not be eligible to get the course completion certificate until all the courses of III, IV, V and VI semesters are successfully completed.

A lateral entry student shall be eligible to get his/her CGPA upon successful completion of the courses of III to VIII semesters within the stipulated time period as per the norms specified in 26.

Any student who has given more than 4 chances for successful completion of I /III semester courses and more than 3 chances for successful completion of II / IV semester courses shall be permitted to attend V/VII semester classes ONLY during the subsequent academic year as the case may be. In simpler terms there shall NOT be any ODDBATCH for any semester. Note: Grade AB should be considered as failed and treated as one head for deciding academic progression. Such rules are also applicable for those students who fail to register for examination (s) of any course in any semester.

16. Grading of performances

Letter grades and grade points allocations:

Based on the performances, each student shall be awarded a final letter grade at the end of the semester for each course. The letter grades and their corresponding grade points are given in Table– XII.

Table – XII: Letter grades and grade points equivalent to Percentage of marks and performances

Percentage of Marks Obtained	Letter Grade	Grade Point	Performance
90.00–100	O	10	Outstanding
80.00-89.99	A	9	Excellent
70.00 -79.99	В	8	Good
60.00 -69.99	С	7	Fair
50.00 -59.99	D	6	Average
Less than 50	F	0	Fail
Absent	AB	0	Fail

A learner who remains absent for any end semester examination shall be assigned a letter grade of AB and a corresponding grade point of zero. He/she should reappear for the said evaluation/examination in due course.

17. The Semester grade point average (SGPA)

The performance of a student in a semester is indicated by a number called 'Semester Grade Point Average' (SGPA). The SGPA is the weighted average of the grade points obtained in all the courses by the student during the semester. For example, if a student takes five courses (Theory/Practical) in a semester with credits C1,C2, C3,C4and C5and the student's grade points in these courses are G1, G2, G3, G4 and G5, respectively, and then students' SGPA is equal to:

$$C_{1}G_{1}+C_{2}G_{2}+C_{3}G_{3}+C_{4}G_{4}+C_{5}G_{5}$$

$$SGPA= C_{1}+C_{2}+C_{3}+C_{4}+C_{5}$$

The SGPA is calculated to two decimal points. It should be noted that, the SGPA for any semester shall take into consideration the F and ABS grade awarded in that semester. For example if a learner has a F or ABS grade in course 4, the SGPA shall then be computed as:

$$C_{1}G_{1}+C_{2}G_{2}+C_{3}G_{3}+C_{4}*ZERO+C_{5}G_{5}$$

$$C_{1}+C_{2}+C_{3}+C_{4}+C_{5}$$

18. Cumulative Grade Point Average (CGPA)

The CGPA is calculated with the SGPA of all the VIII semesters to two decimal points and is indicated in final grade report card/final transcript showing the grades of all VIIIsemestersandtheircourses. The CGPA shall reflect the failed status in case of Fgrade(s), till the course(s) is/are passed. When the course(s) is/are passed by obtaining a pass grade on subsequent examination(s) the CGPA shall only reflect the new grade and not the fail grades earned earlier. The CGPA is calculated as:

$$C_{1}S_{1}+C_{2}S_{2}+C_{3}S_{3}+C_{4}S_{4}+C_{5}S_{5}+C_{6}S_{6}+C_{7}S_{7}+C_{8}S_{8}$$

$$CGPA = C_{1}+C_{2}+C_{3}+C_{4}+C_{5}+C_{6}+C_{7}+C_{8}$$

where C_1 , C_2 , C_3 ,... is the total number of credits for semester I,II,III,... and S_1 , S_2 , S_3 ,... is the SGPA of semester I,II,III,....

19. Declaration of class

The class shall be awarded on the basis of CGPA as follows: First Class with Distinction = CGPA of. 7.50 and above First Class = CGPA of 6.00 to 7.49 Second Class = CGPA of 5.00 to 5.99

20. Project work

All the students shall undertake a project under the supervision of a teacher and submit are port. The area of the project shall directly relate any one of the elective subject opted by the student in semester VIII. The project shall be carried out in group not exceeding 5 in number. The project report shall be submitted in triplicate (typed & bound copy not less than 25

pages).

The internal and external examiner appointed by the University shall evaluate the project at the time of the Practical examinations of other semester(s). Students shall be evaluated in groups for four hours (i.e., about half an hour for a group of five students). The projects shall be evaluated as per the criteria given below.

Evaluation of Dissertation Book: Objective(s) of the work done 15 Marks Methodology adopted 20Marks Results and Discussions 20 Marks Conclusions and Outcomes 20Marks **Total 75 Marks**

Evaluation of Presentation: Presentation of work 25Marks Communication skills 20Marks Question and answer skills 30Marks **Total 75 Marks**

Explanation: The 75 marks assigned to the dissertation book shall be same for all the students in a group. However, the 75 marks assigned for presentation shall be awarded based on the performance of individual students in the given criteria.

21. Industrial training (Desirable)

Every candidate shall be required to work for at least 150 hours spread over four weeks in a Pharmaceutical Industry/Hospital. It includes Production unit, Quality Control department, Quality Assurance department, Analytical laboratory, Chemical manufacturing unit, Pharmaceutical R&D, Hospital (Clinical Pharmacy),Clinical Research Organization, Community Pharmacy, etc. After the Semester – VI and before the commencement of Semester – VII, and shall submit satisfactory report of such work and certificate duly signed by the authority of training organization to the head of the institute.

22. Practice School

In the VII semester, every candidate shall undergo practice school for a period of 150hours evenly distributed throughout the semester. The student shall opt any one of the domains for practice school declared by the program committee from time to time.

At the end of the practice school, every student shall submit a printed report (in triplicate)on the practice school he/she attended (not more than 25 pages). Along with the exams of semester VII, the report submitted by the student, knowledge and skills acquired by the student through practice school shall be evaluated by the subject experts at college level and grade point shall be awarded

23. Award of Ranks

Ranks and Medals shall be awarded on the basis of final CGPA. However, candidates who fail in one or more courses during the B.Pharm program shall not be eligible for award of ranks. Moreover, the candidates should have completed the B.Pharm program in minimum prescribed number of years, (four years) for the award of Ranks.

24. Award of degree

Candidates who fulfill the requirements mentioned above shall be eligible for award of degree during the ensuing convocation.

25. Duration for completion of the program of study

The duration for the completion of the program shall be fixed as double the actual duration of the program and the students have to pass within the said period, otherwise they have to get fresh Registration.

26. Re-admission after break of study

Candidate who seeks re-admission to the program after break of study has to get the approval from the university by paying a condo nation fee.

No condonation is allowed for the candidate who has more than 2 years of breakup period and he/she has to rejoin the program by paying the required fees.

		SEMESTER -	· I						
Course Ti	tle	HUMAN ANATOMY AND PHYSIOLOGY-I							
Course cod	de BP101T	Total credits: 4	L	T	P	S	R	O/F	C
	de Di IVII	Total hours: 45	3	1	0	0	0	0	4
Pre-requis	site Nil	Co-requisite				Nil			
Programn	ne	Bachelor	of Phai	macy	,				
Semester	r	Fall/ I semester of fire	st year	of the	Progr	amme			
	1. Explain th	e gross morphology, structur	e and fu	ınction	ns of va	arious	organs	of the	
	human boo	ły.							
Course	2. Describe the	he various homeostatic mech	anisms	and th	eir imł	oalance	es.		
Objective	Objectives 3. Identify the various tissues and organs of different systems of human body.								
	4. Perform th	4. Perform the various experiments related to special senses and nervous system.							
	5. Appreciate	e coordinated working patter	n of diff	erent o	organs	of eac	h syste	em	
CO1	Understand an	Understand and apply the basic terminology and fundamental knowledge of the							
COI	structure and f	structure and functions of various cells and tissues of the human body.							
	Describe the	Describe the morphology and physiology of skeletal system along with the							
CO2	mechanism of	mechanism of muscle contraction in co-ordination with the joints and skin along with							
		their significance.							
CO3		Understand the composition, function of various body fluids like blood and lymph,							
	and describe si	and describe significance and analyze their relation to disorders.							
CO4	Classify the p	Classify the peripheral nervous system, nerves and explain the morphology and							
CO4	working princ	working principles of special senses.							
	Understand th	Understand the anatomy, physiology of the heart and analyze the parameters to							
CO5	understand and	understand and							
	explain their re	elation to CVS and related di	sorders.						
Unit No		Contont	Conta	ct	Loo	mina (Jutaar	m 0	VI.

Unit-No.	Content	Contact Hour	Learning Outcome	KL
	Introduction to human body			
	Definition and scope of anatomy			
	And physiology, levels of structural			
	organization and body systems, basic life			
	processes, homeostasis, basic			
	anatomical terminology.Cellular level of			
	organization Structure and functions of cell,		Students will be able to	
	transport across cell membrane, cell division,		learn gross morphology,	
I	cell junctions. General principles of cell		structure and functions of	
1	communication, intracellular signaling		various organs of the	
	pathway activation by extracellular signal		human body.	
	molecule, Forms of intracellular		numan body.	
	signaling: a) Contact-dependent b)			
	Paracrinec) Synaptic d) Endocrine	10		1,2,
	Tissue level of organization Classification of	10		3,4
	tissues, structure, location and functions of			
	epithelial, muscular and nervous and			
	connective tissues.			
II	Integumentary system Structure and		Students will be able to	
	functions of skin Skeletal system Divisions of		learn ab bones the joint of	
	skeletal system, types of bone, salient		human body	
	features and functions of bones of axial			
	and appendicular skeletal system.			

	Organization of skeletal muscle, physiology of			
	muscle contraction, neuromuscular junction	10		1,2,
	Joints Structural and functional classification,			3,4
	types of joints movements and its articulation			
III	Body fluids and blood Body fluids,		Students will be able to)
	composition and functions of blood,		learn various homeostation	;
	hemopoeisis, formation of hemoglobin,		mechanisms and their	
	anemia, mechanisms of coagulation, blood		imbalances.	
	grouping, Rah factors, ransfusion, its			
	significance and disorders of blood,	10		1,2,
	Reticuloendothelial system. Lymphatic			3,4
	system Lymphatic organs and tissues,			
	lymphatic vessels, lymph circulation and			
	functions Of lymphatic system			
IV	Peripheral nervous system: Classification of		Students will be able to)
	peripheral nervous system: Structure and		learn about nervous system	1
	functions of sympathetic and parasympathetic		of human body	
	nervous system. Origin and functions of spinal			
	and cranial nerves. Special senses: Structure	8		1,2,
	and functions of eye, ear, nose and tongue and			3,4
	their disorders.			
V	Cardiovascular system Heart – anatomy of		Students will be able to	
	heart, blood circulation, blood vessels,		learn about cardiovascular	
	structure and functions of artery, vein and		system.	
	capillaries, elements of conduction system of			
	F			
	heart and eartbeat, its regulation by			
	_			1,2,
	heart and eartbeat, its regulation by	7		1,2, 3,4
	heart and eartbeat, its regulation by autonomic nervous system, cardiac output,	7		

TEXT BOOKS:

- T1: Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypee brothers' medical publishers, New Delhi.
- T2: Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York

REFERENCE BOOKS:

- R1: Principles of Anatomy and Physiology by Tortora Grabowski. Palmetto, GA, U.S.A.
- R2: Text book of Medical Physiology- Arthur C, Guyton and John.E. Hall. Miamisburg, OH, U.S.A..

RELATIONSHIP BETWEEN COURSE OUTCOMES (CO) AND PROGRAM OUTCOMES

CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome			
1	Understand and apply the basic terminology and fundamental knowledge of the structure and functions of various cells and tissues of the human body.	PO1,PO5,PO6,PO8,PO11			
2	Describe the morphology and physiology of skeletal system along with the mechanism of muscle contraction in coordination with the joints and skin along with their significance.	PO1,PO5,PO6,PO8,PO11			
3	Understand the composition, function of various body fluids like blood and lymph, and describe significance and analyze their relation to disorders.	PO1,PO5,PO6,PO8,PO11			
4	Classify the peripheral nervous system, nerves and explain the morphology and working principles of special senses.	PO1,PO5,PO6,PO8,PO11			
5	Understand the anatomy, physiology of the heart and analyze the parameters to understand and explain their relation to CVS and related disorders.	PO1,PO2,PO5,PO6,PO8,PO11			

			SEMESTE	R – I						
Course	Title		PHARMACEU	UTICA	AL AN	ALY	SIS-I			
Course code		BP102T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 45	3	1	0	0	0	0	4
Pre-req		Nil	Co-requisite				Ni	1		
Progra			Bachele			•				
Seme			Fall/ I semester of f			he P	rogramı	ne		
Cour			of the course student sh							
Object	tives		nd the principles of volu					al anal	ysis	
		_	various volumetric and	electro	ochemi	cal tı	trations			
60	.1	_	analytical skills		1 4 1	•	1 1	1	<u> </u>	
CO	1		ndamentals of Pharmac	eutica	i Anaiy	ysis a	na preai	ct the s	sources of	errors
CO		and impurities	-1:C:+:	4:	f	: 1 1	1			
CO	2	involvement.	classifications, and rea	actions	soi ac	ia-ba	se and	non-a	queous ii	ıranon
CO	2		c principle, method,	and a	nnliggt	ion i	nyalyad		miona ma	thoda
	3		omplexometric gravime	-					irious ille	illous
CO	<u></u>		quate knowledge of base						redov tite	ations
	7		quate knowledge of bas on in pharmaceutical an			anu	cominq	ucs UI	TCGOA HIL	at10118
CO	5		ciple, electrodes used,			centi	cal appl	ication	of	
	5	electrochemical ar	_	una p	JIIGITIG	ccan	сы аррі	reaction	01	
Unit-No.			ontent		Conta	ct	Leari	ning O	utcome	KL
					Hou			8		
	a) ph	armaceutical an	alysis- Definition and			5	Students	will l	oe able	
	scope	·				t	o learn			
	i) Diff	ifferent techniques of analysis				I	Different	techni	ques of	
	ii) Me	thods of expressin	g concentration			ľ	harmac	eutical	analysis,	
	1 ′	imary and secondary standards.				ϵ	errors and	d pharr	nacopoeia	
		_	standardization of							
			al solutions-Oxalic acid	,						1,2,
I			ochloric acid, sodium		10					3
		phate, sulphuric ac	•							
	r .	nganate and cericammonium sulphate								
	1` ′	ors: Sources of er								
		methods of minimizing errors, cy, precision and significant figures								
		• •	arces of impurities							
	' '	licinal agents, limi	-							
			Theories of acid by	ase		(Students	will be	able to	
			of acid base titrations a				earn the			
		, , , , , , , , , , , , , , , , , , ,	tions of strong, weak, a						aqueous	
II			es, neutralization curves				itration.			1,2,
	•		Solvents, acidimetry a		10					3
		-	nd estimation of Sodi							
		ate and Ephedrine								
	Precip	oitation titratio	ons: Mohr's meth	od,		5	Students	will be	able to	
			olhard's, Fajans meth	od,		1	earn pre	cipitati	on	
Ш	estima	tion of sodium chl	loride.					_	exometric	
	_		on: Classification, me				itration,	gravim	etric	
	ion in	dicators, masking	and demasking reager	nts,		г	nalysis			

	estimation of Magnesium sulphate, and calcium			
	gluconate.			1,2,
	Gravimetry: Principle and steps involved in	10		3
	gravimetric analysis. Purity of the precipitate: co-			
	precipitation and post precipitation, Estimation of			
	barium sulphate. Basic Principles, methods and			
	application of diazotisation titration.			
	Redox titrations		Students will be able to	
	(a) Concepts of oxidation and reduction		learn Concepts of	
IV	(b) Types of redox titrations (Principles and		oxidation and reduction	1,2,
1 1	applications) Cerimetry, Iodimetry,	8	and Types of redox	3
	Iodometry, Bromatometry, Dichrometry,		titrations.	
	Titration with potassium iodate			
	Electrochemical methods of analysis		Students will be able to	
	Conductometry- Introduction, Conductivity cell,		learn Electrochemical	
	Conducto metric titrations, applications.		methods of analysis	
	Potentiometry-Electrochemical cell, construction			
	and working of reference (Standard hydrogen, silver			
	chloride electrode and calomel electrode) and			
V	indicator electrodes (metal electrodes and glass			
	electrode), methods to determine end point of			
	potentiometric titration and applications.			1,2,
	Polarography- Principle, Ilkovic equation,	7		3
	construction and working of dropping mercury			
	electrode and rotating platinum electrode,			
	applications			

TEXT BOOKS:

T1: A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone, Press of University of London

T2: A.I. Vogel, Text Book of Quantitative Inorganic analysis

REFERENCE BOOKS:

R1: P. Gundu Rao, Inorganic Pharmaceutical Chemistry.

RELATIONSHIP BETWEEN COURSE OUTCOMES (CO) AND PROGRAM OUTCOMES

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Understand the fundamentals of Pharmaceutical Analysis	PO1,PO2,PO3,PO4,PO7,PO9				
	and predict the sources of errors and impurities.	,PO11				
2	Explain theories, classifications, and reactions of acid-base	PO1,PO2,PO3,PO4,PO11				
	and non-aqueous titration involvement.					
3	Describe the basic principle, method, and application involved in	PO1,PO2,PO3,PO4,PO11				
	various methods of precipitation,					
	complexometric gravimetric, and diazotization titrations.					
4	Demonstrate adequate knowledge of basic principles and	PO1,PO2,PO3,PO4,PO11				
	techniques of redox titrations and their application in					
	pharmaceutical analysis.					
5	Illustrate the principle, electrodes used, and pharmaceutical	PO1,PO2,PO3,PO4 ,PO8,				
	application of electrochemical analysis methods.	PO11				

		SEMESTE	ER – I						
Course Title		Ph	armaceut	tics I					
Course code	BP103T	Total credits: 4	L	T	P	S	R	O/F	C
		Total hours: 45	3	1	0	0	0	0	4
Pre-requisite	Nil	Co-requisite				N	il		
Programme		Bache	lor of Ph	arma	acy				
Semester		Fall/ I semester of	first year	r of t	he Pro	ogram	me		
Course	Upon completion of	f this course the stude	nt should	be a	ble to:				
Objectives	1. Know the hist	ory of profession of p	harmacy						
-	2. Understand th	ne basics of differen	t dosage	form	ns, pha	ırmace	utical i	incompatib	ilities
	and pharmace	utical calculations							
	3. Understand the	e professional way of	handling	the 1	prescri	ption			
	4. Preparation of	various conventional	l dosage f	orms	1				
CO1	Retrieve the historic	cal background and d	evelopme	nt of	the Pl	narmac	y profe	ssion.	
CO2	Elaborate the know	ledge of the pharm	aceutical	calc	culatio	ns for	dosag	e forms,	
	applying mathemati	cal concepts in formu	ılation an	d adn	ninistra	ation.			
CO3	Illustrate the knowle	edge of monophasic a	and bipha	sic li	quid d	osage f	forms		
CO4	Explain and demons	strate the concept of s	supposito	ries a	nd pha	rmace	utical i	ncompatibi	ilities.
CO5		edge to formulate a							
	products proficiently	~					•		
Unit-No.	Coi	ntent	Contact	Lea	arning	Outco	ome		KL
			Hour						
I	Historical bac	ckground and		Stu	dents	will 1	oe able	e to learn	1
	development of	_		His	torical	ba	ickgrou	ınd and	l
	pharmacy: History	y of profession of		dev	elopm	ent o	f prof	fession of	1
	Pharmacy in Ind	ia in relation to		pha	rmacy				
	pharmacy educati	on, industry and							
	organization, Pharm	macy as a career,							
	Pharmacopoeias: I	ntroduction to IP,							
	BP, USP and Extra	Pharmacopoeia.							
	Dosage forms: Intr	oduction to dosage							
	forms, classification	n and definitions	10						1,2,
	Prescription: Defin	nition, Parts of							3
	rescription, handlin	g of Prescription							
	and Errors in prescr	ription.							
	Posology: Definition	on, Factors							
	affecting posology.	Pediatric dose							
	calculations based of	on age, body							
	weight and body su	rface area.							
II	Pharmaceutical ca	lculations:		Stu	dents	will be	e able	to learn	1,2,
	Weights And meas	ures – Imperial &	10	Pha	ırmace	utical	calcula	tions, solid	. 3
	Metric system, Calc			and	l liquid	l dosag	ge form		
	-	s, alligation, proof							
	-	solutions based on							
	freezing point and r								
	Powders: Definition								
	advantages and dis								
	& compound po								
	• •	usting powders,							
	effervescent,	efflorescent a							

	Liquid dosage forms: Advantages			
	and disadvantages of liquid dosage			
	forms. Excipients used in formulation			
	of liquid dosage forms. Solubility			
	enhancement techniques			
III	Monophasic liquids: Definitions and		Students will be able to learn	
	preparations of Gargles, Mouth		Monophasic and biphasic liquid	
	washes, Throat Paint, Eardrops, Nasal		dose	
	drops, Enemas, Syrups, Elixirs,			
	Liniments and Lotions.			
	Biphasic liquids:			
	Suspensions: Definition, advantages			
	and disadvantages, classifications,			
	Preparation of suspensions;			1,2,
	Flocculated and Deflocculated	10		3
	suspension &stability problems and			
	methods to overcome.			
	Emulsions: Definition, classification,			
	emulsifying agent, test for the			
	identification of type of			
	Emulsion, Methods of			
	preparation & stability problems and			
	methods to overcome.			
V	Semisolid dosage forms:	7	Students will be able to learn	1,2,
	Definitions, classification, mechanisms		Semi solid dosage form	3
	and factors influencing dermal			
	penetration of drugs.			
	Preparation of ointments, pastes,			
	creams and gels. Excipients used in			
	semi solid dosage forms. Evaluation of			
	semi solid dosages forms			

TEXT BOOKS:

- T1: H.C. Ansel et al., Pharmaceutical Dosage Form and Drug Delivery System, Lippincott Williams and Walkins, New Delhi.
- T2: Carter S.J., Cooper and Gunn's-Dispensing for Pharmaceutical Students, CBS publishers, New Delhi.

REFERENCE BOOKS:

- R1: Lachmann. Theory and Practice of Industrial Pharmacy, Lea& Fibiger Publisher, The University of Michigan.
- R2: Alfonso R. Gennaro Remington. The Science and Practice of Pharmacy, Lippincott Williams, New Delhi.

RELATIONSHIP BETWEEN COURSE OUTCOMES (CO) AND PROGRAM OUTCOMES

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Retrieve the historical background and development of the	PO1,PO2,PO3,PO6,PO7,PO8				
	Pharmacy profession.	,PO9,PO11				
2	Elaborate the knowledge of the pharmaceutical	PO1,PO2,PO3,PO6,				
	calculations for dosage forms, applying mathematical concepts	PO8,PO9,PO11				
	in formulation and administration.					
3	Illustrate the knowledge of monophasic and biphasic liquid	PO1,PO2,PO3PO4,				
	dosage forms	PO6,PO8,PO9,PO11				
4	Explain and demonstrate the concept of suppositories and	PO1,PO2,PO3PO4,				
	pharmaceutical incompatibilities.	PO6,PO8,PO9,PO11				
5	Apply the knowledge to formulate and evaluate semi-solid	PO1,PO2,PO3PO4,				
	pharmaceutical products proficiently.	PO6,PO8,PO9,PO11				

SEMESTER – I										
Course Title	•									
Course code	BP104T	Total credits: 4	L	T	P	S	R	O/F	C	
		Total hours: 45	3	1	0	0	0	0	4	
Pre-requisite	Nil	Co-requisite	Nil							
Programme			elor of P							
Semester		Fall/ I semester of			f the P	rograi	nme			
		f course student shal								
Course		rces of impurities an	d metho	ds to	detern	nine th	e impu	rities in inor	ganic	
Objectives	drugs and pha								_	
		e medicinal and phar			_		_	_		
CO1		of the Pharmacopoo		impı	urity s	ources	, and	various limi	t	
		g pharmaceutical imp								
CO2		of acids, bases, buff	ers, and	elec	trolyte	s and	summa	arize the nee	d for	
	dental products.									
CO3		oounds like acidifiers	s, antacio	is, ca	thartic	s, and	antımi	crobials play	ın	
	medicine and pharm		C 11'.'	1	. 11		1 .	1		
CO4		e and applications o						ances such as		
G07		orants, emetics, haem						· · · ·	1	
CO5	Describe the impor	tance of radiopharma			explai	in now	radioa	ctivity is mea	isurea.	
Unit-No.	Cor	ntent	Contac Hour	t	L	earnii	ıg Out	come	KL	
	Impurities in phai	maceutical	Hour	Str	idents	will be	able t	o learn the		
	substances: History of Pharmacopoeia,				Students will be able to learn the sources of impurities and quality					
	Sources and types of impurities,					_		ine the		
	principle involved in the limit test for				puritie					
	Chloride, Sulphate, Iron, Arsenic, Lead				armace		_			
	and Heavy metals, modified limit test			1						
I	for Chloride and Sulphate		10							
	General methods of preparation,									
	assay for the compo	ounds super scripted							1.2	
	with asterisk (*), pr	roperties and							1,2,	
	medicinal uses of i	norganic)	
	compounds belong	ing to the following								
	classes									
	Acids, Bases and							o learn the		
	equations and buffe					•		ceutical		
	general, buffers in pharmaceutical systems, preparation, stability, buffered				portan		_			
					-		-	mowledge or	1	
	isotonic solutions,		10				_	nostic		
	tonicity, calculation			_		-		and dental		
	adjusting isotonicity			•				d about		
	Major extra and in				ajor Ex		ntra Ce	ellular		
	electrolytes: Functi	-		ele	ectrolyt	es			1,2,	
		Electrolytes used in							3	
	the replacement the									
		m chloride, Calcium								
	gluconate* and Ora (ORS), Physiologic	•								
	balance.	ai aciu dase								
	varance.									

	Dental products: Dentifrices, role of fluoride in the treatment of dental caries, Desensitizing agents, Calcium carbonate, Sodium fluoride, and Zinc eugenol cement.			
III	Gastrointestinal agents Acidifiers: Ammonium chloride* and Dil. HCl Antacid: Ideal properties of antacids, combinations of antacids, Sodium Bicarbonate*, Aluminium hydroxide gel, Magnesium hydroxide mixture Cathartics: Magnesium sulphate, Sodium or thophosphate, Kaolin and Bentonite Antimicrobials: Mechanism, classification, Potassium permanganate, Boric acid, Hydrogen peroxide*, Chlorinated lime*, Iodine and its preparations	10	Students will be able to learn preparations and assay procedures of gastrointestinal agents, expectorants To understand about mechanism & Classification of Antimicrobials	1,2,
IV	Miscellaneous compounds Expectorants: Potassium iodide, Ammonium chloride*. Emetics: Copper sulphate*, Sodium potassium tartrate Haematinics: Ferrous sulphate*, Ferrous gluconate Poison and Antidote: Sodium thiosulphate*, Activated charcoal, Sodium nitrite Astringents: Zinc Sulphate, Potash Alum	8	Students will be able to gain knowledge about Emetics, understand about Haematinic, Astringents Poison& Antidote	1,2,
V	Radiopharmaceuticals: Radio activity, Measurement of radioactivity, Properties of α, β, γ radiations, Halflife, radio isotopes and study of radioisotopes - Sodium iodide I ¹³¹ , Storage conditions, precautions &pharmaceutical application of radioactive substances.	7	Students will be able to learn the measurement, storage and pharmaceutical applications of radiopharmaceuticals	1,2,

TEXT BOOKS:

T1: A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.

T2: A.I. Vogel, Text Book of Quantitative Inorganic analysis.

REFERENCE BOOKS:

R1: M.L Schroff, Inorganic Pharmaceutical Chemistry.

RELATIONSHIP BETWEEN COURSE OUTCOMES (CO) AND PROGRAM OUTCOMES

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Recall the history of the Pharmacopoeia, the impurity sources, and various limit tests for identifying pharmaceutical impurities.	PO1,PO3,PO7,PO9,PO11				
2	Examine the role of acids, bases, buffers, and electrolytes and summarize the need for dental products.	PO1,PO3,PO7,PO9,PO11				
3	Illustrate how compounds like acidifiers, antacids, cathartics, and antimicrobials play in medicine and pharmaceuticals.	PO1,PO3,PO7,PO9,PO11				
4	Define the relevance and applications of additional miscellaneous substances such as astringents, expectorants, emetics, haematinics, poisons, and antidotes.	PO1,PO3,PO7,PO9,PO11				
5	Describe the importance of radiopharmaceuticals and explain how radioactivity is measured.	PO1,PO3,PO4, PO7,PO9,PO11				

		SEMESTER	. – I						
Course Title	Communication skills								
Course code	BP105T	Total credits: 2	L	T	P	S	R	O/F	C
		Total hours: 30T	2	0	0	0	0	0	2
Pre-requisite	Nil	Co-requisite				N	il		
Programme		Bachelor							
Semester		Fall/ I semester of fin				gramı	me		
Course		f the course the student							
Objectives		ehavioral needs for a	pharma	cist 1	to fund	ction 6	effectiv	ely in the	areas
	of pharmaceutical	-	** 1	1\					
		ectively (Verbal and Nor		ıl)					
		ge the team as a team pla	ayer						
	Develop interview		1						
CO1		ip qualities and essentia			1 41	CC 4	C 1:		
CO1	1 -	munication concept, its	s barrier	s, an	ia the 6	effects	of div	erse	
CO2	communication per	-	, onle ol ox	100.0	n d voni	ana at	vlag of		ation
CO2		come familiar with nonveness, listening abilities,					yies oi	communic	auon.
CO3		erview process, practic					v. alai11	a and	
CO4	presentation technic		e, and	ueve	юр ш	iei vie	w SKIII	s and	
CO5	*	the group discussion 1	nrocess	and	nractic	e inte	rnercor	nal cooper	ative
03	and time manageme	• .	process	and	practic	c inte	грсгзог	iai, coopei	ative,
Unit-No.			Contact	Lea	rning	Outco	me		KL
			Hour	200	········s	oute	,,,,,,		112
I	Communication S	kills: Introduction,		Stud	dents v	vill be	e able	to gain	
	Definition, The Imp	portance of		com	nprehei	nsive ı	ınderst	anding	
	Communication, T	ne Communication		of c	ommu	nicatio	on's sig	nificance,	
	Process – Source, N	Message, Encoding,		proc	cess ef	fective	comm	nunication	
	Channel, Decoding	, Receiver,		and	unders	stand,	identify	and	
	Feedback, Context			ove	rcome	divers	e barrie	ers that	
	Barriers to comm	unication:		hino	der con	nmuni	cation	and	
	Physiological Barri				-		pective		
	Barriers, Cultural E						during		
	· ·	arriers, Interpersonal	7	proc	cess of	comn	nunicat	ion	1,2
	Barriers, Psycholog	gica Barriers,							
	Emotional barriers								
	Perspectives in Co								
	Introduction, Visua	•							
	Language, Other fa								
	perspective. Past Ex	-							
TT	Prejudices, Feelings			C4	14	:11 1	-1.1	. to 1	
II	Elements of Comr		7					to learn	
	Introduction, Face		7		_			tanding of	
	Communication -	·						ation and	
	Body Language (No	on-verbal		me	uses. I	ogain	er ine i	nowledge	
	communication),								

	Verbal Communication, Physical		of the different Communication	
	Communication, Communication		styles and apply them during	
	Styles: Introduction, The		communication.	
	Communication Styles Matrix with			
	example for each -Direct			
	Communication Style, Spirited			
	Communication Style, Systematic			
	Communication Style,			
	Considerate Communication Style			
Ш	Basic Listening Skills: Introduction,		Students will be able to learn the	
	Self-Awareness, Active Listening,		ways of listening and develop	
	Active Listener, Listening in Difficult		listening skills. To gather the	
	Situations		knowledge on writing skills and	
	Effective Written Communication:		apply when required.	
	Introduction, When and When Not to			
	Communication - Complexity of the			
	Topic, Amount of Discussion'			
	Required, Shades of Meaning, Formal	7		1,2
	Communication			
	Writing Effectively: Subject Lines,			
	Put the Main Point First, Know Your			
	Audience, Organization of the Message			
IV	Interview Skills: Purpose of an		Students will be able to learn the	
	interview, Do's and Don'ts of an		process of interview and learn	
	Interview		to face interview To understand	
	Giving Presentations: Dealing		the process presentation skills by	
	with Fears, planning your Presentation,	5	overcoming fears, Effectively	1,2
	Structuring Your Presentation,		planning and	
	Delivering Your Presentation,		structuring presentations, learn to	
	Techniques of Delivery		deliver with confidence	
V	Group Discussion: Introduction,		Students will be able to learn the	
	Communication skills in group		skills to engage effectively in	
	discussion, Do's and Don'ts of group	4	group discussions Through	1,2
	discussion		understanding their purpose.	

TEXT BOOKS:

T1: Basic communication skills for Technology, Andréa. J. Ruther Ford, 2nd Edition, Pearson Education, 2011.

T2: Communication skills, Sanjay Kumar, Pushpalata, 1stEdition, Oxford Press, 2011.

REFERENCE BOOKS:

R1: Organizational Behaviour, Stephen.P. Robbins, 1st Edition, Pearson, 2013.

RELATIONSHIP BETWEEN COURSE OUTCOMES (CO) AND PROGRAM OUTCOMES

CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome			
1	Recognize the communication concept, its barriers, and the	PO1,PO2,PO3,PO4,PO5,PO6			
1	effects of diverse communication perspectives.	,PO8,PO11			
2	Understand and become familiar with nonverbal cues and	PO1,PO2,PO3,PO4,PO5,PO6			
	various styles of communication.	,PO8,PO11			
3	Develop self-awareness, listening abilities, and writing	PO1,PO2,PO3,PO4,PO5,PO6			
3	abilities	,PO8,PO11			
4	Understand the interview process, practice, and develop	PO1,PO2,PO3,PO4,PO5,PO6			
4	interview skills and presentation techniques.	,PO8,PO11			
	Gain knowledge of the group discussion process and practice	PO1,PO2,PO3,PO4,PO5,PO6			
5	interpersonal, cooperative, and time management	,P08,P011			
	skills.	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			

		SEMESTE	R – I									
Course Title	Remedial Biolog	gy										
Course code	BP106RBT	Total credits: 2	L	T	P	S	R	O/F	C			
		Total hours: 30T	2	0	0	0	0	0	2			
Pre-requisite	Nil	Co-requisite				N	il					
Programme		Bache	lor of Pha	ırma	ıcy							
Semester		Fall/ I semester of				gramı	ne					
Course		n of the course, the stude										
Objectives		lassification and salient t			_							
		the basic components of	•	•	•		•					
	reference to											
CO1	_	g world and the morphol										
CO2		fluid, circulation, diges										
CO3		ncepts of the excretory s	ystem, ne	rvou	s syste	m, end	locrine	system, ar	ıd			
	Human Reprodu											
CO4		ts and mineral nutrition a										
CO5	_	nt respiration, growth an							ı			
Unit-No.	•	Content	Contact		Le	arning	Outco	ome	KL			
			Hour									
I		efinition and characters						to learn				
		ms Diversity in the						ing world				
	_	nomial nomenclature					the five					
		of life and basis of		1 7	_			about the				
		alient features of			_		Flower	rıng				
		a, Fungi, Animalia and		piai	nts par	ts						
	Plantae, Virus, N	s Morphology of							1.2			
		flowering plants –Root,	7						1,2,			
	_	ice, flower, leaf, fruit,	,)			
		natomy of Root, stem,										
		ledons &Dicotyledons.										
II	Body fluids and	<u>*</u>		Stu	dents	will be	able t	to learn				
		blood, blood groups,					r under					
	_	lood Composition and						stem and				
	_	ph Human circulatory					• •	Recall and	1,2,			
		of human heart and	7				1 about		3			
	1 *	rdiac cycle, cardiac		Dig	estive	systen	n Recal	l and				
	output and ECG			bett	ter und	lerstan	d about	The				
				resp	oirator	y syste	m					
	Digestion and A	bsorption										
		ry canal and digestive										
	glands Role of digestive enzymes											
		ption and assimilation										
	of digested food											
	Breathing and i	-										
	-	ry system Mechanism										
	_	its regulation Exchange										
	of gases, transpo	-										
	regulation of res	piration Respiratory										

	volumes							
Ш	Excretory products and their		Students will be able to Recall					
	elimination		and better understand about the					
	Modes of excretion Human excretory		Excretory system Recall and	ļ				
	system- structure and function		better understand about the	;				
	Urine formation Rennin angiotens in		Nervous system Recall and better					
	system		understand about The Endocrine					
	Neural control and coordination		system and reproductive system					
	Definition and classification of							
	nervous system Structure of a neuron							
	Generation and conduction of nerve							
	impulse Structure of brain and spinal							
	cord Functions of cerebrum,							
	cerebellum, hypothalamus and medulla							
	oblongata Chemical coordination and	7		1,2,				
	regulation Endocrine glands and their	,		3				
	secretions Functions of hormones							
	secreted by endocrine glands							
	Human reproduction							
	Parts of female reproductive system							
	Parts of male reproductive system							
	Spermatogenesis and Oogenesis							
	Menstrual cycle		Students will be able to learn					
	Plants and mineral nutrition:							
	Essential mineral, macro and		Recall and better understand	1.2				
	micronutrients Nitrogen metabolism,	-	about the Plants and mineral	1,2,				
	Nitrogen cycle, biological nitrogen	5	nutrition.	3				
	fixation		Recall and better understand					
	Photosynthesis		about the Photosynthsis					
	Autotrophic nutrition, photosynthesis,							
	Photosynthetic pigments, Factors							
	Affecting photosynthesis.							
V	Plant respiration: Respiration,		Students will be able to learn					
	glycolysis, fermentation (anaerobic).		Recall and better understand					
	Plant growth and development Phases		about the Plant respiration and					
	and rate of plant growth, Condition of		Plant growth and Development					
	growth, Introduction to plant growth		Recall and better understand					
	regulators		about the cell and tissue					
	Cell - The unit of life	4		1,2,				
	Structure and functions of cell and cell			3				
	organelles. Cell division							
	Tissues							
	Definition, types of tissues, location and							
	functions.							

T1: Text book of Biology by S. B. Gokhale

T2: A Text book of Biology by Dr. Thulajappa and Dr. Seetaram.

REFERENCE BOOKS:

R1: A Text book of Biology by B.V. Sreenivasa Naidu R2: A Text book of Biology by Naidu and Murthy.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Explain the living world and the morphology of	PO1,PO3,PO4,PO5,PO7,PO8
1	Flowering plants.	,РО9
2	Explain the body fluid, circulation, digestion, absorption,	PO1,PO3,PO4,PO5,PO7,PO8
	breathing, and respiration.	,РО9
3	Describe the concepts of the excretory system, nervous	PO1,PO3,PO4,PO5,PO7,PO8
3	system, endocrine system, and Human Reproduction.	,РО9
4	Explain the plants and mineral nutrition and	PO1,PO3,PO4,PO5,PO7,PO8
4	photosynthesis.	,РО9
5	Describe the plant respiration, growth and development,	PO1,PO3,PO4,PO5,PO7,PO8
3	cell, and tissue.	,РО9

		SEMESTE	ZR – I									
Course Title			lial Matl	nema	tics							
Course code	BP106RMT	Total credits: 2	L	T	P	S	R	O/F	C			
		Total hours: 45T	2	0					2			
Pre-requisite	Nil	Co-requisite	 			N	<u>1l</u>					
Programme		Bachelor of Pharmacy Fall/ I semester of first year of the Programme										
Semester	I In an a small stick as	f the course the studen				ogram	me					
Course Objectives	 Know the theory and their application in Pharmacy Solve the different types of problems by applying theory Appreciate the important application of mathematics in Pharmacy Know the theory and its application in Pharmacy. 											
CO1	•	* *		.1								
CO2		ypes of problems by			-							
CO3		ortant application of n				nacy.						
CO4		s of Analytical Geom										
CO5		l concepts and princ	_	_		_		or				
	Pharmaceutical Scie	ences and understandi	Contac		Cimica	1 Pnari	nacy.					
Unit-No.	Cor	ntent	Hour	١	Le	arning	g Outco	me	KL			
I	Partial fraction Int Polynomial, Rationa and Improper fracti fraction, Resolving fraction, Application Fraction in Chemica Pharmacokinetics Logarithms Introdu Theorems/Propertie Common logarithm and Mantissa, work application of logari pharmaceutical prob Function: Real Va Classification of re- Limits and continual Introduction, Limit Definition of limit of	al fractions, Proper ons, Partial g into Partial g into Partial in of Partial al Kinetics and action, Definition, as of logarithms, as, Characteristic and examples, that to solve olems. It was function, al valued function, al valued function, of a function	6	lea Pai		fract: Loga	rithms,	e to	1,2			
II	Matrices and Deter Introduction matrices, Operation Transpose of a matrices, Operation Multiplication, Deteror determinants, Mino Adjoint or adjugate Singular and non-si Inverse of a matrix, of linear of equation method, Cramer's requation and roots of	6				e able termina	to learn nt	1,2				

	Cayley– Hamilton theorem, application			
	of Matrices in solving			
	Pharmacokinetic equations			
III	Calculus Differentiation: Introductions, Derivative of a function, Derivative of a constant, Derivative of a product of a constant and a function, Derivative Of the sum or difference of two functions, Derivative of the product of two functions (product formula), Derivative of the quotient of two functions (Quotient formula) – Without Proof, Derivative of xn w.r.t x, where n is any rational number, Derivative of ex, Derivative of loge x, Derivative of ax, Derivative of trigonometric functions from first principles (without Proof), Successive Differentiation, Conditions for a function to be a maximum or a	6	Students will be able to learn Calculus	1,2
IV	minimum at a point. Application Analytical Geometry Introduction: Signs of the Coordinates, Distance formula, Straight Line: Slope or gradient of a straight line, Conditions for parallelism and perpendicular of two lines, Slope of a line joining two points, Slope Method of substitution, Method of Partial fractions, Integration by parts, definite integral, application	6	Students will be able to learn Analytical Geometry, Integration	1,2
V	Differential Equations: Some basic definitions, Order and degree, quations in separable form, Homogeneous equations, Linear Differential equations, Exact equations, Application in solving Pharmacokinetic equations Laplace Transform: Introduction, Definition, Properties of Laplace transform, Laplace Transforms of elementary functions, Inverse Laplace transforms, Laplace transform of derivatives, Application to solve Linear differential equations, Application in solving Chemical kinetics and Pharmacokinetics equations	06	Students will be able to learn Differential Equations, Laplace Transform	1,2

- T1: Differential Calculus by Shanthinarayan
- T2: Pharmaceutical Mathematics with application to Pharmacy by Panchaksharappa Gowda D.H.

REFERENCE BOOKS:

R1: Higher Engineering Mathematics by Dr.B.S.Grewal.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Know the theory and its application in Pharmacy.	PO1,PO2, PO3,PO4,PO8,PO11
2	Solve the different types of problems by applying theory.	PO1,PO2, PO3,PO4,PO8,PO11
3	Appreciate the important application of mathematics in Pharmacy.	PO1,PO2, PO3,PO4,PO8,PO11
4	Solve different kinds of Analytical Geometrical problems.	PO1,PO2, PO3,PO4,PO8,PO11
5	Apply mathematical concepts and principles to perform Computations for Pharmaceutical Sciences and understanding to help in Clinical Pharmacy.	PO1,PO2,PO3,PO4,PO8,PO11

		SEMESTE	R – I								
Course Title		Human Ana	tomy a	nd 1	Physiolog	gy					
Course code	BP107P	Total credits: 2	L	T	P	S	R	O/F	C		
		Total hours: 4	0	0	4	0	0	0	2		
Pre-requisite	Nil	Co-requisite				N	lil				
Programme		Bachel	lor of P	harı	nacy						
Semester		Fall/ I semester of first year of the Programme									
Course	1. Handle compou										
Objectives	-	2. Study the human organ system.									
		ological parameters.									
CO1		copical evaluation of		ıs tis	sues to u	ınders	tand t	he compone	ents of		
	· ·	and their mechanisms.									
CO2	Identify and unders	tand the axial and	append	dicul	ar bones	s and	their	location,			
	arrangements, and f	unctions of various be	ones.								
CO3	Analyze various he	ematological and card	diovasc	ular	paramete	ers, su	ich as	blood press	sure,		
	heart rate, and pulse	rate, and differentiate	e health	ıy fro	om diseas	sed inc	dividu	als.			
CO4	Identify and under	stand the various arr	angeme	ents	and orga	anizati	on of	different to	issues		
	and organs using me	odels and charts of the	e humai	n bo	dy.						
CO5	Estimate the hemat	ological parameters	by com	pari	ng with l	nealth	y indi	viduals to			
	understand the phys	iology of blood in hea	althy an	ıd di	seased in	dividu	ıals.				
Unit-No.	Con	ntent	Conta	ct	Le	arning	g Out	come	KL		
			Hou	r							
I	1. Study of compou	nd microscope.						to learn	1,2		
	2. Microscopic stu	dy of epithelial and			he differ	_					
	connective tissue			a	nd worki	ng me	chani	sm of			
	_	dy of muscular and		C	Compoun	d mici	oscop	e.			
	nervous tissue			S	tudents	will	be ab	le to learn	1,2		
	4. Identification of	axial bones		p	erform h	emato	logica	1			
	5. Identification of	appendicular bones		e	xperimer	ıts and	corre	late			
	6. Introduction to he	emocytometry.		h	aematolo	gical	param	neters with			
	7. Enumeration of	white blood cell		c	linical co	onditio	ons in	relevance			
	(WBC) count			to	the heal	thcare	;				
	8. Enumeration	of total red blood	4	S	tudents	will b	e able	to learn	1,2		
	corpuscles (RBC) c	ount		N	Aeasure o	ardio	vascul	ar			
	9. Determination of	-		p	arameter	S					
	10. Determination of	I		S	tudents	will b	e able	to learn	1,2		
	11. Estimation of he	-		a	bout the	micros	scopic	study of			
	12. Determination of			e	pithelial,	conne	ective,	muscular			
	13. Determination o	• •		a	nd nervo	ous					
	sedimentation rate ('		T	issue						
	14. Determination	of heart rate and		S	tudents	will b	e able	to learn	1,2		
	pulse rate.	_		Δ	bout axi	al and	apper	ndicular			
	15. Recording of bl	ood pressure.		b	ones.						

T1: Practical workbook of Human Physiology by K. Srinageswari and Rajeev Sharma, Jaypee brother's medical publishers, New Delhi.

REFERENCE BOOKS:

R1: Human Physiology (vol 1 and 2) by Dr. C.C. Chatterrje, Academic Publishers Kolkata.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Perform the microscopical evaluation of various tissues to understand the components of the cellular system and their mechanisms.	PO1,PO2, PO3,PO4,PO5, PO6,PO8,PO9,PO11
2	Identify and understand the axial and appendicular bones and recognize their location, arrangements, and functions of various bones.	PO1,PO2, PO3,PO4,PO5, PO6,PO8,PO9,PO11
3	Analyze various haematological and cardiovascular parameters, such as blood pressure, heart rate, and pulse rate, and differentiate healthy from diseased individuals.	PO1,PO2, PO3,PO4,PO5, PO6,PO8,PO9,PO11
4	Identify and understand the various arrangements and organization of different tissues and organs using models and charts of the human body.	PO1,PO2, PO3,PO4,PO5, PO6,PO8,PO9,PO11
5	Estimate the haematological parameters by comparing with healthy individuals to understand the physiology of blood in healthy and diseased individuals.	PO1,PO2, PO3,PO4,PO5, PO6,PO8,PO9,PO11

	SEMESTER – I											
Course Title	Pharmaceutical Analysis I											
Course code	BP108P	Total credits: 2	L	T	P	S	R	O/F	C			
		Total hours: 4	0	0	4	0	0	0	2			
Pre-requisite	Nil Co-requisite Nil											
Programme		Bachelor of Pharmacy										
Semester		Fall/ I semeste	r of first ye	ar of	the Pr	ogram	me					
Course	Gain hands-on	experience with sta	ndard analy	tical 1	reagent	s, labo	ratory	instruments	, and			
Objectives	glassware.	lassware.										
	_	standard solutions	-		metho	ds.						
	_	ssays through volur	•	sis.								
		hemical methods in	-									
	_	in handling and p	reparing pl	narma	ceutica	l anal	ytical s	samples and	d			
	substances.											
CO1		les to assess organi										
CO2		de, sulfate, Iron, an										
CO3	_	dardize Sodium h		_		d, Sod	lium th	iosulfate,				
		nganate, and ceric										
CO4		ity of Ammoniun										
		te, Hydrogen perox										
CO5		ormality of acids an										
Unit-No.	Co	ntent	Contact		Le	arnin	g Outc	ome	KL			
			Hour									
I	I Limit Test of th	e following						o learn				
	(1) Chloride			the limit of chloride, limit of iron,								
	(2) Sulphate			limit of sulphate, limit of arsenic, preparation and standardization of								
	(3) Iron				_							
	(4) Arsenic	4 . 4 4			dium l	-						
	_	d standardization			_			rdization of				
	of						_	eparation				
	(1) Sodium hyd							ammonium				
	(2) Sulphuric ac				phate S							
	(3) Sodium thic			understand the preparation and standardization of Ferrous								
	(4) Potassium p (5) Ceric ammo	•										
	III Assay of the	_	45		•				6			
	compounds along	•		understand the preparation and standardization of Sodium								
	Standardization							ll able to				
		ride by acid base						ion and				
	titration Ferrous	·				•	•	ion una				
	Cerimetry Coppe				standardization of Hydrogen peroxide Students w							
	Iodometry Calciu			-	_	•		preparation				
		Hydrogen peroxide					_	Sodium				
	by Permanganom				zoate S							
		aqueous titration						tion and				
	Sodium Chloride	_	standardization of Sodium									
	titration IV Dete			hydroxide Students will able to								
	Normality by ele	ectro-analytical						termination				
		tometric titration		of i	Normal	lity by	electro	- analytical				
	of strong acid ag	ainst strong base			thods S			-				

Conductometric titration of strong	understand about the
acid and weak acid against strong	determination of Normality by
base Potentiometric titration of	electro-analytical methods
strong acid against strong base	Students will able to understand
	about the determination of
	Normality by electro-analytical
	Methods

T1:Bentley and Driver's Textbook of Pharmaceutical Chemistry. T2:John H. Kennedy, Analytical chemistry principles.

REFERENCE BOOKS:

R1: P. Gundu Rao, Inorganic Pharmaceutical Chemistry.

	CO PO Mapping								
SN	Course Outcome (CO)	Mapped Program Outcome							
1	Applying principles to assess organic molecules using titration methods.	PO1,PO2,PO3,PO4,PO5,PO 6,PO7,PO8,PO11							
2	Determine chloride, sulfate, Iron, and arsenic content in pharmaceutical substances.	PO1,PO2,PO3,PO4,PO5,PO 6,PO7,PO8,PO11							
3	Prepare and standardize Sodium hydroxide, Sulphuric acid, Sodium thiosulfate, Potassium permanganate, and ceric ammonium sulfate.	PO1,PO2,PO3,PO4,PO5,PO 6,PO7,PO10,PO11							
4	Analyze the purity of Ammonium chloride, ferrous sulfate, Copper sulfate, Calcium gluconate, Hydrogen peroxide, Sodium benzoate, and Sodium Chloride.	PO1,PO2,PO3,PO4,PO5,PO 6,PO7,PO8,PO11							
5	Determine the normality of acids and bases by conductometry and potentiometry.	PO1,PO2,PO3,PO4,PO5,PO 6,PO7,PO8,PO10, PO11							

	SEMESTER – I										
Course Title		Pharmac	eutics I	- Pra	actical						
Course code	BP109P	Total credits: 2	L	T	P	S	R	O/F	C		
		Total hours: 4	0	0	4	0	0	0	2		
Pre-requisite	Nil	Co-requisite	4 51			N	il				
Programme			lor of Ph								
Semester		Fall/ I semester of				_					
	1. Demonstrate proficiency in weigh					hnique	es essei	ntial for			
	prescription dispensing.										
Course	2. Prepare solid powder dosage forms.										
Objectives	3.Estimate drug quantities needed for solutions of specific strengths.										
	4.Prepare, pack, and label various semisolid dosage forms.5.Design suitable liquid dosage forms for given drugs										
G01						1 .1					
CO1		nental calculation for									
CO2	_	ic liquid dose formula							ration.		
CO3		ng skills of making p			forms	as per	require	ements			
CO4		dge of formulating su									
CO5	Prepare semisolid d	osage forms, includir			as per r	equire	ments		ı		
Unit-No.	Con	ntent	Contac	t	Le	arning	Outco	ome	KL		
			Hour				,				
	1. Syrups										
	a) Syrup IP'66	Ć.		Stı	ıdents	will be	e able t	o learn to			
		ompound syrup of Ferrous hosphate BPC'68 xirs a) Piperazine citrate elixir		pre	pare ar	nd disp	ense Si	imple			
	_			prepare and dispense Simple syrup I.P., to prepare and							
				dis	pense I	Piperaz	ine citi	ate elixir,			
	b) Paracetamol pedi			to p	- orepare	and d	ispense				
	Linctus a) Terpin I IP'66 Solutions	Hydrate Linctus		Pai	racetam	ol pae	diatric	syrup, to			
		int (Mandles Daint)		pre	pare ar	nd disp	ense Ic	dine			
	b) Iodine Throat Par	· ·		Th	roat Pa	int, to	prepar	e and			
	a) Strong solution	oi ammomum		dis	pense (Cresol	with so	oap			
I	acetate b) Cresol with soar	n colution		sol	ution, t	o prep	are and	dispense			
1	c) Lugol's solution	`		Ca	lamine	lotion					
	5. Suspensions	1		,to	prepare	e and d	lispens	e Liquid			
	a) Calamine lotion			r			_	repare and			
	b) Magnesium Hy			dis	pense (ORS po	owder ((WHO), to			
	c) Aluminimum H			Г	pare ar	_					
	6. Emulsions a) Tur	•	45			_		prepare	3,4		
	b) Liquid paraffin e	•	13		_			owder, to	3,1		
	7. Powders and Gra			Г	pare ar	_		-			
	a) ORS powder (V						oare an	d dispense			
	b) Effervescent gra	,		Iod	line ga	rgle					
	c) Dusting powder										
	d) Divded powders			+							
	8. Suppositories										
	a) Glycero gelatin	suppository									
	b) Coca butter sup										
	c) Zinc Oxide supp										
	8. Semisolids										
	a) Sulphur ointmer	nt									

b) N	on-staining-iodine ointment with	
m	ethyl salicylate	
c) C	arbopal gel	
9. G	argles and Mouthwashes	
a) Io	odine gargle	
b) C	hlorhexidine mouthwash	

T1: Carter S.J., Cooper and Gunn's. Tutorial Pharmacy, CBS Publications, New Delhi.

REFERENCE BOOKS:

R1: Dilip M. Parikh: Handbook of Pharmaceutical Granulation Technology, Marcel Dekker, INC, New York.

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Analyze the fundamental calculation for calculating the	PO1,PO2,PO3,PO4,PO6,PO8				
1	dosage by the requirements.	,PO9,PO11				
2	Evaluate monophasic liquid dose formulations for both	PO1,PO2,PO3,PO4,PO6,PO8				
	internal and external administration.	,PO9,PO11				
3	Apply the formulating skills of making powder dosage	PO1,PO2,PO3,PO4,PO6,PO8				
3	forms as per requirements	,PO9,PO11				
4	Explain the knowledge of formulating suppositories	PO1,PO2,PO3,PO4,PO6,PO8				
4	Explain the knowledge of formulating suppositories	,PO9,PO11				
5	Prepare semisolid dosage forms, including cosmetics, as	PO1,PO2,PO3,PO4,PO6,PO8				
	per requirements	,PO9,PO11				

		SEMESTI	ER – I						
Course Title		Pharmaceutic	cal Inorga	anic	Chemi	stry			
Course code	BP110P Tota	al credits: 2	L	T	P	S	R	O/F	C
	Tota	al hours: 4	0	0	4	0	0	0	2
Pre-requisite	Nil Co-	requisite			'	N	il		
Programme	ne Bachelor of Pharmacy								
Semester	Fall/ I semester of first year of the Programme								
Course	Conduct purity	tests to assess E	Bentonite j	purit	y.				
Objectives	2. Evaluate the ne	eutralizing capac	ity of Alu	ıminı	ım hyd	roxide	Gel.		
	3. Prepare variou	s inorganic comp	pounds wi	th pl	armac	eutical	signifi	cance.	
		e, sulfate, iron, a					_		
CO1	Apply the theoretically				o the 1	numer	ous ion	s and met	als in
	the various compounds t								
CO2	Identify several inorgar	nic substances u	ısing vari	ous (chemic	als an	d evalı	ate the ca	tions
	and anions.								
CO3	Experiment on the swell	-							
CO4	Outline the importance					inum	hydrox	ide gel and	1
	Identify potassium iodat								
CO5	Formulate various inorg	anic pharmaceu	uticals lik	e B	oric ac	id, Po	otash A	Alum, and	
	Ferrous Sulfate.								
Unit-No.	Content	-	Contact		Lea	arning	g Outco	me	KL
_			Hour	_					
I	I Limit tests for following	-		Stu	dents v	vill be	able to	learn	
	T: : C . C11 : 1				. 1				
	Limit test for Chlorides	-				_	les with		
	Modified limit test for	-		staı	ndards	wheth		n pass or fail	
	Modified limit test for Sulphates	-		staı is re	ndards eported	wheth	er they	pass or fail	
	Modified limit test for Sulphates Limit test for Iron	Chlorides and		star is re Stu	ndards eported dents	wheth . will b	er they be able	pass or fail to learn	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met	Chlorides and		star is re Stu Ide	ndards eported dents ntify	wheth . will beach of	er they be able of the	pass or fail to learn Inorganic	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi	Chlorides and als t test for		star is re Stu Ide sub	ndards eported dents ntify of stance	wheth . will beach of by check	er they be able of the	pass or fail to learn	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi Arsenic II Identification	Chlorides and als t test for test		is re Stu Ide sub	ndards eported dents ntify stance ort ther	wheth will teach of by cl	er they be able of the nemical	pass or fail to learn Inorganic tests and	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi Arsenic II Identification Magnesium hydroxide F	Chlorides and als t test for test errous		star is re Stu Ide sub rep Stu	eported dents ntify estance ort ther dents	wheth will teach of by che must be will to	er they be able of the nemical	e to learn Inorganic tests and	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo	als t test for test errous onate Calcium		star is re Stu Ide sub rep Stu Org	eported dents ntify e stance ort ther dents ganize	wheth . will beach of by chen m will beach of the chen beach of the chen will beach of the chen beach of	er they be able of the nemical be able rity, ca	pass or fail to learn Inorganic tests and to learn pacity and	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha	chlorides and als t test for test ferrous onate Calcium		star is re Stu Ide sub rep Stu Org det	ndards eported dents ntify e stance ort ther dents ganize ermina	wheth . will the cach of the characteristics	er they be able of the nemical be able rity, ca	pass or fail to learn Inorganic tests and to learn pacity and bstance	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha III Test for purity Swel	als t test for test errous onate Calcium ate ling power of	45	star is re Stu Ide sub rep Stu Org det Stu	ndards eported dents ntify e stance ort ther dents ganize ermina dents v	wheth will to by che will to for pution of will be	oe able of the nemical oe able rity, ca the sul able to	e to learn Inorganic tests and to learn pacity and estance learn	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limi Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha III Test for purity Swel Bentonite Neutraliz	als t test for test errous onate Calcium ate ling power of ing capacity of	45	star is re Stu Ide sub rep Stu Org det Stu Pla	ndards eported dents ntify of stance ort ther dents ganize ermina dents v n prepa	wheth will teach of the control of	oe able of the nemical oe able rity, ca able to of Inorg	to learn Inorganic tests and to learn pacity and bstance learn ganic	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limit Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha III Test for purity Swel Bentonite Neutraliz aluminum hydroxide	als t test for test errous onate Calcium ate ling power of ing capacity of e gel	45	star is re Stu Ide sub rep Stu Org det Stu Pla	ndards eported dents ntify of stance ort ther dents ganize ermina dents v n prepa	wheth will teach of the control of	oe able of the nemical oe able rity, ca the sul able to	to learn Inorganic tests and to learn pacity and bstance learn ganic	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limit Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha III Test for purity Swel Bentonite Neutraliz aluminum hydroxide Determination of potassi	als t test for test errous onate Calcium ate ling power of ing capacity of e gel ium iodate and	45	star is re Stu Ide sub rep Stu Org det Stu Pla	ndards eported dents ntify of stance ort ther dents ganize ermina dents v n prepa	wheth will teach of the control of	oe able of the nemical oe able rity, ca able to of Inorg	to learn Inorganic tests and to learn pacity and bstance learn ganic	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limit Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha III Test for purity Swel Bentonite Neutraliz aluminum hydroxide Determination of potassi	als t test for test errous onate Calcium ate ling power of ing capacity of e gel ium iodate and de	45	star is re Stu Ide sub rep Stu Org det Stu Pla	ndards eported dents ntify of stance ort ther dents ganize ermina dents v n prepa	wheth will teach of the control of	oe able of the nemical oe able rity, ca able to of Inorg	to learn Inorganic tests and to learn pacity and bstance learn ganic	
	Modified limit test for Sulphates Limit test for Iron Limit test for Heavy met Limit test for Lead Limit Arsenic II Identification Magnesium hydroxide F sulphate Sodium bicarbo gluconate Copper sulpha III Test for purity Swel Bentonite Neutraliz aluminum hydroxide Determination of potassi	als t test for test errous onate Calcium ate ling power of ing capacity of e gel ium iodate and de anic	45	star is re Stu Ide sub rep Stu Org det Stu Pla	ndards eported dents ntify of stance ort ther dents ganize ermina dents v n prepa	wheth will teach of the control of	oe able of the nemical oe able rity, ca able to of Inorg	to learn Inorganic tests and to learn pacity and bstance learn ganic	

T1: A.H. Beckett & J.B. Stenlake's, Practical Pharmaceutical Chemistry Vol I & II, Stahlone Press of University of London, 4th edition.

REFERENCE BOOKS:

R1: Anand & Chatwal, Inorganic Pharmaceutical Chemistry.

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Apply the theoretically learned limit test principles to the numerous ions and metals in the various compounds used to make pharmaceuticals.	PO1,PO2,PO3,PO5,PO7,PO8 ,PO11				
2	Identify several inorganic substances using various chemicals and evaluate the cations and anions.	PO1,PO2,PO3,PO5,PO7,PO8 ,PO11				
3	Experiment on the swelling capacity of Bentonite and interpret the result.	PO1,PO2,PO3,PO5,PO7,PO8 ,PO11				
4	Outline the importance of the neutralizing capacity of Aluminium hydroxide gel and Identify potassium iodate and iodine in potassium iodide	PO1,PO2,PO3,PO5,PO7,PO8 ,PO11				
5	Formulate various inorganic pharmaceuticals like Boric acid, Potash Alum, and Ferrous Sulfate.	PO1,PO2,PO3,PO5,PO7,PO8 ,PO11				

			SEMESTI	ER – I							
Course Title			Comi	nunic	atio	n ski	lls				
Course code	BP111P		Total credits: 1	L		T	P	S	R	O/F	C
			Total hours: 2	0		0	2	0	0	0	1
Pre-requisite	N	Vil	Co-requisite		•			N	il		
Programme			Bache	elor of	Pha	ırma	ıcy				
Semester			Fall/ I semester of	first y	ear	of t	he Pro	gram	me		
Course	1.	Apply lear	rned skills to craft co	mpelli	ng b	ousin	ess wr	itten c	ommur	ication.	
Objectives		2. Compose formal emails and exhibit good interview etiquette.									
	3.	Communi	cate effectively in bu	isiness	and	heal	lthcare	scena	rios.		
	4.	Analyze la	anguage nuances, pro	onunci	atioı	n, an	d comi	nunica	ation re	quirement	š.
	5.	Prepare ef	fective written mater	ials.							
CO1	Enhance	communic	ation abilities and be	ecome	mor	e ade	ept at n	neeting	g and g	reeting oth	iers.
CO2	Learn and	d use appro	opriate sentence cons	structio	n, p	ronu	nciatio	on, and	l vocab	ulary.	
CO3	Develop l	listening sl	kills and understandi	ng skil	ls b	y exp	osing	variou	is speed	ches	
CO4	Enhance	presentation	on skills and become	e more	ade	pt in	writte	n con	munic	ation. Lea	rn to
	write an e	effective en	nail.								
CO5	Develop	interview-	handling skills, such	as con	nmı	ınica	tion sl	cills (V	erbal a	and nonve	rbal),
	self- pres	entation, n	egotiation skills, and	lactive	list	ening	g.				
Unit-No.		Co	ntent	Con	tact	Lea	rning	Outco	ome		KL
				Но	ur						
			on covering the							learn	
	following	g topics M	eeting People Asking	3		con	nprehe	nsive ı	ınderst	anding of	
	-	•	Friends What did you	1					-	icance,	
	do? Dos a	and Don'ts	s Pronunciations			pro	cess ef	fective	comm	unication,	
	_	the follow							mprehe		
	Pronuncia	ation (Con	sonant Sounds)			und	erstan	ding o	f Pronu	inciations	
	Pronuncia	ation and l	Nouns Pronunciation	ı		sign	nifican	ce, pro	ocess		
I	(Vowel S	Sounds) Ac	lvanced Learning	2				eff	ective		1,2
	Listening	Compreh	ension / Direct and			Pro	nuncia	tion			
	Indirect S	Speech Fig	ures of Speech			To	gain co	omprel	nensive	;	
	Effective	Communi	cation Writing Skills	S		und	erstan	ding o	f Effec	tive	
	Effective	Writing In	nterview Handling			Cor	nmuni	cation	Writin	g Skills	
	Skills E-N	Mail etique	ette			Effe	ective	Writin	g Inter	view	
	Presentat	ion Skills				Har	ndling	Skills	E-Mail	etiquette	
						Pre	sentati	on Ski	11s		

T1: Brilliant-Communication skills, Gill Hasson, 1stEdition, Pearson Life, 2011.

REFERENCE BOOKS:

R1: The Ace of Soft Skills: Attitude, Communication and Etiquette for success, Gopala Swamy Ramesh, 5thEdition, Pearson, 2013.

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Enhance communication abilities and become more adept	PO1,PO2,PO3,PO4,PO5,PO6				
1	at meeting and greeting others.	,PO8,PO11				
2	Learn and use appropriate sentence construction,	PO1,PO2,PO3,PO4,PO5,PO6				
	pronunciation, and vocabulary.	,PO8,PO11				
3	Develop listening skills and understanding skills by	PO1,PO2,PO3,PO4,PO5,PO6				
3	exposing various speeches	,PO8,PO11				
4	Enhance presentation skills and become more adept in	PO1,PO2,PO3,PO4,PO5,PO6				
•	written communication. Learn to write an effective email.	,PO8,PO11				
	Develop interview-handling skills, such as communication skills	PO1,PO2,PO3,PO4,PO5,PO6				
5	(Verbal and nonverbal), self-presentation,	,PO8,PO11				
	negotiation skills, and active listening.	,1 00,1011				

Objectives	ann a	SEMESTER – I											
Pre-requisite Programme Semester Course Objectives 2. Pe 3. Str CO1 Under CO2 Recall their m CO3 Catego CO4 Assess leaf, se CO5 Identif Unit-No. I 1. Int bio a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De coo 5. Mi ide Str	1DDD T	6/											
Programme Semester Course Objectives 2. Pe 3. Stu CO1 Under CO2 Recall their m CO3 Catego CO4 Assess leaf, so leaf, so Unit-No. I 1. Int bio a) Stu b) Se c) Mo d) Pe 2. Stu 3. Stu fru 4. De coo 5. Mi ide Sto	ZKBP TO	otal credits: 1	L	T	P	S	R	O/F		C			
Programme Semester Course Objectives 2. Pe 3. Stu CO1 Under CO2 Recall their m CO3 Catego CO4 Assess leaf, so leaf, so Unit-No. I 1. Int bio a) Stu b) Se c) Mo d) Pe 2. Stu 3. Stu fru 4. De coo 5. Mi ide Sto	Te	otal hours: 2	0	0	2	0	0	0		1			
Semester Course Objectives 2. Pe 3. Str CO1 Under CO2 Recall their m CO3 Catego CO4 Assess leaf, se leaf, se leaf, se co5 Identif Unit-No. I 1. Int bio a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide Str	Nil Co	o-requisite				N	il						
Course Objectives 2. Pe 3. Str CO1 Under CO2 Recall their m CO3 Catego CO4 Assess leaf, se leaf, se Unit-No. I 1. Int bio a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De coo 5. Mi ide Str		Bachelo	or of Pha	arma	ıcy								
Objectives	F	all/ I semester of f	irst year	of t	he Pro	gram	me						
3. Stu CO1 Under CO2 Recall their m CO3 Catego CO4 Assess leaf, se leaf, se Unit-No. I 1. Int bio a) Stu b) Se c) Mo d) Pe 2. Stu 3. Stu fru 4. De coi 5. Mi ide Stu	udy of cell and i												
CO1 Under CO2 Recall their man CO3 Categor CO4 Assess leaf, so CO5 Identife Unit-No. I 1. Into bio a) Strong by Second do Pe 2. Strong Strong CO5 Strong	-	pic study of tissues.											
CO2 Recall their m CO3 Catego CO4 Assess leaf, so leaf, s	udy of blood par												
their m CO3 Catego CO4 Assess leaf, so leaf, so CO5 Identif Unit-No. I 1. Int bio a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De coo 5. Mi ide Str		ng of microscope an	_										
CO3 Catego CO4 Assess leaf, so leaf, so CO5 Identif Unit-No. I 1. Int bio a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide Str		f a cell and its inclu	usions id	lentif	y vario	ous pla	ant pa	rts and	orga	nize			
CO4 Assess leaf, so l	nodifications.												
leaf, se CO5 Identif Unit-No. I 1. Int bic a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide Str		logy of frogs by usin											
I 1. Into bio a) Ste b) Se c) Mod d) Pe 2. Ste 3. Ste fru 4. De cor 5. Mi ide Ste		ical study and iden	tification	of 1	tissues	pertir	nent to	the st	em, r	oot,			
Unit-No. I 1. Int bic a) Str b) Se c) Mod d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide Str	eed, fruit, and flo												
I 1. Int bid a) Str. b) Se c) Md d) Pe 2. Str. 3. Str. fru 4. De cor. 5. Mi ide Str.	fy the bones and	determine blood gr	oup, blo	od pr	essure	and ti	dal vo	lume.					
bic a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide	Conte	ent	Contact		Lea	arning	g Outo	come		KL			
bic a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide			Hour										
a) Str b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De cor 5. Mi ide Ste	troduction to exp	periments in						le to l					
b) Se c) Mo d) Pe 2. Str 3. Str fru 4. De co 5. Mi ide Str	ology				roscop		-	of tis					
c) Mod) Pe 2. Stu 3. Stu fru 4. De co 5. Mi ide Ste	udy of Microsco	•			asuring	,	blood	re	lated				
d) Pe 2. Stu 3. Stu fru 4. De co 5. Mi ide Ste	ection cutting tec	•		para	ameters	5.							
2. Stu 3. Stu fru 4. De cor 5. Mi ide Ste	founting and stain	· ·											
3. Stu fru 4. De co 5. Mi ide Ste	ermanent slide pr	•											
fru 4. De co 5. Mi ide Ste	udy of cell and i												
4. De co. 5. Mi ide Ste	udy of Stem, Ro												
5. Mi ide Ste	*	neir modifications	20							1.0			
5. Mi ide Ste	4. Detailed study of frog by using 30 1,2									1,2			
ide Ste	•	1											
Ste	omputer models	5. Microscopic study and											
	omputer models icroscopic study												
	omputer models licroscopic study entification of ti	ssues pertinent to											
	omputer models licroscopic study entification of ti- em, Root Leaf, s	ssues pertinent to											
	omputer models licroscopic study entification of ti- tem, Root Leaf, s ower	ssues pertinent to seed, fruit and											
	omputer models licroscopic study entification of ti- em, Root Leaf, s ower entification of bo	ssues pertinent to seed, fruit and ones											
9. De	omputer models licroscopic study entification of ti- tem, Root Leaf, s ower	ssues pertinent to seed, fruit and ones blood group											
6. Ide 7. De	omputer models licroscopic study entification of ti- em, Root Leaf, s	ssues pertinent to							Stem, Root Leaf, seed, fruit and flower 6. Identification of bones				

- T1: Practical human anatomy and physiology. by S.R.Kale and R.R.Kale.
- T2: A Manual of pharmaceutical biology practical by S.B.Gokhale, C.K.Kokate and S.P.Shriwastava.

REFERENCE BOOKS:

R1: Biology practical manual according to National core curriculum. Biology forum of Karnataka. Prof .M.J.H.Shafi.

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Understand the handling of microscope and permanent slide preparation techniques.	PO1,PO2,PO5,PO7,PO8,PO9				
2	Recall the structure of a cell and its inclusions identify various plant parts and organize their modifications.	PO1,PO2,PO5,PO7,PO8,PO9				
3	Categorize the physiology of frogs by using computer models.	PO1,PO2,PO5,PO7,PO8,PO9				
4	Assess the microscopical study and identification of tissues pertinent to the stem, root, leaf, seed, fruit, and flower.	PO1,PO2,PO5,PO7,PO8,PO9				
5	Identify the bones and determine blood group, blood pressure and tidal volume.	PO1,PO2,PO5,PO7,PO8,PO9				

	SEMESTER – II									
Course Ti	ourse Title HUMAN ANATOMY AND PHYSIOLOGY-II									
Course co	ode BP 201T	Total credits: 4	L	T P S R O/F					C	
		Total hours: 45T	3	1	0	0	0	0	4	
Pre-requis		Co-requisite		_		N	il			
	Programme Bachelor of Pharmacy									
	SemesterFall/ II semester of first year of the programmeCourse1. Explain various human body organs' gross morphology, structure, and function									
Course Objective		the various homeostatic							ons.	
Objective		he various tissues and or								
		the hematological tests l	_				• •		on.	
		clotting time, and record					_			
	volume.	,	•		ĺ			•	•	
	5. Apprecia	te coordinated working p	pattern o	f diffe	erent o	rgans	of each	system		
	6. Apprecia	te the interlinked mecha	nisms in	the n	nainter	nance o	of norm	al function	ing	
	`	asis) of human body.								
CO1		atomy and physiology o	of the ce	ntral	nervou	ıs syst	em, ne	rve tracts,	and	
~~~	reflex actions		4							
CO2		e functions, secretion,	_						ın the	
CO2		tract, its disorders, and t							and	
CO3	their disorders.	natomy and physiology	or the	resp	iratory	and	urinary	systems	and	
CO4		arious endocrine glands	and inter	nret 1	the dif	ferent	hormo	nes their		
C04		heir pathological conditi		prec	tiic dii	iciciit	HOTHIO	nes, men		
CO5		atomy and physiology o		le an	d fema	ale rep	roducti	ive system	s and	
	_	processes related to the i				_		•		
Unit-No.	Co	ontent	Contac	t	Le	arning	g Outco	ome	KL	
			Hour							
	Nervous system							o learn		
	Organization of nerv	•					tructur			
	neuroglia, classificat					•	hysiolo ociated	~		
	r -	ibre, electrophysiology,								
	action potential, Nerve impulse, rece	ntorg gymanga			-	_	-	system, vledge of		
	neurotransmitters.	piors, synapse,			_			y anatomy		
		eam: Maningas				-	, kidile n, urine		'n	
	ventricles of brain a	entral nervous system: Meninges,		_			irition i		1,2,	
I		structure and functions	10					disorders	3	
	of brain (cerebrum, l						s well a			
	stem, cerebellum), s					-	of resp			
	structure, functions	,-			_	-	_	function,		
	and efferent nerve tr			1 -		•	lation,			
	and enterent herve tr	dots, reflex detivity)		1 ^			pacities	-		
					isport,			ciency		
					_		ration	-		
						ion me				
	Digestive system						be abl	e to		
	Anatomy of GI Tra	ct with enecial	6						1,2,	
	mutoffly of Of Tru	ct with special		lear	1 11				1,4,	
П	reference to anatom	-				classif	ication,	,	3	

	stance to reconstation of the stance of			
	stomach, regulation of acid production through parasympathetic nervous system, pepsin role in protein digestion) small intestine 54 and large intestine, anatomy and functions of salivary glands, pancreas and liver, movements of GIT, digestion and absorption of nutrients and disorders of GIT.  Energetics Formation and role of ATP, Creatinine Phosphate and BMR.		action, and the structure and functions of key endocrine glands such as the pituitary, thyroid, parathyroid, adrenal, pancreas, pineal, and thymus; further, they should be proficient in recognizing and explaining disorders associated with these glands.  Students will be able to learn	
Ш	Respiratory system Anatomy of respiratory system with special reference to anatomy of lungs, mechanism of respiration, regulation of respiration Lung Volumes and capacities transport of respiratory gases, artificial respiration, and resuscitation methods.  Urinary system Anatomy of urinary tract with special reference to anatomy of kidney and nephrons, functions of kidney and urinary tract, physiology of urine formation, micturition reflex and role of kidneys in acid base balance, role of RAS in kidney and disorders of kidney.	10	the intricate anatomy and functions of the gastrointestinal tract, including the stomach, small and large intestine, salivary glands, pancreas, and liver; furthermore, they should grasp the movements of the GI tract, processes of digestion and nutrient absorption, and be able to identify and describe disorders associated with the gastrointestinal system. Additionally, students should have a comprehensive knowledge of energetics, encompassing the formation and roles of ATP, creatinine phosphate, and basal metabolic rate (BMR).	1,2, 3
IV	Endocrine system Classification of hormones, mechanism of hormone action, structure and functions of pituitary gland, thyroid gland, parathyroid gland, adrenal gland, pancreas, pineal gland, thymus and their disorders	10	Students will be able to learn the anatomy and functions of the Male and female reproductive systems, including the roles of sex hormones and the physiological processes of menstruation, fertilization, spermatogenesis, oogenesis, pregnancy, and parturition. Additionally, students should possess a foundational knowledge of genetics, encompassing chromosomes, genes, DNA, protein synthesis, and the principles of genetic inheritance patterns	1,2,
V	Reproductive system Anatomy of male and female reproductive system, Functions of male and female reproductive system, sex hormones,	9	Students will be able to learn neural organization, ncompassing the structure and function of neurons, neuroglia, and nerve	1,2,

physiology of menstruation, fertilization,	fibers, while also grasping key
spermatogenesis, oogenesis, pregnancy	concepts such as
and parturition	electrophysiology, action
Introduction to genetics	potential, nerve impulse,
Chromosomes, genes and DNA, protein	receptors, synapses, and
synthesis, genetic pattern of inheritance	neurotransmitters; additionally,
	they should exhibit knowledge of
	the central nervous system,
	including the meninges,
	ventricles, cerebrospinal fluid,
	and the structural and functional
	aspects of the brain and spinal
	cord.

T1: Textbook of Practical Physiology by C.L. Ghai, Jaypeebrothers medical publishers, New Delhi. T2: Practical workbook of Human Physiology by K. Srinageswari and Rajeev Sharma, Jaypee brother's medical publishers, New Delhi.

### **REFERENCE BOOKS:**

- R1: Text book of Medical Physiology- Arthur C,Guyton and John.E. Hall. Miamisburg, OH, U.S.A. R2: Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MI USA
- R3: Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York.
- R4: Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypeebrothers medical publishers, New Delhi.

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
	Explain the anatomy and physiology of the central nervous system,					
1	nerve tracts, and reflex actions the anatomy and physiology of	PO1,PO5,PO6,PO8,PO11				
	various organs of the human body using models, charts, etc.					
	Understand the functions, secretion, digestion, and absorption of					
2	nutrients in the gastrointestinal tract, as well as its disorders, along	PO1,PO3,PO5,PO6,PO8,PO11				
	with the roles of ATP, creatinine, and BMR.					
3	Understand the anatomy and physiology of the respiratory	PO1,PO3,PO5,PO6,PO8,PO11				
3	and urinary systems, along with their disorders.	101,103,103,100,100,1011				
	Describe the various endocrine glands and interpreting the					
4	different hormones and their functions, along with the	PO1,PO3,PO5,PO6,PO8,PO11				
	pathological conditions.					
	Explain the anatomy and physiology of the male and female					
5	reproductive systems, along with the various life processes related	PO1,PO3,PO5,PO6,PO8,PO11				
	to the reproductive system.					

Course Ti	SEMESTER – II										
	tle PH	ARMACEUTICA	L ORG	ANI	C CHI	EMIST	ΓRY-I				
Course co	de BP 202T	<b>Total credits: 4</b>	L	T	P	S	R	O/F	C		
		Total hours: 45T	3	1	0	0	0	0	4		
Pre-requis		Co-requisite				N	il				
Programn			or of P								
Semester		Fall/ II semester of					ıme				
		1 1									
Course		2. Write the structure, name, and the type of isomerism of the organic compound									
Objective	20	3. Write the reaction, name the reaction and the orientation of the reactions  4. Account for the reactivity/stability of compounds									
		<ul><li>4. Account for the reactivity/stability of compounds</li><li>5. Identify/confirm the identification of organic compound</li></ul>									
							1	1f			
CO1	Classify Organic co structural isomerism	_		C no	menci	ature a	ına ma	ike use of			
	Relate Alkanes, Alk	-		nec o	nd Lit	iliza I	El and	F2 reaction	anc.		
CO2	with mechanism, imp										
CO2	Alder reaction	ortance of Addition	ii reactic	113 W I	iui iiici	ciiaiiisi	11 410112	3 WILLI DICI	3		
	Categorize and Justit	fy SN1 and SN2 rea	actions	with	mecha	nism a	nd ster	eochemistr	·V		
CO3	Outline the structures								΄,		
	Qualitative tests	s una uses of some c	ompou.	145, 4	114 1 111	ary 20 c	one one i	ouseu on			
	Discuss some named	d reactions that in	volve	Aldeh	vdes	and K	etones	with			
CO4	Qualitative Analysis				•						
	Describe the Import	Describe the Importance of the Acidity of Carboxylic acids and the Basicity of									
CO5	Amines, Analyze the	Amines, Analyze them by Qualitative Analysis with structures and uses of some									
	compounds										
Unit-No.	Conten	t	Contac	t	Le	arning	<b>Outco</b>	ome	KL		
			Hour	G.							
	Cl: 64:	-4						o learn			
	Classification, nomencla						of Org				
	<b>isomerism</b> Classification Compounds Common and							1			
	_ *				npoun						
T	of nomanalatura of organ	•		IUP	AC sy	stems	of nom	nenclature			
	of nomenclature of organ	ic compounds (up		IUP of c	PAC sy organic	stems comp	of nom ounds (	nenclature (Up to 10			
	to 10 Carbons open chain	ic compounds (up and carbocyclic	7	IUP of c Car	PAC sy organic bons c	stems comp pen cl	of nom ounds ( nain and	nenclature (Up to 10	12		
	to 10 Carbons open chain compounds) Structural is	ic compounds (up and carbocyclic	7	IUP of c Car carl	PAC sy organic bons c pocycl	stems compopen chic com	of nomounds (  ounds (  nain and  pounds	nenclature (Up to 10 d s) To	1,2		
	to 10 Carbons open chain	ic compounds (up and carbocyclic	7	IUP of c Car carl	PAC sy organic bons o ocycl erstan	vstems c comp open ch ic com d struct	of nomounds (  nain and pounds of tural in the second seco	nenclature (Up to 10	1,2		
	to 10 Carbons open chain compounds) Structural is organic compounds	and carbocyclic omerisms in	7	IUP of c Car carl und in o	PAC sy organic bons c occycl erstan organic	e compopen chic com d structum	of nomounds (nain and pounds etural in ounds	nenclature (Up to 10 d s) To somerism	1,2		
	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and	ic compounds (up and carbocyclic omerisms in	7	IUP of c Car cart und in o	PAC sy organic bons o pocycl erstan- organic dents	e component comp	of nomounds (nain and pounds etural in ounds et able t	nenclature (Up to 10 d s) To somerism	1,2		
	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio	and carbocyclic omerisms in  Conjugated on in alkanes,	7	IUP of c Car carb und in o Stu SP3	PAC sy organic bons coocycl erstan- organic dents hybri	e component comp	of nomounds (nain and pounds tural in ounds eable to all in all	nenclature (Up to 10 d s) To somerism to learn kanes,	1,2		
	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes,	and carbocyclic omerisms in  Conjugated on in alkanes, uses of paraffins.	7	IUF of of ce Cart cart und in o Stu SP3 Hal	PAC sy organic bons cocycl derstanderganic dents hybri ogena	e compopen chic compopen distruction of	of nomounds (nain and pounds etural in ounds et able t	nenclature (Up to 10 d s) To somerism to learn kanes,	1,2		
	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio	and carbocyclic omerisms in  Conjugated on in alkanes, uses of paraffins.  hybridization in	7	IUP of co Car cart und in o Stu SP3 Hal uses	PAC sy organic bons cocycl erstanderganic dents by hybric ogenation	e component comp	of nomounds (nain and pounds etural in ounds et able to ound e	nenclature (Up to 10 d s) To somerism to learn kanes,	1,2		
	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes, Stabilities of alkenes, SP2	and carbocyclic omerisms in  Conjugated on in alkanes, uses of paraffins. hybridization in ons – kinetics,	7	IUP of co Car carl und in o Stu SP3 Hal uses	PAC syorganic bons of pocyclerstand dents by hybric ogenants of paunders	e component of the comp	of nomounds (nain and pounds etural in ounds et able to ound e	nenclature (Up to 10 d s) To somerism to learn kanes, s, and illities of	1,2		
II	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes, Stabilities of alkenes, SP2 alkenes E1 and E2 reactions	and carbocyclic omerisms in  Conjugated on in alkanes, uses of paraffins. hybridization in ons – kinetics, al halides,	7	of control of carbon ca	PAC syorganic bons of pocyclerstand dents by hybric ogenants of paunders	e component of the comp	of nomounds (nain and pounds etural in ounds etural in ounds etural in ounds etural in ounds etural in alkane	nenclature (Up to 10 d s) To somerism to learn kanes, s, and illities of	1,2		
II	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes, Stabilities of alkenes, SP2 alkenes E1 and E2 reaction order of reactivity of alky	and carbocyclic omerisms in  I Conjugated on in alkanes, uses of paraffins. 2 hybridization in ons – kinetics, vI halides, attions, Saytzeffs		IUP of co Car carl und in o Stu SP3 Hal uses To alke	PAC sy organic bons of pocycl erstan- organic dents is hybri- ogena- s of pa unders enes, S	e compopen chic compopen chic compopen chic compopen chic compopen chic compopen will be dization of the compopen chican	of nomounds (nain and pounds etural in ounds etural in ounds etural in ounds etural in ounds etural in alkane	nenclature (Up to 10 d s) To somerism to learn kanes, s, and illities of tion in			
п	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes, Stabilities of alkenes, SP2 alkenes E1 and E2 reaction order of reactivity of alky rearrangement of carboca	and carbocyclic omerisms in  Conjugated on in alkanes, uses of paraffins. hybridization in ons – kinetics, which halides, ations, Saytzeffs s. E1 verses E2	7	IUP of co Car carb und in o Stu SP3 Hal uses To alke alke	PAC sy organic bons of boocycl erstander organic dents hybri ogenants s of pa unders enes, Senes unders	e component comp	of nomounds (nain and pounds etural in ounds etural in ounds etural in ounds etural in ounds etural in alkane	nenclature (Up to 10 d s) To somerism to learn kanes, s, and illities of tion in	1,2		
II	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes, Stabilities of alkenes, SP2 alkenes E1 and E2 reaction order of reactivity of alky rearrangement of carboca orientation and evidences	I Conjugated on in alkanes, uses of paraffins. 2 hybridization in ons – kinetics, v1 halides, ations, Saytzeffs s. E1 verses E2 ong E1 and E2		IUP of co Car carl und in o Stu SP3 Hal uses To alke alke reac	PAC sy organic bons of pocycl erstan- organic dents is hybri- ogena- s of pa unders enes, S enes unders ettions	e component of com	of nomounds (nain and pounds etural in ounds etural in ounds etural in ounds etural in ounds etural in alkane etural in alkan	nenclature (Up to 10 d s) To somerism to learn kanes, s, and illities of tion in			
II	to 10 Carbons open chain compounds) Structural is organic compounds  Alkanes*, Alkenes* and dienes* SP3 hybridizatio Halogenation of alkanes, Stabilities of alkenes, SP2 alkenes E1 and E2 reaction order of reactivity of alky rearrangement of carboca orientation and evidences reactions, Factors affecting	and carbocyclic omerisms in  Conjugated on in alkanes, uses of paraffins. hybridization in ons – kinetics, which halides, ations, Saytzeffs is. E1 verses E2 ong E1 and E2 ectrophilic		IUP of co Car carl und in o Stu SP3 Hal uses To alke alke reac	PAC sy organic bons of pocycl erstan- organic dents is hybri- ogena- s of pa unders enes, S enes unders ettions	e component of com	of nomounds (nain and pounds etural in ounds etural in ounds etural in ounds etural in ounds etural in alkane etural in alkane etural in alkane etural in alkane etural in etura	nenclature (Up to 10 d s) To somerism to learn kanes, s, and illities of tion in			

	Markownikoff's orientation, free radical		Saytzeffs orientation, and	
	addition reactions of alkenes, Anti		evidence. E1 verses E2 reactions,	
	Markownikoff's orientation. Stability of		Factors affecting E1 and E2	
	conjugated dienes, Diel-Alder,		reactions. Ozonolysis,	
			•	
	electrophilic addition, free radical addition		electrophilic addition reactions of	
	reactions of conjugated dienes, allylic		alkenes, Markownikoff's	
	rearrangement		orientation, free radical addition	
			reactions of alkenes, Anti	
			Markownikoff's orientation.	
			To understand the Stability of	
			conjugated dienes, Diel-Alder,	
			electrophilic addition, free radical	
			addition reactions of conjugated	
			dienes, allylic rearrangement	
III	Alkyl halides* SN1 and SN2 reactions		Students will be able to learn	
	-kinetics, order of reactivity of alkyl		SN1 and SN2 reactions - kinetics,	
	halides, stereochemistry and rearrangement		order of reactivity of alkyl halides,	
	of carbocations. SN1 versus SN2		stereochemistry, and	
	reactions, Factors affecting SN1 and		rearrangement of carbocations.	
	SN2 reactions Structure and usesof ethyl		To understand SN1 versus SN2	
	chloride, Chloroform, trichloroethylene,		reactions, Factors affecting SN1	
	tetrachloroethylene, dichloromethane,		and SN2 reactions Structure	
	tetrachloromethane and iodoform.		and uses of ethyl chloride,	
	Alcohols*- Qualitative tests, Structure		Chloroform, trichloroethylene,	2,3,
	and uses of Ethyl alcohol, Methyl alcohol,	10	tetrachloroethylene,	4
	chlorobutanol, Cetosteryl alcohol,	_	dichloromethane,	
	Benzyl alcohol, Glycerol, Propylene		tetrachloromethane and iodoform.	
			To understand Qualitative tests,	
			Structure, and uses of Ethyl	
			alcohol, Methyl alcohol, chloro-	
			butanol, Cetosteryl alcohol,	
			Benzyl alcohol, Glycerol,	
			Propylene glycol	
IV	Carbonyl compounds* (Aldehydes and		Students will be able to learn	
1 1 1	ketones) Nucleophilic addition,			
	, <u>.</u>		Nucleophilic addition,	
	Electromeric effect, aldol condensation,		Electromeric effect, To understand	
	Crossed Aldol condensation,		aldol condensation, Crossed Aldol	
	Cannizzaro reaction, Crossed Cannizzaro		condensation, Cannizzaro	1.2
	reaction, Benzoin condensation, Perkin	10	reaction, Crossed Cannizzaro	1,3,
	condensation, qualitative tests, Structure	10	reaction, Benzoin condensation,	4
	and uses of Formaldehyde, Paraldehyde,		Perkin condensation.	
	Acetone, Chloral hydrate, Hexamine,		To understand qualitative tests,	
	Benzaldehyde, Vanilin, Cinnamaldehyde.		Structure, and uses of	
			Formaldehyde, Paraldehyde,	
			Acetone, Chloral hydrate,	
			examine, Benzaldehyde, Vanilin,	
			Cinnamaldehyde	
V	Carboxylic acids* Acidity of carboxylic		Students will be able to learn	
	acids, effect of substituents on acidity,		the acidity of carboxylic acids,	
	inductive effect and qualitative tests for		the effect of substituents on	

carboxylic acids ,amide and ester		acidity, inductive effect, and	
Structure and Uses of Acetic acid, Lactic		qualitative tests for carboxylic	
acid, Tartaric acid, Citric acid, Succinic		acids, amide, and ester. To	
acid. Oxalic acid, Salicylic acid, Benzoic		understand the structure and	
acid, Benzyl benzoate, Dimethyl phthalate,		Uses of Acetic acid, Lactic acid,	
Methyl salicylate and Acetyl salicylic acid	8	Tartaric acid, Citric acid, and	2,3,
Aliphatic amines* Basicity, effect of		Succinic acid. Oxalic acid,	4
substituent on Basicity. Qualitative test,		Salicylic acid, Benzoic acid,	
Structure and uses of Ethanolamine,		Benzyl benzoate, Dimethyl	
Ethylenediamine, Amphetamine		phthalate, Methyl salicylate and	
		Acetyl salicylic acid To	
		understand the Basicity, the	
		effect of substituent on Basicity	
		of amines. To understand	
		Qualitative tests, Structure, and	
		uses of Ethanolamine,	
		Ethylenediamine, Amphetamine	

T1: Organic Chemistry by Morrison and Boyd.

T2: Textbook of Organic Chemistry by B.S. Bahl & ArunBahl.

## **REFERENCE BOOKS:**

R1: Organic Chemistry by I.L. Finar, Volume-I. R2: Organic Chemistry by P.L. Soni.

R3: Reaction and reaction mechanism by Chatwal. R4: Organic Chemistry by Clayden.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Classify Organic compounds based on IUPAC nomenclature and make use of structural isomerism in Organic compounds	PO1,PO3,PO8,PO11
2	Relate Alkanes, Alkenes, and Conjugated dienes and Utilize E1 and E2 reactions with mechanism, importance of Addition reactions with mechanism along with Diel's Alder reaction	PO1,PO3,PO11
3	Categorize and Justify SN1 and SN2 reactions, with mechanism and stereochemistry, Outline the structures and uses of some compounds, and Analyze alcohols based on Qualitative tests	PO1,PO3,PO11
4	Discuss some named reactions that involve Aldehydes and Ketones with Qualitative Analysis and distinguish some aldehydes with structures and uses	PO1,PO3,PO8,PO11
5	Describe the Importance of the Acidity of Carboxylic acids and the Basicity of Amines, Analyze them by Qualitative Analysis with structures and uses of some compounds	PO1,PO3,PO8,PO11

		SEMESTE	R – II						
Course Titl	e	BIC	CHEM	IISTR	RY				
Course code	e BP 203T	Total credits: 4	L	T	P	S	R	O/F	C
		Total hours: 45T	3	1	0	0	0	0	4
Pre-requisit	e Nil	Co-requisite				N	il		
Programme		Bach	elor of l	Pharn	nacy				
Semester		Fall/ II semester o				ogran	nme		
Course	1 -	pletion of course stude							
Objectives		d the catalytic role of	-		_		-		
		new drugs, and the the	_		_			-	
	conditions		.4:	C 41		. 1:		1 41	
		d the genetic organiz				_	genome	e and the	
CO1		of DNA in synthesizin	-	_			aiool in	nnortonoo	
COI		and applications of va				, biolog	gicai in	проглапсе	,
CO2	-	lamentals of metabolis				ne inv	olved :	in the	
	metabolism of b		m, proc		and Ste	ра ши	orveu 1	iii tiic	
CO3		tabolism of nutrient mo	olecules	in phy	vsiolog	ical an	d natho	ological	
	conditions.		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	p	, 210108		- Panii	21081001	
CO4		ımmalian genome's g	genetic	organ	nization	and	the D	NA's fu	nctions
		RNAs and proteins.		U					
CO5		catalytic role of enzy	mes, the	impo	rtance	of enz	yme in	hibitors in	the
	design of new dr	rugs, and the therapeuti	c and di	agnos	tic app	licatio	ns of e	nzymes.	
Unit-	Con	tent	Conta	ct	Le	arning	g Outc	ome	KL
No.			Hou	r					
	nolecules							to learn	
l I	· · · · · · · · · · · · · · · · · · ·	tion, chemical nature			princi	ples of	chemi	istry in	
		arbohydrate, lipids,		bic	ology				
	eic acids, amino ac	ids and proteins.							
	energetics	4	0						2.2
	cept of free energy,		8						2,3
	gonic reaction, Ref gy, enthalpy and er	ationship between free							
l l	ential. Energy rich c	= -							
Γ		al significances of ATP							
	cyclic								
	bohydrate metabo	lism		Stı	udents	will be	e able 1	to learn	
	colysis Pathway, en			the	metab	olism	of nutr	ient	
sign	ificance Citric acid	cycle- Pathway,		mo	lecules	s in ph	ysiolog	gical and	
ene	getics and significa	nce HMP shunt and		pat	thologi	cal cor	nditions	5	
	ignificance; Glucos	•							
	• , ,	deficiency Glycogen	10						2,3
	•	nd glycogen storage							
	, ,	eogenesis - Pathway							
	~	rmonal regulation of							
	d glucose level and	Diabetes mellitus							<u> </u>
	ogical oxidation	(ETC) 1:							
	etron transport chair								
mec	nanism. Oxidative j	phosphorylation & its							

	mechanism and substrate phosphorylation			
	Inhibitors ETC and oxidative phosphorylation			
	/ Uncouples			
Ш	Lipid metabolism β-Oxidation of saturated fatty acid (Palmitic acid) Formation and utilization of ketone bodies; ketoacidosis De novo synthesis of fatty acids (Palmitic acid) Biological significance of cholesterol and conversion of cholesterol into bile acids, steroid hormone and vitamin D Disorders of lipid metabolism: Hypercholesterolemia, atherosclerosis, fatty liver and obesity.  Amino acid metabolism General reactions of amino acid metabolism: Transamination, domination & decarboxylation, urea cycle and its disorders Catabolism of phenylalanine and tyrosine and their metabolic disorders (Phenyketonuria, Albinism, alkeptonuria, tyrosinemia) Synthesis and significance of biological substances; 5- HT, melatonin, dopamine, noradrenaline, adrenaline Catabolism of	10	Students will be able to learn the metabolism of nutrient molecules in physiological and pathological conditions	2,3
	heme; hyperbilirubinemia and jaundice			
IV	Nucleic acid metabolism and genetic information transfer Biosynthesis of purine and pyrimidine nucleotides Catabolism of purine nucleotides and Hyperuricemia and Gout disease Organization of mammalian genome Structure of DNA and RNA and their functions DNA replication (semi conservative model) Transcription or RNA synthesis Genetic code, Translation or Protein synthesis and inhibitors	10	Students will be able to learn the genetic organization of mammalian genome and functions of DNA in the synthesi of RNAs and proteins.	s 2,3
V	Enzymes Introduction, properties, nomenclature and IUB classification of enzymes Enzyme kinetics (Michaelis plot, Line Weaver Burke plot) Enzyme inhibitors with examples Regulation of enzymes: enzyme induction and repression, allosteric enzymes regulation Therapeutic and diagnostic applications of enzymes and isoenzymes Coenzymes—Structure and biochemical functions	7	Students will be able to learn the catalytic role of enzymes, importance of enzyme inhibitors in design of new drugs, therapeutic and diagnostic applications of enzymes.	2,3

- T1: Principles of Biochemistry by Lehninger.
- T2: Harper's Biochemistry by Robert K. Murry, Daryl K. Granner and Victor W. Rodwell. T3: Biochemistry by D. Satyanarayan and U.Chakrapani.

T4: Textbook of Biochemistry by Rama Rao. T5: Textbook of Biochemistry by Deb.

## **REFERENCE BOOKS:**

- R1: Biochemistry, C.B.Powar & G.R.Chatwal, Himalaya publishing house.
- R2: L. Stryer, Text Book of Bio Chemistry. W.H. Freemann & Co. Ltd. 6th Edition. R3: West, Edward, Text Book of Biochemistry; Freeman and company, Sanfransisco.
- R4: E.E.Conn and PK Stumpf, Outlines of Biochemistry; John Wiley and sons, New York

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
	Understand the basics of chemistry, function, classification,	PO1,PO2,PO3,PO4,PO6,PO7,
1	biological importance, qualitative tests, and applications of	PO8,PO9,PO11
	various biomolecules.	
	Clarify the fundamentals of metabolism, process, and steps	PO1,PO2,PO3,PO4,PO6,PO7,
2	involved in the metabolism of biomolecules.	PO8,PO9,PO11
2	Describe the metabolism of nutrient molecules in	PO1,PO2,PO3,PO4,PO6,PO7,
3	physiological and pathological conditions.	PO8,PO9,PO11
4	Analyze the mammalian genome's genetic organization and	PO1,PO2,PO3,PO4,PO6,PO7,
4	the DNA's functions in synthesizing RNAs and proteins.	PO8,PO9,PO11
	Elaborate on the catalytic role of enzymes, the importance of	PO1,PO2,PO3,PO4,PO6,PO7,
5	enzyme inhibitors in the design of new drugs, and the	PO8,PO9,PO11
	therapeutic and diagnostic applications of enzymes.	

				SEMESTE	R – II								
Course	Title			PATH	ЮРНУ	YSI	OLO	GY					
Course	code	BP 20	4T	<b>Total credits: 4</b>	L	T		P	S	R	O/F		С
				Total hours: 45T	3		1	0	0	0	0		4
Pre-req	uisite		Nil	Co-requisite					N	il			
Progra	mme				elor of								
Seme					of first year of the programme								
Cour			1 1										
Object	tives	<ol> <li>Describe the etiology and pathogenesis of the selected disease states.</li> <li>Name the signs and symptoms of the diseases; and</li> </ol>											
							ses; a	and					
GO				e complications of the					1		1 1:	.1	
CO	1			principles of cell inj	ury, ad	lapta	ation	and n	nechan	ısm ın	volved 1	n th	ie
CO	2			mation and repair	1				C 41	14-	1 1:		- C
CO	2	1		logy, pathogenesis, r, respiratory, and ren			ıgem	ent of	tne	selecte	a aisea	ses	OI
CO	3			r, respiratory, and renath physiological aspe			mata	logica	1 digaa	cec ce1	ected di	CAC	CAC
	J	1	_	nervous, and gastroin				_	ı uista	.scs, 801	ccica al	sca	363
CO	<u> </u>			ogy, pathogenesis an					natitie	cance	er hone	es e	and
	7		y the etion Disorders	ogy, pathogenesis an	a man	agei	iiiCiii	OI IIC	patitis	, cance	or, bone	٠٥, ١	ıııd
CO5 Explain the complications and pa				olications and path 1	hysiol	ogic	cal c	onditi	ons of	f selec	ted infe	ectio	ous
		_	-	smitted diseases	,	- 6							
Unit-			Cont		Cont	act	ct Learning Outcome					KL	
No.					Hot			Ü					
I	Basic	princip	oles of Cell	injury and			Stu	dents	will b	e ablo	e to lea	rn	
	Adapt	ation:	Introductio	n, definitions,			prin	ciples	of Ce	ll inju	ry To ga	ain	
	1		_	its and Types of			kno	wledg	e on c	ellular	adaptati	on	
	1	-		es of cellular injury,			To		lerstan			sic	
	1		•	brane damage,				hanisı		volved		the	
	1		_	Ribosome damage,			r		of in	flamma	ation a	nd	
			U // I	nology of cell injury			repa	iir					
	1 -		changes (A										
	1			a, Metaphase,									
	1 -		_	, Intra cellular									
	1			ion, Enzyme leakage &Alkalosis,	1.0								2.2
				asic mechanism	10	,							2,3
	1	-		of inflammation									
	and re		ne process	vi iiiiaiiiiiativii									
	1	_	Clinical s	gns of inflammation									
	1			mmation, Mechanisn									
	1			lteration in vascula	l l								
				d flow, migration o									
	r			inflammation, Basic									
	1		wound hea										
	skin, P	athoph	nysiology o	f Atherosclerosis									
П	Cardi	ovascu	lar System	: Hypertension,			Stu	dents	will be	e able t	to learn		
	_			ischemic heart			gros	ss path	physi	ology o	of		
			•	dial infarction,							To gair	ı	
				iosclerosis)				_		e gross	_		
	Respi	ratory	system: As	sthma, Chronic	10	)	phy	siolog	y of re	nal dis	eases. T	0	2,3

	obstructive airways diseases.		understand the gross path	
	Renal system: Acute and chronic renal		physiology of respiratory	
	failure		diseases	
Ш	Haematological Diseases:		Students will be able to learn	
	Iron deficiency, megaloblasticanemia (Vit		disease condition and clinical	
	B12 and folic acid), sickle cell anemia,		aspects of blood disorders To	
	thalasemia, hereditary acquired anemia,		gain knowledge on pathology of	
	hemophilia Endocrine system: Diabetes,		endocrine disease To understand	
	thyroid diseases, disorders of sex hormones		the path physiology of	
	Nervous system:	10	psychiatric disorders To gain	2,3
	Epilepsy, Parkinson's disease, stroke,		knowledge on pathological	
	psychiatric disorders: depression,		aspects of Gastrointestinal	
	schizophrenia and Alzheimer's disease.		system	
	Gastrointestinal system: Peptic Ulcer			
IV	Inflammatory bowel diseases, jaundice,		Students will be able to learn	
	hepatitis (A,B,C,D,E,F) alcoholic liver		path physiology of Inflammatory	
	disease. Disease of bones and joints:		bowel diseases To gain	
	Rheumatoid arthritis, osteoporosis and gout		knowledge on jaundice, hepatitis,	
	Principles of cancer: classification, etiology		To understand the path	
	and pathogenesis of cancer	8	physiology of joint disorders and	2,3
	Diseases of bones and joints:		cancer	
	Rheumatoid Arthritis, Osteoporosis, Gout			
	Principles of Cancer: Classification,			
	etiology and pathogenesis of			
V	Infectious diseases:		Students will be able to learn	
	Meningitis, Typhoid, Leprosy, Tuberculosis		path physiology of infectious	
	Urinary tract infections	7	disease and sexually transmitted	2,3
	Sexually transmitted diseases: AIDS,		diseases	
	Syphilis, Gonorrhea			

- T1: Vinay Kumar, Abul K. Abas, Jon C. Aster; Robbins & Cotran Pathologic Basis of Disease; South Asia edition; India; Elsevier; 2014.
- T2: Harsh Mohan; Text book of Pathology; 6th edition; India; Jaypee Publications; 2010.

## **REFERENCE BOOKS:**

- R1: Laurence B, Bruce C, Bjorn K.; Goodman Gilman's The Pharmacological Basis of Therapeutics; 12th edition; New York; McGraw-Hill; 2011.
- R2: Best, Charles Herbert 1899-1978; Taylor, Norman Burke 1885-1972; West, John B (John Burnard); Best and Taylor's Physiological basis of medical practice; 12th ed; united states;
- R3: William and Wilkins, Baltimore;1991 [1990 printing].

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Discuss the basic principles of cell injury, adaptation and mechanism involved in the process of inflammation and Repair	PO1,PO3,PO4,PO6,PO8,PO9, PO11
2	Describe the aetiology, pathogenesis, and management of the selected diseases of the cardiovascular, respiratory, and renal system	PO1,PO2,PO3,PO4,PO5,PO6, PO7,PO8,PO9,PO11
3	Understand the path physiological aspects of haematological diseases, selected diseases of the endocrine, nervous, and gastrointestinal system	PO1,PO3,PO4,PO6,PO8,PO9, PO11
4	Clarify the aetiology, pathogenesis and management of hepatitis, cancer, bones, and joint disorders	PO1,PO2,PO3,PO4,PO5,PO6, PO7,PO8,PO9,PO11
5	Explain the complications and path physiological conditions of selected infectious and sexually transmitted Diseases	PO1,PO3,PO4,PO5,PO6,PO7, PO8,PO9,PO11

	SEMESTER – II											
Course	Title		COMPUTER APP	LICAT	ON	S IN PH	IARM	ACY				
Course	code	BP 205T	Total credits: 3		T	P	S	R	O/F	C		
			Total hours: 30T	3	0	0	0	0	0	3		
Pre-rec	_	Nil	Co-requisite				N	il				
Progra				elor of F		•						
Seme			Fall/ II semester o					nme				
Cou		_	1. Upon completion of the course the student shall be able to									
Objec	tives	2. Know the various types of application of computers in pharmacy										
			3. Know the various types of databases									
			various applications of									
CC			comprehend the variou							ions.		
CC	)2		apply the concepts			on syst	tems a	and so	ftware			
			plemented in health car									
CC	03		s of HTML, XML, CS			rammin	g langi	uages a	and an			
<del>-</del>			web servers and server	•								
CO	<b>)</b> 4		nputers in pharmacy inc		_							
			nical pharmacy, e-preso						ection			
CC	<u>)5</u>		ta by applying comput									
Unit-		Con	tent	Contac	et	Le	arning	Outco	ome	KL		
No.				Hour								
I		•	nary number system,			udents						
		al number system						-	tems To			
	1 -	n, Hexadecimal n	*			derstan						
		rsion decimal to binary, binary to				nversio						
			y etc, binary addition,		binary arithmetic of addition and							
	1		ne's complement		subtraction To understand the concepts of an information							
	ľ	s complement me	•	6		_	or an ir	norma	uon	1,2		
	1	lication, binary			Sy	stem						
		pt of Information										
			gathering, requirement, data flow diagrams,									
		• •	input/output design,									
	r	-	ning and managing the									
	Project	• •	ining and managing the									
II	-		roduction to HTML,		St	udents	will be	ahle 1	to learn	1		
11		O	nming languages,						nderstand	1		
	1	uction to web ser				e conce	_					
			o databases, MYSQL,	6		nguages	•	•	•			
			cy Drug database			ncept of						
			, .			derstan						
						tabase r		_				
Ш	Applic	cation of comp	uters in Pharmacy	6		udents			•	1		
		_	age and retrieval,						armacy			
	_		hematical model in				_	_	e data of			
	Drug d	design, Hospital	and Clinical		m	edicatio	n infor	mation	, records			
	_	acy, Electronic I			an	d docur	nents,	sedate				
		-	s, barcode medicine		ad	ministra	ation.	Го аррі	raise the			
	identif	ication and auto	mated dispensing of		ap	plicatio	ns of c	omput	ers in			
	drugs,	mobile technolo	gy and adherence		ph	armacy	such a	ıs drug				

	monitoring Diagnostic System, Lab-	information services,	
	diagnostic System, Patient Monitoring	pharmacokinetics, mathematical	
	System, Pharma	model in drug design, hospital	
	Information System	and clinical pharmacy etc.,	
IV	Bioinformatics: Introduction, Objective of	Students will be able to learn	1
	Bioinformatics, Bioinformatics Databases,	the applications of computers in	
	Concept of Bioinformatics, Impact of	pharmacy such as drug	
	Bioinformatics in Vaccine Discovery	information services,	
		pharmacokinetics, mathematical	
		model in drug design, hospital	
		and clinical pharmacy etc.,	
V	Computers as data analysis in Preclinical	Students will be able to learn	1
	development: Chromatographic dada	the application of computers	
	analysis(CDS), Laboratory Information	clinical data analysis To	
	management System (LIMS) and Text	understand the concept of	
	Information Management System(TIMS)	Laboratory Information	
		Management System To	
		understand the concept of Text	
		Information Management System	

T1: Understand data analysis to analyze pharmaceutical product development using computers.

## **REFERENCE BOOKS:**

- R1: Computer Application in Pharmaceutical Research and Development –Sean Ekins Wiley-Interscience, A John Wi
- R2: Bioinformatics (Concept, Skills and Applications) S.C.Rastogi CBS Publishers and Distributors, 4596/1- A, 11 Darya
- R3: Microsoft office Access 2003, Application Development Using VBA, SQL Server, DAP and InfoPath Cary N.Prague Delhi 110002

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Remember and comprehend the various number systems	PO1,PO2,PO3,PO4,PO5,PO6,P					
1	used in computer applications.	O7,PO8,PO9,PO11					
2	Understand and apply the concepts of information systems and	PO1,PO2,PO3,PO4,PO5,PO6,P					
2	software technologies implemented in health care system	O7,PO8,PO9,PO11					
	Apply the basics of HTML, XML, CSS, and programming	PO1,PO2,PO3,PO4,PO5,PO6,P					
3	languages and an introduction to web servers and server	07,P08,P09,P011					
	products.	07,108,109,1011					
	Describe of computers in pharmacy include drug data	PO1,PO2,PO3,PO4,PO5,PO6,P					
4	storage, pharmacokinetics, hospital, and clinical pharmacy, e-	07,P08,P09,P011					
	prescribing, and barcode medicine detection	07,108,109,1011					
5	Evaluation of data by applying computers in experimental	PO1,PO2,PO3,PO4,PO5,PO6,P					
3	Development	O7,PO8,PO9,PO11					

SEMESTER – II													
Course	Course Title ENVIRONMENTAL SCIENCES												
Course code		BP 206T	Total credits: 3		T	P		S	R	O/F	C		
			Total hours: 30T	3	0	0		0	0	0	3		
Pre-requisite Nil			A .										
Progra				lor of I									
	Semester Fall/ II semester of first year of the programme												
		<ol> <li>Upon completion of the course, the student shall be able to:</li> <li>Create awareness about environmental problems among learners.</li> </ol>											
Objec	cuves		c knowledge about the	_				_		me			
		_	attitude of concern fo					ameu	proore	71118.			
		_	armers to participate in					ction	and en	vironment	a1		
		improvemen	• •	CIIVIIO	111110	iiiai pi	Oic.	Ction	and cn	VIIOIIIICIII	uı		
		_	lls to help the concern	ed indi	vidu	als ide	ntif	v and	solve o	environme	ntal		
		problems.						) 11111					
		_	ain harmony with Nat	ure									
CO	)1		onmental problems an		arnei	rs.							
CO	)2	Outline basic know	wledge about the envi	ronmer	nt an	d its al	lied	prob	lems.				
CO	)3	Develop an attitud	e of concern for the e										
CO		_	~ ~	nmental protection, fostering sustainable action.									
CO	)5	_	_	dividuals in identifying and solving									
	1	environmental pro									1		
Unit-		Conte	nt	Conta		]	Lea	rning	g Outco	ome	KL		
No.	TE1 . ) (	- 1,-1 1-		Hour		14 1		•11 1					
I	1	ultidisciplinary nature of nmental studies Natural Resources  Students will be able to lear Create awareness about											
		nmental studies in able and non- rene											
		able and non- tene		- 1			_		ns among ness about	1,2			
		resources; b) Wate	10						iess about	1,2			
		al resources; d) Foo	10	- 1			_		titude of				
		· ·	urces; f) Land resources: Role of						_				
		ividual in conserva			ľ	concern for the environment.							
	resources												
II													
	-	are and function of	•		E	Ecosyst	em	and t	heir fu	nctions			
	Introdu	action, types, chara	acteristic features,		a	nd val	ues	Effec	tively	apply			
	structu	re and function of	the ecosystems:		b	asic p	inc	iples	of the 1	natural and	l		
	Forest	ecosystem; Grassla	and ecosystem;		S	ocial s	cier	ices to	o curre	nt issues			
		ecosystem; Aquat	•		o	of natu	al r	esour	ces and	d the	1,2,		
		s, streams, lakes, ri	vers, oceans,	10		nviron					3		
	estuari	es)				-			dge abo				
										riately use			
										natural			
										evant to			
										ources			
										dentify			
						-			al issue				
									s and tl d be ab				
									u be ab				
					a	uuress	ıne	se iss	ues in	all			

			informed and thoughtful manner	
Ш	Environmental Pollution: Air pollution;		Students will be able to learn	
	Water pollution; Soil pollution		environmental pollutants Know	
			about the different types of host	
			of the pollutions Effect of	1,2,
		10	pollution in environment as well	3
			as human healthTo know about	
			the precautionary measures of	
			environmental pollutions	

T1: Y.K. Sing, Environmental Science, New Age International Pvt, Publishers, Bangalore T2: De A.K., Environmental Chemistry, Wiley Eastern Ltd.

### **REFERENCE BOOKS:**

- R1: Agarwal, K.C. 2001 Environmental Biology, Nidi Publ. Ltd. Bikaner.
- R2: BharuchaErach, The Biodiversity of India, Mapin Publishing Pvt. Ltd., Ahmedabad 380 013, India,
- R3: Brunner R.C., 1989, Hazardous Waste Incineration, McGraw Hill Inc. 480p R4: Clark R.S., Marine Pollution, Clanderson Press Oxford
- R5: Cunningham, W.P. Cooper, T.H. Gorhani, E & Hepworth, M.T. 2001, Environmental Encyclopedia, Jaico Publ. House, Mumbai, 1196.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Understand environmental problems among learners.	PO1,PO4,PO7,PO10,PO11					
2	Outline basic knowledge about the environment and its allied problems.	PO1,PO4,PO7,PO10,PO11					
3	Develop an attitude of concern for the environment.	PO1,PO4,PO7,PO10,PO11					
4	Outline proactive engagement in environmental protection, fostering sustainable action.	PO1,PO3,PO4,PO7,PO10,PO11					
5	Acquire skills to help the concerned individuals in identifying and solving environmental problems.	PO1,PO3,PO4,PO7,PO10,PO11					

SEMESTER – II											
Course	Title		Human Anatomy And Physiology								
Course code		BP 207P	<b>Total credits: 2</b>	L	Т	P	S	R	O/F	С	
			Total hours: 4	0	0	4	0	0	0	2	
Pre-requisite		Nil	Co-requisite					Nil			
Progra	mme		Bach	elor of	Pha	rmacy					
Semes	ter		Fall/ II semester	of first y	ear	of the	progr	amme			
Cour	se	1. Identify hu									
Object	ives	2. Conduct various blood estimations.									
		3. Measure human heart rate, blood pressure, body temperature, BMI, and blood									
		oxygen levels using diverse methods.									
			lung capacities usin								
CO	1	Understand the ar	natomy and physiolog	gy of vai	ious	s humar	body	organs	using mo	dels,	
		charts, etc.									
CO	2		nanent slides of vital								
CO	3		ent types of lung cap	-							
CO	4		gical reflex testing, bo							ing, etc.	
CO	5		lain family planning			1 0			tests.		
Unit-		Conte	Contac	t L	earning	g Outc	ome		KL		
No.				Hour							
I	1		mentary and special						to learn	3,4,5	
	S	enses using specin		- 1		_		ts related			
		o study the r		to special senses and nervous							
		sing specimen, mo		1 -	ystem, v			ostatic			
		o demonstrate the			nechanis						
	l	eurological exami		- 1		_		phology,			
		o demonstrate the			tructure						
	1	lfactory nerve	0			arious o	rgans	of the h	numan		
	ı		ferent types of taste.		bo	ody					
		o demonstrate the	•								
	1	o demonstrate the	•								
			dal volume and vital								
		apacity. tudy of digestive,	ragniratory	4							
		ardiovascular syst	•	4							
		•	ns with the help of								
	l	nodels, charts and	•								
		tecording of basal	•								
	l	~	nning devices and								
		regnancy diagnosi	•								
1 -		Demonstration of to									
	l	ell analyser									
			f vital organs and								
		onads.	5								
			locrine system								
		·	· · · · · · · · · · · · · · · · · · ·								
	,										
			•								
		edback mechanism									
	13. P g 14. T u 15. Re	ermanent slides of onads. To study the end sing specimen, mo cording of body to emonstrate positive	docrine system dels, etc emperature To e and negative								

T1: Textbook of Practical Physiology by C.L. Ghai, Jaypeebrothers medical publishers, New Delhi. T2: Practical workbook of Human Physiology by K. Srinageswari and Rajeev Sharma, Jaypee brother's medical publishers, New Delhi.

### **REFERENCE BOOKS:**

- R1: Text book of Medical Physiology- Arthur C,Guyton and John.E. Hall. Miamisburg, OH, U.S.A. R2: Physiological basis of Medical Practice-Best and Tailor. Williams & Wilkins Co, Riverview, MLUSA
- R3: Anatomy and Physiology in Health and Illness by Kathleen J.W. Wilson, Churchill Livingstone, New York.
- R4: Essentials of Medical Physiology by K. Sembulingam and P. Sembulingam. Jaypeebrothers medical publishers, New Delhi.

CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome			
1	Understand the anatomy and physiology of various human	PO1,PO2,PO3,PO4,PO6,,PO8			
1	body organs using models, charts, etc.	.0O9,PO11			
2	Analyze the permanent slides of vital organs and gonads	PO1,PO2,PO3,PO4,PO6,PO1			
	Analyze the permanent shoes of vital organis and gonads	0,PO8.0O9,PO11			
3	Discuss the different types of lung capacity tests and their	PO1,PO2,PO3,PO4,PO5,			
	estimation.	PO6,PO10,PO8.0O9,PO11			
1	Perform neurological reflex testing, body temperature	PO1,PO2,PO3,PO4,PO5,PO6,			
•	measurement, BMR recording, etc.	PO10,PO8.0O9,PO11			
5	Describe and explain family planning devices and	PO1,PO2,PO3,PO4,PO6,PO1			
3	pregnancy diagnosis tests.	0,PO8.0O9,PO11			

SEMESTER – II										
Course Title	P	HARMACEUTICA	AL OR	GANI	C CHI	EMIST	ΓRY-I			
Course code		otal credits: 2		T	P	S	R	O/F	C	
		otal hours: 4	0	0	4	0	0	0	2	
Pre-requisite		Co-requisite				N	<u>il</u>			
Programme			elor of l							
Semester		Fall/ II semester o	f first y	ear of	f the pr	ogran	ıme			
Course		nic compounds.	. ,	C			1			
Objectives	I	e melting and boilin			-	_	nas.			
		lecular models of va		-	_	ounas.				
CO1		minary tests and ele				a a dina	an faty	musta sala 4		
CO1	Identify standard la								or	
	chemical handling,		nowied	ge oi	the pre	ıımına	ry tests	s used in		
CO2	qualitative organic and Explain the princip	•	action o	f ovrte	a alama	anta tla	مادردوس	Laggaiana	) a tast	
CO2	and describe the sig								s test	
CO3	Apply solubility te									
COS	compounds in diffe			•				_	ng in	
	the identification Pr		пстргст	tile ie.	suits to	make	IIIIOIII	ica accisio	)115 III	
CO4	Analyze and differe		nnounds	s hase	d on 1	their fi	unction	al groups		
004	through systemic qu	_	проини	o oase	u on i	inen i	unction	ar groups	,	
CO5		<u>*</u>	onfirm	the ide	entity o	of the 11	nknow	n compou	nd	
		Synthesize suitable solid derivatives, confirm the identity of the unknown compound hrough melting point/boiling point determination, and construct various molecular								
	models.			.011, 011				11010001011		
Unit-	Conten	t	Conta	ct	Le	arning	Outco	ome	KL	
No.			Hou	r			,			
I Syst	ematic qualitative anal	ysis of unknown		Stu	ıdents	will be	e able t	o learn		
	nic compounds like			Lal	borator	y Glas	sware ı	usage and		
Pre	liminary test: Color, o	dour,		Safety Measurements with						
alip	phatic/aromatic compo	unds, saturation		precautions in the ,laboratory.						
l l	unsaturation, etc.			I		-		olor, odor,		
l l	tection of elements lik	•						pounds,		
	phur and Halogen by l	Lassaigne's test					nsatura			
	ubility test							ments like		
l I	nctional group test like				_	_		Halogen		
	ea, Carbohydrates, Am	-			Lassai	_		1 :1:		
	s, Aldehydes and Ketones, Alcohols,						ia Calii	hility of		
	-		4		unders			•	3,4	
l l	ers, Aromatic and Hal	ogenated	4	var	ious or	ganic	compo	unds.	3,4	
Hy	ers, Aromatic and Hal drocarbons, Nitro com	ogenated	4	var To	rious or unders	ganic (stand th	compone func	unds. tional		
Hy-	ers, Aromatic and Hal drocarbons, Nitro com ilides.	ogenated pounds and	4	var To gro	rious or unders oup test	ganic of stand the stand the	compoi ne func Phenols	unds. tional s, Amides/		
Hy An Me	ers, Aromatic and Haldrocarbons, Nitro comilides.  Iting point/Boiling point/Boiling	ogenated pounds and	4	var To gro Ure	rious or unders oup test ea, Car	ganic of tand the stand the stand the stand the standard	compone func Phenols rates, A	unds. tional s, Amides/ amines,		
Hy An Me cor	ers, Aromatic and Hal drocarbons, Nitro com ilides. lting point/Boiling poin pounds	ogenated pounds and int of organic	4	var To gro Uro Car	ious or unders oup test ea, Car rboxyli	ganic of stand the stand the stand the stand the standard	compoune func Phenols rates, A s, Aldel	unds. tional s, Amides/ mines, hydes and		
Hy An Me cor Ide	ers, Aromatic and Haldrocarbons, Nitro comilides. Iting point/Boiling pointpounds antification of the unkn	ogenated pounds and int of organic own compound	4	var To gro Uro Ca: Ke	unders oup test ea, Car rboxyli tones,	rganic of stand the stand the standard bohydr ic acids Alcoho	compone func Phenols rates, A s, Aldel ols, Este	unds. tional s, Amides/ mines, hydes and ers,		
Hy An Me con Ide fro	ers, Aromatic and Haldrocarbons, Nitro comilides. Iting point/Boiling pointpounds Intification of the unknown the literature using the	ogenated pounds and int of organic own compound	4	var To gro Uro Ca: Ke Aro	unders unders oup test ea, Car rboxyli tones, A	rganic of stand the stand the stand stand bohydr ic acids Alcoho and H	compone ne func Phenols rates, A s, Aldel ols, Este Jalogen	unds. tional s, Amides/ mines, hydes and ers,		
Hy An Me cor Ide fro	ers, Aromatic and Haldrocarbons, Nitro comilides. Iting point/Boiling pointpounds Intification of the unknown the literature using the ling point.	ogenated pounds and int of organic own compound melting point/	4	var To gro Uro Ca: Ke Aro	rious or unders oup test ea, Car rboxyli tones, a omatic drocarl	rganic of stand the stand the bohydric acids Alcohomand H bons, 1	compone func Phenols rates, A s, Aldel ols, Este Ialogen Nitro	unds. tional s, Amides/ amines, hydes and ers, ated		
Hy An Me con Ide fro boi Pre	ers, Aromatic and Haldrocarbons, Nitro comilides. Iting point/Boiling pointpounds Intification of the unknown the literature using a ling point.  In paration of the literature of the literature using a ling point.	ogenated pounds and int of organic own compound melting point/ derivatives and	4	var To gro Uro Ca: Ke Aro	rious or unders oup test ea, Car rboxyli tones, a omatic drocarl	rganic of stand the stand the bohydric acids Alcohomand H bons, 1	compone ne func Phenols rates, A s, Aldel ols, Este Jalogen	unds. tional s, Amides/ amines, hydes and ers, ated		
Hy An Me cor Ide fro boi Pre conf	ers, Aromatic and Haldrocarbons, Nitro comilides. Iting point/Boiling pointpounds Intification of the unknown the literature using the ling point. In paration of the unknown of the unknown the literature using the literature unknown the unknown the literature using the unknown the literature using the	ogenated pounds and int of organic own compound melting point/ derivatives and own compound	4	var To gro Uro Ca: Ke Aro	rious or unders oup test ea, Car rboxyli tones, a omatic drocarl	rganic of stand the stand the bohydric acids Alcohomand H bons, 1	compone func Phenols rates, A s, Aldel ols, Este Ialogen Nitro	unds. tional s, Amides/ amines, hydes and ers, ated		
Hy An Me con Ide fro boi Pre conf by m	ers, Aromatic and Haldrocarbons, Nitro comilides. Iting point/Boiling pointpounds Intification of the unknown the literature using a ling point.  In paration of the literature of the literature using a ling point.	ogenated pounds and int of organic own compound melting point/ derivatives and nown compound oint.	4	var To gro Uro Ca Ke Aro Hy	rious or unders oup test ea, Car rboxyli tones, a omatic drocar mpound	rganic of stand the stand the bohydra a cacids Alcoho and H bons, 1	compone func Phenols rates, A s, Aldel ols, Este Ialogen Nitro	unds. tional s, Amides/ amines, hydes and ers, ated		

Preparation of suitable solid derivatives	compounds.
from organic compounds	To identify the unknown
Construction of molecular models	compounds from the literature
	using MP/BP. Preparation of the
	derivatives and confirmation of
	the unknown compound by
	melting point/ boiling point.
	Preparation of suitable solid
	derivatives from organic
	compounds

T1: Practical Organic Chemistry by Mann and Saunders T2: Vogel's textbook Practical Organic Chemistry.

### **REFERENCE BOOKS:**

- R1: Introduction to Organic Laboratory techniques by Pavia, Lampman and Kriz. R2: Advanced Practical organic chemistry by N.K.Vishnoi.
- R3: Reaction and reaction mechanism by Ahluwaliah/Chatwal.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Identify standard laboratory equipment and its functions, describe safety protocols for chemical handling, and demonstrate knowledge of the preliminary tests used in qualitative organic analysis.	PO1,PO2,PO3,PO5,PO8,PO1 0 ,PO11
2	Explain the principles behind the detection of extra elements through Lassaigne's test and describe the significance of these tests in organic compound identification.	PO1,PO2,PO3,PO5,PO8,PO1 0 ,PO11
3	Apply solubility tests to determine the solubility characteristics of organic compounds in different solvents and interpret the results to make informed decisions in the identification process	PO1,PO2,PO3,PO5,PO8,PO1 0 ,PO11
4	Analyze and differentiate organic compounds based on their functional groups, through systemic qualitative analysis.	PO1,PO2,PO3,PO5,PO7,PO8, PO10 ,PO11
5	Synthesize suitable solid derivatives, confirm the identity of the unknown compound through melting point/boiling point determination, and construct various molecular models.	PO1,PO2,PO3,PO5,PO7, PO8,PO10 ,PO11

			SEMESTE	R – II						
Course	Title		BIO	CHEN	1ISTR	Y				
Course Title BIOCHEMISTRY  Course code BP 209P Total credits: 2 L T P S R O/F							O/F	С		
			Total hours: 4	0	0	4	0	0	0	2
Pre-req	uisite	Nil	Co-requisite				N	il		
Prograi	mme			lor of l						
Semes	ster		Fall/ II semester of	-		_	ogran	ıme		
Cour			arious amino acids and	_		mples.				
Objec	tive	1	ely analyze carbohydra	_						
3. Detect abnormal constituents in urine samples.										
CO			reagents, and qualitativ							
CO			sts are applicable for dif							
CO			erent constituents presen							
CO			nstituents quantitatively							
CO	5	Create experime	ents on biological sampl	es inclu	ıding I	Enzyme	es biol	ogical	samples.	
Unit-		Cor	tent	Conta	ct	Le	arning	<b>Outco</b>	ome	KL
No.				Hou						
I	-	tative analysis o							to learn	
	`		actose, Maltose,	qualitative analysis of carbohydrates, Proteins,						
		ose and starch)	<b></b>			•				
			r Proteins (albumin and		red	ucing s	sugars	(DN	ISA	
	Casei	*	of roducing sugars		and	ntitati	wa ana	train of	Phlood	
	-		of reducing sugars Proteins (Biuret		_	antitati otinino		•	and serum	
	meth		riotenis (Diuret			al chole		i sugai	and scrun	L
		,	f urine for abnormal		iou	ai Ciioi	csicioi			
		ituents	i di ine foi donomidi		Pre	enaratio	on of b	uffer so	olution	
		mination of bloo	od creatinine			l measi				
	Deter	mination of blo	od sugar	4				1		3,4
			m total cholesterol		catalytic role of enzymes,					
	Prepa	aration of buffer	solution and		imj	portanc	e of e	nzyme	inhibitors	
	meas	urement of pH				design				
	-	•	ydrolysis of starch			rapeuti		_		
			vary amylase activity		app	olicatio	ns of e	nzyme	s.	
			emperature on Salivary							
	_	ase activity.								
			bstrate concentration							
	on sa	livary amylase a	ctivity.							

- T1: Practical Biochemistry by R.C. Gupta and S. Bhargavan.
- T2: Practical Biochemistry for Medical students by Rajagopal and Ramakrishna.

### **REFERENCE BOOKS:**

R1: Introduction of Practical Biochemistry by David T. Plummer. (3rd Edition). R2: Practical Biochemistry by Harold Varley.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Relate different reagents, and qualitative tests to be used.	PO1,PO2,PO3,PO4,PO5,PO6, PO7,PO8,PO9,PO10 ,PO11
2	Choose what tests are applicable for different biological	PO1,PO2,PO3,PO4,PO5,PO6,
	samples.	PO7,PO8,PO9,PO10 ,PO11
3	Categorize different constituents present in biological	PO1,PO2,PO3,PO4,PO5,PO6,
	samples.	PO7,PO8,PO9,PO10 ,PO11
4	Estimate the constituents quantitatively in biological samples	PO1,PO2,PO3,PO4,PO5,PO6,
4	like blood, and urine.	PO7,PO8,PO9,PO10 ,PO11
_	Create experiments on biological samples including Enzymes	PO1,PO2,PO3,PO4,PO5,PO6,
3	biological samples.	PO7,PO8,PO9,PO10 ,PO11

		SEMESTE	R – II						
Course Title		COMPUTER APPI	LICATIO	)NS I	N PH	IARM.	ACY		
Course code	BP 210P	Total credits: 2	L	T	P	S	R	O/F	C
		Total hours: 2	0	0	2	0	0	0	1
Pre-requisite	Nil	Co-requisite				Ni	il		
Programme			elor of Ph		_				
Semester		Fall/ II semester of	f first yea	r of t	the pr	ogran	ıme		
Course	1	data using Excel.							
Objectives		nta tables using SQL.			,.				
	^	computer applications in	•	•	cation	1.			
CO1		d modify databases in M							
CO1		ess to store and retrieve					d	timadia	
CO2		and print from the patien			nic de	sign, a	na mui	umedia.	
CO3	_	-			:6		1		
CO4		relopment of an HTML-l							~ 4 ~
CO5	XML Webpage	e conversion of tables of	л шиогт	auon,	, mqu	mies, I	omis,	anu report	s 10
Unit- No.	1 6	ntent	Contact	Τ	ΙΛ	arnina	Outco	me.	KL
OIII - 140.	Cu.	ntent	Hour		LC	ai iiiiig	Guice	Jille	KL
I Desig	m a questionnai	re using a word	Hour	Stuc	donts	will b	na ahla	e to learn	
about Creat inforr Retrice adver Creat , gene Creat patier Using Desig delete datab Gene from Creat Drug MS A 0. C MS A 1. I	a particular dise a HTML web mation.  eve the information was effects using ing mailing label in the a database in a formation was access and modify the ase rating report an patient database ing invoice table information storaccess.	page to show personal tion of a drug and its online tools els Using Label Wizard MS WORD MS Access to store the with the required fields Access to view, add, e patient record in the d printing the report e e using – MS Access rage and retrieval using rking with queries in es, Queries, Forms	4	data com To prog web data prince To I MS Wizinfor designated ir eco General data in M store	anamunio gras gramm serv base ciples earn Wo ard rmatio gned -frien ng, d rds erate nlessl base, IS Ace and	p weening lawer fur mana create rd us and gon from the letting and y from the create ccess,	and of find of	chnologies, concepts, lities, and at system g labels in the Label the patient database ess with a r viewing, modifying the reports a patient voice table efficiently	3,4

- T1: Computer Application in Pharmacy William E.Fassett –Lea and Febiger, 600South Washington Square USA, (215) 922-1330.
- T2: Computer Application in Pharmaceutical Research and Development –Sean Ekins John Willey and Sons, INC., Publication.
- T3: Bioinformatics (Concept, Skills and Applications) S.C.Rastogi CBS Publishers and Distributors, 4596/1 110 002(INDIA).

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Apply MS Access to store and retrieve drug information	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8,PO9 ,PO11
2	Understand and apply knowledge of the Internet, graphic design, and multimedia.	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8,PO9 ,PO11
3	Create reports and print from the patient database	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8,PO9 ,PO11
4	Design and development of an HTML-based personal information web page	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8,PO9 ,PO11
5	Understand the conversion of tables of information, inquiries, forms, and reports to XML webpages	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8,PO9,PO11

			SEMEST	ER – II	I							
Course	Title		Pharmaceut	tical Or	gani	c Chen	nistry-	-II				
Course	code	BP301T	Total credits: 4	L	T	P	S	R	O/F	C		
			Total hours: 45	3	1	0	0	0	0	4		
Pre-req		Nil	Co-requisite					Nil				
Progra	mme		Bac	helor of	Pha	rmacy						
Seme	ster		Fall/ III semester	of first	t yea	r of the	prog	ramm	e			
Cou	rse	1. Write the structure, name, and the type of isomerism of the organic compound.										
Objec	tives	2. Write the reaction, name the reaction and orientation of reactions.										
			reactivity/stability of	compou	nds.							
		4. Prepare orga	nic compounds.									
CO	1		l analyze the stru						•			
		1	the reactions of ben			-						
		and their effects	on benzene, and ex	plain th	e str	ucture	and ap	oplicat	ions of chen	nicals	s ii	
			l and practical contex									
CO	2	1 ^	atile reactivity of phe									
		•	s, highlighting their				_	•		•		
			ibstituents on thei		tivity	and	perfo	orm q	qualitative 1	tests	to	
			m from other Compo									
CO	3		understanding of the		•				_			
			nisms, and recognizing	_	_		-				_	
			fat and oil quality, purity, and composition, while understanding the principles behind their									
		letermination										
CO	4	Analyze and demonstrate their knowledge through the synthesis and reactions of										
		polynuclear hydrocarbons. Evaluate the structure and medicinal uses of listed										
		r	rocarbons and their d	erivativ	es, as	sess th	eir pot	ential	applications	in the	e	
		field of medicine										
CO	5	1	anes' synthesis techr				•		0	•		
			shed theories, and jus	stify the	react	ivity of	f cyclo	propai	ne and cyclol	buten	ne	
		molecules.		T								
Unit-		Cont	ent	Conta		]	Learn	ing Ou	itcome	ŀ	<b>K</b> L	
No.	_			Hour		• .	*** *					
Ι		ene and its deriva							to learn			
			and other evidences			-	-		and other			
			ructure of benzene,				_		the structur	e		
		tal picture, resona				benzei		•				
		natic characters, H							, aromatic			
		tions of benzene	,						l's rule.			
	_	nonation, halogen	· · · · · · · · · · · · · · · · · · ·						ections of			
		lel crafts alkylatio	•						, sulphonation			
		ations, Friedel cra	•	10		_			ity, Friedel	$ \frac{3}{2}$	3,4	
		tituents, effect of				afts alk	-		-		,	
		ivity and orientat							afts acylation	n.		
			ompounds towards			pact o						
		rophilic substituti							tion of	1		
		cture and uses of l	DD1, Saccharin,						ene compour	nds		
	RHC	and Chloramine					electro	philic s	substitution			
						action.	1		DDT			
									DDT,			
					Sa	ccharit	ı, BHC	$\mathcal{I}$ and $\mathcal{O}$	Chloramine			

II	Phenols - Acidity of phenols, effect of substituents on acidity, qualitative tests, Structure and uses of phenol, cresols, resorcinol, naphthols  Aromatic Amines - Basicity of amines, effect of substituents on basicity, and synthetic uses of aryl diazonium salts  Aromatic Acids - Acidity, effect of substituents on acidity and important reactions of benzoic acid.	10	Students will be able to learn acidity of phenols, consider how substituents affect acidity, aromatic acids, assess how substituents affectacidity, and list some of benzoic acid's key reactions.	4
III	Fats and Oils  a. Fatty acids – reactions. b. Hydrolysis, Hydrogenation,    Saponification and Rancidity of oils,    Drying oils. c. Analytical constants – Acid value,    Saponification value, Ester value, Iodine    value, Acetyl value, Reichert Meissl    (RM) value – significance and principle    involved in their determination.	10	Students will be able to learn how fatty acids react. Analyze the stability of fats and oils using the processes of hydrolysis, hydrogenation, saponification, rancidity, and drying relevance of the underlying concept behind the following analytical constants for fats and oils: Acid value, Saponification value, Ester value, Iodine value, Acetyl value, and Reichert Meissl (RM) value	3
IV	Polynuclear hydrocarbons:  a. Synthesis, reactions b. Structure and medicinal uses of Naphthalene, Phenanthrene, Anthracene, Diphenylmethane, Triphenylmethane and their derivatives	8	Students will be able to learn Polynuclear Hydrocarbons' Synthesis and Reactions. structure and therapeutic applications of phenanthrene, anthracene, diphenylmethane, triphenylmethane, and its derivatives.	4,
V	Cyclo alkanes  Stabilities — Baeyer's strain theory, limitation of Baeyer's strain theory, Coulson and Moffitt's modification, Sachse Mohr's theory (Theory of strainless rings), reactions of cyclopropane and cyclobutane only		Students will be able to learn general preparation and reaction processes for cycloalkanes.  Explain how Baeyer's strain theory explains the stability of cyclo alkanes and discuss its limits, as well as those of Coulson and Moffitt's modification, Sachse Mohr's theory (theory of strainless rings), and reactions involvingsolely cyclopropane and cyclobutane.	5

- T1: Organic Chemistry by Morrison and Boyd.
- T2: Textbook of Organic Chemistry by B.S. Bahl & ArunBahl.

### **REFERENCE BOOKS:**

- R1: Organic Chemistry by I.L. Finar, Volume-I. R2: Organic Chemistry by P.L. Soni.
- R3: Reaction and reaction mechanism by Chatwal. R4: Organic Chemistry by Clayden.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand and analyze the structural derivation, demonstrate a comprehensive understanding of the reactions of benzene, Identify and describe the different substituents and their effects on benzene, and explain the structure and applications of chemicals in various industrial and practical contexts.	PO1,PO3,PO4,PO11
2	Explore the versatile reactivity of phenols, aromatic amines, and aromatic acids in diverse organic reactions, highlighting their crucial role in organic synthesis and analyze the influence of substituents on their reactivity and perform qualitative tests to differentiate them from other Compounds	PO1,PO3,PO4,PO11
3	Demonstrate an understanding of the chemistry of fats and oils, evaluating their chemical mechanisms, and recognizing the importance of analytical constants in assessing fat and oil quality, purity, and composition, while understanding the principles behind their determination	PO1,PO2,,PO3,PO4,P O10,PO11
4	Analyze and demonstrate their knowledge through the synthesis and reactions of polynuclear hydrocarbons. Evaluate the structure and medicinal uses of listed polynuclear hydrocarbons and their derivatives, assess their potential applications in the field of medicine	PO1,PO2, PO3,PO4,PO11
5	Assess cycloalkanes' synthesis techniques and reactivity, examine ring stability based on established theories, and justify the reactivity of cyclopropane and cyclobutene molecules.	PO1,PO3,PO4,PO11

			SEMESTE	ER – III						
Course	Title		Physic	al Phar	mace	eutics l	[			
Course	code	BP302T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 45	3	1	0	0	0	0	4
Pre-req	uisite	Nil	Co-requisite			•		Nil		
Progra	Programme Bachelor of Pharmacy									
Seme	ster		Fall/ III semester	of third	year	of the	prog	ramme	9	
Cou	rse	. Understand va	rious physicochemical	properti	ies of	f drug 1	molec	ules in	designing	the
Objec	tives	dosage forms								
		Know the prin	ciples of chemical kind	etics & u	se th	em for	stabi	lity tes	ting	
			on of the expiry date of							
			the use of physicoche	mical pr	oper	ties in	deve	loping	and evalu	ating
		dosage forms.								
CO	1		mechanisms of solute	e-solvent	inte	eraction	ns an	d solu	bility of o	different
		drug Molecules								
CO	2		olications of different p	physicoc	hemi	cal pro	operti	es of d	rug molec	ules as
		well as states of								
CO	3	_	and practical understan	_		undan	nental	s and t	heories rel	ated to
60			and its measurement te			1	· ·		1 C	
CO	)4	_	e diverse intermolecu	ilar force	es co	ntribu	ting to	o com	plex forma	ation
CO	\ <u></u>	and their practi	* *	<b>.</b>	<b>:</b>			4: 1	1 CC	
CO	)5	_	Acquire knowledge of the procedures for preparing pharmaceutical buffers, pH, and neir Significance							
Unit-				Cantaa	4	Ι.,	<b>:</b>	~ O4a		I/I
No.		Con	tent	Contac Hour	١	Le	агшп	g Outc	come	KL
I I	Saluh	ility of drugs: S	olubility expressions,	Hour	Stu	ıdents	will ł	ne ahle	to learn	
•			solvent interactions,						nificance	
			eters, solvation &		- 1	_		_	ence, and	
	1		e approach to the		1 ^				ons in drug	<u>o</u>
		s influencing sol				_		_	opment.	
			biological systems.						1	
	Solubi	ility of gas in liq	uids, solubility of	10						4
	liquid	s in liquids, (Bin								
	solutio	ons) Raoult's lav	v, real solutions.							
	Partial	lly miscible liqui	ds, Critical solution							
	_		cations. Distribution							
		s limitations and								
II			properties of matter:						to learn	
		_	es in the state of			states				
		, latent heats, va						_	operties,	
		_	int, eutectic mixtures,	10					inciples	
			ers, relative humidity,	10				sicoche		3
	_		id crystals, glassy		beł	navior	ot dru	g mole	ecules.	
		solid- crystallin	e, amorpnous &							
	polymorphism.									
	molec	_	roperties of drug ive index, optical	1						
			ve index, optical stant, dipole moment,							
			determinations and							
	applic		acterimianons and							
	applic	ut10113								

III	Surface and interfacial phenomenon:		Students will be able to learn	
	Liquid interface, surface & interfacial		the stability and performance of	
	tensions, surface free energy, measurement		pharmaceutical products and	
	of surface & interfacial tensions, spreading	8	develop innovative solutions in	4
	coefficient, adsorption at liquid interfaces,		the field of pharmaceutical	
	surface active agents, HLB Scale,		science	
	solubilisation, detergency,			
	adsorption at solid interface.			
IV	Complexation and protein binding:		Students will be able to learn	
	Introduction, Classification of		drug protein interactions, predict	
	Complexation, Applications, methods of		potential drug interactions, and	
	analysis, protein binding, Complexation and	8	consider the implications of	3
	drug action, crystalline structures of		protein binding on drug	
	complexes and thermodynamic treatment of		pharmacokinetics and	
	stability constants.		pharmacodynamics.	
V	pH, buffers and Isotonic solutions:		Students will be able to learn	
	Sorensen's pH scale, pH determination		pH, buffers, and isotonic	
	(electrometric and calorimetric),		solutions to real-world	
	applications of buffers, buffer equation,	7	scenarios, such as medical	1
	buffer capacity, buffers in pharmaceutical		treatments, pharmaceutical	
	and biological systems, buffered isotonic		industry, research, laboratory	
	solutions.		procedures etc.	

- T1: Text book of Physical Pharmaceutics by C.V.S. Subramanyam, VallabhPrakashan.
- T2: Physical Pharmacy by S.P Agarwal and Rajesh Khana, CBS Publishers and distributors PVT Ltd; 2nd Edition.

### **REFERENCE BOOKS:**

- R1: Physical Pharmacy by Alfred Martin.
- R2: Tutorial Pharmacy by Cooper and Gunn, CBS Publishers and distributors PVT Ltd.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the mechanisms of solute-solvent interactions and solubility of different drug molecules	PO1,PO2,PO3,PO4,PO6,PO8PO11
2	Understand applications of different physicochemical properties of drug molecules as well as states of matter	PO1,PO2,PO3,PO4,PO6,PO8PO11
3	Gain expertise and practical understanding of the fundamentals and theories related to surface tension and its measurement techniques.	PO1,PO2,PO3,PO4,PO6,PO8PO11
4	Comprehend the diverse intermolecular forces contributing to complex formation and their practical applications	PO1,PO2,PO3,PO4,PO6,PO8PO11
5	Acquire knowledge of the procedures for preparing pharmaceutical buffers, pH, and their significance	PO1,PO2,PO3,PO4,PO6,PO8PO11

			SEMESTE	R – III						
Course '	Course Title Pharmaceutical Microbiology									
Course	code	BP303T	Total credits: 4	L	T	T P S R O/F				С
			Total hours: 45	3	1	0	0	0	0	4
Pre-requ	iisite	Nil	Co-requisite			•	Ni	il		•
Progran	nme		Bache	lor of Pl	harn	nacy				
Semes	ter	]	Fall/ III semester of	Third y	ear	of the p	rograr	nme		
Cours	se	1. Understand me	ethods of identificatio	n, cultiva	ation	and pres	servati	on of v	arious	
Objecti	ives	microorganism	ıs							
		2. To understand	the importance and i	mplemei	ntati	on of ste	rilizati	on in p	pharmaceur	tical
		processing and	l industry							
		3. Learn sterility to	esting of pharmaceuti	cal prod	ucts	•				
		4. Carried out mice	robiological standard	ization c	of Ph	armaceu	ticals.			
		5. Understand the	cell culture technolog	gy and its	s app	olication	s in ph	armace	eutical indu	ıstries
CO1		Understand method	s of identification, cu	ltivation	, and	d preserv	ation o	of vario	ous	
		microorganisms								
CO2		l .	mportance and imple	mentatio	n of	f steriliza	ıtion ir	pharn	naceutical	
		processing and indu								
CO3	3		ng of pharmaceutical	•						
CO4	•	Carried out microbi	ological standardizat	ion of Ph	narm	aceutical	ls.			
COS	5	Understand the cell	culture technology as	nd its app	plica	tions in	pharm	aceutic	cal industri	es.
Unit-		Conte	nt	Contac	et	Lea	arning	Outco	ome	KL
No.				Hour						
I	Intro	duction, history of	microbiology, its		S	tudents	will be	able 1	to learn	
		thes, scope and its in	-		hi	istory, br	anches	s and s	cope of	
	1	duction to Prokaryot			- 1	microbiology To understand the				
		of ultra-structure a			st	ructure a	ınd cla	ssifica	tion of	
		fication of bacteria,							different	
	_	rements, raw materi				quireme				
		a and physical parar	0			nd cultiva				
	I_	th curve, isolation a	-	10		acteria T				2,3
		ods for pure cultures			- 1	strumen				
		obes, quantitative m			^	plication		fferent		
		rial growth (total &	,		m	icroscop	es			
	1	of different types								
		scopy, dark field mi	croscopy and							
	1	ron microscopy.								
II			eria using staining			tudents			to learn	
			ram's &Acid fast			aining te	•			
		ng) and biochemica	` /			lentificat				
	1		procedure, merits,						tion and	3,4
			tions of physical,			s differei			nd	
		-	tion and mechanical		in	strumen	ts utili:	zed		
		od of sterilization.	C							
		nation of the efficien	icy of sterilization							
	metho		1 1							
	1 ^ ^	oment's employed in	-							
777		zation. Sterility indi				4 1 1	•11 •			
Ш	_	of morphology, class							to learn	
	repro	duction/replication a	and cultivation		S	tructure a	and cla	ssitica	tion	

	Fungi and Viruses.		of	
	Classification and mode of action of		fungi and viruses To understand	l l
	disinfectants Factors influencing		different disinfectant and their	
	disinfection, antiseptics and their		significance and application To	
	evaluation. For bacteriostatic and	10	understand the sterility	2,3
	bactericidal actions Evaluation of bactericidal		testing methods for Quality	
	& Bacteriostatic. Sterility testing of products		Control of Pharmaceuticals	
	(solids, liquids, Ophthalmic and othersterile			
	according to IP, BP and USP.			
IV	Designing of aseptic area, laminar flow		Students will be able to learn	
	equipments; study of different sources of		aseptic area for microbiological	
	contamination in an aseptic area and		studies and different equipment	
	methods of prevention, clean area		used in aseptic area To	
	classification.	8	understand methods and	1
	Principles and methods of different		application of microbiological	
	microbiological assay. Methods for		assays for quality control and	
	standardization of antibiotics, vitamins and		assurance of pharmaceutical	
	amino acids. Assessment of a new antibiotic.		products like antibiotics.	
V	Types of spoilage, factors affecting the		Students will be able to learn	
	microbial spoilage of pharmaceutical		Microbiological contaminations	
	products, sources and types of microbial		and ways to prevent them. To	
	contaminants, assessment of microbial		understand the microbiological	
	contamination and spoilage.		stability testing for	
	Preservation of pharmaceutical products		pharmaceuticals. To understand	
	using antimicrobial agents, evaluation of	7	concept of cell culture and its	2,3
	microbial stability of formulations.		application in pharmaceutical	
	Growth of animal cells in culture, general		industry and drug development	
	procedure for cell culture, Primary,		Research	
	established and transformed cell cultures.			
	Application of cell cultures in			
	pharmaceutical industry and research.			

T1: Pharmaceutical Microbiology – AshutoshKar, New Age International (P) Ltd., Publishers, New Delhi, India

T2: A textbook of Pharmaceutical Microbiology – Dr.Kuntal Das, NiraliPrakashan, Pune – 411005.

### **REFERENCE BOOKS:**

R1: W.B. Hugo and A.D. Russel: Pharmaceutical Microbiology, Blackwell Scientific publications, Oxford London.

R2: Prescott and Dunn., Industrial Microbiology, 4th edition, CBS Publishers & Distributors, Delhi.

R3: Review of Microbiology & Immunology - Apurba Sankar Sastry, Jaypee Brothers Medical Publishers

(P) Limited, New Delhi – 110002.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand methods of identification, cultivation, and	PO1,PO2,PO3,PO4,PO6,PO7,
1	preservation of various microorganisms	PO9,PO10, PO11
2	To understand the importance and implementation of	PO1,PO2,PO3,PO4,PO6,PO7,
2	sterilization in pharmaceutical processing and industry	PO9,PO10, PO11
3	Learn sterility testing of pharmaceutical products.	PO1,PO2,PO3,PO4,PO6,PO7,
3	Learn stermty testing of pharmaceutical products.	PO9,PO10, PO11
1	Carried out microbiological standardization of	PO1,PO2,PO3,PO4,PO6,PO7,
-	Pharmaceuticals.	PO9,PO10, PO11
_	Understand the cell culture technology and its applications	PO1,PO2,PO3,PO4,PO6,PO7,
3	in pharmaceutical industries.	PO9,PO10, PO11

	SEMESTER – III									
Course	rrse Title Pharmaceutical Engineering									
Course	code	BP304T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 45T	3	1	0	0	0	0	4
Pre-req	uisite	Nil	Co-requisite				N	il	•	
Progra	mme		Bachelo	or of Ph	arma	acy				
Semes	ster		Fall/ III semester of 7	Third ye	ar o	f the p	rogra	mme		
Cour	·se	1. To know vario	ous unit operations used	d in Phar	mace	eutical	indus	tries.		
Object	ives	2. To understand	the material handling	techniqu	es.					
		3. To perform va	rious processes involve	ed in pha	ırmac	ceutica	al man	ufactur	ing process	s.
		4. To carry out va	arious tests to prevent	environn	nenta	ıl pollı	ıtion.			
		5. To appreciate	and comprehend signif	icance o	f pla	nt lay	out de	sign fo	r optimum	use of
		resources.								
CO	1	Understand the f	fundamental concepts of	of size re	duct	ion ar	id sepa	aration	and their	
		pharmacological a	pplications.							
CO	2	Describe the fund	amental concepts, met	hods, ar	nd ap	plicat	ions o	f heat	transfer,	
		evaporation, and d	istillation in pharmace	utical pro	epara	tions.				
CO	3	Demonstrate the b	asic concepts and phar	maceuti	cal ap	pplicat	tions o	f dryin	g and mixi	ng.
CO	4	Explain the theorie	es, principles, and facto	ors that i	mpac	t filter	ing an	d centr	rifugation.	
CO	5	Assess the constru	ction of the facility, co	rrosion i	ssue	s, and	metho	ds for p	preventing	
		corrosion.								
Unit-		Conto	ent	Contact	t	Le	arning	g Outco	ome	KL
No.				Hour						
I	Flow	of fluids: Types of	f manometers,		Stu	dents	will	be abl	e to learn	i
	Reyn	olds number and it	s significance,		sign	ifican	ce of t	he flow	v of fluids,	,
	Berno	oulli's theorem and	lits applications,		size	reduc	tion a	nd size	separation	
	Energ	gy losses, Orifice n	neter, Venturimeter,							
	Pitot	tube and Rotomete	r.							
	Size 1	Reduction: Objecti	ves, Mechanisms &							
	Laws	governing size red	luction, factors							
		ting size reduction,	• •							
		ruction, working, u								
		rits of Hammer mi								
	_	gy mill, Edge runne	r mill & end runner	10						1,2
	mill.									
			ives, applications &							
		anism of size separ								
		-	eves, size separation							
		-	working, uses, merits							
		lemerits of Sieve sh	*							
-		rator, Air separator, Bag filter &								
		ation tank.			G.		•11			
II		•	ives, applications &	10				be abl		2.2
			s. Fourier's law, Heat	10			_		ransfer,	2,3
		•	onvection & radiation.					evapo		
		nterchangers & hea			and	uistilla	alion to	echniqu	ie	
	_	ration: Objectives,								
		s influencing evapo								
		_	other heat process.							
	princ	ipies, construction,	working, uses, merits							

				1
	and demerits of Steam jacketed kettle,			
	horizontal tube evaporator, climbing film			
	evaporator, forced circulation evaporator,			
	multiple effect evaporator& Economy of			
	multiple effect evaporator.			
	Distillation: Basic Principles and			
	methodology of simple distillation, flash			
	distillation, fractional distillation, distillation			
	under reduced pressure, steam distillation &			
	molecular distillation			
III	Drying: Objectives, applications &		Students will be able to learn	
	mechanism of drying process, measurements		principle involves in drying and	
	& applications of Equilibrium Moisture content,		mixing.	
	rate of drying curve. principles,			
	construction, working, uses, merits and demerits			
	of Tray dryer, drum dryer spray dryer,			
	fluidized bed dryer, vacuum dryer, freeze			
	dryer.			
	• Mixing: Objectives, applications &			
	factors affecting mixing, Difference	10		3
	between solid and liquid mixing,			
	mechanism of solid mixing, liquids			
	mixing and semisolids mixing. Principles,			
	Construction, Working, uses, Merits and			
	Demerits of Double cone blender, twin			
	shell blender, ribbon blender, Sigma			
	blade mixer, planetary mixers, Propellers,			
	Turbines, Paddles & Silverson Emulsifier			
IV	• Filtration: Objectives, applications,		Students will be able to learn	
1 1	Theories & Factors influencing filtration,		principles and uses of filtration	
	filter aids, filter medias. Principle,		and centrifugation.	L
	Construction, Working, Uses, Merits and		and centification.	
	demerits of plate & frame filter, filter leaf,			
	_			
	rotary drum filter, Meta filter & Cartridge filter, membrane filters and Seidtz filter.	O		2.2
	0	8		2,3
	• Centritugation: Objectives, principle & applications of Centrifugation, principles,			
	construction, working, uses, merits and			
	demerits of Perforated basket centrifuge,			
	Non-perforated basket centrifuge, semi			
<b>T</b> 7	continuous centrifuge & super centrifuge.		G. I	
V	Materials of pharmaceutical plant		Students will be able to learn	
	construction, Corrosion and its prevention:		Materials of pharmaceutical plant	
	Factors affecting during materials selected for		construction, Corrosion and its	3
	Pharmaceutical plant construction, Theories		prevention	_
	of corrosion, types of corrosion and there	7		3
	prevention. Ferrous and nonferrous metals,			
	inorganic and organic non metals, basic of			1
	material handling systems.			

T1: Introduction to chemical engineering – Walter L Badger & Julius Banchero, Latest edition. T2: Solid phase extraction, Principles, techniques and applications by Nigel J.K. Simpson- Latest edition.

T3: Unit operation of chemical engineering – Mcabe Smith, Latest edition.

### **REFERENCE BOOKS:**

R1:Pharmaceutical engineering principles and practices – C.V.S Subrahmanyam et al., Latest edition.

R2:Remington practice of pharmacy- Martin, Latest edition.

R3T:heory and practice of industrial pharmacy by Lachmann, Latest edition. R4: Physical pharmaceutics-C.V.S Subrahmanyam et al., Latest edition.

R5: Cooper and Gunn's Tutorial pharmacy, S.J. Carter, Latest edition.

	CO PO Mapping								
SN	Course Outcome (CO)	Mapped Program Outcome							
1	Understand the fundamental concepts of size reduction and	PO1,PO3,PO4,PO6,PO8,,PO							
1	separation and their pharmacological applications.	10, PO11							
	Describe the fundamental concepts, methods, and applications of	PO1,PO3,PO4,PO6,PO8,,PO							
2	heat transfer, evaporation, and distillation in pharmaceutical	101,1 03,1 04,1 00,1 06,,1 0							
	preparations.	10,1011							
3	Demonstrate the basic concepts and pharmaceutical applications of	PO1,PO3,PO4,PO6,PO8,,PO							
3	drying and mixing.	10, PO11							
4	Explain the theories, principles, and factors that impact filtering	PO1,PO3,PO4,PO6,PO8,,PO							
4	and centrifugation.	10, PO11							
5	Assess the construction of the facility, corrosion issues, and	PO1,PO3,PO4,PO6,PO8,,PO							
3	methods for preventing corrosion.	10, PO11							

			SEMESTE								
Course Title Pharmaceutical Organic Chemistry -II (Practical)							_				
Course	code	BP305P	Total credits: 2	L	T	P	S	R	O/F	C	
			Total hours: 4	0	0	4	0	0	0	2	
Pre-requ	uisite	Nil	Co-requisite					Nil			
Progran	nme		Bache	elor of F	Pharn	nacy					
Semes	ter		Fall/ III semester o	f third y	year (	of the	progr	amme			
Cour	se	1. Measu	re the yield and purity o	f synthe	sized	organ	ic con	pounds	S.		
Object	ives	2. Design	various classes of mone	o-substit	tuted	benzei	nes fro	m diaz	onium salt	s.	
CO	1	-	d laboratory equipmen	t and th	neir f	unctio	ns, ar	d desc	ribe safet	y	
		protocols for ch	•								
CO2	2		underlying principles	of steam	disti	llation	and r	ecrystal	llization		
		_	purification process.								
CO	3		nificance of these cons			_	•	• • •	•		
		_	ats and oils, and employ				-				
			mphasizing the relation	ship bet	ween	the pr	inciple	es and p	oractical		
		applications									
CO ₂	1		lamental principles and								
		_	blished procedures, inc	luding c	alcula	ating re	eactan	t quanti	ties and re	action	
		conditions									
CO	5	_	ty and purity of synthe	_	arma	ceutica	al con	pounds	s based on	L	
			nd characterization tech								
Unit-		Cor	ntent	Contac		Le	arnin	g Outc	ome	KL	
No.				Hour							
1	Exp	eriments	Involving						to learn		
		Laboratory Tec	-	Experiments Involving							
		Recrystallization		Laboratory Techniques							
		Steam distillation		Standardization of Reagents Preparation of Compounds							
			ollowing Oil Values		Pre	eparati	on of (	Compo	unds		
	`	•	zation of Reagents):								
		Acid value	1								
		Saponification v	alue								
		Iodine value	1								
		reparation of Co	-								
		Benzanilide / Ph	•								
		benzoate/Acetan		4						2.4	
		reaction.	Aniline by acylation	4						3,4	
		2,4,6- omogniling/Porc	bromoacetanilide from								
		ne/Acetanilide b									
		nination) reaction									
	`	5-Nitrosalicylic									
3.			ene from Salicylic								
		acid/Nitro benze	•								
		reaction.	no oy maanon								
4.			m Benzyl chloride by								
		oxidation reaction									
			alicylic acid from								
		alkylbenzoate/ al	•								
		airyiochzoate/ al	kyroancyraic by								

hydrolysis reaction.	
6. 1-Phenylazo-2-naphthol from Aniline	
by diazotization and coupling	
reactions.	
7. Benzil from Benzoin by	
oxidation reaction.	
8. Dibenzalacetone from Benzaldehyde	
by Claisen-Schmidt reaction.	
9. Cinnamic acid from Benzaldehyde by	
Perkin reaction.	
10. P Iodobenzoic acid from P –	
aminobenzoic acid.	

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Identify standard laboratory equipment and their functions,	PO1,PO2,PO3,PO4,PO6,PO9,,P						
1	and describe safety protocols for chemical handling	O10, PO11						
2	Demonstrate the underlying principles of steam distillation and	PO1,PO2,PO3,PO4,PO6,PO9,,P						
	recrystallization techniques in the purification process.	O10, PO11						
3	Evaluate the significance of these constants in assessing the quality, purity, and composition of fats and oils, and employ Appropriate laboratory methods for their determination, emphasizing the relationship between the principles and practical Applications	PO1,PO2,PO3,PO4,PO6,PO9,,P O10, PO11						
4	Explain the fundamental principles and perform pharmaceutical synthesis experiments according to established procedures, including calculating reactant quantities and reaction conditions	PO1,PO2,PO3,PO4,PO6,PO9,P O10, PO11						
5	Assess the quality and purity of synthesized pharmaceutical compounds based on analytical data and characterization techniques.	PO1,PO2,PO3,PO4,PO6,PO9,,P O10, PO11						

	SEMESTER III									
Course	Title		Physical Phar	rmaceut	ics –	I (Pra	ctical)			
Course	code	BP306P	Total credits: 2	L	T	C				
			Total hours: 4	0	0	4	0	0	0	2
Pre-requ	iisite	Nil	Co-requisite				ľ	Vil	1	
Progran	nme		Bache	elor of P	harn	nacy				
Semest	ter		Fall/ III semester o	f third y	ear (	of the	progra	mme		
Cours	se	1. Measure	the HLB value of sur	factants.						
Objecti	ves	2. Estimate	adsorption capacity o	f adsorb	ents.					
		3. Assess dr	ug solubility in differ	ent solve	ents.					
CO1	-	Understand the cor	ncept of solubility of	drug mol	ecule	es				
CO2	2	Describe the effect	of pKa and pH value	in form	ulatio	on, dev	elopm	ent of	dosage for	n
CO3	}	Understand the kn	owledge of hydropho	obicity a	nd n	nembra	ne per	meabi	lity of drug	g
		molecules by deter	mining partition coef	ficient						
CO4	l	Understand the co	ncept of CMC and	demonst	rate t	the var	rious d	etermi	ining meth	ods of
		Surface Tension								
CO5	;	Learn about the say	ponification method b	y evalua	ting	the HI	LB valu	ie		
Unit-		Conte	ent	Contac	t	Le	arning	Outc	ome	KL
No.				Hour						
I	1. I	Determination of th	e solubility of a drug		Stu	ıdents	will b	e able	to learn	
	a	at room temperature	<b>e</b> .		the	physi	cochen	nical fo	eatures of	
	2. I	Determination of pl	Ka value by half		col	loidal	and dis	persec	d systems	
	1	neutralization/Hend		in 1	raw ma	aterials	are so	)		
	6	equation.		imj	portan	t in the	pharn	naceutical		
	3. I	Determination of p	artition coefficient		sci	ences				
		of benzoic acid in b								
			artition coefficient							
	(	of iodine in CCl4 ar	nd water.							
		Determination of %	•							
	1	NaCl in a solution u	sing phenol-water							
		system by CST met								
		Determination of su								
	_	given liquids by dro	p count and drop	4						3,4
		weight method.								
		Determination of H								
		Lipophilic Balance)								
		surfactant by sapon								
	_	Determination of Fr								
		Langmuir constants	using activated							
		charcoal.	5.5 d - 5 dd							
		Determination of cr								
		concentration of sur								
		Determination of st	•							
		anddonor-acceptor i								
		_	y solubility method.							
			tability constant and							
		_	o of Cupric- Glycine							
	(	complex by pH titra	ttion method.							

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Understand the concept of solubility of drug molecules	PO1,PO2,PO3,PO4,PO7,PO9,PO10, PO11						
2	Describe the effect of pKa and pH value in formulation, development of dosage form	PO1,PO2,PO3,PO4,PO7,PO9,PO10, PO11						
3	Understand the knowledge of hydrophobicity and membrane permeability of drug molecules by determining partition coefficient	PO1,PO2,PO3,PO4,PO7,PO9,PO10, PO11						
4	Understand the concept of CMC and demonstrate the various determining methods of Surface tension	PO1,PO2,PO3,PO4,PO7,PO9,PO10, PO11						
5	Learn about the saponification method by evaluating the HLB Value	PO1,PO2,PO3,PO4,PO7,PO9,PO10, PO11						

			SEMEST	ER III						
Course	Title		Pharmaceutic	utical Microbiology (Practical)						
Course	code	BP307P	Total credits: 2	L	T	P	S	R	O/F	C
			Total hours: 4	0	0	4	0	0	0	2
Pre-req		Nil	Co-requisite					Nil		
Prograi				elor of I						
Semes		1 11 .:0	Fall/ III semester							
Cour			y microorganisms rele aseptic techniques in h	_				crobiol	logy.	
Object	ives	11.	n microbial isolation,	_		_		erizatio	on technia	ies
CO	1		techniques of sterilizat			, шта с	114140	<u> </u>	on coominge	
CO			rious staining methods		e, gra	m stai	ning a	nd acid	-fast staini	ng.
CO			working, construction							_
	1	used in Microbi	ology		• •					
CO	4	Understand the r	nethods of isolation of	f pure cu	lture	of bac	teria.			
CO	5		robial assay of antibion	tics.						
Unit-		Cont	ent	Contac	t   ¯	Le	arnin	g Outc	ome	KL
No.	1	T . 1 .:	1 . 1 . 0 1:00	Hour	G.	1 .	•11. 1		, ,	
	1.		and study of different d processing in			i <b>aents</b> nciple			to learn	
			microbiology, e.g.,			_		nt type	of	
		-	ator, laminar flow,			ining.	1111010	ш турс	01	
			autoclave, hot air			······································				
		-	p freezer, refrigerator,							
			ised in experimental							
		microbiology.	-							
	2.	Sterilization of	of glassware,							
			nd sterilization of							
		media.								
	3.	ū	of bacteria and							
		preparations.	ent stabs and slants	4						2.4
	4.		ods - Simple, Gram	4						3,4
	٦.	_	acid-fast staining							
		•	n with practical).							
	5.	,								
		_	ns by multiple streak							
		plate techniqu								
		techniques.								
	6.	$\mathcal{C}$	•							
		•	cup plate method							
	_	and other met								
	7.	•	mination by Hanging							
	0	drop method.	<b></b>							
	8. 9.	· ·	g of pharmaceuticals. al analysis of water.							
		D. Biochemical t	•							
	11	J. Diochemical (	.cow.							

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Recall different techniques of sterilization	PO1,PO2,PO3,PO4,PO6,PO7, PO9,PO10, PO11					
2	Demonstrate various staining methods – simple, gram staining and acid-fast staining.	PO1,PO2,PO3,PO4,PO6,PO7, PO9,PO10, PO11					
3	Understand the working, construction and application of different instruments used in microbiology	PO1,PO2,PO3,PO4,PO6,PO7, PO9,PO10, PO11					
4	Understand the methods of isolation of pure culture of bacteria.	PO1,PO2,PO3,PO4,PO6,PO7, PO9,PO10, PO11					
5	Perform the microbial assay of antibiotics.	PO1,PO2,PO3,PO4,PO6,PO7, PO9,PO10, PO11					

	SEMESTER III								
Course Title		Pharmaceutic	al Engin	eerii	ng (Pr				
Course code	BP308P	Total credits: 2	L	T	7 7 0 11 0/1			С	
		Total hours: 4	0	0		0	0	0	2
Pre-requisite		Co-requisite				1	Nil		
Programme		Bachelor of Pharmacy							
Semester	1 5	Fall/ III semester o							
Course	1	e physical properties		iacei	itical f	ormula	tions.		
Objectives	· ·	of powder characteri		. 1		:1		11_	4
601	7	effects of distinct fa	_			_		_	nı.
CO1 CO2		l principles of pharm les underlying pharn					ocesse	S.	
CO2		peration of pharmace					ma a m t		
CO3		t of process variables					ment.		
CO4	• •	s to investigate and in		_			duat a	nolity	
Unit-No.	Cont		Contact			arning			KL
Unit-No.	Com	tent	Hour		Le	arming	Guice	ome	KL
I 1	l. Determination of	fradiation constant	Hour	Stu	ıdants	will h	a ahla	to learn	
•		painted and painted			termina				
	glass.	pannea ana pannea			ıstant o				
	•	n – To calculate the			ainted				
	efficiency of stea			_	am dis	_		8	
] 3	3. To determine the			То	detern	nine th	e over	all heat	
	transfer coefficie	ent by heat		trai	nsfer c	oeffici	ent by	heat	
	exchanger.			exc	hange	r.			
	4. Construction of o	drying curves (for		Co	nstruct	ion of	drying	curves	
	calcium carbona	te and starch).		De	termin	ation o	f mois	ture	
	5. Determination of	f moisture content		cor	ntent ai	nd loss	on dry	ying.	
	and loss on drying	~		Determination of humidity of					
	<ol><li>Determination of</li></ol>				From		ddry b	ulb	
		y bulb temperatures			nperatu				
	3. Use of Dewpoint				e of De				3,4
	Description of C				scriptio				
	working, and app				rking,				
		Machinery such as thine, fluidized bed			e analy	-	sievin	g Size	
	coater, fluid ener				calcul		unifor	mity	
	dehumidifier.	gy mmi,						e by using	
1	10. Size analysis by	sieving – To			uble C	_	•	e oy asing	
		tribution of tablet				21			
	granulations – C								
	various size free								
		etic and logarithmic							
	probability plots.								
1	1. Size reduction:								
1	2. To verify the law	vs of size reduction							
	using ball mill.								
1	3. Determining Kic	-							
	Bond's coefficie	-							
	requirement, and	critical speed of							

Ball Mill.	
14. Demonstration of colloid mill,	
planetary mixer, fluidized bed	
dryer, freeze dryer and such other	
major equipment.	
15. Factors affecting rate of filtration	
and evaporation (surface area,	
concentration, thickness /viscosity).	
16. To study the effect of time on the	
Rate of Crystallization.	
17. To calculate the uniformity Index	
for given sample by using Double	e
Cone Blender.	

	CO PO Mapping						
SN	Course Outcome (CO)	<b>Mapped Program Outcome</b>					
1	Recall fundamental principles of pharmaceutical manufacturing	PO1,PO2,PO3,PO4,PO6,PO9,					
1	processes.	PO10, PO11					
2	Explain the principles underlying pharmaceutical unit operations.	PO1,PO2,PO3,PO4,PO6,PO9,					
	Explain the principles underlying pharmaceutical unit operations.	PO10, PO11					
3	Demonstrate the operation of pharmaceutical manufacturing	PO1,PO2,PO3,PO4,PO6,PO9,					
3	equipment.	PO10, PO11					
4	Analyze the impact of process variables on drug product	PO1,PO2,PO3,PO4,PO6,PO9,					
4	quality.	PO10, PO11					
5	Design experiments to investigate and improve pharmaceutical	PO1,PO2,PO3,PO4,PO6,PO9,					
3	product quality.	PO10, PO11					

			SEMESTE	R – IV						
Course	Title		Pharmaceutica	al Organ	ic C	hemisti	y –III			
Course	code	BP401T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 45T	3	1	0	0	0	0	4
Pre-req		Nil	Co-requisite				Ni	il		
Progra				elor of P						
Seme			Fall/ IV semester							
Cou			the methods of prepara							
Object	tives	_	stereo chemical aspects	of organ	nic co	ompoun	ds and	stereo	chemical	
		reactions		4.		0				
			edicinal uses and other					_		
CO	1		-chemical features inclu	ıdıng coı	ntorn	nation a	nd ster	eo-ele	ctronic effe	cts of
		organic molecul								
CO	2	_	basic experimental pri	inciples of	of geo	ometric	al isom	nerism	and atrop	
		isomerism								
CO	3		ctures and synthesis of	simple fi	ve-m	ember l	neteroc	yclic o	organic	
		compounds.						•	1	
CO	4		actures and synthesis of	essentia	l six	- memb	ered o	rganıc	heterocycl	1C
		Compounds	1 1 : 0		•					
CO	5		d mechanisms for com		<u> </u>			•		
Unit-		Con	tent	Contac	et	Lea	arning	Outco	ome	KL
No.	G.		1:	Hour	G.		•••			
		-	al isomerism— Optical						to learn	
		y, enantiomerism			ereo-che					
		ompounds Eleme		- 1	luding reoelec					
		and achiral mole						5 01		
Ι		clature of optica	10	018	ganic m	oiccuic	55.		1,2	
	_	•	em of nomenclature of ons of chiral molecules	10						1,2
	_		and resolution of							
			metric synthesis:							
		and absolute	inicule synthesis.							
	•		Nomenclature of		Sti	udents	will he	ahle 1	to learn	
			is Trans, EZ, Syn Anti			ometric				
	-	s) Methods of de	*			mencla				
	1 -	uration of geome				merism				
	_	•	ism in Ethane, n-		- 1	ethods o		_		
П			ne. Stereo isomerism		coı	nfigurat	ion of	geome	etrical	
			s (Atropisomerism)	10		_		-	ormational	1,2
	_	nditions for option	· -		Isc	merism	ı in alk	anes.		
		-	eoselective reactions		Ste	ereoisor	nerism	in Bip	henyl	
		•						_	specific	
						d Stereo			_	
	Hetero	eyclic compoun	ds: Nomenclature		Stı	udents	will be	able	to learn	
	and cla	ssification Synth	esis, reactions and		Не	terocyc	lic Co	mpoun	ds.	
	Medic	inal uses of follow	ving		No	mencla	ture 8	apply	y this to	
Ш	compo	ounds/derivatives	Pyrrole, Furan, and	10		ming of				1,2
	Thiopl	nene Relative are	omaticity and		Со	mpoun	ds. Cla	ssifica	tion of	
	reactiv	ity of Pyrrole, Fu	aran and Thiophene		Не	terocyc	lic Cor	npoun	ds.	
	1				Sv	nthesize	e reacti	one ac	sociated	Ī

			with heterocycles. Medicinal	
			Compounds of these heterocycles.	
			Relative Aromaticity and	
			Reactivity of these heterocycles	
	Synthesis, reactions and medicinal uses of		Students will be able to learn	
	following compounds/derivatives Pyrazole,		Heterocyclic Compounds.	
	Imidazole, Oxazole and Thiazole. Pyridine,		synthesize the reactions	
IV	Quinoline, Isoquinoline, Acridine and Indole.		associated with Heterocyclic	
	Basicity of pyridine Synthesis and medicinal	8	compounds . structures and	1,2
	uses of Pyrimidine, Purine, azepines and their		synthesis of basic six- membered	
	derivatives		organic heterocyclic compounds	
	Reactions of synthetic importance Metal		Students will be able to learn	
	Hydride reduction (NaBH4 and LiAlH4),		Some Important Organic	
	Clemmensen reduction, Birch reduction,		reactions. Reactions of Synthetic	
$\mathbf{V}$	Wolff Kishner reduction. Oppenauer-	7	Importance. the applications of	1,2
	oxidation and Dakin reaction. Beckmanns		some Organic Compounds.	
	rearrangement and Schmidt rearrangement.			
	Claisen-Schmidt condensation			

T1:A text book of organic chemistry – Arun Bahl, B.S. Bahl. T2: Heterocyclic Chemistry by Raj K. Bansal.

### **REFERENCE BOOKS:**

R1 Organic Chemistry by Morrison and Boyd 5. R2 Organic chemistry by I.L. Finar, Volume-I & II. <a href="https://www.carewellpharma.in/bpharmacy/notes/4th-sem/pharmaceutical-organic-chemistry-3">https://www.carewellpharma.in/bpharmacy/notes/4th-sem/pharmaceutical-organic-chemistry-3</a>

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Illustrate Stereo-chemical features including conformation and	PO1,PO2,PO3,PO7, PO8,
1	stereo-electronic effects of organic molecules.	PO11
2	Comprehend the basic experimental principles of geometrical	PO1,PO2,PO3,PO7, PO8,
	isomerism and atropisomerism	PO11
3	Outline the structures and synthesis of simple five-member	PO1,PO2,PO3,PO7, PO8,
3	heterocyclic organic compounds.	PO11
4	Describe the structures and synthesis of essential six- membered	PO1,PO2,PO3,PO7, PO8,
4	organic heterocyclic compounds	PO11
5	Describe detailed mechanisms for common naming reactions	PO1,PO2,PO3,PO7, PO8,
3	Describe detailed mechanisms for common naming reactions	PO11

			SEMESTE	ER – IV						
Cours	rse Title MEDICINALCHEMISTRY-I									
Cour	se code	BP402T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 45T	3	0	2	0	0	0	4
	equisite	Nil	Co-requisite	1 40			N	il		
	ramme			elor of P						
	ester		Fall/ IV semester							
	urse		e chemistry of drugs of							C 1
Obje	ectives		e drug metabolic path actural Activity Relati	•						i arugs
			mical synthesis of son	- '	AK)	or unit	ereni c	1888 01	urugs	
	<b>O</b> 1		asic principles of med		mist	<b>17</b> 1/				
	02		e of disease or disord				sm of	action	for drugs a	ecting
	02	on ANS	e of disease of disord	aci ana t	.10 111	Conami	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	action	Tor drugs t	cting
C	O3		e chemistry of phar	maceutic	als r	elates	to the	eir pha	rmacologic	cal
	<b>-</b>	and pharmacokin	•		1	- 1	un	P.10		
C	O4		ral activity relationsh	ip (SAR)	betv	veen va	rious	medica	tion classes	s and
			ivity of drugs acting o							
C	O5	Understand the	synthesis of various	drug prod	lucts	with c	lassifi	cation	based on	
		chemical structur	e.							
Unit-		Conte	nt	Contact	;	Le	arning	g Outco	ome	KL
No.				Hour						
I			Chemistry History						o learn	
		elopment of medic		r	sicoch					
	1 -	chemical propertie						_	action and	
		al action Ionizatio			_		_	etabolism		
		Coefficient, Hydbinding, Chelation		ın r	Medicir	nai cne	emistry			
		and Geometrical i		10						1,2
	_		etabolism principles-	10						1,2
	_	and Phase-II.	cutoonsin principles-							
			etabolism including							
		hemical aspects.	8							
П	1		mic Nervous System		Stu	idents v	will be	able to	o learn the	
	Adrener	gic Neurotransmi	tters:		dru	gs actii	ng on A	ANS i.	e.,	
	Biosyntl	hesis and cataboli	sm of catecholamine.		Syr	npatho	mimet	ic agen	ıts	
			pha & Beta) and the							
	distribut			10						1,2
		nomimetic agents:								
		-	Direct acting- Nor-							
		rine, Epinephrine,								
	_	ne, Methyldopa, C								
		nine, Isoprotereno nol* Bitolterol N								
		nol*, Bitolterol, N azoline and Xylor	-							
	-	· ·	droxyamphetamine,							
			nexedrine. Agents							
		xed mechanism: E	-							
	Metaran		1,							
		gic Antagonists:	Alpha adrenergic							
		<u> </u>	. 6		_1					1

	1.11 T.1		1	
	blockers: Tolazoline*, Phentolamine,			
	Phenoxybenzamine, Prazosin,			
	Dihydroergotamine, Methysergide.			
	Beta adrenergic blockers: SAR of beta			
	blockers, Propranolol*, Metibranolol, Atenolol,			
	Betazolol, Bisoprolol, Esmolol, Metoprolol,			
	Labetolol, Carvedilol			
Ш	Cholinergic neurotransmitters:	10	Students will be able to learn the	1,2
	Biosynthesis and catabolism of acetylcholine.		drugs acting on	
	Cholinergic receptors (Muscarinic & Nicotinic)		ANS i.e., Parasympathomimetic	
	and their distribution.		agents	
	Para-sympathomimetic agents: SAR of			
	Para- sympathomimetic agents			
	Direct acting agents: Acetylcholine,			
	Carbachol*, Bethanechol, Methacholine,			
	Pilocarpine. Indirect acting/ Cholinesterase			
	inhibitors (Reversible & Irreversible):			
	Physostigmine, Neostigmine*, Pyridostigmine,			
	Edrophonium chloride, Tacrine			
	hydrochloride, Ambenonium chloride,			
	Isofluorphate, Echothiophate iodide,			
	Parathione, Malathion.			
	Cholinesterase reactivator: Pralidoxime			
	chloride. Cholinergic Blocking agents: SAR of			
	cholinolytic agents Solanaceous alkaloids and			
	analogues: Atropine sulphate, Hyoscyamine			
	sulphate, Scopolamine hydrobromide,			
	Homatropine hydrobromide, Ipratropium			
	bromide*.			
	Synthetic cholinergic blocking agents:			
	Tropicamide, Cyclopentolate hydrochloride,			
	Clidinium bromide, Dicyclomine			
	hydrochloride*, Glycopyrrolate, Methantheline			
	bromide, Propantheline bromide, Benztropine			
	mesylate, Orphenadrine citrate, Biperidine			
	hydrochloride, Procyclidine hydrochloride*,			
	Tridihexethyl chloride, Isopropamide iodide,			
	Ethopropazine hydrochloride.			
IV	Drugs acting on Central Nervous System:	8	Students will be able to learn	1,2
	A. Sedatives and Hypnotics:		the drugs acting on CNS	
	Benzodiazepines: SAR of Benzodiazepines,			
	Chlordiazepoxide, Diazepam*, Oxazepam,			
	Chlorazepate, Lorazepam, Alprazolam,			
	Zolpidem Barbiturtes: SAR of barbiturates,			
	Barbital*, Phenobarbital, Mephobarbital,			
	Amobarbital, Butabarbital, Pentobarbital,			
	Secobarbital Miscelleneous: Amides &			
	imides: Glutethmide. Alcohol & their			
	carbamate derivatives: Meprobomate,			
	Ethchlorvynol. Aldehyde & their derivatives:			
	Ememoryhor. Aldenyde & men derivatives:			1

	m: 10 ti n 111 1		I	
	Triclofos sodium, Paraldehyde.			
	<b>B.</b> Antipsychotics Phenothiazeines: SAR of			
	Phenothiazeines – Promazine hydrochloride,			
	Chlorpromazine hydrochloride*,			
	Triflupromazine, Thioridazine hydrochloride,			
	Piperacetazine hydrochloride,			
	Prochlorperazine maleate, Trifluoperazine			
	hydrochloride. Ring Analogues of			
	henothiazeines: Chlorprothixene, Thiothixene,			
	Loxapine succinate, Clozapine. Fluro			
	buterophenones: Haloperidol, Droperidol,			
	Risperidone. Beta amino ketones: Molindone			
	hydrochloride. Benzamides: Sulpieride.			
	C. Anticonvulsants: SAR of Anticonvulsants,			
	mechanism of anticonvulsant action			
	Barbiturates: Phenobarbitone, Methabarbital.			
	Hydantoins: Phenytoin*, Mephenytoin,			
	Ethotoin Oxazolidine diones:			
	Trimethadione, Paramethadione Succinimides:			
	Phensuximide, Methsuximide, Ethosuximide*			
	Urea and monoacylureas: Phenacemide,			
	Carbamazepine* Benzodiazepines:			
	Clonazepam Miscellaneous: Primidone,			
	Valproic acid, Gabapentin, Felbamate			
V	Drugs acting on Central Nervous System:		Students will be able to learn	
,	General anesthetics: Inhalation anesthetics:		the SAR of the drugs acting on	
	Halothane*, Methoxy flurane, Enflurane,		ANS and CNS	
	Sevoflurane, Isoflurane, Desflurane. Ultra			
	short acting barbitutrates: Methohexital			
	sodium*, Thiamylal sodium, Thiopental			
	sodium.			
	Dissociative anesthetics: Ketamine			
	hydrochloride.* Narcotic and non-narcotic			
	analgesics Morphine and related drugs: SAR			
	of Morphineanalogues, Morphine sulphate,			
	Codeine, Meperidine hydrochloride,			
	Anilerdine hydrochloride, Diphenoxylate			
	hydrochloride, Loperamide hydrochloride,			
	1			
	Fentanyl citrate*, Methadone hydrochloride*,			
	Propoxyphene hydrochloride, Pentazocine,	7		1.2
	Levorphanol tartarate.	7		1,2
	Narcotic antagonists: Nalorphine			
	hydrochloride, Levallorphantartarate, Naloxone			
	hydrochloride. Anti-inflammatory agents:			
	Sodium salicylate, Aspirin, Mefenamic acid*,			
	Meclofenamate, Indomethacin, Sulindac,			
	Tolmetin, Zomepriac, Diclofenac, Ketorolac,			
	Ibuprofen*, Naproxen, Piroxicam, Phenacetin,			
	Acetaminophen, Antipyrine, Phenylbutazone.			

T1: Medicinal Chemistry – I by Dr. Sanjay G. Walode – Nirali Prakash Publication T2: Medicinal Chemistry – I by K.G. Bothara - Nirali Prakash Publication

T3: Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry. T4: Foye's Principles of Medicinal Chemistry.

T5: Burger's Medicinal Chemistry, Vol I to IV. T6: Martindale's extra pharmacopoeia.

T7: The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1-5.

### **REFERENCE BOOKS:**

R1: Remington's Pharmaceutical Sciences.

R2: Introduction to principles of drug design- Smith and Williams. R3: Organic Chemistry by I.L. Finar, Vol. II.

R4: Indian Pharmacopoeia.

R5: Text book of practical organic chemistry- A.I.Vogel.

https://www.carewellpharma.in/bpharmacy/notes/4th-sem/medicinal-chemistry-1

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Understand the basic principles of medicinal chemistry	PO1,PO3,PO8, PO11					
2	Identify the cause of disease or disorder and the mechanism	PO1,PO2,PO3,PO6, PO8,					
	of action for drugs acting on ANS	PO11					
3	Illustrate how the chemistry of pharmaceuticals relates to their pharmacological and pharmacokinetic profiles.	PO1,PO2,PO3,PO6, PO8,					
3	their pharmacological and pharmacokinetic profiles.	PO11					
	Recall the structural activity relationship (SAR) between	PO1,PO2,PO3,PO6, PO8,					
4	various medication classes and the biological activity of	PO11					
	drugs acting on CNS	1011					
5	Understand the synthesis of various drug products with	PO1,PO2,PO3,PO6, PO8,					
3	classification based on chemical structure.	PO11					

	SEMESTER – IV											
Cour	ourse Title Physical Pharmaceutics-II											
Course code				L	T	P	S	R	O/F		С	
			Total hours: 45T	3	1	0	0	0	0		4	
Pre-r	equisite	Nil	Co-requisite					Nil	1			
Prog	ramme		Bache	lor (	of Phai	rmacy						
Sen	nester		Fall/ IV semester of	of 2 ⁿ	d year	of the	prog	ramme				
Co	urse	1. Understa	nd various physicochemica	l pro	opertie	s of dr	rug m	olecule	s in design	ning	the	
Obje	ectives	dosage f	orms									
			e principles of chemical kin			use the	em fo	r stabil	ity testing	and		
			ation of expiry date of form									
			trate use of physicochemica	al pro	opertie	s in de	velop	oing and	l evaluatin	g do	sage	
		forms.										
C	CO1		he physicochemical prope						ispersed sy	ster	n, and	
			nd electrical properties in	•								
C	CO2		sic understanding of the rh			flow	prope	erties of	solids and	l flu	ids	
_			ication in pharmaceutical s									
C	<b>CO3</b>	_	sic knowledge of pharmace		_					-	tems	
	10.4		f surfactants and interfacial	•		•						
C	CO4		ne micromeritics and size d	ıstrı	butions	s of va	rious	pharma	aceutical d	osag	ţе	
	105		ir applications		4:1	1:	.4:	- C		1	1	
C	CO5		Understand the chemical kinetics, pharmaceutical applications of various physical and chemical behaviors of dosage forms and their accelerated stability testing procedure,									
			_	ia tr	neir ac	ceierai	ea st	ability	testing pro	ocea	lure,	
Unit-		r	gradation, prevention, etc.  Content		Contact	4	Lagr	mina C	···taama		KL	
No.		,	Content		Jontaci Hour	L	Leai	ning C	Outcome		KL	
I 110.	Callaid	al dispersions	: Classification of dispersed		Hour	Stude	ente v	vill he	able to lea	rn		
-		_	al characteristics, size &	•					significanc			
	1 -	of colloidal pa					of coll	-	,			
	_	s & comparative account of their general							maceutical			
		s. Optical, kinetic & electrical properties.			5			d the pi			1,2	
	_		coacervation, peptization &					•	g formulat	ion		
		ve action.				_		pment				
II	Rheolog	gy: Newtonian	systems, law of flow,			Stude	ents v	vill be a	able to lea	rn		
	kinemat	tic viscosity, effect of temperature, non-				Newto	onian	and no	n-newtoni	an		
	Newton	ian systems, pseudoplastic, dilatant, plastic				flow,	visco	sity and	d the			
		ppy, thixotropy in formulation,							oles govern	ning		
		nation of viscosity, capillary, falling			10	defor	matio	n of so	lids		1,2	
	_	rotational viscometers Deformation of										
		Plastic and ela										
	_		n, Elastic Modulus									
III		-	spension, interfacial		10				able to lear	rn	1,2	
	r -	perties of suspended particles, settling in				1 -			lity and			
	suspensions, formulation of floce			25	1 *			erformance of Coarse ispersion and develop				
			ions. Emulsions and theori									
		Isification, microemulsion and multiple						suspen the fie				
		ons; Stability of emulsions, preservation of						n the 116 tical sc				
		ns, rheological properties of emulsions and				huatu	naceu	iicai sc	ichice.			
IV		on formulation by HLB method. neretics: Particle size and distribution, mean				Stude	nte -	vill ba	able to lea	rr		
l V	pviicrom	erenes: Partic	ie size and distribution, me	an		Stude	ents v	viii be a	adie to lea	L U		

	particle size, number and weight distribution, particle number, methods for determining particle size by different methods, counting and separation method, particle shape, specific surface, methods for determining surface area, permeability, adsorption, derived properties of powders, porosity, packing arrangement, densities, bulkiness & flow properties.	10	assess Micromeritics, determination methods of particle size, and properties of powders and its pharmaceutical application.	1,2
V	Drug stability: Reaction kinetics: zero, pseudo- zero, first & second order, units of basic rate constants, determination of reaction order. Physical and chemical factors influencing the Chemical degradation of pharmaceutical product: temperature, solvent, ionic strength, dielectric constant, specific & general acid base catalysis, Simple numerical problems. Stabilization of medicinal agents against common reactions like hydrolysis & oxidation. Accelerated stability testing in expiration Dating of pharmaceutical dosage forms. Photolytic degradation and its prevention	10	Students will be able to learn Kinetics of dosage forms, accelerated stability study and in Expiration dating of pharmaceutical dosage forms as well as the photolytic degradation and its prevention.	1,2

T1: Text book of Physical Pharmaceutics by C.V.S. Subramanyam, Vallabh Prakashan.

T2: Physical Pharmacy by S.P Agarwal and Rajesh Khana, CBS Publishers and distributors PVT Ltd; 2nd Edition.

### **REFERENCE BOOKS:**

R1: Physical Pharmacy by Alfred Martin

R2: Liberman H.A, Lachman C, Pharmaceutical dosage forms. Disperse systems, volume 1, 2,3. Marcel Dekkar Inc.

https://www.carewellpharma.in/bpharmacy/notes/4th-sem/physical-pharmaceutics-2#google_vignette

CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome			
1	Understand the physicochemical properties of colloidal systems, dispersed system, and their kinetic and electrical properties in pharmaceutical sciences.	PO1 ,PO2,PO3, PO4, PO11			
2	Develop a basic understanding of the rheology and flow properties of solids and fluids and their application in pharmaceutical sciences.	PO1 ,PO2,PO3, PO4, PO11			
3	Develop a basic knowledge of pharmaceutical suspensions, emulsions and HLB systems and the role of surfactants and interfacial phenomena in pharmaceutical sciences.	PO1 ,PO2,PO3, PO4, PO11			
4	Understand the micromerities and size distributions of various pharmaceutical dosage forms and their applications	PO1 ,PO2,PO3, PO4, PO11			
5	Understand the chemical kinetics, pharmaceutical applications of various physical and chemical behaviors of dosage forms and their accelerated stability testing procedure, photolytic degradation, prevention, etc.	PO1 ,PO2,PO3, PO4, PO11			

			SEMESTE	R – IV							
Course	Title	Pharmacology-I									
Course code		BP404T	Total credits: 4	L	T	P	P S R O/F		O/F	C	
			Total hours: 45T	3	1	0	0	0	0	4	
Pre-requisite		Nil	Co-requisite				N	il			
Progra				elor of P							
Seme	ster	Fall/ IV semester of 2 nd year of the programme									
Cou			ne pharmacological ac								
Objec	tives	2. Explain the mechanism of drug action at organ system/sub cellular/ macromolecular									
		levels.									
		3. Apply basic pharmacological knowledge to prevent and treat various diseases.									
			ffect of drugs on anim	•		_					
			orrelation of pharmaco								
CO	)1		emonstrate the Pharm						-	entia	
			ites of drug administra								
CO	2		oly the Principles and			of drug	action	n and I	Receptor tl	neori	
			dverse drug reaction n								
CO			drug action on the au							n.	
CO			s action on the central								
CO	5		herapy to different di	isease co	nditi	ons and	d unde	erstand	drug addi	ction	
		drug abuse, tolera	1								
Unit-		Cont	Contac	t	Learning Outcome			ome	KI		
No.				Hour							
I		al Pharmacology				Students will be able to learn					
		oduction to Phar			various drug sources and comprehend the essential drugs concept. Recognize and explain addiction, tolerance, dependence,						
		tion, historical lan									
	Г	acology, nature ar									
		ial drugs concept and routes of drug							-		
		istration, Agonists, antagonists(						-	erasy, and	1.	
		etitive and non competitive), spare		8			_		and the	1,2	
	_	ors, addiction, tolerance, dependence,				cesses					
		hylaxis, idiosyncrasy,allergy.b.				sorption					
		acokinetics- Membrane transport,			me	tabolis	m, and	excret	cion		
	_	tion, distribution, metabolism and ion of drugs. Enzyme induction,									
		e inhibition, kinet	ics of elimination		C4-	- d a4a	:11 h.	ahla 4			
II		al Pharmacology	Dringinles and						to learn	,	
		rmacodynamics-			princij g actio			nanisms of			
		nisms of drug acti assification of rece			•		•				
		ors. drug receptors	12		receptor theories and the classification of receptors.			1,2			
	_		s, G-protein–coupled	12		plore di		_	018.	1,2	
		ors, ion channel re				eraction	_	_			
	_	embrane enzyme	•			nsducti		_	ns		
		embrane JAK-S	_						ctions at		
		or and receptors th	-			pharm					
	_	ription factors, do	-		LIIC	իսայու	acorii)	10	· U1.		
		nship, therapeutic	•								
			ors modifying drug								
	action.	-	oro mourrying drug								
	action.	•									

	b. Adverse drug reactions. c. Drug interactions (pharmacokinetic and pharmacodynamic)			
III	pharmacodynamic)  d. Drug discovery and clinical evaluation of new drugs -Drug discovery phase, preclinical evaluation phase, clinical trial phase, phases of clinical trials and pharmacovigilance.  2. Pharmacology of drugs acting on peripheral nervous system a. Organization and function of ANS. b. Neurohumoral transmission, co-transmission and classification of neurotransmitters. c. Parasympathomimetics, Parasympatholytics,		Students will be able to learn the organization and fundamental functions of the autonomic nervous system. Comprehend the mechanisms of neurohumoral transmission and co- ransmission.	
	Sympathomimetics, sympatholytics. d. Neuromuscular blocking agents and skeletal muscle relaxants (peripheral). e. Local anesthetic agents. f. Drugs used in myasthenia gravis and glaucoma	10	Understand the pharmacology of parasympathomimetics and their effects. Understand the pharmacology of sympatholytics and their clinical applications	1,2
IV	Pharmacology of drugs acting on central nervous system Neurohumoral transmission in the C.N.S. special emphasis on importance of various neurotransmitters like with GABA, Glutamate, Glycine, serotonin, dopamine. General anesthetics and pre-anesthetics. Sedatives, hypnotics and centrally acting muscle relaxants. Anti-epileptics Alcohols and disulfiram	8	the significance of various neurotransmitters in the CNS, including GABA, glutamate, glycine, serotonin, and dopamine. Comprehend the mechanisms of neurohumoral transmission in the CNS. Learn about the pharmacology, mechanisms of action, and clinical uses of general anesthetics and preanesthetics.	1,2
V	Pharmacology of drugs acting on central nervous system  a. Psychopharmacological agents: Antipsychotics, antidepressants, antianxiety agents, anti-manics and hallucinogens.  b. Drugs used in Parkinsons disease and Alzheimer's disease.  c. CNS stimulants and nootropics. d. Opioid analgesics and antagonists e. Drug addiction, drug abuse, tolerance and dependence.	7	Students will be able to learn the pharmacology, mechanisms, and therapeutic uses of CNS stimulants.  Comprehend the pharmacological actions and potential cognitive benefits of nootropic agents.  Learn about the pharmacology, mechanisms of action, and clinical applications of opioid analgesics.	1,2

T1: Goodman and Gilman's, The Pharmacological Basis of Therapeutics, 13th Edition (2017).

### **REFERENCE BOOKS:**

R1: Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology Churchil

Livingstone Elsevier, 9th Edition (2019).

- R2: Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata Mc Graw Hill, 12th Edition (2012).
- R3: Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A. K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins, 9th Edition (2008).
- R4: Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews- Pharmacology, 8th Edition (2022).
- R5: K. D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi, 8th Edition (2019).
- R6: Sharma H. L., Sharma K. K., Principles of Pharmacology, Paras medical publisher, 2nd Edition (2012).
- R7: Modern Pharmacology with clinical Applications, by Charles R. Craig & Robert, 6th Edition (2003).
- R8: Ghosh MN. Fundamentals of Experimental Pharmacology. Hilton & Company, Kolkata, 7th Edition (2019).
- R9: Kulkarni SK. Handbook of experimental pharmacology. Vallabh Prakashan (2014). . <a href="https://www.cartercenter.org/resources/pdfs/health/ephti/library/lecture_notes/health_science_student_s/Pharmacology.pdf">https://www.cartercenter.org/resources/pdfs/health/ephti/library/lecture_notes/health_science_student_s/Pharmacology.pdf</a>

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
	Understand and demonstrate the Pharmacology, nature, and					
1	source of drugs, the essential drug concept, routes of drug	PO1,PO3,PO6,PO9,PO11				
	administration, and the Pharmacokinetics of drugs.					
	Analyze and Apply the Principles and mechanisms of drug					
2	action and Receptor theories and learn about adverse drug	PO1,PO3,PO6,PO9,PO11				
	reaction management.					
3	Understand about drug action on the autonomous nervous	PO1,PO3,PO6,PO9,PO11				
3	system and its mechanism.	101,103,100,103,1011				
4	Explain the drug's action on the central nervous system and	PO1,PO3,PO6,PO9,PO11				
<b>-</b>	its mechanism.	101,103,100,103,1011				
	Apply the drug therapy to different disease conditions and					
5	understand drug addiction, drug abuse, tolerance, and	PO1,PO3,PO6,PO9,PO11				
	dependence.					

	SEMESTER – IV										
Course	Title	]	PHARMACOGNOS	SY AND	PHY	YTOCI	HEMI:	STRY	I		
Course	code	BP405T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 45T	3	1	0	0	0	0	4	
Pre-rec		Nil	Co-requisite				N	il			
Progra				elor of							
Seme			Fall/ IV semester								
Cou			techniques in the cul		•			erude d	rugs		
Objec	tives		crude drugs, their us				re				
			aluation techniques for			_	.•	C 1			
			4. To carry out the microscopic and morphological evaluation of crude drugs								
CO	<b>)</b> 1	Understand the arrangement of crude drugs from natural sources and their									
CO		tandardization techniques.									
CO	)2	llustrate the factors for cultivating healthy medicinal plants and explain their collection and commercial conservation.									
CC				14		1 :4:	: c: -		1		
CO			mportance of plant tis							gnosy.	
CO	<b>74</b>		strate the various tylditional and modern s	-		-	ciad011	nes an	u meir		
CC	) <del>5</del>		t plant fibers, source				litas f	rom no	ture and		
	<i>)</i> 5	_	-	_	шаг у	Пістанс	incs i	10111 11a	iture, and		
Unit-			erapeutic agents from marine sourc			Ιρ	arnina	Outce	nme.	KL	
No.		Conte	Content			Learning Outcome					
I	Introd	uction to Pharm	acognosy:	Hour		udents	will be	e able 1	to learn		
		finition, history,	•						l drugs.		
	1 '	pment of Pharma	-			unders			_		
	1	•	Animals, Marine &		tec	chnique	s for n	atural c	lrugs		
	l .	culture	·			•					
	(b) Or	ganized drugs, un	organized drugs								
	(dried	latex, dried juices	, dried extracts,								
	gums a	and mucilages, old	eoresins and oleo-								
	gum -r	esins).									
	Classif	ication of drugs:	Alphabetical,								
	_	ological, taxonom									
	1 -	acological, chemo	and sero								
	taxono			10						1,2	
		y control of Dru									
	_		drugs of natural								
	_	Evaluation by org	•								
		copic, physical, c									
	_	cal methods and									
	1	tative microscopy	_								
	including lycopodium spore method, leafconstants, camera lucida and diagram										
			o scale with camera								
	lucida.										
П		ation, Collection	, Processing and		St	udents	will be	e able 1	to learn		
		e of drugs of nat	_			Students will be able to learn the factors affecting medicinal					
	_	ation and Collecti						_	tion, and		
		origin Factors in	~		r	mmerci					
		tion of medicinal	-	10						1,2	

	hormones and their applications. Polyploidy, mutation and hybridization with reference to medicinal plants			
III	Plant tissue culture: Historical development of plant tissue culture, types of cultures, Nutritional requirements, growth and their maintenance. Applications of plant tissue culture in pharmacognosy. Edible vaccines	7	Students will be able to learn Of advance techniques for cultivation.	1,2
IV	Pharmacognosy in various systems of medicine: Role of Pharmacognosy in allopathy and traditional systems of medicine namely, Ayurveda, Unani, Siddha, Homeopathy and Chinese systems of medicine. Introduction to secondary metabolites: Definition, classification, properties and test for identification of Alkaloids, Glycosides, Flavonoids, Tannins, Volatile oil and Resins	10	Students will be able to learn the utilization of natural drugs in different system of medicine.  To know the chemical nature of the natural drugs.	1,2
V	following drugs Plant Products: Fibers - Cotton, Jute, Hemp Hallucinogens, Teratogens, Natural allergens Primary metabolites: General introduction, detailed study with respect to chemistry, sources, preparation, evaluation, preservation, storage, therapeutic used and commercial utility as Pharmaceutical Aids and/or Medicines for the following Primary metabolites: Carbohydrates: Acacia, Agar, Tragacanth, Honey Proteins and Enzymes: Gelatin, casein, proteolytic enzymes (Papain, bromelain, serratiopeptidase, urokinase, streptokinase, pepsin). Lipids(Waxes, fats, fixed oils): Castor oil, Chaulmoogra oil, Wool Fat, Bees Wax Marine Drugs: Novel medicinal agents from marine sources	8	Students will be able to learn about additive products of natural sources.	1,2

T1: Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhlae (2007), 37th Edition, Nirali Prakashan, New Delhi.

T2: Text Book of Pharmacognosy by T.E. Wallis.

## **REFERENCE BOOKS:**

R1: W.C.Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Sounders & Co., London, 2009. R2: Tyler, V.E., Brady, L.R. and Robbers, J.E., Pharmacognosy, 9th Edn., Lea and Febiger, Philadelphia, 1988.

- R3: Mohammad Ali. Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi.
- R4: Essentials of Pharmacognosy, Dr.SH. Ansari, IInd edition, Birla publications, New Delhi, 2007
- R5: Anatomy of Crude Drugs by M.A. Iyengar
- R6: Practical Pharmacognosy: C.K. Kokate, Purohit, Gokhlae

R7: Pharmacognosy for diploma in pharmacy: Mohidul Islam, Dr. Faruk Alam. <a href="https://www.carewellpharma.in/bpharmacy/notes/4th-sem/pharmacognosy-1">https://www.carewellpharma.in/bpharmacy/notes/4th-sem/pharmacognosy-1</a>

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the arrangement of crude drugs from natural sources and their standardization techniques.	PO1,PO3,PO10,PO11
2	Illustrate the factors for cultivating healthy medicinal plants and explain their collection and commercial conservation.	PO1,PO3,PO8,PO7, PO10,PO11
3	Understand the importance of plant tissue culture and its significance in pharmacognosy.	PO1,PO4,PO7,PO10,PO11
4	Describe to illustrate the various types of secondary metabolites and their utilization in traditional and modern systems of medicine.	PO1
5	Explain different plant fibers, sources of primary metabolites from nature, and therapeutic agents from marine sources.	PO1

	SEMESTER IV											
Cours	se Title		Medicinal	Chemis	try –	I (Pra	ctical)					
Cours	se code	BP406P	<b>Total credits: 2</b>	L	T	P	S	R	O/F	C		
			Total hours: 4	0	0	4	0	0	0	2		
Pre-re	equisite	Nil	Co-requisite	Nil								
Progr	Programme Ba				chelor of Pharmacy							
Semester Fall/ IV semest				of thire	l year	r of the	progi	amme				
Co	urse	4. Analyz	ze crude drugs using n	nicrosco	pic a	nd cher	nical n	nethods				
Obje	ectives	5. Detern	nine crude drug purity	using q	uantit	ative m	nicrosc	opic m	ethods.			
C	Synthesize the selected drugs or drugs intermediate as per the synthetic scheme.											
C	Prepare standard solutions as per the volumetric monograph.											
C	O3	Carry out the ass	say procedure to check	k the pu	ity of	f the se	lected	drugs.				
C	O4	Determine the pa	artition coefficient of	any sele	cted o	drugs.						
C	O5	Determine the pl	hysicochemical prope	erties of drugs and draw their importance.								
Unit-		Conte	ent	Conta	et	Le	arning	g Outco	ome	KL		
No.				Hour								
I	Prepara	ation of drugs/ i	ntermediates	4	Stı	ıdents	will b	e able t	to learn	1,2		
		zole 1,3-oxazole			Ste	ereo-ch	emical	feature	es			
	Benztria	zole 2,3-dipheny	d quinoxaline		inc	luding	confo	mation	and			
	Benzoca	ine Phenytoin Pl	nenothiazine		ste	reoelec	tronic	effects	of organi	c		
	Barbitur	ate Assay of dru	<b>igs</b> Chlorpromazine		molecules.							
		-	Ibuprofen Aspirin	The students should be able								
	Furosen	nide <b>Determinat</b>	ion of Partition		gain knowledge about							
	coeffici	ent for any two	drugs		Ge	ometri	cal Iso	merism	l <b>.</b>			

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Synthesize the selected drugs or drugs intermediate as per	PO1,PO2,PO3,PO5,PO7,PO8,
1	the synthetic scheme.	PO11
2	Prepare standard solutions as per the volumetric monograph.	PO1,PO2,PO3,PO5,PO7,PO8,
_ <u></u>	r repare standard solutions as per the volumetric monograph.	PO11
2	Carry out the assay procedure to check the purity of the	PO1,PO2,PO3,PO5,PO7,PO8,
3	selected drugs.	PO11
4	Determine the partition coefficient of any selected drugs.	PO1,PO2,PO3,PO5,PO7,PO8,
4	Determine the partition coefficient of any selected drugs.	PO11
5	Determine the physicochemical properties of drugs and draw	PO1,PO2,PO3,PO5,PO7,PO8,
3	their importance.	PO11

	SEMESTER IV  Course Title Physical Pharmaceutics- II (Practical)													
						II (Pr	,	)						
Cour	se code	BP407P	Total credits: 2	L	T	P	S	R	O/F	C				
			Total hours: 4	0	0	4	0	0	0	2				
	equisite	Nil	Co-requisite				ľ	Nil						
	ramme			helor of										
	ester	Fall/ III semester of third year of the programme												
	urse	Measure reaction rate constants and half-lives of various order reactions.												
Obje	ectives		1 & 1 1											
		3. Assess fluid viscosity  Understand various partials size analysis methods for formulation development												
C	01	Understand various particle size analysis methods for formulation, development												
		of a pharmaceutical dosage form												
C	<b>O2</b>	Understand different physicochemical characteristics of drugs and excipients for												
		designing pharmaceutical dosage form												
	03	Elaborate on the mechanism and application of various types of viscometers.												
C	<b>O</b> 4	Describe the physical stability, settling tendencies, and efficacy of various suspension												
	0.	agents  Illustrate the fundamentals of chemical kinetics and apply them to determine the												
C	<b>O</b> 5			cal kineti	cs ai	nd app	ly ther	n to de	etermine tl	ne				
TT *4	T	expiration date of		<u> </u>		т	•	0.4		TZT				
Unit- No.		Conte	ent	Contact Hour		Le	arning	<b>Outco</b>	ome	KL				
110.	1 Do	tarmination of no	поиг	C4	donta	will be	abla 4	a laawn						
		_	nrticle size, particle ng sieving method		Students will be able to learn Determination of particle size,									
			article size, particle			particle size distribution using								
		e distribution usi	•						ination of					
		thod			ticle si									
		termination of bu				_								
		nsity, and porosit	distribution using Microscopic method Determination of bulk											
		termine the angle		density, true density, and porosity										
		_	nt on angle of repose											
		termination of vi			Determine the angle of repose and influence of lubricant on									
		ng Ostwald's vis	•			gle of re		Tublica	ant on					
		•	edimentation volume		_		_	f visco	sity of					
		th effect of differ				iid usir			on or					
		ents	ent suspending		_	comete	-		ion of					
I	_		edimentation volume	4					with effect	1,2				
			ent concentrations of						agents					
		ingle suspending					_	_	nentation					
			scosity of semisolid						lifferent					
		using Brookfield	•			centra								
		-							510					
		first order	rmination of reaction rate constant suspending agent Determination of viscosity of											
			action rate constant						okfield					
		second order	Combant			comete	•	5.5						
		celerated stabilit	v studies					of reac	tion rate					
	11. /10		, 2000100											
					constant for first order  Determination of reaction rate									
						istant f								
						celerate								
					AU	cicial	cu sia0	mity St	uares					

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand various particle size analysis methods for formulation, development of a pharmaceutical dosage form	PO1,PO2,PO3,PO4,PO,PO11
2	Understand different physicochemical characteristics of drugs and excipients for designing pharmaceutical dosage form	PO1,PO2,PO3,PO4,PO,PO11
3	Elaborate on the mechanism and application of various types of viscometers.	PO1,PO2,PO3,PO4,PO,PO11
4	Describe the physical stability, settling tendencies, and efficacy of various suspension agents	PO1,PO2,PO3,PO4,PO,PO11
5	Illustrate the fundamentals of chemical kinetics and apply them to determine the expiration date of a formulation	PO1,PO2,PO3,PO4,PO,PO11

	SEMESTER IV												
Cou	rse Tit	de	PHARMAC	COLO	GY-I (	Pra	ctical)						
Cou	rse coo	de BP408P	Total credits: 2	L	T	P	S	R	O/F	C			
			Total hours: 4	0	0	4	0	0	0	2			
	requisi		Co-requisite					Nil					
	gramn				Pharn	•							
	mester		Fall/ III semester of		•				;				
	ourse	l l	e of laboratory animals a		CPCSE	EA g	uideline	S					
	jective		fects of drugs using soft										
•	C <b>O</b> 1				nd routes of drug administration and								
			dy fluids in experimenta										
	CO2		CPCSEA guidelines and	d study	of co	mmo	only use	ed instr	uments in				
		experimental pha											
	C <b>O3</b>		ct of drugs on animals b	-		_							
	C <b>O</b> 4	1	chanisms of drug action			_							
	C <b>O</b> 5		cess of new drug discov	ery an						T			
Unit-		Co	ontent		Conta	- 1	Lear	ning O	utcome	KL			
No.	1 7				Hou			*** *					
I		-	imental pharmacology		4	L		s will b	be able to	3,4,			
		*	nmonly used instruments in experimental				earn			5,6			
	•	oharmacology Study of common laboratory animals				Introduction to experimental							
		*	•				_						
			ratory animals as per				harmac		.1				
		CPCSEA guidelines	eachailannach blaad vrithd				Commo nstrume	-	ea				
			techniques: blood withd paration, anesthetics, an										
		uthanasia used for a		u			xperim		Study of				
			nmar studies ites of drug administrati	on in		Г	ommor		-				
		nice/rats	ites of drug administrati	OH III			nimals	1 14001	atory				
			ly of effect of hepatic microsomal enzyme			Maintenance of							
		•	obarbitone sleeping time				aborato		•				
		nice	ocuronone steeping time	• 111	animals as per								
			iary motility of frog		CPCSEA								
		sophagus	, , ,				uidelin		nmon				
		Effect of drugs on ral	obit eye						niques:				
		_	scle relaxants using rot	a-rod			olood v	•	•				
		pparatus	Č			s	erum a	and pla	sma				
			comotor activity using					_	esthetics,				
	a	ctophotometer				a	nd eutl	nanasia	used for				
	12. A	anticonvulsant effec	t of drugs by MES and F	PTZ		a	nimal s	tudies	Study of				
	n	nethod				ć	lifferent	routes	s of drug				
	13. S	tudy of stereotype a	nd anti-catatonic activit	y of		a	dminist	ration	in				
	d	rugs on rats/mice				r	nice/rat	s Study	of effect				
	14. Study of anxiolytic activity of drugs using							of hepa					
	ra	ats/mice				microsomal enzyme							
		•	etics by different method		inducers on the								
		•	iques and animal exper			r	henoba	rbitone	e sleeping				
		•	lated experiments by so	oftware		t	ime						
	and vi	deos.				i	n mice						

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Summarize the Pre-clinical Pharmacology and routes of drug administration and collection of body fluids in experimental animals.	PO1, PO2, PO6, PO7, PO8, PO11
2	Summarize the CPCSEA guidelines and study of commonly used instruments in experimental pharmacology.	PO1, PO2, PO6, PO7, PO8, PO10, PO11
3	Identify the effect of drugs on animals by simulated experiments	PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO11
4	Illustrate the mechanisms of drug action and its biological effects.	PO1,PO2, PO6, PO7, PO8, PO11
5	Develop the process of new drug discovery and development of a drug.	PO1, PO2, PO6, PO7, PO8, PO11

			SEMEST	TER IV								
Cour	se Title	PHAI	RMACOGNOSY A	ND PHY	TOC	CHEM	ISTR	Y I (Pr	actical)			
Cour	se code	BP409P	Total credits: 2	L	T	P	S	R	O/F	C		
			Total hours: 4	0	0	4	0	0	0	2		
Pre-re	equisite	Nil	Co-requisite				]	Nil				
Prog	ramme		Bachelor of Pharmacy									
Sen	nester		Fall/ III semester of third year of the programme									
	ourse											
Obje	ectives		extraction techniques									
		3. Identify unorganized drugs through qualitative chemical tests.										
	CO1	Understand the identification of crude drugs utilizing chemical evaluation methods.										
	CO2	Utilize physical evaluation methods to assess the quality and purity of crude drugs.										
C	CO3	Describe to explicate linear measurements for crude drug identification.										
C	<b>CO4</b>	Understand the illustration of quality control methods for the standardization of herbal										
		drugs.										
C	CO5	Define to exhibit t	the significance of the						rude drugs.			
Unit-		Conte	nt	Contac	ct Le	arning	Outc	ome		KL		
No.				Hour								
I	1	. Analysis of cru	• •	4					to learn	3,4,		
			Tragacanth, Acacia,			alysis		_		5		
		Agar, Gelatin,			emical		_					
		Castor oil		Acacia, Agar, Gelatin, Starch,								
	2		of stomatal number			Honey, Castor oil Determination of stomatal number and index						
		and index										
	3		of vein-islet number,	Determination of vein-isle								
			nation, and palisade						nination,			
		ratio	0		and palisade ratio Determination of size of starch grains, calcium							
	4	. Determination						_				
		-	oxalate crystals by			alate cr	-		_			
	_	eyepiece micro							tion of			
	3	width	of fiber length and			er leng			an of			
			of number of starch	Determination of number of starch grains by Lycopodium								
	0					_	-					
		grains by Lycop method	oodium spore						nation of			
	7	. Determination	of ach value		ash value Determination of extractive values of crude drugs							
			of extractive values	Determination of moisture								
	0	of crude drugs	or canachive values	content of crude drugs								
	Ω	. Determination	I - I									
		of crude drugs	or moisture content			d foam		ı swel	inig mucx			
	1	0. Determination	of swelling index		and	a ioaiii	mg					
			or swerring much									
		and foaming										

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the identification of crude drugs utilizing	PO1,PO2,PO3,PO5,PO8,PO1
1	chemical evaluation methods.	1
2	Utilize physical evaluation methods to assess the quality and	PO1,PO2,PO3,PO5,PO8,PO1
	purity of crude drugs.	1
3	Describe to explicate linear measurements for crude drug	PO1,PO2,PO3,PO5,PO8,PO1
3	identification.	1
4	Understand the illustration of quality control methods for the	PO1,PO2,PO3,PO5,PO8,PO1
-	standardization of herbal drugs.	1
5	Define to exhibit the significance of the physiochemical	PO1,PO2,PO3,PO4,
3	evaluation of crude drugs.	PO5,PO8,PO11

	SEMESTER – V												
Course	Title		Medicinal	Chemist	ry – l	I (The	ory)						
Course	code	BP501T	Total credits: 4	L	T	P	S	R	O/F	C			
			Total hours: 45T	3	1	0	0	0	0	4			
Pre-requ	uisite	Nil	Co-requisite				N	il					
Prograi				elor of I									
Semes			Fall/V semester o				rograi	nme					
Cour		_ ^	on of the course the stud										
Object	ives		I the chemistry of drugs		_	•		_	-	C			
			I the drug metabolic path	hways, a	dvers	e effect	s, and	therape	utic value	01			
		drugs 3. Know the Structural Activity Relationship of different classes of drugs											
		<ul><li>3. Know the Structural Activity Relationship of different classes of drugs</li><li>4. Study the chemical synthesis of selected drugs</li></ul>											
CO	1	·	classification, nomeno			nicture	activit	v relati	ionchin				
CO	L		· ·	-				•	•				
		concerning their mechanism of action of various antihistamines, proton pump inhibitors, and anti-neoplastic agents.											
CO	2		mical aspects, synthes		e of	action	. and	medic	inal bene	fits of			
			agents, including diuret										
		anti-hypertensi	-	,	0 0	,,			,				
CO	3	Explain the sy	nthetic methods, primar	y structi	ıral r	equiren	nents,	pharma	cophoric				
		features, and structural activity relationships for various classes of medicinal agents											
		used as anti- arrhythmics, anti-hyperlipidemics, coagulants, and drugs used in congestive											
		heart failure.											
CO			Describe hormones' role, structure, and biological and therapeutic significance.  Utilize the structural aspects and synthesis of agents for treating diabetes and										
CO	5			nthesis	of ag	ents fo	or trea	ting di	iabetes ar	ıd			
		local anesthesis				Learning Outcome KL							
Unit-		Со	ntent	Contac	t	Le	arning	g Outco	ome	KL			
No.	A 4°1	• , • •	4 II. 4 .	Hour	C.	1 4	11 1	11 / 1	1				
I		istaminic agen	distribution in the					ble to 1	iearn nechanism				
	_	an body	distribution in the						nic agents				
		ntagonists: Dip	henhydramine					e chemi	_	,			
		chloride*, Dime							drug and				
	-	laminescuccinat	-		-				al Activity				
	-		Diphenylphyraline						nt class of				
		chloride, Tripel			dru		1						
	hydro	chloride, Chlor	cyclizine hydrochloride,										
	Mecli	zine hydrochlor	ride,	10						2,3			
	Bucli	zine hydrochlo	ride, Chlorpheniramine										
		-	hydrochloride*,										
		daminetartarate											
			eprazine tartrate,										
		-	rochloride, Azatidine										
			Loratadine, Cetirizine,										
		cetrazine Cromo	•										
	H ₂ -ar Ranit	_	netidine*, Famotidine,										
		iain. cic Proton pum	un inhihitars										
		-	razole, Rabeprazole,										
	_	_	razore, raocprazore,										
	1 amo	Pantoprazole											

	Auti naculastia agantsi		1	
	Anti-neoplastic agents:			
	Alkylating agents: Meclorethamine*,			
	Cyclophosphamide, Melphalan,			
	Chlorambucil, Busulfan, Thiotepa			
	Antimetabolites: Mercaptopurine*,			
	Thioguanine, Fluorouracil, Floxuridine,			
	Cytarabine, Methotrexate*, Azathioprine			
	Antibiotics: Dactinomycin, Daunorubicin,			
	Doxorubicin, Bleomycin			
	Plant products: Etoposide,			
	Vinblastinsulphate, Vincristinsulphat			
	Miscellaneous: Cisplatin, Mitotane.			
II	Anti-anginal:	10	Student will be able to learn the	
	Vasodilators: Amyl nitrite, Nitroglycerin*,		classification, uses & mechanism	
	Pentaerythritoltetranitrate,		of action of antianginal agents. To	
	Isosorbidedinitrite*, Dipyridamole.		understand the chemical synthesis	
	Calcium channel blockers:		of selected drugs and understand	
	Verapamil, Bepridil hydrochloride,		The Structural Activity	
	Diltiazem hydrochloride, Nifedipine,		Relationship of different class of	
	Amlodipine, Felodipine, Nicardipine,		drugs	
	Nimodipine.			
	Diuretics:			
	Carbonic anhydrase inhibitors:			
	Acetazolamide, Methazolamide,			
	Dichlorphenamide. Thiazides:			
	Chlorthiazide, Hydrochlorothiazide,			
	Hydroflumethiazide, Cyclothiazide,			
	Loop diuretics: Furosemide, Bumetanide,			
	Ethacrynic acid.Potassium sparing			
	Diuretics: Spironolactone, Triamterene,			
	Amiloride. Osmotic Diuretics: Mannitol			
	Anti-hypertensive Agents: Timolol,			
	Captopril, Lisinopril, Enalapril, Benazepril			
	hydrochloride, Quinapril hydrochloride,			
	Methyldopate hydrochloride, Clonidine			
	hydrochloride, Guanethidinemonosulphate,			
	Guanabenz acetate, Sodium nitroprusside,			
	Diazoxide, Minoxidil, Reserpine,			
	Hydralazine hydrochloride			
III	Anti-arrhythmic Drugs: Quinidine		Student will be able to learn the	
	sulphate, Procainamide hydrochloride,		classification, uses & mechanism	
	Disopyramide phosphate, Phenytoin		of action of Anti-Arrhythmic	
	sodium, Lidocaine hydrochloride, Tocainide		Drugs, Anti-Hyperlipidemic	
	hydrochloride, Mexiletine hydrochloride,		Agents, Coagulant and	
	Lorcainide hydrochloride, Amiodarone,		Anticoagulants & Drugs used in	
	Sotalol.		Congestive Heart Failure agents	
	Anti-hyperlipidemicagents: Clofibrate,		To understand the chemical	
	Lovastatin, Cholesteramine and Cholestipol	10	synthesis of selected drugs. and to	2,3,
	Coagulant & Anticoagulants Menadione,		understand the Structural Activity	4
	Acetomenadione, Warfarin, Anisindione,		Relationship of different class of	

	clopidogrel		drugs.	
	Drugs used in Congestive Heart Failure:			
	Digoxin, Digitoxin, Nesiritide, Bosentan,			
	Tezosentan.			
IV	Drugs acting on Endocrine system		Student will be able to learn the	
	Nomenclature, Stereochemistry and		classification, uses & mechanism	
	metabolism of steroids		of action of Drugs acting on	
	Sex hormones: Testosterone, Nandralone,		Endocrine System, Drugs for	
	Progestrones, Oestriol, Oestradiol,		Erectile Dysfunction, Oral	
	Oestrione, Diethyl stilbestrol.		Contraceptives, Corticosteroids &	
	Drugs for erectile dysfunction: Sildenafil,		Thyroid and Anti Thyroid Drugs	
	Tadalafil.		To understand the chemical	
	Oral contraceptives: Mifepristone,	8	synthesis of selected drugs and	2,3,
	Norgestril, Levonorgestrol		understand the Structural Activity	4
	Corticosteroids: Cortisone, Hydrocortisone,		Relationship of different class of	
	Prednisolone, Betamethasone,		drugs.	
	Dexamethasone			
	Thyroid and antithyroid drugs: L-			
	Thyroxine, L-Thyronine, Propylthiouracil,			
	Methimazole.			
V	Antidiabetic agents: Insulin and its		Student will be able to learn the	
	preparations Sulfonylureas: Tolbutamide		classification, uses & mechanism	
	Chlorpropamide, Glipizide, Glimepiride.		of action of Antidiabetic Agents	
	Biguanides: Metformin. Thiazolidinediones:		& Local Anaesthetics To	
	Pioglitazone, Rosiglitazone. Meglitinides:		understand the chemical synthesis	
	Repaglinide, Nateglinide. Glucosidase		of selected drugs and understand	
	inhibitors: Acrabose, Voglibose.		the Structural Activity	
	Local Anesthetics: SAR of Local	7	Relationship of different class of	2,3,
	anesthetics		drugs.	4
	Benzoic Acid derivatives; Cocaine,			
	Hexylcaine, Meprylcaine, Cyclomethycaine,			
	Piperocaine.			
	Amino Benzoic acid derivatives:			
	Benzocaine, Butamben, Procaine,			
	Butacaine, Propoxycaine, Tetracaine,			
	Benoxinate.			
	Lidocaine/Anilide derivatives: Lignocaine,			
	Mepivacaine, Prilocaine, Etidocaine.			
	Miscellaneous: Phenacaine, Diperodon,			
	Dibucaine.			

T1: Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry. T2: Foye's Principles of Medicinal Chemistry.

## **REFERENCE BOOKS:**

R1: Burger's Medicinal Chemistry, Vol I to IV.

R2: Introduction to principles of drug design- Smith and Williams.

R3: The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1to 5.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the classification, nomenclature, and structure- activity relationship concerning their mechanism of action of various antihistamines, proton pump inhibitors, and anti- neoplastic agents.	PO1,PO2,PO3,PO6,PO8,PO11
2	Recall the chemical aspects, synthesis, mode of action, and medicinal benefits of cardiovascular agents, including diuretics, anti-aging, calcium channel blockers, and other anti-hypertensives.	PO1,PO2,PO3,PO6,PO8,PO11
3	Explain the synthetic methods, primary structural requirements, pharmacophoric features, and structural activity relationships for various classes of medicinal agents used as anti-arrhythmics, anti-hyperlipidemics, coagulants, and drugs used in congestive heart failure.	PO1,PO2,PO3,PO6,PO8,PO11
4	Describe hormones' role, structure, and biological and therapeutic significance.	PO1,PO2,PO3,PO6,PO8,PO11
5	Utilize the structural aspects and synthesis of agents for treating diabetes and local anesthesia drugs.	PO1,PO2,PO3,PO6,PO8,PO11

SEMESTER – V												
Course		Industrial Pharmacy I (Theory)  BP502T Total credits: 4 L T P S R O/F C										
Course	code	BP502T	Total credits: 4	L		T P S R O/F						
			Total hours: 45T	3	1	0	0	0	0	4		
Pre-requ		Nil	Co-requisite				N	il				
	rogramme Bachelor of Pharmacy Semester Fall/ V semester of third year of the programme											
Semes							rograi	mme				
Cour			of the course the st									
Object	ives		rious pharmaceutica	_								
			s considerations in t	•		•			•			
CO			lid, liquid, and semi									
CO			arious pharmaceutic									
CO			considerations in th				naceu	tical do	sage forms	•		
CO		_	lity of pharmaceutic	_					1			
CO ₄		***	of formulations of C									
CO	)		and official require	rements for	or Pa	ackaging	Mate	erials S	science and	1		
Unit-		quality control te		Contact		Т.		· Or-4:	2220	KL		
No.		Conte	ent	Contact Hour		Le	arnınş	g Outco	ome	KL		
I I	Drofo	rmulation Studi	es: Introduction to	Hour	C+	udente v	rill ha	able to	learn the	+		
1			and objectives,			ed of pro						
	r	_	ical characteristics		110	cu or pro		ilation	study.			
	1	g substances.	irear characteristics									
		a. Physical prope	erties: Physical									
	'	form (crystal &	•									
		particle size, si	- ′									
		-	ibility profile (pKa,									
		pH, partition of										
		polymorphism	,,									
	1		<i>perties:</i> Hydrolysis,	7						1,2		
		oxidation, redu	ction,									
		racemisation, p	olymerization									
		BCS classifica	tion of drugs & its									
		significant App	olication of									
		preformulation	considerations in									
		the developme	nt of solid, liquid									
		-	eral dosage forms									
		and its impact	on stability of									
		dosage forms.										
П	Table					udents w						
		troduction, ideal		4.0					terization	1.0		
		olets, classificatio		10	of	tablets a	and ora	al liquio	ds.	1,2		
		Excipients, Formulation of tablets,										
	granulation methods, compression and processing problems. Equipments											
			Dienis. Equipments									
		d tablet tooling.	es of coating									
		ablet coating: Typ ating materials, for	~									
		ating materials, in										
		ating composition ating, equipment										
	CO	anng, equipment	emproyed and									

	Asfasta in acating		T	
	defects in coating.			
	c. Quality control tests: In process and			
	finished product tests			
	Liquid orals: Formulation and			
	manufacturing consideration of syrups			
	and elixirs suspensions and emulsions;			
	Filling and packaging; evaluation of			
	liquid orals official in pharmacopoeia			
III	Capsules:		Students will be able to learn	
	a. Hard gelatin capsules: Introduction,		formulation and characterization	
	Production of hard gelatin capsule		of capsules and pellets.	
	shells. size of capsules, Filling,			
	finishing and special techniques of			
	formulation of hard gelatin capsules,			
	manufacturing defects. In process			
	and final product quality control tests			
	for capsules.			
	b. Soft gelatin capsules: Nature of shell			
	and capsule content, size of capsules,	8		1,2
	importance of base adsorption and			
	minim/gram factors, production, in			
	process and final product quality			
	control tests. Packing, storage and			
	stability testing of soft gelatin			
	capsules and their applications.			
	Pellets: Introduction, formulation			
	requirements, pelletization process,			
	equipments for manufacture of pellets			
IV	Parenteral Products:		Students will be able to learn	
1 4	a. Definition, types, advantages and		formulation and characterization	
	limitations. Preformulation factors		of parenteral and ophthalmic	
	and essential requirements,		preparations.	
	•		preparations.	
	vehicles, additives, importance of			
	isotonicity	10		1,2
	b. Production procedure, production	10		1,2
	facilities and controls, aseptic			
	processing			
	c. Formulation of injections, sterile			
	powders, large volume parenterals			
	and lyophilized products.			
	d. Containers and closures selection,			
	filling and sealing of ampoules,			
	vials and infusion fluids. Quality			
	control tests of parenteral products.			
	Ophthalmic Preparations: Introduction,			
	formulation considerations; formulation			
	of eye drops, eye ointments and eye			
	lotions; methods of preparation; labeling,			
	containers; evaluation of ophthalmic			
	preparations			
	r ^		ı	

V	Cosmetics: Formulation and preparation		Students will be able to learn	n
	of the following cosmetic preparations:		formulation and characterization	n
	lipsticks, shampoos, cold cream and		of cosmetic preparation and	d
	vanishing cream, tooth pastes, hair dyes		aerosols.s	
	and sunscreens.			
	Pharmaceutical Aerosols: Definition,			
	propellants, containers, valves, types of			
	aerosol systems; formulation and			
	manufacture of aerosols; Evaluation of			
	aerosols; Quality control and stability	10		1,2,
	studies.			3,4
	Packaging Materials Science: Materials			
	used for packaging of pharmaceutical			
	products, factors influencing choice of			
	containers, legal and official requirements			
	for containers, stability aspects of			
	packaging materials, quality control tests.			

T1: Theory and Practice of Industrial Pharmacy by Liberman & Lachman.

### **REFERENCE BOOKS:**

R1: Pharmaceutical dosage forms-Tablets, volume1-3byH.A. Liberman, Leon Lachman & J.B.Schwartz.

R2: Pharmaceutical dosage form-Parenteral medication vol-1 & 2 by Liberman & Lachman. R3:

Pharmaceutical dosage form disperse system VOL-1 by Liberman & Lachman.

R4: Modern Pharmaceutics by Gilbert S.Banker& C.T.Rhodes,3rd Edition.

R5: Remington: The Science and Practice of Pharmacy, 20th edition Pharmaceutical Science (RPS).

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the various pharmaceutical dosage forms and their manufacturing techniques.	PO1,PO2,PO6,PO10,PO11
2	Describe various considerations in the development of pharmaceutical dosage forms.	PO1,PO2,PO3,PO6,PO10,PO 11
3	Evaluate the quality of pharmaceutical dosage form	PO1,PO2,PO3,PO6,PO11
4	Apply the basics of formulations of Cosmetics and Pharmaceutical Aerosols.	PO1,PO2,PO6,PO10,PO11
5	Categorize legal and official requirements for Packaging Materials Science and quality control tests.	PO1,PO2,PO3,PO6,PO10,PO 11

			SEMEST	ER – V							
Cour	se Title		Pharm	acology-	II (T	heory)					
Cour	se code	BP503T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 45T	3	1	0	0	0	0	4	
Pre-re	equisite	Nil	Co-requisite				Ni	il			
Prog	ramme			elor of P		•					
Sen	nester		Fall/V semester of					nme			
Co	urse		on of this course the stu								
Obje	ectives		the mechanism of drug	g action a	nd its	releva	nce in t	the trea	tment of		
		different dis									
			e isolation of different	organs/tis	ssues	from tl	ne labo	ratory a	animals by		
		simulated e	•								
			e the various receptor a		_		_	_			
			he correlation of pharm								
C	01		hemodynamic and ele		_	•			•		
	102		chanism of action of dr								
C	O2	1 ^	chanism of action of		t dru	gs actii	ng on	blood	and blood	-	
	102		and the urinary system		. C 1''	· ·		1. 1	1.4 1 1		
	03	•	aluate the mechanism of							-	
C	<b>O</b> 4		ic concepts in endoc	_				-		iate	
	O5		he mechanism of action of different drugs acting on the endocrine system.								
C	.05	Utilize the principles and applications of bio assay, compare different types of bioassays, and illustrate the mechanism of action of various drugs acting on the female									
		reproductive system.									
Unit-	1	Cont		Contac	<i>t</i>	Ια	arnina	Qutco	)ma	KL	
No.		Cont	CIII	Hour	١	LC	ai mng	Guice	Jille	KL	
<u>I</u>	Pharma	acology of drugs	acting on cardio	Hour	Stu	idents v	vill be	able to	learn		
•		r system	acting on cardio						gestive		
		action to hemody	namic and			-	-		rtensive		
	1	hysiology of hea						• •			
		used in congestiv		10	drugs, Anti-anginal drugs, Anti- arrhythmic drugs,					1,2	
		ypertensive drug			Antihyperlipidemic drugs.						
	1 2	nginal drugs.					•				
			Anti-hyperlipidemic								
	drugs										
II	Pharma	acology of Drug	s Acting on		Stu	idents v	vill be	able to	learn		
		•	: a. Drugs used in the			•	-		herapy of		
			natinics, coagulants,					_	ulants and	1	
		~	brinolytics and anti-	10		_			ytics and	1,2	
	Г	-	volume expanders.		antiplatelet drugs, Plasma volum						
			s Acting on Urinary		^		, Diure	etics an	d Anti-		
	System: a. Diuretics. b. Anti-diuretics.					retics.					
III		ids and related o	O	10					to learn		
			oids and classification						ssification,		
			their antagonists.			stamine					
		staglandins, Thro	mboxanes and			agonist		-			
		kotrienes.	inin and Call at B						kotrienes,		
	_		inin and Substance P.			giotens		-			
			flammatory agents						idal anti-		
	I. Anti	i-gout drugs Anti	rneumatic drugs		ınt	ıammat	ory ago	ents, A	nti-gout		

			drugs, Antirheumatic drugs	
IV	Pharmacology of drugs acting on endocrine		Students will be able to learn	
	system		about basic concepts in endocrine	
	a. Basic concepts in endocrine pharmacology.		pharmacology, Anterior Pituitary	
	b. Anterior Pituitary hormones- analogues		hormones- analogues and their	
	and their inhibitors.		inhibitors, Thyroid hormones-	
	c. Thyroid hormones- analogues and their		analogues and their inhibitors,	
	inhibitors.	8	Hormones regulating plasma	1,2
	d. Hormones regulating plasma calcium		calcium level- Parathormone,	
	level- Parathormone, Calcitonin and		Calcitonin and Vitamin-D,	
	Vitamin-D.		Insulin, Oral Hypoglycemic	
	e. Insulin, Oral Hypoglycemic agents and		agents and glucagon, ACTH and	
	glucagon.		corticosteroids.	
	f. ACTH and corticosteroids.			
$\mathbf{V}$	Pharmacology of drugs acting on endocrine		Students will be able to learn	
	system		about androgens and Anabolic	
	a. Androgens and Anabolic steroids.		steroids, Estrogens, progesterone	
	b. Estrogens, progesterone and oral		and oral contraceptives, Drugs	
	contraceptives.		acting on the uterus	
	c. Drugs acting on the uterus.			
	Bioassay	7		1,2
	a. Principles and applications of bioassay.			
	b. Types of bioassay			
	c. Bioassay of insulin, oxytocin, vasopressin,			
	ACTH, d-tubocurarine, digitalis, histamine and 5-HT			

T1: Goodman and Gilman's, The Pharmacological Basis of Therapeutics, 13th Edition (2017).

### **REFERENCE BOOKS:**

R1: Rang H. P., Dale M. M., Ritter J. M., Flower R. J., Rang and Dale's Pharmacology Churchil Livingstone Elsevier, 9th Edition (2019).

R2: Katzung B. G., Masters S. B., Trevor A. J., Basic and clinical pharmacology, Tata McGrawHill, 12th Edition (2012).

R3: Marry Anne K. K., Lloyd Yee Y., Brian K. A., Robbin L.C., Joseph G. B., Wayne A. K., Bradley R.W., Applied Therapeutics, The Clinical use of Drugs, The Point Lippincott Williams & Wilkins, 9th Edition (2008).

R4: Mycek M.J, Gelnet S.B and Perper M.M. Lippincott's Illustrated Reviews- Pharmacology, 8th Edition (2022).

R5: K. D. Tripathi.Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi, 8th Edition (2019).

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Understand the hemodynamic and electrophysiology of the heart and analyze and estimate the mechanism of action of drugs that affect cardiovascular systems.	PO6,PO8,PO9,PO11						
2	Explain the mechanism of action of different drugs acting on blood and blood-forming organs and the urinary system.	PO6,PO8,PO9,PO11						
3	Analyze and evaluate the mechanism of action of different autocoids and related drugs.	PO6,PO8,PO9,PO11						
4	Discuss the basic concepts in endocrine pharmacology and analyze and evaluate the mechanism of action of different drugs acting on the endocrine system.	PO6,PO8,PO9,PO11						
5	Utilize the principles and applications of bioassay, compare different types of bioassays, and illustrate the mechanism of action of various drugs acting on the female reproductive system.	PO1,PO3,PO6,PO8,PO9,PO1 1						

SEMESTER – V											
Cour	se Title	Pharmacognosy And Phytochemistry II (Theory)									
Cour	rse code	BP504T	Total credits: 4	LI	1	P	S	R	O/F	C	
			Total hours: 45T	3 1		0	0	0	0	4	
Pre-r	equisite	Nil	Nil Co-requisite Nil								
Prog	ramme		Bachelor of Pharmacy								
Sen	nester		Fall/V semester of th	ird yea	r o	of the p	rogra	mme			
Co	ourse		etion of the course, the studen								
Obj	ectives		the modern extraction techni-	ques, ch	ara	acteriza	ition, a	ınd ider	ntification	of the	
			rugs and phytoconstituents								
			stand the preparation and dev	-	nt (	of herb	al forn	nulation	1.		
			stand the herbal drug interact								
	~~.		out isolation and identification								
	C <b>O</b> 1	1	the formation of secondary	metabo	)lite	es by s	severa	ı metal	bolic path	ways	
	701	in plants.		1	1.	1.					
(	C <b>O2</b>		composition, chemistry, che						rapeutic u	ses,	
	203		eial applications of medicinal	•					romiona al.	omia=1	
	C <b>O3</b>	_	method of isolation, ident from plant sources.	1110at101	1,	and at	naiysis	oi v	arious ch	emical	
	CO4		industrial production, esting	nation	040	d neili	zation	of ac	cential ak	emicol	
	JU4		from the plant.	nanon,	an	ia utili.	Zation	or es	Schilai Ch	Cillical	
(	CO5		he various extraction techniq	ues and	th	e role o	of spec	troscor	ny and		
	203		phic techniques in the isola				•	•	•		
Unit-			Content	Contac				g Out		KL	
No.				Hour				9			
I	Metabol	ic pathways	in higher plants and their	•	S	Students	will	be able	e to learn		
	determi				a	bout th	e syntl	netic pa	ıthways		
	Brief st	udy of basic 1	netabolic pathways and								
	formati	ion of different secondary metabolites									
	_	-	ays- Shikimic acid pathway,	7						2,3	
			d Amino acid pathway.								
			of radioactive isotopes in the								
	_		genetic studies.		-	v. 4					
II			composition, chemistry &						to learn		
		*	ources, therapeutic uses and						of natural		
	metaboli		ons of following secondary		-	roducts liseases		_			
			uwolfia, Belladonna, Opium,					•	on of nts from		
			id Flavonoids:	14		aw mate		пропс	iits iioiii	2,3	
		-	iu i iavonolus.	17	1.0	a vv 111al	<b>.</b> 1141.			2,5	
	Lignans, Tea, Ruta Steroids, Cardiac (		vcosides								
	& Triterpenoids: Liquorice, Dioscorea, Digitalis										
	Volatile oils: Mentha, Clove, Cinnamon, Fennel,										
	Coriander,										
	Tannins: Catechu, Pterocarpus										
	Resins: Benzoin, Guggul, Ginger, Asafoetida,										
	Myrrh, C	Colophony									
	-		loes, Bitter Almond								
		Other Terpe									
	Naphtha	quinones:G	entian, Artemisia, taxus,								

	carotenoids			
III	Isolation, Identification and Analysis of	6	Students will be able to	2,3
	Phytoconstituents		learn	,4,
	a. Terpenoids: Menthol, Citral, Artemisin		about the synthetic processes	5
	b. Glycosides: Glycyrhetinic acid &Rutin		for	
	c. Alkaloids: Atropine, Quinine, Reserpine,		production of phytoconstituents.	
	Caffeine			
	d. Resins: Podophyllotoxin, Curcumin			
IV	Industrial production, estimation and utilization of		Students will be able to learn	
	the following phytoconstituents: Forskolin,		about modern techniques of	
	Sennoside, Artemisinin, Diosgenin, Digoxin,	10	isolation of phytoconstituents.	2,3
	Atropine, Podophyllotoxin, Caffeine, Taxol,			
	Vincristine and Vinblastine			
V	Basics of Phytochemistry		Students will be able to learn	
	Modern methods of extraction, application of latest		about various and extraction	
	techniques like Spectroscopy, chromatography and	8	isolation technique.	2,3
	electrophoresis in the isolation, purification and			
	identification of			
	crude drugs.			

T1: Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhlae (2007), 37th Edition, NiraliPrakashan, New Delhi.

T2: Text Book of Pharmacognosy by T.E. Wallis.

### **REFERENCE BOOKS:**

R1: W.C.Evans, Trease and Evans Pharmacognosy, 16th edition, W.B. Sounders & Co., London, 2009.

R2: Tyler, V.E., Brady, L.R. and Robbers, J.E., Pharmacognosy, 9th Edn., Lea and Febiger, Philadelphia, 1988.

R3: Mohammad Ali.Pharmacognosy and Phytochemistry, CBS Publishers & Distribution, New Delhi. R4: Essentials of Pharmacognosy, Dr.SH.Ansari, IInd edition, Birla publications, New Delhi, 2007 R5: Anatomy of Crude Drugs by M.A.Iyengar

R6: Practical Pharmacognosy: C.K. Kokate, Purohit, Gokhlae.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the formation of secondary metabolites by several metabolic pathways in plants.	PO1,PO4,PO11
2	Describe the composition, chemistry, chemical classes, bio sources, therapeutic uses, and commercial applications of medicinal plant secondary metabolites.	PO1,PO4,PO7,PO11
3	Explain the method of isolation, identification, and analysis of various chemical constituents from plant sources.	PO1,PO2,PO4, PO7,PO11
4	Describe the industrial production, estimation, and utilization of essential chemical constituents from the plant.	PO1,PO4,PO7, PO11
5	Understand the various extraction techniques and the role of spectroscopy and chromatographic techniques in the isolation, purification, and identification of phytoconstituents.	PO1,PO4,PO11

			SEMESTI	ER – V								
Cours	e Title		Pharmaceutica	l Jurispr	uden	ce (Th	eory)					
Cours	e code	BP505T	Total credits: 4	L	T	P	S	R	O/F	С		
			Total hours: 45T	3	1	0	0	0	0	4		
Pre-re	quisite	Nil	Co-requisite				N	il	1			
Progr	amme		Bache	lor of Ph	arma	ıcy						
Semo	ester		Fall/V semester of	third yea	r of	the pr	ogram	me				
Cou	ırse	Upon completion o	f the course, the stude									
Objec	ctives	1. Pharmaceutical	legislation and their in	mplication	ns in o	develo	ping a	nd mar	keting			
		pharmaceuticals	•									
		2. Various Indian pharmaceutical Acts and Laws										
		3. The regulatory a	authorities and agencie	es governi	ng th	e man	ufactu	re and	sale of			
		pharmaceuticals	<b>;</b>									
		4. The code of ethi	cs during the pharmac	ceutical pr	actic	e						
CO	<b>D1</b>	Explain the schedul	les and functioning of	various c	omm	ittees	in the	Drug a	nd Cosmeti	c Act,		
		rules, and Indian ph	narmaceutical Acts.									
CO	<b>D2</b>		atory authorities and	_	_	_			cture and s	ale of		
			nts and packaging gui									
CO	<b>D3</b>	Understand about	Understand about the production, processing, and control of narcotic and psychotropic									
		drugs.	rugs.									
CO	<b>04</b>	Describe the salient Features of the Drugs and Magic Remedies Act and its Rules, the										
			elty to Animals Act19	960 and tl	ne Na	ational	Phari	naceut	ical Pricing	g		
		Authority										
CO	<b>O5</b>		of ethics for pharma	•			•			luding		
		international laws, as prescribed by the Pharmacy Council of India from time to time.										
Unit-		Conte	nt	Contact		Le	arning	g Outco	ome	KL		
No.				Hour								
I	•	tives, Definitions,	O	10					e to learn	′′		
		edules to the Act a				_	•		nvolved in	4		
	_	_	ses of drugs and		1 ^		i manu	facturi	ng of			
		tics prohibited from	• •		drug	gs.						
		•	offences and penalties.									
	1	facture of drugs – Pr										
		facture and sale of c	•									
	1	ense for manufacture	cense and conditions									
	1		test, examination and									
		is, manufacture of	•									
	1	e and repacking lice	-									
II			et, 1940 and its rules	10	Stud	dents	will b	e able	e to learn	2,3,		
11	_		chedule G, H, M, N,	10					drugs and			
		, V, X, Y, Part XII I			I	metics			8			
			Wholesale, Retail									
	` ′	and Restricted licen	· ·									
	penalties Labeling &Packing of drugs-											
	General labeling requirements and speciments											
		for drugs and cosn	-									
		tted colors.										
	Offen	ces and penalties.										
	Admir	nistration of the Act	and Rules – Drugs									

	Tashuisal Advissors David Castual dosas		
	Technical Advisory Board, Central drugs		
	Laboratory, Drugs Consultative Committee,		
	Government drug analysts, Licensing		
	authorities, controlling authorities, Drugs		
	Inspectors		
III	Pharmacy Act –1948: Objectives,	10	Students will be able to learn 2,3,
	Definitions, Pharmacy Council of India; its		about provisions of Pharmacy act. 4,5
	constitution and functions, Education		
	Regulations, State and Joint state pharmacy		
	councils; constitution and functions,		
	Registration of Pharmacists, Offences and		
	Penalties		
	Medicinal and Toilet Preparation Act		
	-1955: Objectives, Definitions, Licensing,		
	Manufacture In bond and Outside bond,		
	Export of alcoholic preparations,		
	Manufacture of Ayurvedic, Homeopathic,		
	Patent & Proprietary Preparations. Offences		
	and Penalties.		
	Narcotic Drugs and Psychotropic substances		
	Act-1985 and Rules: Objectives, Definitions,		
	Authorities and Officers, Constitution and		
	Functions of narcotic & Psychotropic		
	Consultative Committee,		
	National Fund for Controlling the Drug		
	Abuse, Prohibition, Control and Regulation,		
	opium poppy cultivation and production of		
	poppy straw, manufacture, sale and export of		
	opium, Offences and Penalties		
IV	Study of Salient Features of Drugs and	8	Students will be able to learn 2,3
	Magic Remedies Act and its rules:		About provisions of magic
	Objectives, Definitions, Prohibition of certain		remedies act, prevention of animal
	advertisements, Classes of Exempted		cruelty act and DPCO 2013.
	advertisements, Offences and Penalties		
	Prevention of Cruelty to animals Act-1960:		
	Objectives, Definitions, Institutional Animal		
	Ethics Committee, CPCSEA guidelines for		
	Breeding and Stocking of Animals,		
	Performance of Experiments, Transfer and		
	acquisition of animals for experiment,		
	Records, Power to suspend or revoke		
	registration, Offences and Penalties		
	National Pharmaceutical Pricing Authority:		
	Drugs Price Control Order (DPCO)- 2013.		
	Objectives, Definitions, Sale prices of bulk		
	drugs, Retail price of formulations, Retail		
	price and ceiling price of scheduled		
	formulations, National List of Essential		
	Medicines (NLEM)		
V	Pharmaceutical Legislations – A brief		Students will be able to learn
	208		35 8615 15 160111

review, Introduction, Study of drugs enquiry		about the pharmaceutical ethics.	
committee, Health survey and development			
committee, Hathi committee and Mudaliar			
committee			
Code of Pharmaceutical ethics Definition,			
Pharmacist in relation to his job, trade,	7		2,3
medical profession and his profession,			
Pharmacist's oath			
Medical Termination of Pregnancy Act			
Right to Information Act Introduction to			
Intellectual Property Rights (IPR)			

- T1: Forensic Pharmacy by B. Suresh.
- T2: Text book of Forensic Pharmacy by B.M. Mithal. T3: Hand book of drug law-by M.L. Mehra.
- T4: A text book of Forensic Pharmacy by N.K. Jain.
- T5: Drugs and Cosmetics Act/Rules by Govt. of India publications.

## **REFERENCE BOOKS:**

- R1: Medicinal and Toilet preparations act 1955 by Govt. of India publications.
- R2: Narcotic drugs and psychotropic substances act by Govt. of India publications. R3: Drugs and Magic Remedies act by Govt. of India publication.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Explain the schedules and functioning of various committees in the Drug and Cosmetic Act, rules, and Indian pharmaceutical Acts.	PO1,PO3,PO4,PO5,PO11
2	Illustrate the regulatory authorities and agencies governing the manufacture and sale of labelling requirements and packaging guidelines for drugs and cosmetics	PO1,PO3,PO4,PO7,PO10,PO 11
3	Understand about the production, processing, and control of narcotic and psychotropic drugs.	PO1,PO2,PO4,PO5,PO7,PO1
4	Describe the salient Features of the Drugs and Magic Remedies Act and its Rules, the Prevention of Cruelty to Animals Act1960 and the National Pharmaceutical Pricing Authority	PO1,PO3,PO4,PO5,PO10,PO 11
5	Describe the code of ethics for pharmaceutical practice and explain other laws, including international laws, as prescribed by the Pharmacy Council of India from time to time.	PO1,PO3,PO4,PO5,PO11

			SEMESTEI	R - V									
Cours	se Title		Industrial P	harma	cy I (I	Pract	tical)						
Cours	se code	BP506P	Total credits: 2	L	T	P	S	R	O/F	С			
			Total hours: 4	0	0	4	0	0	0	2			
Pre-re	equisite	Nil	Co-requisite				N	il		-			
Progr	ramme		Bachel	or of I	Pharm	acy							
Sem	ester		Fall/V semester of	hird y	ear of	f the	progran	nme					
Co	urse	1. Demonstrate	e proficiency in pre-form	ılation	studie	s.							
Obje	ectives	1	and GLP while performi					-	•				
		_	lls in formulating and cor	_									
C	<b>O</b> 1		drugs' numerous physico	chemi	ical pa	rame	eters to l	e disc	ussed befo	re pre-			
		formulation.											
C	<b>O2</b>		e manufacturing and eve	aluatio	n of n	nulti	ple solic	d dosag	ge forms s	uch as			
		_	tablets, capsules, pills, and powder										
C	O3	Comprehend and execute the principles related to producing and assessing parenteral											
		pharmaceutical formulations, including injectable and ophthalmic solutions.											
C	O4	1 .	Gain proficiency in applying knowledge to produce semisolid pharmaceutical formulations										
		like creams and	<u> </u>										
	O5	Assemble the nu	Assemble the numerous techniques and analyze pharmaceutical packaging materials.										
Unit-			Content			ıtact	Lear	ning O	utcome	KL			
No.	1 D	C 1.4:	1: D 4 1 A			our	G. 1 .	'11 1	11 /				
1			dies on Paracetamol, Asp	ırın, or		Students will be able to learn about about							
	1	y other drug	lustion of Domostomal to	hlata									
		-	aluation of Paracetamol ta aluation of Aspirin tablets				preform prepare		study and				
		•	film coating of tablets/gr				pharmac						
		•	aluation of Tetracycline c				formulat						
		•	ium Gluconate injection	арзате	3		ioiiiiaiai	.1011.					
		-	orbic Acid injection										
		ality control test of (as per IP) marketed tablets											
	-	d capsules	(as per 11 ) marinered a										
		•	drops and eye ointments										
			ms (cold/vanishing cream	)									
		-	s containers (as per IP)	•	4	4				3,4			

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the drugs' numerous physicochemical parameters to be discussed before pre-formulation.	PO1,PO2,PO3,PO4,PO9,PO11
2	Demonstrate the manufacturing and evaluation of multiple solid dosage forms such as tablets, capsules, pills, and powder	PO1,PO2,PO3,PO4,PO9,PO11
3	Comprehend and execute the principles related to producing and assessing parenteral pharmaceutical formulations, including injectable and ophthalmic solutions.	PO1,PO2,PO3,PO4,PO7,PO9,P O11
	Gain proficiency in applying knowledge to produce semisolid pharmaceutical formulations like creams and gels.	PO1,PO2,PO3,PO4,PO7,PO9,P O11
5	Assemble the numerous techniques and analyze pharmaceutical packaging materials.	PO1,PO2,PO3,PO4,PO7,PO8,P O9,PO11

			SEMEST								
	se Title		Pharm						T _		
Cour	se code	BP507P	Total credits: 2	L	T	P	S	R	O/F	C	
			Total hours: 4	0	0	4	0	0	0	2	
	equisite	Nil	Co-requisite					Nil			
	ramme		Bachelor of Pharmacy								
	nester	1 0	Fall/V semester of third year of the programme								
	ourse ectives	_	1. Operate lab equipment according to SOPs for preclinical experimentation.								
	CO1		2. Conduct bioassay experiments in different tissues  Explain in-vitro pharmacology, bioassay, and the physiological salt solution used								
	.01	for performing		issay,	ana t	ne pn	y 51010	gicai s	an solutio	II us	
(	CO2		bioassay of insulin, ox	vtocir	ı. vaso	pressi	1. AC	TH. d-	tubocurari	ne.	
			amine, and 5-HT softwar							,	
(	CO3	_	e-Response Curves usin						rations and	obse	
		the effect of	agonist and antagonist so	ftware	e-based	l simul	ated e	xperime	ents.		
C	CO4		efficacy of analgesic a				ory dı	rugs in	in-vivo an	imal	
			software-based simulate								
C	CO5	_	armacology with related				and a	pprecia	te the new	er	
ET •4	1	targets of se	veral disease conditions f	or trea	atment.				•	17	
Unit-			Content				ntact Iour		arning itcome	K	
No. I	1. Int	roduction to in	-vitro pharmacology and	nhvei	ologica		lour		ts will be		
1		t solutions.	-vitro pharmacology and	physi	ologica	"		able to			
			n isolated frog heart.					about v			
		_	blood pressure and hear	t rate	of dog.			proced	ure to		
		-	ly of diuretic activity of drugs using rats/mice.					evaluat	te the		
	5. DI	RC of acetylcho	line using frog rectus abo	domin	is			pharma	acological		
		iscle.							of drugs on		
			gmine and atropine on D						ory animal		
		•	ng frog rectus abdominis	muscl	le and r	at		in softv	ware.		
		um respectively		1							
		oassay of nistar	mine using guinea pig ile	um by	/						
		U	ocin using rat uterine horn	, by			4			3,4,	
		erpolation met	_	- ~ <i>j</i>			•			5, <del>1</del> ,	
		_	onin using rat fundus stri	by tl	hree-po	int				, ,	
		assay.	,		•						
	10. Bi	oassay of acety	lcholine using rat ileum/	colon	by four	:-					
	_	int bioassay.									
		termination of									
	_	azosin using rat	• •								
		· •	l's plot method).	sia :1 -	11100						
			PD2 value using guinea pgens and spasmolytics us								
		iect of spasmog unum.	gens and spasmorytics us	mg rai	υυπ						
			y activity of drugs using	carrac	geenan-	_					
		luced paw-eder			5- 511MII						
		-	of drugs using central a	nd							
		rinharal mathac									

peripheral methods.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Explain in-vitro pharmacology, bioassay, and the physiological salt solution used for performing bioassay.	PO1,PO2,PO3,PO4,PO5,PO6, PO8,PO9,PO11
2	Demonstrate bioassay of insulin, oxytocin, vasopressin, ACTH, d-tubocurarine, digitalis, histamine, and 5-HT software-based simulated experiments.	PO1,PO2,PO3,PO4,PO5,PO6, PO8,PO9,PO11
3	Prepare Dose-Response Curves using suitable isolated tissue preparations and observe the effect of agonist and antagonist software-based simulated experiments.	PO1,PO2,PO3,PO4,PO5,PO6, PO8,PO9,PO11
4	Evaluate the efficacy of analgesic and anti-inflammatory drugs in in-vivo animal models using software-based simulated experiments.	PO1,PO2,PO3,PO4,PO5,PO6, PO8,PO9,PO11
5	Correlate pharmacology with related medical sciences and appreciate the newer targets of several disease conditions for treatment.	PO1,PO2,PO3,PO4,PO5,PO6, PO8,PO9,PO11

				SEME	ESTER –	V						
Cou	rse T	itle		Pharmacognos	y And Ph	ytoch	emisti	y II (	Practi	ical)		
Cou	rse co	ode	BP508P	Total credits: 2	L	T	P	S	R	(	O/F	С
				Total hours: 4	0	0	4	0	0		0	2
Pre-	requi	site	Nil	Co-requisite			•	•	Nil	•		
Prog	gram	me	Bachelor of Pharmacy									
Se	meste	er	Fall/ V semester of third year of the programme									
C	ourse	•	1. Identify pl	nytoconstituents in co	rude drug	S.						
Ob	jectiv	es	2. Prepare he	erbal formulations.								
				ample purity compar								
	CO1		Perform the n	norphological and mi	icroscopi	cal eva	aluatio	n tech	nique	s in the	e identifi	cation
			of crude drugs									
(	CO2		Perform the ex	straction of phytocor	stituents	from o	crude d	lrugs.				
(	CO3		Accomplish th	ne isolation of phytoc	constituer	nts froi	n crud	e drug	gs.			
(	CO4		Apply various	chromatographic me	ethods to	evalua	ate herl	oal ex	tracts	and ph	ytoconst	ituents.
(	CO5		Demonstrate t	he extraction of vola	tile oils a	nd che	mical	tests f	or cru	de drug	gs	
Unit-				Content			Co	ntact	Lear	ning C	Outcome	KL
No.								our				
I				logy, and powder cha				ours		nts wi		
				ion of: Cinchona, Ci	nnamon,	Senna		per			n about	3,4,5,6
				ennel, and Coriander			pra	ctical	•	ation of		
			_	isolation & detection						constit	uents	
		_	-	ine from tea dust b. I	_				and th			
			_	ine from Belladonna	d. Senno	sides				fication	•	
			n Senna						l l	_	ethod,	
	3.	_	_	rs by paper chromato	graphy						thods of	
	4.		of herbal extr							constit	uents,	
				tile oils and detection	n of				use of			
			oconstituents b	*						natogra	phic	
	6.		•	rugs by chemical te		foetid	a		techni	ques.		
		ii.Be	enzoin iii. Cole	ophony iv.Aloes v.N	Myrrh							

T1: Text book of Pharmacognosy by C.K. Kokate, Purohit, Gokhlae (2007), 37th Edition, Nirali Prakashan, New Delhi.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Perform the morphological and microscopical evaluation techniques in the identification of crude drugs.	PO1,PO7,PO8,PO11
2	Perform the extraction of phytoconstituents from crude drugs.	PO1,PO2,PO3,PO4,PO7,PO8, PO10,PO11
3	Accomplish the isolation of phytoconstituents from crude drugs.	PO1,PO2,PO3,PO4,PO7,PO8, PO11
4	Apply various chromatographic methods to evaluate herbal extracts and phytoconstituents.	PO1,PO2,PO3,PO4,PO7,PO8, PO11
5	Demonstrate the extraction of volatile oils and chemical tests for crude drugs	PO1,PO2,PO3,PO4,PO7,PO8, PO10,PO11

		SEMESTER – VI									
Cou	rse Title		MEDIC	INAL CI	HEM	ISTRY	<b>7-III</b>				
Cou	ırse code	BP 601T	Total credits: 4	L	T	P	S	R	O/F		C
			Total hours: 45T	3	1	0	0	0	0		4
Pre-	requisite	Nil	Co-requisite			-	N	Vil		-	
	gramme	Bachelor of Pharmacy									
	emester	Fall/ II semester of first year of the programme									
C	Course	Understand the importance of drug design and different drug design techniques.									
Ob	jectives		tand the chemistry of d	_	_			_	-	•	
	J		3. Know the metabolism, adverse effects, and therapeutic value of drugs.								
		4. Know the importance of SAR of drugs.									
	CO1		lustrate the history a			ent of	Anti	biotics	with stru	ctu	res,
			and classification		•						
	CO2		Summarize and Organize the classification, synthesis, and SAR of Antibiotics,								
			Quinolines and Miscel		, -J	,	,			- •	- 7
	CO3		Categorize Antituber		ents	Ouina	olones	Antiv	viral agent	s v	with
		_	chanism of action and	_	, 1 - 100 9	~ min		,	agont	- <b>'</b>	
	CO4	1 7			rotos	zoal. A	ntheln	nintics	and		
	CO4	Sulfonamides	Distinguish and Prioritize Antifungal, Antiprotozoal, Anthelmintics and								
		structures including Mechanism of action and SAR									
	CO5		Decide and Compile various drug design, physicochemical properties, Docking								
	CO3	techniques an	-		-		ui pro	percies, D	OCI	ung	
Unit-		Cont	· .	Contact				g Outc	ome	Ti	KL
No.		Cont	CIIC	Hour		L	.41 11111	g Out	ome	'	IXL
I	Antibiotic	s Historical bac	ekoround	10	То	under	stand	about	History of	f 4	4,5
_			emistry, Structure						•		.,c
			hemical degradation		Development of Antibiotics.  To Understand and Remember th						
			rtant products of the		Degradation products with						
	following	_	products or and		structures. To Remember the						
		antibiotics: Per	nicillin.					d under			
	ľ		mase inhibitors,			chanis					
	1 ^ ^	ams Aminogly	· ·						size		
		cin, Neomyci		To Classify and synthesize various classes of Antibiotics.							
		•	line,Oxytetracycline,						icleus of		
	1		cycline, Doxycycline					nalospo			
II		s Historical back		10					Macrolides	,   ,	4,5
			emistry, Structure						Understand		<i>y-</i>
			mical degradation			_	•		degradation		
	-	-	rtant products of the					rolides	·		
		_	lide: Erythromycin								
	_		nycin. Miscellaneous:		structures. To Remember the classification and understand						
		•	lamycin. Prodrugs:		Mechanism of action. To Classify					7	
	_		lication of prodrugs						s classes of		
	design. Antimalarials: Etiology of malaria.				Chloramphenicol, Chloroquia						
	_	s: SAR, Quinin		and Pamaquine, Prodrugs. To							
	-	ne*, Amodiaqı			ply S	_	for	nucleus o	f		
	_	, Pamaquine*,	-					nolines			
	^	•	ne. Biguanides and				,				
	•		guanilpamoate,								
	-		s: Pyrimethamine,								
<u></u>	5 Gamini.	CIIalie Ou	j								

	Artesunete, Artemether, Atovoquone.			
Ш	Anti-tubercular Agents Synthetic anti		To gain knowledge about Anti-	
	tubercular agents: Isoniozid, Ethionamide,		tubercular agents. To understand	
	Ethambutol, Pyrazinamide, Para amino		mechanism of action with	
	salicylic acid. Anti tubercular antibiotics:		synthesis and classification of	
	Rifampicin, Rifabutin,		Antitubercular agents. To	
	Cycloserine, Streptomycine, Capreomycin		Understand and Remember, about	
	sulphate. Urinary tract anti-infective agents		classification, synthesis mechanism	
	Quinolones: SAR of quinolones,		of Quinolones. To Remember the	
	Nalidixic Acid, Norfloxacin, Enoxacin,		synthesis of Nitrofurantoin. To	
	Ciprofloxacin, Ofloxacin, Lomefloxacin,		analyze the classification with	
	Sparfloxacin, Gatifloxacin, Moxifloxacin	10	structures and synthesis of	4,5
	Miscellaneous: Furazolidine, Nitrofurantoin,	10	Antiviral agents.	,,,,,
	Methanamine. Antiviral agents: Amantadine		i intervitor organis	
	hydrochloride, Rimantadine hydrochloride,			
	Idoxuridinetrifluoride, Acyclovir,			
	Gancyclovir, Zidovudine, Didanosine,			
	Zalcitabine, Lamivudine, Loviride,			
	Delavirding, Ribavirin, Saquinavir, Indinavir,			
	Ritonavir.			
IV	Antifungal agents: Antifungal antibiotics:		To gain knowledge about	
	Amphotericin-B, Nystatin, Natamycin,		Antifungal agents. To understand	
	Griseofulvin. Synthetic Antifungal agents:		Mechanism of action with	
	Clotrimazole, Econazole, Butoconazole,		synthesis and classification. To	
	Oxiconazole Tioconozole, Miconazole,		Understand and Remember, about	
	Ketoconazole, Terconazole, Itraconazole,		classification, synthesis mechanism	
	Fluconazole, Naftifine hydrochloride,		of Anthelmintics. To analyze	
	Tolnaftate. Anti-protozoal Agents:		SAR of Suphonamides with	
	Metronidazole, Tinidazole, Ornidazole,	8	synthesis and structures and	4,5
	Diloxanide, Iodoquinol,		classification. To Remember the	
	PentamidineIsethionate, Atovaquone,		synthesis of Trimethoprim and	
	Eflornithine. Anthelmintics:		Dapsone	
	Diethylcarbamazine citrate, Thiabendazole,			
	Mebendazole, Albendazole, Niclosamide,			
	Oxamniquine, Praziquantal,			
	Ivermectin. Sulphonamides and Sulfones			
	Historical development, chemistry,			
	classification and SAR of Sulfonamides:			
	Sulphamethizole, Sulfisoxazole,			
	Sulphamethizine, Sulfacetamide,			
	Sulphapyridine, Sulfamethoxaole,			
	Sulphadiazine, Mefenide acetate,			
	Sulfasalazine. Folatereductase inhibitors:			
	Trimethoprim, Cotrimoxazole. Sulfones:			
	Dapsone.			
$\mathbf{V}$	Introduction to Drug Design Various		To Understand and Remember,	
	approaches used in drug design.		About Various approaches for	
	Physicochemical parameters used in		drug design. To analyze and	
	quantitative structure activity relationship		evaluate the physicochemical	
	(QSAR) such as partition co efficient, ammet's		parameters like Hansch analysis	

electronic parameter, Tafts steric Parameter		and concepts of combinatorial
and Hansch analysis.	7	chemistry. To develop and create 4,5
Pharmacophore modeling and docking		different agents by combinatorial
techniques. Combinatorial Chemistry:		chemistry and Docking of drugs
Concept and applications chemistry: solid		on to various proteins
phase and solution phase synthesis. of		
Combinatorial		

T1: A text book of Medicinal chemistry – Vol 1 and Vol 2 Surendra Nath Pandey Latest Edition. T2: A text book of Medicinal Chemistry – Vol 1 and Vol 2 Kadam Mahadik Latest Edition.

## **REFERENCE BOOKS:**

R1: Wilson and Giswold's Organic medicinal and Pharmaceutical Chemistry. R2: Foye's Principles of Medicinal Chemistry.

R3: Introduction to principles of drug design- Smith and Williams. R4: Organic Chemistry by I.L. Finar, Vol. II.

R5: The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1-5.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Relate and Illustrate the history and development of Antibiotics with structures, degradation, and classification	PO1,PO4,PO8,PO9,PO10,PO11					
2	Summarize and Organize the classification, synthesis, and SAR of Antibiotics, antimalarials, Quinolines and Miscellaneous	PO1,PO2,PO3,PO4,PO8,PO9,PO10 ,PO11					
3	Organize and Categorize Antitubercular agents, Quinolones, Antiviral agents with synthesis, Mechanism of action and SAR.	PO1,PO2,PO3,PO4,PO8,PO9,PO10 ,PO11					
4	Distinguish and Prioritize Antifungal, Antiprotozoal, Anthelmintics and Sulfonamides structures including Mechanism of action and SAR	PO1,PO2,PO3,PO4,PO8,PO9,PO10 ,PO11					
5	Decide and Compile various drug design, physicochemical properties, Docking techniques and Combinatorial synthesis with Applications	PO1,PO2,PO3,PO4,PO8,PO9,PO10 ,PO11					

			SEMESTI	ER – VI								
Cou	rse Title		PHA	ARMACOLOGY-III								
Cou	rse code	BP 602T	Total credits: 4	L	T	P	S	R	O/F		C	
			Total hours: 45T	3	1	0	0	0	0		4	
Pre-requisite		Nil	Co-requisite				Ni	<u>il</u>				
	gramme		Bachelor of Pharmacy									
	mester	1 77 1	Fall/ II semester of first year of the programme  1. Understand the mechanism of drug action and its relevance in the treatment of									
	Course		stand the mechanism of one of the control of the co	drug actioi	n and	its rel	evance	in the	treatment	10		
Ob	jectives		rehend the principles of t	tovicology	and	traatm	ant of	vorious	noiconin	œ		
		_	ciate the correlation of pl						_	gs		
	CO1		d contrast the specific p									
	COI		inctions among member						_	,		
	CO2		e pharmacological and							an	d	
	~ <b>~</b>	non-infectiou		201061			4148	5° 101 1		w11		
	CO3		principles and several ty	pes of To	xicity	·.						
	CO4		pharmacology with other	_			•					
	CO5		path physiology, symp					ons of	various	dis	seas	
		conditions.					•					
Unit-		Cor	ntent	Contact	Lea	rning	Outco	me			KL	
No.				Hour								
Ι	Pharmace	ology of drug	s acting on Respiratory		Τοι	ınders	tand th	e drug				
	system						-	ry system				
		asthmatic drug		and	gastro	intestii	nal trac	t				
	_	s used in the m										
	_	ctorants and ar										
		decongestants										
	1	ratory stimula										
		cology of drug	10							4,5		
		ntestinal Tract										
		lcer agents.										
			ion and diarrhoea.									
			and suppressants.									
	1	stants and carn ics and anti-en										
II		otherapy	neucs.		To	rain lo	nowlad	ge abo	ut the			
11		1.0	of chemotherapy.		_	-		-	otherapy			
		namides and c			Pilai	macol	ogy of	CHCIII	листару			
			lins, cephalosporins,	10							4,5	
			nacrolides, quinolones	10							.,.	
		luoroquinolins										
		Aminoglycosi	•									
III	Chemoth		10	То я	get the	knowl	edge al	out	+	4,5		
		ıbercular agen	ats		_			ic agent			•	
		eprotic agents					•	Č				
		ungal agents										
		riral drugs e.A	nthelmintics									
	f. Antin	nalarial drugs										
	a Antio	maahia aganta			1							

g. Antiamoebic agents

Chemotherapy

To know about the chemotherapy

Urinary tract infections and sexually transmitted diseases. Chemotherapy of malignancy.  4. Immunopharmacology a. Immunostimulants b. Immunosuppressant Protein drugs, monoclonal antibodies, target drug antigen, biosimilars	8	for UTI, sexually transmitted diseases, malignancy and also know about the Immunopharmacology	4,5
V Principles of toxicology a. Definition and basic knowledge of subacute and chronic toxicity. b. Definition and basic knowledge of genotoxicity, carcinogenicity, terate and mutagenicity c. General principles of treatment of poisoning d. Clinical symptoms and management barbiturates, morphine, and organophosphorus compound and lemercury and arsenic poisoning. e. Chronopharmacology f. Definition of rhythm and cycles. g. Biological clock and their sign leading to chronotherapy.	ogenicity t of 7 ead,	To gather the knowledge about The toxicology and Chronopharmacology	4,5

T1: K. D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.

- T2: Rang & Dale's Pharmacology, Elsevier.
- T3: Lippincott Illustrated Reviews: Pharmacology.
- T4: Goodman and Gilman's, The Pharmacological Basis of Therapeutics.

### **REFERENCE BOOKS:**

- R1: PHARMACOLOGY III, by Dr. SACHIN V. TEMBHURNE.
- R2: Pharmacology-III, by Dr. Shaik Harun Rasheed, SIA Publishers & Distributors Pvt Ltd.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Compare and contrast the specific pharmacology of the major classes of drugs, essential distinctions among members of each class, the risks, and benefits, etc.	PO1,PO3,PO6,PO7,PO8,PO9, PO11
2	Elaborate the pharmacological and toxicological effects of drugs for infectious and non-infectious diseases.	PO1,PO3,PO6,PO7,PO8,PO9, PO11
3	Evaluate the principles and several types of Toxicity.	PO1,PO6,PO7,PO8,PO9,PO1
4	Correlate of pharmacology with other biomedical sciences.	PO1,PO3,PO6,PO7,PO8,PO9, PO11
5	Describe the path physiology, symptoms, and treatment options of various disease conditions.	PO1,PO3,PO6,PO7,PO8,PO9, PO11

SEMESTER – VI										
Course	Title		HERBAL	DRUG 1	TECH	HNOLO	GY			
Course	code	BP 603T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 45T	3	1	0	0	0	0	4
Pre-rec		Nil	Co-requisite				N	il		
Progra				nelor of						
Seme			Fall/ II semester							
Cou			d raw material as a sou	urce of h	erbal	drugs fi	om cu	ltivatio	n to herbal	drug
Objec	tives	product.					0.1			
			WHO and ICH guidel						ugs.	
		1	herbal cosmetics, natu			s, nutrac	eutica	ls.		
			e patenting of herbal d			1 1	1	1 1	. 11	1 1
CC	)1		inition of herbs, herba	l medicii	ne, ne	erbal me	dicina	I produ	cts, and her	bal
CC	<u> </u>	drug preparation.		.1411.	t.:C	1	1 41	1 1.	1. 1	1
CO	)2	1 * * *	wledge of nutraceutica		•		•			
			tions of specific herbs		ging	anment	s like	aiabete	s, ardiovaso	cular
CC	13		strointestinal disorders materials of herbal		704:	n harkal	000	otics :	noludina 41	nai#
	,,		ions, and applications							
CC	<u> </u>		nderstanding of WHO						-	
	7		protocols for herbal dru		II gu	ildelilles	to de	velop t	ina impiem	CIII
CC	)5			_	ng P	ractices	(GMF	) in pro	oducing her	hal
	,,	Illustrate the significance of Good Manufacturing Practices (GMP) in producing herbal drugs.								our
Unit-		Cont	ent	Contac	t	Le	arning	g Outco	ome	KL
No.				Hour				9		
I	Herbs	as raw material	s Definition of herb,		Не	erb, Hert	al For	mulatio	n, Modern	
	herbal	medicine, herba	l medicinal product,		cu]	ltivation	, Holis	stic dru	gs	
	herbal	drug preparatio	on Source of Herbs							
	Selecti	on, identification	n and authentication							
	of herl	oal materials Pro	cessing of herbal raw							
	materi	al Biodynamic	Agriculture Good							
	agricul	ltural practices	in cultivation of							
	medici	nal plants includ	ing Organic farming.							
		_	gement in medicinal							4,5
	r	-	ioinsecticides. Indian							
	1 -		a) Basic principles							
		•	, Siddha, Unani and							
		opathy b)	Preparation and							
	1	•	urvedic formulations							
			, Ghutika, Churna,							
II		and Bhasma	ıl aspects, Market,		V.		1140000	uticola	and Herb-	
111			types of products						and nero-	4,5
	_	•	Health benefits and	_ ′	וטו	ug inter	activili	3		4,5
			s in ailments like							
			, Cancer, Irritable							
		syndrome and va								
		•	y of following herbs							
			, Chicory, Ginger,							
			ey, Amla, Ginseng,	1						

	Ashwagandha, Spirulina Herbal-Drug and			
	Herb-Food Interactions: General introduction			
	to interaction and classification. Study of			
	following drugs and their possible side			
	effects and interactions: Hypercium, kava-			
	kava, Ginkobiloba, Ginseng, Garlic, Pepper			
	& Ephedra.			
III	Herbal Cosmetics Sources and description		Materials and methods involved	
	of raw materials of herbal origin used via,		in manufacturing of Herbal	
	fixed oils, waxes, gums colours, perfumes,		cosmetics, Herbal excipients-	
	protective agents, bleaching agents,		importance and use, Preparations	
	antioxidants in products such as skin care,		of Herbal formulations	
	hair care and oral hygiene products. Herbal			
	excipients: Herbal Excipients – Significance			
	of substances of natural origin as excipients	10		4,5
	colorants, sweeteners, binders, diluents,			,
	viscosity builders, disintegrants, flavors&			
	perfumes. Herbal formulations :			
	Conventional herbal formulations like			
	syrups, mixtures and tablets and Novel			
	dosage forms like phytosomes			
IV	Evaluation of Drugs WHO & ICH		Who and ICH guidelines, IPR in	
1	guidelines for the assessment of herbal		herbal drug, Regulations in	
	drugs Stability testing of herbal drugs.		manufacturing herbal drug.	
	Patenting and Regulatory requirements of		manuracturing nervar drug.	
	natural products: a) Definition of the terms:			
	Patent, IPR, Farmers right, Breeder's right,			
	Bioprospecting and Biopiracy b) Patenting	10		4,5
	aspects of Traditional Knowledge and	10		7,5
	Natural Products. Case study of Curcuma &			
	Neem. Regulatory Issues - Regulations in			
	India (ASU DTAB, ASU DCC), Regulation			
	of manufacture of ASU drugs - Schedule Z			
V	of Drugs & Cosmetics Act for ASU drugs.  General Introduction to Herbal Industry		Overview of Herbal Industry,	
·			•	
	Herbal drugs industry: Present scope and	7	introduction of objectives and	
	future prospects. A brief account of plant based industries and institutions involved in	7	components of Schedule T.	4,5
	work on medicinal and aromatic plants in			
	India. Schedule T – Good Manufacturing			
	Practice of Indian systems of medicine			
	Components of GMP (Schedule – T) and its			
	objectives Infrastructural requirements,			
	working space, storage area, machinery and			
	equipments, standard operating procedures,			
	health and hygiene, documentation and			

T1: Textbook of Pharmacognosy by Trease& Evans.

T2: Textbook of Pharmacognosy by Tyler, Brady & Robber. T3: Pharmacognosy by Kokate, Purohit and

#### Gokhale.

T4: Essential of Pharmacognosy by Dr. S .H.Ansari.

# **REFERENCE BOOKS:**

R1: Pharmacognosy & Phytochemistry by V.D.Rangari.

R2: Pharmacopoeial standards for Ayurvedic Formulation (Council of Research in Indian Medicine & Homeopathy).

R3:Mukherjee, P.W. Quality Control of Herbal Drugs: An Approach to Evaluation of Botanicals. Business Horizons Publishers, New Delhi, India, 2002.

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Describe the definition of herbs, herbal medicine, herbal medicinal products, and herbal drug preparation.	PO1,PO11						
2	Apply their knowledge of nutraceuticals to identify and analyze the health benefits and potential applications of specific herbs in managing ailments like diabetes, cardiovascular diseases, and gastrointestinal disorders.	PO1,PO10,PO11						
3	Analyze the raw materials of herbal origin used in herbal cosmetics, including their properties, functions, and applications in skincare, hair care, and oral hygiene products.	PO1,PO3,PO7,PO9,PO10,PO 11						
4	Integrate their understanding of WHO and ICH guidelines to develop and implement stability testing protocols for herbal drugs.	PO1,PO7,PO8,PO9,PO11						
5	Illustrate the significance of Good Manufacturing Practices	PO1,PO7,PO8,PO9,PO10,PO						
3	(GMP) in producing herbal drugs.	11						

SEMESTER – VI										
Course '	Title		BIOPHARMACEUT	ICS ANI	) PH	ARMA	COK	INETI	CS	
Course	code	BP 604T	Total credits: 4	L	T	P	S	R	O/F	C
_		2	Total hours: 45T	3	1	0	0	0	0	4
Pre-requ		Nil	Co-requisite				N	il		
Progran				helor of I						
Semest		1 111	Fall/ II semester							. 1 41
Cours		1. Understa significa	and the basic concepts i	n biophar	mace	eutics ai	na pna	гтасок	inetics ar	ia their
Objecti	ives	_	nce. ma drug concentration-	time data	to ca	lculate	the nl	narmacc	kinetic	
		_	ers to describe the kine				_			iem
		_	n, and elimination.	iics of are	ig ao.	sorption	1, 4150	Toution	, 11100001	13111,
			rstand the concepts of b	oioavailah	ility :	and bio	eaniva	alence c	of drug nr	oducts
			r significance.	. 10	1110)				1 01 08 P1	5 <b>44.75 (</b> 5
			and various pharmacoki	inetic para	amete	ers, thei	r signi	ficance	&applica	itions
CO1	1		e basic concepts in b							
		significance.	_	_						
CO2	2	Use plasma dr	ug concentration-time	data to cal	lculat	e the pl	narma	cokinet	ic parame	ters to
		describe the ki	netics of drug absorption	on, distrib	ution	ı, metab	olism	, excret	ion, and	
		elimination.								
CO3	3		cepts of bioavailability	y and bio	oequi	valence	of c	lrug pr	oducts ar	nd their
		significance.								
CO4			us pharmacokinetic par		heir s	significa	ance &	applica	ations	
COS	5		oncepts of Non-linearity					<u> </u>		
Unit- No.		Co	ntent	Contact Hour		Lea	arnınş	g Outco	ome	KL
I I	Absor	ption; Mechan	ieme of drug	Hour	Stu	dents w	ill be	able to	learn	-
1		ption, weenan	•					of drug		
			orption though GIT,			nan bod	-	or drug	55 111	
	1		om Non per oral		110411		-9			
	1 -	_	, Distribution Tissue							
			s, binding of drugs,	10						4,5
	Г		drug distribution,							
	plasm	a and tissue pro	otein binding of							
	drugs,	factors affecti	ng protein-drug							
	bindin	g. Kinetics of	protein binding,							
		<u> </u>	of protein binding							
II		•	etabolism and basic					able to		
		-	olic pathways renal					n of dru	gs in	
		_	actors affecting renal		hun	nan bod	ly			. =
		_	enal clearance, Non	10						4,5
		_	excretion of drugs							
		ailability and Bioequivalence:								
		tion and Objectives of all ability, absolute and relative								
		ailability, ab ailability,	solute and relative measurement of							
		•	ritro drug dissolution							
	model	•	~							
		,	udies, methods to							
	_	ce the dissolu	•							
	FIIII	and andiona	Tares wild							

	bioavailability of poorly soluble drugs.			
III	Pharmacokinetics: Definition and introduction to Pharmacokinetics, Compartment models, Non compartment models, physiological models, One compartment open model. (a). Intravenous Injection (Bolus) (b). Intravenous infusion and (c) Extra vascular administrations.  Pharmacokinetics parameters - KE, t1/2,Vd,AUC,Ka, Clt and CLR-definitions methods of eliminations, understanding of their significance and application	10	Students will be able to learn about pharmacokinetics parameters	4,5
IV	Multicompartment models: Two compartment open model. IV bolus Kinetics of multiple dosing, steady state drug levels, calculation of loading and mainetnance doses and their significance in clinical settings.	8	Students will be able to learn about compartment model	4,5
V	Nonlinear Pharmacokinetics: a. Introduction, b. Factors causing Nonlinearity. c. Michaelis-menton method of estimating parameters, Explanation with example of drugs	7	Students will be able to learn about nonlinear pharmacokinetics	4,5

T1: Bio pharmaceutics and Pharmacokinetics-A Treatise, By M. Brahmankar and Sunil B.Jaiswal, Vallabh Prakashan Pitampura, Delhi

T2: Text Book of Biopharmaceutics and Pharmacokinetics; By Robert F Notari

# **REFERENCE BOOKS:**

R1: Applied biopharmaceutics and pharmacokinetics, Leon Shargel and Andrew B.C.YU 4th edition, Prentice-Hall International USA

R2: Text book of Biopharmaceutics and Clinical Pharmacokinetics by, Milo R3: Pharmacokinetics: By Milo Glbaldi Donald, R. Mercel Dekker

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Understand the basic concepts in biopharmaceutics and pharmacokinetics and their significance.	PO1,PO2,PO4,PO6,PO9,PO1 0,PO11					
2	Use plasma drug concentration-time data to calculate the pharmacokinetic parameters to describe the kinetics of drug absorption, distribution, metabolism, excretion, and elimination.	PO1,PO2,PO9,PO10,PO11					
3	Apply the concepts of bioavailability and bioequivalence of drug products and their significance.	PO1,PO2,PO9,PO10,PO11					
4	Analyze various pharmacokinetic parameters, their significance & applications	PO1,PO2,PO4,PO6,PO9,PO1 0,PO11					
5	Illustrate the concepts of Non-linearity.	PO1,PO2,PO4,PO6,PO9,PO1 0,PO11					

SEMESTER – VI											
	se Title		PHARMACEU								
Cour	se code	BP 605T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 45T	3	1					4	
	equisite	Nil	Co-requisite				N	il			
	ramme	Bachelor of Pharmacy									
	nester		Fall/ II semester								
	ourse		anding the importance	of Immob	ilized	l enzyr	nes in	Pharn	naceutical		
Obj	ectives	Industrie		•			0 1				
			engineering application	•			t phan	maceu	iticals		
		_	nce of Monoclonal anti				41.	1			
	701		ate the use of microorg								
	CO1		e importance of Immob								
	CO2		engineering principles	_		_		iceum	cais.		
	CO3		gnificance of Monoclor					2010 =	raduation		
'	CO4		oorganisms in ferments gn and its various contr		noiog	gy and .	iarge-s	scare p	nouuciion		
	CO5		gn and its various cond lood products and their		n Dr	ocessi.	10 and	Store	ge.		
Unit-	J <b>U</b> 3	Cont	•	Contact			arning		_	KL	
No.		Cont	CIII	Hour		Le	ai 11111 <u>8</u>	z Out	Come	KL	
I	Brief int	roduction to Ri	otechnology with	Hour	Und	lerstan	ding t	he ha	sic concepts		
1		e to Pharmaceu							nderstanding		
			- Methods of enzyme				_	-	Immobilized		
		ization and app			ymes	in		armaceutical			
		ors- Working a		1 -	ıstries						
		rs in Pharmace									
	Brief int	roduction to Pro	10						4,5		
	Use of m	nicrobes in indu	stry. Production of								
	Enzymes	s- General cons	sideration - Amylase,								
			ipase, Protease,								
		_	ciples of genetic								
	engineer										
II		cloning vector							pt of vectors		
		eases and DNA	~						y. Apply the		
			nnology. Application			_	_		n production		
	_	c engineering i		10	of p	harma	ceutica	als		15	
			technology and	10						4,5	
	_	on ii) Vaccines	he production of:								
	1	ones-Insulin.	- nepautis- D								
	1		OC.								
Ш	d) Brief introduction to PC  III Types of immunity- humoral immunity,				Und	lerstand	d abou	t the h	numan		
	cellular in		,	10					ents will	4,5	
		re of Immunog	lobulins						luction of	.,2	
	-	ure and Function				cines a		-			
			ctions, Immune								
	,		une suppressions.								
			ne preparation of								
			oxoids, viral vaccine,								
		xins, serum-im									
L	I			1							

				1
	derivatives and other products relative to immunity.			
	e) Storage conditions and stability of official			
	vaccines			
	f) Hybridoma technology- Production,			
	Purification and Applications			
	g) Blood products and Plasma Substituties.			
IV	Immuno blotting techniques- ELISA,		Learn about microbial genetics	
	Western blotting, Southern blotting.			
	Genetic organization of Eukaryotes and			
	Prokaryotes			
	Microbial genetics including transformation,			
	transduction, conjugation, plasmids and	8		4,5
	transposons.			
	Introduction to Microbial biotransformation			
	and applications.			
	Mutation: Types of mutation/mutants			
V	Fermentation methods and general		Understanding of fermentation	
	requirements, study of media, equipments,		technique	
	sterilization methods, aeration process,		1	
	stirring.			
	Large scale production fermenter design and			
	its various controls.	7		4,5
	Study of the production of - penicillins, citric	•		1,50
	acid, Vitamin B12, Glutamic acid,			
	Griseofulvin,			
	Blood Products: Collection, Processing and			
	Storage of whole human blood, dried			
	human			
	plasma, plasma Substituties.			
	piasina, piasina buositutios.			

T1: Pharmaceutical Biotechnology by Prof. Chandrakant Kokare T2: Pharmaceutical Biotechnology by Ravi Kumar Madelia

#### **REFERENCE BOOKS:**

R1: B.R.Glick and , J.J.Pasternak: Molecular Biotechnology: Principles and Applications Of Recombinant DNA: ASM Press Washington D.C.

R2: RA Goldshyet.al.,: Kuby Immunology. R3: J.W.Goding: Monoclonal Antibodies.

R4: J.M.Walker and, E.B.Gingold: Molecular Biology and Biotechnology by Royal Society of Chemistry.

R5: Zaborsky: Immobilized Enzymes, CRCPress, Degraland, Ohio.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the basic concepts in biopharmaceutics and	PO1,PO2,PO3,PO4.PO5,PO6,PO8,
1	pharmacokinetics and their significance.	PO9,PO10,PO11
	Use plasma drug concentration-time data to calculate the	
2	pharmacokinetic parameters to describe the kinetics of drug	PO1,PO2,PO3,PO4.PO5,PO6,PO8,
2	absorption, distribution, metabolism, excretion, and	PO9,PO10,PO11
	elimination.	
2	Apply the concepts of bioavailability and bioequivalence of	PO1,PO2,PO3,PO4.PO5,PO6,PO8,
3	drug products and their significance.	PO9,PO10,PO11
4	Analyze various pharmacokinetic parameters, their	PO1,PO2,PO3,PO4.PO5,PO6,PO8,
4	significance & applications	PO8,PO9,PO10,PO11
5	Illustrate the concepts of Non-linearity.	PO1,PO2,PO3,PO4.PO5,PO6,PO8,
	inustrate the concepts of Non-Intearity.	PO9,PO10,PO11

SEMESTER – VI												
Cou	rse Title		PHARMACEU	TICAL	QU	JALI	ITY	ASSU	JRA	NC	E	
Cou	rse code	BP 606T	Total credits: 4	L	T	Γ	P	S	]	R	O/F	C
			Total hours: 45T	3	1	1	0	0		0	0	4
Pre-i	requisite	Nil	Co-requisite						Nil			'
Prog	gramme		Ba	chelor o	f P	Phari	macy	7				
Sei	mester		Fall/ II semesto	er of firs	t y	ear o	of the	prog	grai	nme	e	
C	ourse	1. Unders	tanding the importanc									al
Obj	jectives	Industri	ies					•				
	•	2. Genetic	e engineering applicati	ons abou	ıt p	rodu	ction	of pl	narn	nace	uticals	
		3. Importance of Monoclonal antibodies in Industries										
		_	iate the use of microon						echr	olo	gy	
	C <b>O</b> 1		ne importance of Immo									ies.
	C <b>O2</b>		e engineering principle									
	C <b>O3</b>		gnificance of Monocle									
	C <b>O</b> 4	-	croorganisms in ferme							e-sc	ale prod	uction
			ign and its various cor				0)		-8		F-30	
	C <b>O</b> 5		blood products and the		tio	n. Pr	roces	sing a	and	Stor	age.	
Unit-		Cont		Contac				<u> </u>				KL
No.				Hour					8			
I	Ouality A	Assurance and	Quality Management		J	Unde	erstan	d the	cGl	MP a	aspects in	1
	-		d concept of Quality					eutic			-	
	_	Quality assurat			1					J		
		Management (										
	-	philosophies										
	1	participants, p										
			verview of QSEM,									
		ial emphasis o	•									
	_	-	y testing guidelines	10								3,4,5
	-		D): Definition,									
	1		QbD program, tools									
			: Overview, Benefits,									
			istration NABL									
			es and procedures									
II			nnel: Personnel		J	Utiliz	ze	the	kne	owle	edge f	or
	_	_	g, hygiene and		r	maint					mportan	ce
	_	records. Premi						entatio			•	
	construct	ion and plant l	layout, maintenance,									3,4,5
		_	tal control, utilities	10								
	and main	tenance of ste	rile areas, control of									
	contamin	ation. Equipm	ents and raw									
			selection, purchase									
	specificat	ions, maintena	ance, purchase									
	specificat	tions and main	tenance of stores for									
III	Quality	Control: Qua	lity control test for		Ţ	Unde	erstan	d the	res	pon	sibilities	
	1	s, rubber clo						QC de		_		
		Packing mate				_		-	-			
	1	_	General Provisions,									
		•	nnel, Facilities,	10								3,4,5

	Equipment, Testing Facilities Operation, Test and Control Articles, Protocol for Conduct of a Nonclinical Laboratory Study, Records and Reports, Disqualification of Testing Facilities			
IV	Complaints: Complaints and evaluation of complaints, Handling of return good, recalling and waste disposal. Document maintenance in pharmaceutical industry: Batch Formula Record, Master Formula Record, SOP, Quality audit, Quality Review and Quality documentation, Reports and documents, distribution records.	8	Understand the scope of quality certifications applicable to pharmaceutical industries ractices.	4,5
V	Calibration and Validation: Introduction, definition and general principles of calibration, qualification and validation, importance and scope of validation, types of validation, validation master plan. Calibration of pH meter, Qualification of UV-Visible spectrophotometer, General principles of Analytical method Validation.  Warehousing: Good warehousing practice, materials management	7	Understand and summarize the general Principles of validation concepts.	3,4

T1: Quality Assurance Guide by organization of Pharmaceutical Products of India. T2: Good Laboratory Practice Regulations, 2nd Edition, Sandy Weinberg Vol. 69.

# **REFERENCE BOOKS:**

R1: The International Pharmacopoeia – Vol I, II, III, IV- General Methods of Analysis and Quality specification for Pharmaceutical Substances, Excipients and Dosage forms

R2: Good laboratory Practices – Marcel Deckker Series R3: ICH guidelines, ISO 9000 and 14000 guidelines

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the importance of Immobilized enzymes in	PO1,PO3,PO5,PO6,PO9,PO10,PO1
1	Pharmaceutical Industries.	1
2	Apply genetic engineering principles to the production of	PO1,PO2,PO6,PO9,PO10,PO11
	pharmaceuticals.	101,102,100,109,1010,1011
3	Explain the significance of Monoclonal antibodies in	PO1,PO5,PO6,PO7,PO9,PO10,PO1
3	Industries	1
	Describe microorganisms in fermentation technology and	PO1,PO2,PO3.PO5,PO6,PO8,PO9,
4	large-scale production fermenter design and its various	PO10,PO11
	controls.	1010,1011
5	Illustrate the blood products and their Collection, Processing	PO1.PO5,PO6PO7,,PO8,PO9,PO10
3	and Storage.	,PO11

			SEMEST	ER – VI							
Course	e Title		MEDICI	NAL CH	EMI	STRY-	· III				
Course	e code	BP 607P	Total credits: 2	L	T	P	S	R	O/F	С	
			Total hours: 4	0	0	4	0	0	0	2	
Pre-rec		Nil	Co-requisite				N	il			
Progra				helor of I							
Seme			Fall/ II semester								
Course		_	are medicinally importa	_							
Objectives			<ol> <li>Characterize prepared compounds using physicochemical methods.</li> <li>Illustrate structures and reactions using chem draw/chem sketch</li> </ol>								
CC	)1	Identify and intermediates.	relate the glassware	and equi	ipme	nt syn	thesiz	ing va	rious drugs	and	
CC	)2	Demonstrate a	and Develop the method	l for Assa	y of c	drugs					
CC	)3		lyze the results obtained				s usin	g the m	nicrowave m	ethod.	
CC	)4		d justify the structures a								
CC	)5	Develop physi	icochemical properties,	Drug like	ness	and Lip	oinski	RO5 in	Drug design	n.	
Unit- No.		Cor	ntent	Contact Hour		Le	arnin	g Outc	ome	KL	
I	senses Intermed Hydrox Chlorol Assay of Chlorol Penicill Importation Microwstructur Determ Propert Moleculand accontent	To understand about Sylater integration of drugs and mediates 1 Sulphanilamide 2 7-roxy, 4-methyl coumarin 3 probutanol 4 Triphenyl imidazole 5 utamide 6 Hexamine loroquine 3 Metronidazole 4 Dapsone 5 propheniramine maleate 6 Benzyl cillin I Preparation of medicinally ortant compounds or intermediates by rowave irradiation technique Drawing stures and reactions using chem draw® remination of physicochemical erties such as logP, clogP, MR, ceular weight, Hydrogen bond donors acceptors for class of drugs course ent using drug design software Drug iness screening (Lipinskies RO5)  Contact Hour  To understand about Sy Intermediates. To Remember To understand about A drugs. To Remember To be carried out To Apply principle, Analyze and the Results obtained.  To understand about A drugs. To Remember To understand about A drugs. To Remember To Earlied out To Apply principle, Analyze and the Results obtained.  To understand about A drugs. To Remember To understand about A drugs. To Remember To Earlied out To Apply principle, Analyze and the Results obtained.  To understand about A drugs. To Remember To understand about A drugs. To Remember To Understand about A drugs. To Remember To Earlied out To Apply principle, Analyze and the Results obtained.  To understand about A drugs. To Remember To Understand about Sylotate to earried out To Apply and Analyze and the Results obtained.  To understand about A drugs. To Remember To Understand about Sylotate to Earlied out To Apply and Analyze the for drug likenes using Ro5 To Evaluate the sylotate to the Understand about Sylotate to the				bout Asmber To Apply ze and sout Sying Michael react nalyze sults obnember to draw nalyze and na Remember to find proper yee the susing see the steet to ste	ember the Apply, he Results say of itrations to the Evaluate enthesis of crowave. ions carried and tained. To what w structures. the valuate the uning them. aber what nd enties. To structures Lipinski	4,5			

T1: Text book of practical organic chemistry- A.I.Vogel.

T2: The Organic Chemistry of Drug Synthesis by Lednicer, Vol. 1-5.

# **REFERENCE BOOKS:**

R1: Organic Chemistry by I.L. Finar, Vol. II.

R2: Introduction to principles of drug design- Smith and Williams. R3: Martindale's extra pharmacopoeia.

R4: Indian Pharmacopoeia Latest edition R5: Remington's Pharmaceutical Sciences.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Identify and relate the glassware and equipment synthesizing	PO1,PO2,PO3.PO4,PO5,PO6,PO7,
1	various drugs and intermediates.	PO8,PO9,PO10,PO11
2	Demonstrate and Develop the method for Assay of drugs	PO1,PO2,PO3.PO4,PO5,PO6,PO7,
	Demonstrate and Develop the method for Assay of drugs	PO8,PO9,PO10,PO11
3	Build and analyze the results obtained by synthesizing drugs	PO1,PO2,PO3.PO4,PO5,PO6,PO7,
3	using the microwave method.	PO8,PO9,PO10,PO11
4	Categorize and justify the structures and reactions drawn	PO1,PO2,PO3.PO4,PO5,PO6,PO7,
•	using Chem Draw.	PO8,PO9,PO10,PO11
5	Develop physicochemical properties, Drug likeness and	PO1,PO2,PO3.PO4,PO5,PO6,PO7,
	Lipinski RO5 in Drug design.	PO8,PO9,PO10,PO11

SEMESTER – VI													
Course	Title		PHAR	RMACO	LOC	SY-III							
Course	code	BP 608P	Total credits: 2	L	T	P	S	R	O/F	C			
			Total hours: 4	0	0	4	0	0	0	2			
Pre-req		Nil	Co-requisite				]	Nil					
Progra		Bachelor of Pharmacy											
Seme	ster	Fall/ II semester of first year of the programme											
Cou		Conduct skillful tissue analysis.											
Object		2. Operate lab equipment according to SOPs for preclinical experimentation											
CO	1	Understand the principles of bioassay and its types including advantages and											
		lisadvantages.											
CO	2	1	n concentrations of v	arious d	lrugs	using	suitab	le isola	ated tissue				
		r -	different bioassays.										
CO			ect of drugs on labor										
CO	4	1	us pharmacological	effects (	analg	gesic, l	ocomo	tion, r	nuscle rela	ıxant,			
		etc.) in animals us											
CO	5		portance, methods, a										
Unit-		Conte	nt	Contac	t	Le	arning	g Outc	ome	KL			
No.				Hour									
I		ose calculation in p	harmacological	4					mportance				
		xperiments			_			ion and	5,6				
		antiallergic activity	by mast cell			evaluate the effect of mast cell							
		tabilization assay				stabilizer using Ex-Pharm							
		tudy of anti-ulcer a				tware							
		sing pylorus ligan				To evaluate the antiulcer and							
		nodel and NSAIDS	S induced ulcer		gastrointestinal motility using Ex Pharm software								
		nodel.											
		tudy of effect of di	•			To evaluate the different agonist and antagonists effect using Ex-							
	_	astrointestinal moti				_			•				
		ffect of agonist and	antagonists on						stimate				
	_	uinea pig ileum Estimation of serum	biochemical					_	rameters d Insulin				
			semi- autoanalyser						sing Ex-				
	_	Effect of saline purg	•						sing Ex-				
		ntestine	ative on nog		Pharm software  To test pyrogens and determine								
		nsulin hypoglycemi	c effect in rabbit			oral L	_		actermine				
		est for pyrogens (r				determ			skin				
		Determination of act	,						e irritation				
		LD50) of a drug fro	· · · · · · · · · · · · · · · · · · ·					-	kinetic				
	,	, ,	tute skin irritation /			ametei	•						
	c	orrosion of a test su	ıbstance		dif	ferent l	oiostat	istics n	nethods				
	12. D	etermination of ac	cute eye irritation /										
	corrosion of a test substance												
	13. C	Calculation of pharm	nacokinetic										
	р	arameters from a gi	ven data										
	_	iostatistics methods											
	p	harmacology( stud	ent's t test,										
	A	NOVA) Biostatist	ics methods in										
	e	xperimental pharm	acology (Chi square										
	te	est, Wilcoxon Signo	ed Rank test)										

T1: K. D. Tripathi. Essentials of Medical Pharmacology, JAYPEE Brothers Medical Publishers (P) Ltd, New Delhi.

T2: Rang & Dale's Pharmacology, Elsevier.

T3: Lippincott Illustrated Reviews: Pharmacology.

T4: Goodman and Gilman's, The Pharmacological Basis of Therapeutics.

# **REFERENCE BOOKS:**

R1: PHARMACOLOGY – III, by Dr. SACHIN V. TEMBHURNE.

R2: Pharmacology-III, by Dr.Shaik Harun Rasheed, SIA Publishers & Distributors Pvt Ltd.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the principles of bioassay and its types including	PO1,PO2,PO5,PO6,PO7,PO8,PO9,
1	advantages and disadvantages.	PO10,PO11
2	Describe unknown concentrations of various drugs using Suitable isolated tissue preparations using different bioassays.	PO1,PO2,PO3.PO4,PO5,PO6,PO7, PO8,PO9,PO11
3	Understand the effect of drugs on laboratory animals and toxicity studies.	PO1,PO2,,PO5,PO6,PO7,PO8,PO9, PO11
4	Evaluate the various pharmacological effects (analgesic, locomotion, muscle relaxant, etc.) in animals using the software.	PO1,PO2,PO3.PO4,PO5,PO6,PO7, PO8,PO9,PO11
5	Understand the importance, methods, and application of	PO1,PO2,PO3.PO4,PO5,PO6,PO7,
	biostatistics in pharmacology.	PO8,PO9,PO11

			SEMESTI	ER – VI								
Course	Title		HERBAL I		ССН	NOLO	GY					
Course	code	BP 609P	Total credits: 2	L	T	P	S	R	O/F	С		
			Total hours: 4	0	0	4	0	0	0	2		
Pre-rec	uisite	Nil	Nil Co-requisite Nil									
Progra	mme	Bachelor of Pharmacy										
Seme	ster	Fall/ II semester of first year of the programme										
Cou	rse	Perform phytochemical screening of crude drug.										
Objec	tives	2. Determine phytochemical content in crude drug										
CC	)1	Understand the pr	rinciples and techniqu	es involve	ed in	prelim	inary p	hytoch	nemical scr	eening		
		of crude drugs.										
CC	)2	Explain the meth	odology and principle	es for dete	rmir	ing the	e alcoh	ol cont	tent of Asa	va and		
		Arista.										
CC	)3		aluate the use of e	excipients	of	natura	ıl orig	gin in j	pharmaceu	tical		
		formulations.										
CC	)4		ability to incorpora									
			tions like creams, loti							eness.		
CO	<b>)</b> 5	111	e of Pharmacopoeial	•			•					
			oal extracts into formu		e sy	rups, n	nixture	s, and t	tablets and			
			npliance with the stan	dards.								
TT *4		Content Contact Learning Outcome KI										
Unit-		Conto	ent			Le	arning	g Outco	ome	KL		
No.	1 7			Contact Hour						KL		
		o perform prelim	inary phytochemical		Me	thods o	of Cher	nical E	ome Evaluation	KL		
No.	sc	o perform prelim	inary phytochemical drugs.		Me Exc	thods o	of Cher Evalua	mical E	Evaluation	KL		
No.	2. D	o perform prelim creening of crude of retermination of th	inary phytochemical		Me Exc Ma	thods or cipient nufacto	of Chei Evalua	nical E	Evaluation	KL		
No.	2. D	o perform prelim creening of crude of etermination of the sava and Arista	inary phytochemical drugs. e alcohol content of	Hour	Me Exc Ma	thods o	of Chei Evalua	mical E	Evaluation	KL		
No.	2. D A 3. E	o perform prelime reening of crude of the crude of the sava and Arista valuation of excip	inary phytochemical drugs. e alcohol content of ients of natural origin	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation	KL		
No.	2. D A 3. E 4. In	o perform prelime or crude of crude of the creening of crude of the creenination of the crude and Arista or corporation of preservation of pre	inary phytochemical drugs. e alcohol content of ients of natural origin epared and	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation	KL		
No.	2. D A 3. E 4. In	o perform prelime reening of crude of the crude and Arista valuation of excipacorporation of presentation of p	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation	KL		
No.	2. D A 3. E 4. In st fc	o perform prelime creening of crude of etermination of the sava and Arista valuation of excipation corporation of presentation of presentations like creening or presentations are presented to the presentation of presentations are presented to the presentation of the pr	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation	KL		
No.	2. D A 3. E 4. In st fc	o perform prelime reening of crude of the sava and Arista valuation of excipactorporation of presentant productions like cruampoos and their	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation.	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation	KL		
No.	2. D A 3. E 4. In st fc sk 5. In	o perform prelime reening of crude of etermination of the sava and Arista valuation of excipation of presentations like crumulations and their acorporation of presentation of	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			
No.	2. D A 3. E 4. In st fc sh 5. In	o perform prelime reening of crude of etermination of the sava and Arista valuation of excipation of presentations like crumulations like crumulations and their acorporation of presentation	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and t in formulations like	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation	<b>KL</b> 3,4		
No.	2. D A 3. E 4. In st fc sk 5. In st sy	o perform prelime reening of crude of etermination of the sava and Arista valuation of excipation of presentations like crumulations like crumpoos and their acorporation of presentation of presentations, mixtures and ardized extractorups, mixtures and extractorups, mixtures and extractorups.	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and t in formulations like d tablets and their	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			
No.	2. D A 3. E 4. In st fc sh 5. In st sy ev	o perform prelime reening of crude of corporation of present of corporation of corporation of crude of corporation of crude of corporation of crude of corporation of crude of corporation of corporation of crude of corporation of	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and t in formulations like d tablets and their	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			
No.	2. D A 3. E 4. In st fc sh 5. In st sy ev	o perform prelime reening of crude of etermination of the sava and Arista valuation of excipation of presentations like crumulations like crumpoos and their acorporation of presentation of presentations, mixtures and ardized extractorups, mixtures and extractorups, mixtures and extractorups.	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and t in formulations like d tablets and their harmacopoeial	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			
No.	2. D A 3. E 4. In st fc sh 5. In st sy ev re 6. M	o perform prelime reening of crude of etermination of the sava and Arista valuation of excipation of presentations like crumulations like crumulations and their acorporation of presentation of presentation of presentations, mixtures and valuation as per Plesquirements.	inary phytochemical drugs.  e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation.  epared and t in formulations like d tablets and their narmacopoeial s of herbal drugs	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			
No.	2. D A 3. E 4. In st fc sh 5. In st sy ev fc 6. M	o perform prelime reening of crude and Arista valuation of excipation of present and ardized extraction of present and ardized extraction of present and ardized extraction of present and ardized extractions, mixtures and valuation as per Place of the company of the crude of t	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and t in formulations like d tablets and their narmacopoeial s of herbal drugs acopoeias	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			
No.	2. D A 3. E 4. In st fc sh 5. In st sy ev re 6. M fr 7. D	o perform prelime reening of crude of etermination of the sava and Arista valuation of excipation of presentations like crumulations like crumulations and their acorporation of presentation as per Plesquirements.  In the property of the presentation of presentation as per Plesquirements.  In the property of the presentation of presentation as per Plesquirements.	inary phytochemical drugs. e alcohol content of ients of natural origin epared and t in cosmetic reams, lotions and evaluation. epared and t in formulations like d tablets and their narmacopoeial s of herbal drugs acopoeias ldehyde content	Hour	Me Exc Ma	thods or cipient nufacto	of Chei Evalua	mical E	Evaluation			

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the principles and techniques involved in preliminary phytochemical screening of crude drugs.	PO1 ,PO8, PO11
2	Explain the methodology and principles for determining the alcohol content of Asava and Arista.	PO1,PO2,PO3.PO4,PO6,PO7,PO8, PO11
3	Analyze and evaluate the use of excipients of natural origin in pharmaceutical formulations.	PO1,PO2,PO3.PO4,PO6,PO7,PO8, PO11
4	Demonstrate the ability to incorporate prepared and standardized herbal extracts into cosmetic formulations like creams, lotions, and shampoos, and evaluate their effectiveness.	PO1,PO2,PO3.PO4,PO5,PO6,PO7, PO8,PO9,PO11
5	Apply knowledge of Pharmacopoeial requirements to incorporate prepared and standardized herbal extracts into formulations like syrups, mixtures, and tablets and evaluate their compliance with the standards.	PO1,PO2,PO3.PO4,PO5,PO6,PO7, PO8,PO9,PO11

	SEMESTER – VII													
Cour	se Title		Instrumental M	<b>Tethods</b>	Of A	nalysis	(Theo	ry)						
Cour	se code	BP701T	Total credits: 4	L	T	P	S	R	O/F	С				
			Total hours: 45T	3	1	0	0	0	0	4				
Pre-r	equisite	Nil	Co-requisite				N	il						
	ramme	Bachelor of Pharmacy												
	nester	VII semester of Fourth year of the programme												
	ourse	Upon completion of the course the student shall be able to												
Obj	ectives	1. Understand the interaction of matter with electromagnetic radiations and its												
		applications in drug analysis  2. Understand the abromate graphic concretion and analysis of drugs												
		2. Understand the chromatographic separation and analysis of drugs.												
		1	3. Perform quantitative & qualitative analysis of drugs using various analytical											
	701	instrument		.1.1	4	4 - 4	1	1'	4: CIN	7				
	C <b>O</b> 1		basic theoretical princ		strum	ientatio	n and a	арриса	uons of U v	'				
	CO2	•	photometer and fluoring principles, instrumer		nd or	mlianti	one of	f ID an	nostrogoons:	,				
	.02		try, AAS and AES, Ne		•	. •		•	есповсору	,				
	CO3		lyze the principles an						matogra <b>n</b> h	v				
	.03					ition or	Colum	ini Cinc	matograph	у,				
	CO4	r -	aper chromatography, TLC, and electrophoresis Evaluate the principles, theory, and instrumentation of HPLC and gas chromatography											
1 1 7				olved in gel, ion exchange, and affinity										
	chromatography.				,		6-,							
Unit-		Conto	*	Contac	t	Learning Outcome								
No.				Hour										
I	UV Visib	le Spectroscop	y Electronic	10	Un	Understand the principles and								
	transition	s, chromophore	s, auxochromes,		Ins	Instrumentation of UV-Visible								
	spectral s	hifts, solvent ef	fect on absorption		spe	ectrosco	py and	d Fluori	imetry					
	1 -		rt's law, derivation											
			ntation: Sources of											
		_	ectors, sample cells,											
			otomultiplier tube,											
		aic cell, Silicon												
			tometric titrations,											
	_	mponent and mu	ilticomponent											
	analysis.	otany Theorem Co	maamta of simplet											
		•	ncepts of singlet, onic states, internal											
		-	, factors affecting											
			Instrumentation											
	and applic													
II			iction, fundamental	10	Un	derstan	d the p	orincipl	es and	3,4,				
	_		lyatomic molecules,			trument	_	_		5				
		-	affecting vibrations.		spe	ectrosco	ру, F	lame P	hotometry,					
	Instrumer	ntation: Sources	of radiation,		Āto	omic al	osorpti	on spe	ctroscopy					
	waveleng	th selectors, det	ectors - Golay cell,		and	d Nephe	eloturb	idomet	ry					
		r, Thermocoupl												
	-	ric detector, and												
			ciple, interferences,											
		ntation, and app												
	Atomic A	Absorption Spe	ctroscopy Principle,											

	interferences, instrumentation, and			
	applications.			
	Nepheloturbidometry Principle,			
	Instrumentation, and applications.			
III	Introduction to Chromatography Adsorption	10	Understand the principles and	3,4,
	and partition column chromatography:		instrumentations of various	5
	Methodology, advantages, disadvantages, and		chromatographic studies	
	applications.			
	Thin layer chromatography: Introduction,			
	principle, methodology, Rf values, advantages,			
	disadvantages, and applications.			
	Paper chromatography: Introduction,			
	methodology, development techniques,			
	advantages, disadvantages, and applications.			
	Electrophoresis: Introduction, factors			
	affecting electrophoretic mobility, techniques			
	of paper, gel, capillary electrophoresis,			
	applications.			
IV	Gas Chromatography: Introduction,	8	Understand the principles and	3,4,
	Theory, instrumentation, and applications.		Instrumentations of GC and HPLC	5
			chromatographic studies	
V	Ion Exchange Chromatography:	7	Understand the principles and	3,4,
	Introduction, classification, ion exchange		instrumentations of Ion exchange	5
	resins, properties, mechanism of ion exchange		chromatography, Gel	
	process, factors affecting ion exchange,		chromatography and Affinity	
	methodology, and applications.		chromatography	
	Gel Chromatography: Introduction, theory,			
	instrumentation, and applications.			
	Affinity Chromatography: Introduction,			
	theory, instrumentation, and applications.			

T1: Instrumental Methods of Chemical Analysis by B.K Sharma. T2: Organic spectroscopy by Y.R Sharma.

T3: Text book of Pharmaceutical Analysis by Kenneth A. Connors.

T4: Vogel's Text book of Quantitative Chemical Analysis by A.I. Vogel. T5: Organic Chemistry by I. L. Finar.

# **REFERENCE BOOKS:**

R1: Organic spectroscopy by William Kemp.

R2: Quantitative Analysis of Drugs by D. C. Garrett.

R3: Quantitative Analysis of Drugs in Pharmaceutical Formulations by P. D. Sethi. R4:

Spectrophotometric identification of Organic Compounds by Silverstein.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the basic theoretical principles, instrumentation and applications of UV Visible spectrophotometer and fluorimeter	PO1,PO2,PO3,PO4,O6,PO8,PO9, PO11
2	Remember the principles, instrumentation and applications of IR spectroscopy, flame photometry, AAS and AES, Nephelometry and Turbidimetry.	PO1,PO2,PO3,PO4,O6,PO8,PO9, PO11
3	Apply and analyze the principles and Instrumentation of column chromatography, paper chromatography, TLC, and electrophoresis	PO1,PO2,PO3,PO4,O6,PO7,PO8, PO9,PO11
4	Evaluate the principles, theory, and instrumentation of HPLC and gas chromatography	PO1,PO2,PO3,PO4,O6,PO7,PO8, PO9,PO11
5	Apply the theory and principle involved in gel, ion exchange, and affinity chromatography.	PO1,PO2,PO3,PO4,O6,PO7,PO8, PO9,PO11

				SEMEST	TER – V	Π							
Course	e Title			Industr	ial Phar	macy	II (Th	eory)					
Course	e code	BP702T	Total c	redits: 4	L	T	P	$\overline{S}$	R	O/F	С		
			Total l	ours: 45T	3	1	0	0	0	0	4		
Pre-rec	quisite	Nil	Co-req	uisite				N	il	1	ı		
Progra	amme		Bachelor of Pharmacy										
Seme	ester		1	VII semester o	f Fourth	year	of the	progra	mme				
Cou	rse	Upon completion of the course, the student shall be able to:											
Objec	ctives	1. Know the	Know the process of pilot plant and scale up of pharmaceutical dosage forms										
		2. Understand the process of technology transfer from lab scale to commercial batch											
		3. Know different Laws and Acts that regulate the pharmaceutical industry											
		4. Understa	nd the a	pproval proces	s and reg	ulatoı	y requ	irement	s for di	rug products	8		
CC	<b>)</b> 1	Recognize th	e releva	ance of manpov	wer and s	pace	needs	in pilot	plant s	cale-up. Ex	plain		
		the necessity	of choo	sing adequate i	aw mate	rials f	or scal	e-up.					
CC	)2	Description of	of WHO	O-recommende	d nomen	clatur	e and	technol	ogy tra	nsfer proce	dure,		
		as well as qua	ality risl	k management o	during R	&D-to	o-produ	action to	echnolo	gy transfer.			
CC	)3	Apply regula	itory kn	owledge for dr	ug appro	val ar	d IND	applica	ation ar	nd assess cli	inical		
		study manage	ement a	nd FDA data su	ıbmissior	ıs.							
CC	)4	Set the QbD	and Six	x Sigma to pha	rmaceuti	cal p	ocesse	s. Asse	ss out-	of-specifica	tions		
		Set the QbD and Six Sigma to pharmaceutical processes. Assess out-of-specifications OOS) and change control implementation.											
CC	)5	nvestigate fresh drug approval procedures and Certificate of Pharmaceutical Product											
		relevance and	relevance and utilize Indian regulatory standards to guarantee pharmaceutical										
		compliance.											
Unit-		Coi	ntent		Contac	t	L	earning	g Outc	ome	KL		
No.					Hour								
I	Pilot		ıle-Up	Techniques:					_	ificance o			
	Genera			<ul> <li>including</li> </ul>									
	1 -	_		requirements,		1 1			_	Explain the	1		
	1 ~	-						portance of selecting appropriate					
	<b>^</b>	1	siderations for solids, 10 raw materials for pilot plant sca						1 1				
	1 ^		solids, and relevant upApply pilot plant					nt scale-u _l	p 5				
	1	entation. SUP	_			techniques for solids, liquid orals, and semi-solids.							
**		ction to platfo									1		
II	Techno		evelopr			1 1	plain		termin	<i></i>			
	Transfo			idelines for			hnolog	•	ansfer	protocols	3		
		logy Transfer					_		_	delines	_		
		logy Transfer				1 1	plemen		qualit	•			
		Ianagement,					nagem 1sfer.		-	echnology	7		
		luction (Proce			10		nster. nsfer			technology			
	(API,	ng), Granular Excipients	•	ed Products,	10				•	o process, aspects.	3,4,		
	` ′	ing Materials		·		pac	ragiiig	5, and C	caming	aspects.			
	_	es and Equip		·									
	and	Validation,	Qual	•									
		cal Method	-	•									
	Regulat												
	_	•		al Aspects and									
				T Agencies in									
		*		ΓΙFAC, BCIL,									
		SIDBI, TT Re		,									
	יייייייייייייייייייייייייייייייייייייי	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	iaica								1		

	Documentation - Confidentiality			
	Agreement, Licensing, MoUs, Legal Issues			
Ш	Regulatory Affairs: Introduction,	10	Understand the roles of approved	3,4,
	Historical overview of Regulatory		regulatory bodies and agencies for	5
	Affairs, Regulatory authorities, Role of		technology transfer Analyze	
	Regulatory Affairs department,		practical aspects and case studies	
	Responsibility of Regulatory Affairs		related to technology	
	Professionals		commercialization Identify	
	Regulatory Requirements for Drug		technology transfer agencies in	
	Approval: Drug Development Teams,		India and their functions.	
	Non-Clinical Drug Development,			
	Pharmacology, Drug Metabolism and			
	Toxicology, General considerations of			
	Investigational New Drug (IND)			
	Application, Investigator's Brochure			
	(IB) and New Drug Application (NDA),			
	Clinical research / BE studies, Clinical			
	Research Protocols, Biostatistics in			
	Pharmaceutical Product Development,			
	Data Presentation for FDA Submissions,			
	Management of Clinical Studies			
IV	Quality Management Systems: Quality	8	Understand the concepts of	3,4,
	Management & Certifications		Quality and Total Quality	5
	Quality management concepts: Concept		Management (TQM) Apply	
	of Quality, Total Quality Management		Quality by Design (QbD) and Six	
	(TQM), Quality by Design (QbD), Six		Sigma concepts to improve	
	Sigma concept, Out of Specifications		pharmaceutical - Apply concepts	
	(OOS), Change control		of OOS and Change Control in	
	Introduction to ISO 9000 series of		maintaining product quality	
	quality systems standards, ISO 14000,		Understand ISO 9000 series of	
	NABL (National Accreditation Board for		quality systems standards and ISO	
	Testing and Calibration Laboratories),		14000 Recall the concept of	
	GLP (Good Laboratory Practice)		NABL and GLP in	
			pharmaceutical manufacturing	
V	Indian Regulatory Requirements:	7	Recall the organization and	3,4,
	Central Drug Standard Control		responsibilities of CDSCO and	5
	Organization (CDSCO) and State		State Licensing Authority	
	Licensing Authority		Understand the importance of	
	Organization and Responsibilities:		Certificate of Pharmaceutical	
	Central Drug Standard Control		Product (COPP) Comprehend the	,
	Organization (CDSCO) and State		regulatory requirements and	
	Licensing Authority roles and		approval procedures for new drugs	
	responsibilities.		in India.	
	Certificate of Pharmaceutical Product			
	(COPP): Overview and significance of the			
	Certificate of Pharmaceutical Product			
	(COPP) in regulatory processes.			
	Regulatory Requirements and Approval			
	Procedures for New Drugs: Detailed			
	procedures and requirements for			

obtaining regulatory approval for		
new drugs in India.		

T1: Douglas J Pisano and David S. Mantus. Text book of FDA Regulatory Affairs A Guide for Prescription Drugs, Medical Devices, and Biologics' Second Edition.

# **REFERENCE BOOKS:**

R1: Regulatory Affairs brought by learning plus, inc. available at http://www.cgmp.com/ra.htm. 2. R2: Regulatory Affairs from Wikipedia, the free encyclopedia modified on 7th April available at http://en.wikipedia.org/wiki/Regulatory_ Affairs.

	CO PO Mapping	
SN	Course Outcome (CO)	<b>Mapped Program Outcome</b>
1	Recognize the relevance of manpower and space needs in pilot plant scale-up. Explain the necessity of choosing adequate raw materials for scale-up.	PO1,PO2,PO3,PO4,PO6,PO1 0,PO11
2	Description of WHO-recommended nomenclature and technology transfer procedure, as well as quality risk management during R&D-to-production technology transfer.	PO1,PO2,PO5,PO11
3	Apply regulatory knowledge for drug approval and IND application and assess clinical study management and FDA data submissions.	PO1,PO2,PO3,POPO11
4	Set the QbD and Six Sigma to pharmaceutical processes.  Assess out-of-specifications (OOS) and change control implementation.	PO1,PO2,PO5,PO10,PO11
5	Investigate fresh drug approval procedures and Certificate of Pharmaceutical Product relevance and utilize Indian regulatory standards to guarantee pharmaceutical compliance.	PO1,PO2,PO11

	SEMESTER – VII												
Cour	se Title		Pha	rmacy Pi	racti	ce (T	heory)	)					
Cour	se code	BP703T	<b>Total credits: 4</b>	I		T	P	S	R	O/F	C		
			Total hours: 45	T 3	3	1	0	0	0	0	4		
Pre-re	equisite	Nil	Co-requisite					Ni	il				
	ramme			Bachelor (									
Sem	ester			r of Fourth year of the programme									
	urse			he student shall be able to									
Obje	ectives		s drug distribution			•							
			ne pharmacy store	_				-					
			drug therapy of	the patier	nt thi	rough	n medi	cation	chart 1	review and	1		
		clinical revie			۱	maal 4	ha mat	ianta					
			cation history integrelated problems		Cou	nser	ne pau	ients					
			ssess adverse drug										
			ected laboratory re	-		toring	o narai	neters	in ther	aneutics) o	f		
		specific disea	· · · · · · · · · · · · · · · · · · ·	(40 1	5 111		5 F W1			r - 335) 0			
		1 *	aceutical care ser	vices									
		_	ounseling in a con		harm	acy.							
		10. Appreciate	the concept of rat	ional drug	ther	ару.							
C	<b>O</b> 1	Understand the	hospital's organiz	zation, th	e ho	spita	l phai	rmacy,	and a	any advers	se		
			nd apply this know										
C	<b>O2</b>						view, apply in the community						
		r	gement, and under		_			•	-	_			
			ication adherence								•		
C	O3		distribution syste		_		_			_			
			Medication adhemacy managemen		anen	ı med	aicano	n nisu	ory into	erview, and	a		
	O4				limn	leme	ntation	n of cli	nical P	harmacy s	and		
	04	OTC drugs.	the budget prepar	ration and implementation of clinical Pharmacy, and									
C	O5	Understand drug store management and analysis, inventory control, investigational drug											
		_	ty to interpret clin		-		-	Commo	,,	, erganionar (	an an 6		
Unit-		Content	· .	Contact				arning	Outco	ome	KL		
No.									,				
I	a) Hosp	ital and Its Orga	nization	7		Kno	w vai	rious (	drug d	istribution	3,4,		
	Definiti	on andClassifica	tion of			Met	hods i	n a hos	spital. l	Know	5		
	_	l: Primary, Secon	·			Г	-		_	ement and			
		ary hospitals; Cl					-			ntify drug			
		d on clinical and r	ion-clinical				_			letect and			
	basis		e II '. I						ug read				
	_	zation Structure view of the organ	-		Know about Community Pharmacy.								
						1 IIai	ппасу.						
	structure and medical staff involved, their roles, and functions.												
		ital Pharmacy a											
	-	anization Definit											
	_	ctions of Hospita											
		tions of hospital 1	•										
	inclu	ding dispensing,	compounding,										
	and p	patient care.											

	Organization Structure, Location,			
	_			
	Layout, and Staff Requirements:			
	Layout and organizational			
	requirements of hospital pharmacies,			
	roles of pharmacists and support staff.			
	Responsibilities and Functions of			
	Hospital Pharmacists: Role and			
	responsibilities of hospital pharmacists			
	in patient care, medication			
	management, and ensuring drug			
	safety.			
	c) Adverse Drug Reactions			
	Classifications of Adverse Drug			
	Reactions: Excessive			
	pharmacological effects, secondary			
	pharmacological effects, idiosyncrasy,			
	allergic reactions, genetically			
	determined toxicity, toxicity following			
	sudden withdrawal of drugs. <b>Drug</b>			
	Interactions: Beneficial interactions,			
	adverse interactions, pharmacokinetic			
	drug interactions. Methods for			
	detecting drug interactions, including			
	spontaneous case reports and record			
	linkage studies.			
	Adverse Drug Reaction Reporting and			
	Management: Procedures for			
	reporting and managing adverse drug			
	reactions in healthcare settings.			
	d) Community Pharmacy			
	Organization and Structure of			
	Retail and Wholesale Drug Stores:			
	Types and designs of retail and			
	wholesale drug stores, including legal			
	requirements for establishment and			
	maintenance.			
	Dispensing of Proprietary Products:			
	Processes and regulations for			
	dispensing proprietary products in			
	community pharmacies.			
	Maintenance of Records of Retail and			
	Wholesale Drug Stores:			
	Requirements and practices for			
	maintaining records in retail and			
	wholesale drug stores.			
II	a) Drug Distribution System in a	10	Drug distribution system in a	3,4,5
	Hospital Dispensing of Drugs to		hospital. Monitor drug therapy of	
	Inpatients: Methods and procedures		patient through medication chart	
	for dispensing drugs to hospitalized		review and clinical review.	
	patients.		Monitor drug therapy of patient	
	, A			<u> </u>

#### **Types of Drug Distribution**

**Systems:** Overview of different systems used for distributing drugs within hospitals.

# **Charging Policy and Labeling:**

Policies and practices related to drug charges and labeling in hospital settings. **Dispensing of Drugs to Ambulatory Patients:** Processes involved in dispensing medications to patients who are not hospitalized.

# **Dispensing of Controlled Drugs:**

Guidelines and procedures for dispensing controlled substances in hospitals.

# b) Hospital Formulary Definition:

Explanation of what constitutes a hospital formulary and its purpose.

# **Contents of Hospital Formulary:**

Types of medications included, guidelines for inclusion, and categories of drugs.

# Differentiation of Hospital Formulary and Drug List:

Comparison between a hospital formulary and a general drug list.

**Preparation and Revision:** Steps involved in preparing and updating the hospital formulary.

# Addition and Deletion of Drugs from Hospital Formulary:

Procedures for adding new drugs and removing existing ones from the formulary.

# c) Therapeutic Drug Monitoring Need for Therapeutic Drug Monitoring:

Reasons why monitoring drug levels in patients is essential.

# Factors to Consider during Therapeutic Drug Monitoring:

Key considerations and parameters monitored during the process.

# Indian Scenario for Therapeutic Drug Monitoring: Specific

considerations and practices related to therapeutic drug monitoring in India.

# d) Medication Adherence Causes of Medication Non- Adherence:

Factors contributing to patients not following prescribed medication

through medication chart review and clinical review

		ragimang			
		regimens.  Pharmacist Role in Medication			
		Adherence: Ways pharmacists can			
		assist patients in adhering to their			
		medication schedules.			
		Monitoring of Patient Medication			
		Adherence: Methods and strategies			
		used to monitor and improve patient			
		adherence to medications.			
	e)	Patient Medication History			
		Interview Need for the Patient			
		Medication History Interview:			
		Importance of gathering			
		comprehensive medication history			
		from patients.			
		Medication Interview Forms: Tools			
		and forms used to conduct medication			
		history interviews effectively.			
	<b>f</b> )	Community Pharmacy Management			
		Financial, Materials, Staff, and			
		Infrastructure Requirements: Key			
		requirements and considerations for			
		managing a community pharmacy			
		effectively.	10	77 1 1 1	2.4
III	<b>a</b> )		10	Know about pharmacy and	3,4,
		C		41	_
		Committee Organization and		therapeutic committee, Know	5
		Functions: Structure and roles of		about Role of pharmacist in the	5
		<b>Functions:</b> Structure and roles of the Pharmacy and Therapeutic		_	5
		<b>Functions:</b> Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre:		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre:  Role and functions of the Drug and Poison Information Centre in		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre:  Role and functions of the Drug and Poison Information Centre in providing information and guidance.		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre:  Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information:		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information: Various sources used for obtaining		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre:  Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information:  Various sources used for obtaining drug-related information.		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information: Various sources used for obtaining drug-related information.  Computerized Services and		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information: Various sources used for obtaining drug-related information.  Computerized Services and Storage Retrieval of Information:		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information: Various sources used for obtaining drug-related information.  Computerized Services and  Storage Retrieval of Information: Use of computerized systems for		about Role of pharmacist in the	5
		Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and Poison Information and guidance.  Sources of Drug Information: Various sources used for obtaining drug-related information.  Computerized Services and Storage Retrieval of Information: Use of computerized systems for managing and retrieving drug		about Role of pharmacist in the	5
	h	Functions: Structure and roles of the Pharmacy and Therapeutic Committee in hospital settings.  Policies of the Pharmacy and Therapeutic Committee: Guidelines for including drugs into the formulary, managing inpatient and outpatient prescriptions, automatic stop orders, and preparing emergency drug lists.  Drug Information Services Drug and Poison Information Centre: Role and functions of the Drug and Poison Information Centre in providing information and guidance.  Sources of Drug Information: Various sources used for obtaining drug-related information.  Computerized Services and  Storage Retrieval of Information: Use of computerized systems for		about Role of pharmacist in the	5

1	Preparation and Implementation:			
	Implementation Budget		and implementation in pharmacy	5
IV	a) Budget Preparation and	8	Know about Budget preparation	3,4,
	patients.			
	interacting with prescribers and			
	communication practices when			
	Communication Skills: Effective			
	prescriptions.			
	requirements for handling			
	medication orders and legal			
	Medication Order: Interpretation of			
	Communication Skills Prescribed			
	d) Prescribed Medication Order and			
	health education initiatives.			
	communication and community			
	involvement in internal			
	Importance of pharmacist			
	and Community Health Education:			
	Interdepartmental Communication			
	Role of Pharmacist in			
	pharmacy practice.			
	standards applicable to community			
	Pharmacy: Ethical guidelines and			
	Code of Ethics for Community			
	clinics.			
	pharmacists to nursing homes and			
	Support and services provided by			
	Services to Nursing Homes/Clinics:			
	initiatives.			
	hospital and external educational			
	programs conducted within the			
	Programs: Overview of training			
	Internal and External Training			
	educational and training programs.			
	pharmacists in conducting			
	Program: Responsibilities of			
	the Education and Training			
	the Hospital Role of Pharmacist in			
	c) Education and Training Program in			
	specialized patient counseling.			
	pharmacists play a crucial role in			
	Pharmacist: Instances where			
	Special Cases that Require the			
	sessions.			
	followed during patient counseling			
	Counseling: Process and procedures			
	Steps Involved in Patient			
	and its importance.			
	what constitutes patient counseling			
	Patient Counseling: Explanation of			

<u> </u>	D		1
	Process and steps involved in		
	preparing and implementing		
	budgets.		
(b)	Clinical Pharmacy Introduction to		
	Clinical Pharmacy: Overview and		
	introduction to the field of clinical		
	pharmacy.		
	Concept of Clinical Pharmacy:		
	Definition and principles underlying		
	clinical pharmacy practice.		
	Functions and Responsibilities of		
	Clinical Pharmacist: Roles and		
	duties of clinical pharmacists in		
	healthcare settings.		
	Drug Therapy Monitoring:		
	Methods and procedures for		
	monitoring drug therapy		
	effectiveness:		
•	Medication chart review		
•	Clinical review		
•	Pharmacist intervention		
•	Ward round participation		
•	Medication history and		
	pharmaceutical care.		
	osing Pattern and Drug Therapy		
	sed on Pharmacokinetic & Disease		
	attern: Considerations and		
	ethodologies for determining dosing		
*	tterns based on pharmacokinetic		
	inciples and disease characteristics.		
(c)	Over-the-Counter (OTC) Sales		
	Introduction and Sale of Over-		
	the- Counter (OTC) Medications:		
	Overview of OTC medications and		
	their sale practices. Rational Use		
	of Common Over-the- Counter		
	Medications: Guidelines and		
	principles for the appropriate and		
	rational use of common OTC		
	medications.		
	rugstore Management and Inventory	Perform Drug store management3,4	4,
	ontrol Organization of Drug Store:	and inventory control in hospitals.5	
	ructure and management practices	Interpret selected laboratory	
	ithin a drug store setting.	results (as monitoring parameters	
	ypes of Materials Stocked and	in therapeutics) of specific disease	
	orage Conditions: Various categories	states	
	materials stocked in drug stores and		
_	propriate storage conditions.		
	urchase and Inventory Control:		
Pr	rinciples of inventory management,		

including purchase procedures, purchase orders, procurement, and stocking practices.

Economic Order Quantity (EOQ) and Reorder Quantity Level: Concepts and calculations related to EOQ and reorder quantity levels in inventory management. Methods Used for the Analysis of Drug

**Expenditure:** Approaches and techniques for analyzing and managing drug expenditure.

Investigational Use of Drugs Description and Principles Involved:

Overview of investigational drug use and underlying principles.

Classification, Control, and Identification: Methods and regulatory aspects related to classifying, controlling, and identifying investigational drugs.

Role of Hospital Pharmacist and Advisory Committee: Involvement of hospital pharmacists and advisory committees in managing investigational drug use.

c) Interpretation of Clinical Laboratory Tests Blood Chemistry, Hematology, and Urinalysis: Understanding and interpretation of clinical laboratory tests related to blood chemistry, hematology, and urinalysis.

#### **TEXT BOOKS:**

T1: Merchant S.H. and Dr. J.S.Quadry. A textbook of hospital pharmacy, 4th ed. Ahmadabad: B.S. Shah Prakakshan; 2001. 2

T2: Parthasarathi G, Karin Nyfort-Hansen, Milap C Nahata. A textbook of Clinical Pharmacy Practiceessential concepts and skills, 1 st ed. Chennai: Orient Longman Private Limited; 2004.

#### **REFERENCE BOOKS:**

R1: William E. Hassan. Hospital pharmacy, 5th ed. Philadelphia: Lea & Febiger; 1986. R2: Tipnis Bajaj. Hospital Pharmacy, 1st ed. Maharashtra: Career Publications; 2008.

R3: Scott LT. Basic skills in interpreting laboratory data, 4thed. American Society of Health System Pharmacists Inc; 2009.

R4: Parmar N.S. Health Education and Community Pharmacy, 18th ed. India: CBS Publishers & Distributers; 2008.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the hospital's organization, the hospital pharmacy, and any adverse drug reactions, and apply this knowledge to the community pharmacy.	PO1,PO2,PO3,PO4,PO5,PO6,PO 7,PO8,PO19,PO11
2	Learn about the patient medication history interview, apply in the community pharmacy management, and understand the hospital formulary, therapeutic drug monitoring, medication adherence, and the hospital drug distribution system.	PO1,PO2,PO3,PO4,PO5,PO6,PO 7,PO8,PO19,PO11
3	Apply the drug distribution system in a hospital, hospital formulary, therapeutic drug monitoring, Medication adherence, patient medication history interview, and community pharmacy management.	PO1,PO2,PO3,PO4,PO5,PO6,PO 7,PO8,PO19,PO11
4	Learn and apply the budget preparation and implementation of clinical Pharmacy, and OTC drugs.	PO1,PO2,PO3,PO4,PO5,PO6,PO 7,PO8,PO19,PO11
5	Understand drug store management and analysis, inventory control, investigational drug use, and the ability to interpret clinical laboratory tests.	PO1,PO2,PO3,PO4,PO5,PO6,PO 7,PO8,PO19,PO11

	SEMESTER – VII										
Course	e Title		Novel Drug	Delivery	Syst	ems (T	heory)	1			
Cours	e code	BP704T	Total credits: 4	L	T	P	S	R	<b>O</b> /I	F	C
			Total hours: 45T	3	1	0	0	0	0		4
Pre-requisite Nil Co-requisite							N	il			
Progra				helor of							
Semo			VII semester of				rograr	nme			
Cou		1 * *	on of the course studer								
Objec	ctives		nd various approaches	s for the	develo	opment	of nov	el drug	g delive	ry	
		systems.				1 1		C .1			
			nd the criteria for sele		-	_	•		develo	pmer	it of
	<b>\</b> 1		delivery systems, thei						41 1	1	1
CO	)1		basics of controlled	drug de	livery	system	ns and	appiy	the kn	iowie	age
CO	)2	in formulation d					of mia		-aulatia		
	)2		formulation and drugelivery, and implants.	release	шесп	iamsim (	or mic	roencaj	psurauc	ш,	
CO	)3	U	concept of drug del	iverv ev	ctema	through	th the	transd	ermol	and	Nana
	JS		es, their application, a	, ,		_	•		CHIII	anu	114110
CO	)4	-	concept of targeted dr			a uisauv	umage	υ.			
CO			delivery system for the		•	riis					
Unit-		Cont		Contac			arning	g Outco	ome		KL
No.		Cont	Ciit	Hour		EC.		, oute			111
I	Contro	olled Drug Deliv	very Systems:	11041	Stu	idents	will	be	able	to	
		_	gy/definitions, and			derstand					
		le, advantages, d			dru		livery	•	tems,	its	
		-	lates. Approaches to			ssificati	•	•	pplicat	ions.	
		_	e formulations based		The	ey will	also le	earn the	e utiliza	ation	
	on diffi	usion, dissolution	n, and ion exchange		of	polym	ers i	n pha	rmaceu	ıtical	
	princip	les. Physicocher	nical and biological	10	for	mulatio	n deve	lopme	nt.		3,4
	propert	ies of drugs rele	vant to controlled								
	release	formulations.									
	Polymo	ers: Introduction	, classification,								
	r -		and application of								
	Γ -		tion of controlled								
		drug delivery sy	stems.								
II		encapsulation	o			derstan			ncept	of	
		-	efinition, advantages,			croenca	_			_	
			pheres/microcapsules,			ivery	and 1	implan	table	drug	
	_	articles. Methods			dei	ivery.					
		apsulation and a		10							2.4
		al Drug Deliver		10							3,4
	Introduction to mucosal drug delive systems. Principles of bioadhesion										
	/mucoadhesion, concepts, advantages,										
		_	ucosal permeability								
		-	erations of buccal								
		y systems.									
	-	rtable Drug Del	ivery Systems								
	_	_	able drug delivery								
		-	sadvantages. Concept								
<u> </u>		, , , , , ,	<u> </u>	l							

	of implants and osmotic pump technology.			
III	Transdermal Drug Delivery Systems Introduction to transdermal drug delivery systems. Permeation through the skin, factors affecting permeation, permeation enhancers. Basic components of TDDS and formulation approaches.  Gastroretentive Drug Delivery Systems Introduction to gastroretentive drug delivery systems. Advantages and disadvantages. Approaches for GRDDS including floating systems, high-density systems, inflatable systems, gastroadhesive systems, and their applications.  Nasopulmonary Drug Delivery System Introduction to nasopulmonary drug delivery systems. Overview of nasal and pulmonary routes of drug delivery. Formulation of inhalers (dry powder and metered dose), nasal sprays, nebulizers.	10	Learn about the techniques involved in improving the delivery of drugs through the skin. Students will understand the concept of localized drug delivery to the GIT and lungs.	
IV	Targeted Drug Delivery Concepts and approaches to targeted drug delivery.  Advantages and disadvantages of targeted drug delivery systems.  Introduction to: Liposomes, Niosomes, Nanoparticles, Monoclonal antibodies  Overview of their applications in drug delivery.	8	Students will be able to learn the techniques involved in targeted drug delivery.	
V	Ocular Drug Delivery Systems Introduction to oculardrug delivery systems. Overview of intraocular barriers and methods to overcome them. Preliminary study of ocular formulations and ocuserts. Intrauterine Drug Delivery Systems Introduction to intrauterine drug delivery systems. Advantages and disadvantages of using intrauterine devices (IUDs). Development of IUDs and their applications.	7	Know about delivery of drugs to eyes and uterine cavity	3,4

T1: N.K. Jain, Controlled and Novel Drug Delivery, CBS Publishers & Distributors, New Delhi, First edition 1997 (reprint in 2001).

T2:. S.P. Vyas and R.K. Khar, Controlled Drug Delivery -concepts and advances, Vallabh Prakashan, New Delhi, First edition 2002.

#### **REFERENCE BOOKS:**

R1: Y W. Chien, Novel Drug Delivery Systems, 2nd edition, revised and expanded, Marcel Dekker, Inc., New York, 1992.

R2: Robinson, J. R., Lee V. H. L, Controlled Drug Delivery Systems, Marcel Dekker, Inc., New York, 1992.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the basics of controlled drug delivery systems	PO1,PO2,PO3,PO6,PO11
1	and apply the knowledge in formulation development.	101,102,103,100,1011
2	Understand the formulation and drug release mechanism of	PO1,PO2,PO3,PO6,PO11
	microencapsulation, mucosal drug delivery, and implants.	101,102,103,100,1011
	Understand the concept of drug delivery systems through the	
3	transdermal and Nano pulmonary routes, their application,	PO1,PO2,PO3,PO6,PO11
	advantages, and disadvantages.	
4	Understand the concept of targeted drug delivery.	PO1,PO2,PO3,PO6,PO11
5	Develop a drug delivery system for the eyes and uterus.	PO1,PO2,PO3,PO6,PO11

			SEMEST	ER – V	I					
Course	Title		INSTRUMENTAL N	ИЕТНО	DDS	OF AN	ALYS	IS (Pra	actical)	
Course	code	BP705P	Total credits: 2	L	T	P	S	R	O/F	C
			Total hours: 4	0	0	) 4	0	0	0	2
Pre-rec	quisite	Nil	Co-requisite					Nil		•
Progra	mme		Ba	chelor	of P	harmac	y			
Seme	ster		VII semester o	f Fourt	h ye	ear of the	e prog	ramme	9	
Course O	bjectives	1. Co	mpare operational tecl	nniques	of U	JV, HPL	C fluo	imeter	, flame ph	otometer,
		etc.								
		2. De	velop basic practical s	kills usi	ng ii	nstrumen	tal tec	hniques	S.	
		3. Co	rrelate quantitative ar	ıd qualit	ativ	e drug a	nalysi	s using	g various a	ınalytical
		ins	truments.							
CO			he fundamental and th							
CO	)2	Apply practi	cal knowledge of va	rious in	stru	mentatio	n tech	niques	for drug	and
		excipient ana	*							
CO			absorption maxima,				, and	unknov	vn concen	tration
			ample by using UV-vi							
CO	)4	Analyze and	estimate the Sodium a	and pota	ssiu	ım ions b	y using	g Flame	e photome	try.
CO	<b>)</b> 5	Evaluate the	organic compounds/ar	nino aci	ls/p	lant pign	nents b	y using	various	
		chromatogra	phic and spectroscopic	cal techr	iqu	es.				
Unit-		Con	tent	Contac	et	Le	arning	g Outco	ome	KL
No.				Hour						
I	1.	Determination	on of absorption		U	Jnderstar	nd the	qualitat	tive and	
		maxima and	effect of solvents on		q	_l uantitati	ve ana	lysis of	drugs by	
		absorption n	naxima of organic		V	arious sp	ectros	copic to	echnique,	
		compounds.				luorimeti	-			
	2.		of dextrose by			hotomet			0	
		colorimetry.			tı	urbidome	etry me	ethods.		
	3.	Estimation of	of sulfanilamide by							
		colorimetry.								
	4.		is estimation of							
		ibuprofen ar	nd paracetamol by UV							
		spectroscopy								
	5.		acetamol by UV-							
		Spectrophoto								3,4,
	6.		of quinine sulfate by							5,6
		fluorimetry.								
	7.	Study of que	•							
		fluorescence								
	8.		on of sodium by							
		flame photor	· ·							
	9.		on of potassium by							
		flame photo	*							
	10.		on of chlorides and							
		sulphates by								
		nepheloturb	· · · · · · · · · · · · · · · · · · ·							
	11.	_	of amino acids by							
		paper chrom								
	12.	-	f sugars by thin layer							
		chromatogra	phy.							

13. Separation of plant pigments by		
column chromatography.		
14. Demonstration experiment on		
HPLC.		
Demonstration experiment on Gas		
Chromatography.		

T1: Practical Pharmaceutical Chemistry by A.H. Beckett and J.B. Stenlake T2: Quantitative Analysis of Drugs by D. C. Garrett

# **REFERENCE BOOKS:**

R1: Quantitative Analysis of Drugs in Pharmaceutical Formulations by P. D. Sethi R2: Spectrophotometric identification of Organic Compounds by Silverstein

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the fundamental and theoretical concepts of	PO1,PO2,PO3,PO4,PO5,PO6,P
_	instrumentation techniques	O7,PO8,PO19,PO11
2	Apply practical knowledge of various instrumentation	PO1,PO2,PO3,PO4,PO5,PO6,P
	techniques for drug and excipient analysis.	O7,PO8,PO19,PO11
	Evaluate the absorption maxima, effect of absorption, and	PO1,PO2,PO3,PO4,PO5,PO6,P
3	unknown concentration of the drug sample by using UV-visible	07,P08,P019,P011
	spectroscopy.	07,108,1019,1011
4	Analyze and estimate the Sodium and potassium ions by using	PO1,PO2,PO3,PO4,PO5,PO6,P
4	Flame photometry.	O7,PO8,PO19,PO11
5	Evaluate the organic compounds/amino acids/plant pigments by	PO1,PO2,PO3,PO4,PO5,PO6,P
3	using various chromatographic and spectroscopical techniques.	O7,PO8,PO19,PO11

	SEMESTER – VIII										
Course T			IOSTATISITCS AND R		EARCH	I MET		OLO	GY	_	
Course c	ode	BP801T	Total credits: 4	L	T	P	S	R	O/F	<b>C</b>	
			Total hours: 45T	3	1	0 0 0 0					
Pre-requi		Nil	Co-requisite				N	il			
Program			Bachelor of Pharmacy								
Semeste			Fall/VIII semester of								
Course		_	ion of M.S. Excel, SPSS,						-		
Objectiv			ow the various statistical		_		e statis	stical p	roblems		
CO1		* *	tical techniques in solving pasic concepts of statistics		•		d	• d oo o	fficients va	. d	
CO2			istributions, sample popul							zu.	
CO ₂			ta obtained by using diff			•				f	
COS		clinical Trials	a obtained by using diff	CICII	и поп р	aranne	uic ic	sis and	u piiases o	L	
CO4			ledge of different softwa	res	such as	Evce	1 SPS	S MIN	NITAR for	the	
		data acquired.	reage of different softwa	03	sucii as	LACC	., 51 5	٠, ١٧١١١	11111 IOI	uic	
CO5		_	stical design for the exper	imer	nts and a	nalysi	is.				
Unit-No.		•	Content		Contact			ing Ou	itcome	KL	
					Hour			8			
I	Intro	duction: Statisti	cs, Biostatistics, Frequenc	y	10	Introd	luction	ı of Bi	ostatics,	1,2,	
	distrib	oution	•			centra	al tend	ency a	nd	3,4,	
	Meas	ures of central t	endency:			Corre	lation			5	
-	Mean	, Median, Mode-	Pharmaceutical example	es							
	Meas	ures of dispersion	n: Dispersion, Range,								
			armaceutical problems								
			on, Karl Pearson's								
1			ntion, Multiple correlation	-							
		naceuticals exam	•			~					
	_		fitting by the method	- 1	10	_ ~			ributions,	1,2,	
		-	the lines $y=a+bx$ and $x$	=a					errors and	3,4,	
1	•		ion, standard error of			paran	netric 1	tests.		5	
1	_	ssion– Pharmace	uticat Examples. n of probability, Binomial								
1		*	istribution, Poisson's								
1			s – problems Sample,								
1			ple, small sample, Null								
1	_		hypothesis, sampling,								
1	• •		ypes of sampling, Error-I								
1			andard error of mean (SEI								
1	• •	rmaceutical exan	,								
1			t(Sample, Pooled or								
1			, ANOVA, (One way an	d							
1	-		ificance difference								
		• • • • • • • • • • • • • • • • • • • •	ts: Wilcoxon Rank Sun	ı		Evalu	ate th	e data	a obtained		
	Test,	Mann-Whitney	U test, Kruskal-Wallis t	est,		by usi	ing di	fferent	non		
	Friedı	man Test.				paran	netric t	ests an	d phases of	f	
	Intro	duction to Rese	earch: Need for research,			clinic	al trial	S.		1,2,	
	Need	for design of Exp	periments, Experiential							3,4,	
	_	n Technique, pla	_							5	
	Grap	<b>hs:</b> Histogram, I	Pie Chart, Cubic Graph,								

	response surface plot, Counter Plot graph			
	<b>Designing the methodology:</b> Sample size			
	determination and Power of a study, Report	10		
	writing and presentation of data, Protocol,			
	Cohorts studies, Observational studies,			
	Experimental studies, Designing clinical trial,			
	various phases.			
IV	Blocking and confounding system for Two-level		Solve using different	1,2,
	factorials.		software (Excel, SPSS,	3,4,
	Regression modeling: Hypothesis testing in Simple		MINITAB) for the data	5
	and Multiple regression models. <b>Introduction to</b>		acquired.	
	Practical components of Industrial and			
	Clinical Trials Problems:			
	Statistical Analysis Using Excel, SPSS,	8		
	MINITAB®, DESIGN OF EXPERIMENTS, R –			
	Online Statistical Software's to Industrial and			
	Clinical trial approach.			
V	Design and Analysis of experiments:		Develop a statistical design	1,2,
	<b>Factorial Design:</b> Definition, 2 ² , 2 ³ design.		for the experiments and	3,4,
	Advantage of factorial design.		analysis	5
	Response Surface methodology:	7		
	Central composite design, Historical design,			
	Optimization Techniques			

T1: Biostatistics and Research Methodology, Dr. Vinod Kumar Bais, PEE VEE.

# **REFERENCE BOOKS:**

R1: Pharmaceutical statistics- Practical and clinical applications, Sanford Bolton, publisher Marcel Dekker Inc. New York.

R2: Design and Analysis of Experiments –PHI, Learning Private Limited, R. Pannerselvam. R3: Sharp, Lester W. Fundamentals of Cytology. 1st edition. McGraw Hill Company; 1943.

CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Understand the basic concepts of statistics, mean, median, mode,	PO1,PO2,PO3,PO4,PO5,PO6,				
1	and coefficients used.	PO7,PO8,,PO11				
2	Categorize the distributions, sample population errors, and	PO1,PO2,PO3,PO4,PO5,PO6,				
	parametric tests.	PO7,PO8,,PO11				
3	Evaluate the data obtained using different non-parametric tests	PO1,PO2,PO3,PO4,PO5,PO6,				
3	and phases of clinical trials.	PO7,PO8,,PO11				
4	Apply the knowledge of software such as Excel, SPSS, and	PO1,PO2,PO3,PO4,PO5,PO6,				
4	MINITAB for the data acquired.	PO7,PO8,,PO11				
5	Explain the statistical design for the experiments and analysis.	PO1,PO2,PO3,PO4,PO5,PO6,				
	Explain the statistical design for the experiments and analysis.	PO7,PO8,,PO11				

			SEMESTER – VIII										
Course													
Course	e code	BP 802T	Total credits: 4	L		P	S	R	O/F	C			
			Total hours: 45T	3	3 1	0	0	0	0	4			
Pre-rec		Nil	Co-requisite				N	il					
Progra					of Phar								
Seme		Fall/VIII semester of first year of the programme											
Cou		1. Acquire high consciousness/realization of current issues related to health and											
Objectives		pharmaceutical problems within the country and worldwide											
		2. Have a critical way of thinking based on current healthcare development.											
		3. Evaluate alternative ways of solving problems related to health and											
CO1		pharmaceutical issues.											
CO1 CO2		Identifying the fundamental concept, health concept, disease, education, hygiene practice.											
CC		Categorize some diseases' symptoms and preventive medications.  Determine the objectives, functioning, and consequences of national health programmes.											
CC			Evaluating some of India's national health programmes in relation to WHO's role										
CC		_	cance of health education is	•	_					d			
		urban areas		5011	.oom un	•0111111	iii y	. 51 / 100	I GIGI GII				
Unit-		<u> </u>	Content	(	Contact	L	earnii	ng Out	come	KL			
No.					Hour			8					
I	Conce	pt of healt	h and disease: Definition,		10	Recogni	ze the	concep	ts and	1,2,			
	concep	ts and eva	aluation of public health.			evaluati	on of p	oublic h	ealth.	3,4			
		-	e concept of prevention	of prevention				Understand the concept of					
	and co	ontrol of	disease, social causes of			prevention and control of							
			l problems of the sick.			disease.							
	Social and health education: Food in					Illustrate sociocultural factors							
	relation to nutrition and health, Balanced					and its relation with health.							
	diet, Nutritional deficiencies, Vitamin					Identify avoidable habits for							
	deficiencies, Malnutrition and its prevention.					personal hygiene and health. Understand malnutrition and its							
	Sociology and health: Socio cultural factors												
	related to health and disease, Impact of urbanization on health and disease, Poverty					prevention.  Recognize the community							
	and health					services in rural, urban and							
	Hygiene and health: Personal hygiene and					school health.							
	health care; avoidable habits					Identify avoidable habits for							
		•				personal							
II	l .		cine: General principles of		10	Identify	and u	ındersta	and the	1,2,			
	r .		ontrol of diseases such as			_			strategies	3,4			
			bola virus, influenza, acute						ocial and				
	respiratory infections, malaria, chicken guine		iea,		preventive pharmacy								
	dengue, lymphatic filariasis, pneumonia,												
	1		betes mellitus, cancer, drug										
TTT		Idiction-drug substance abuse.			Illumentary and the second of								
III National health programs			_			Illustrate sociocultural factors and its relation with health.							
	functioning and outcome of the following:												
	HIV AND AIDS control programme, TB, Integrated disease surveillance program (IDSP),			P).	Identify and understand the general measures and strategies								
	National leprosy control programme, National				to be followed in social and								
			gram, National programme			preventi				1,2,			
		-	ntrol of deafness, Universal		-	1	120	- , •		3,4			
L	-		,	1		l				1 ′			

	immunization programme, National programme			
	for control of blindness, Pulse polio programme.			
IV	National health intervention programme for		Identify and understand the	
	mother and child, National family welfare		general measures and strategies	
	programme, National tobacco control		to be followed in social and	
	programme, National Malaria Prevention		preventive pharmacy	
	Program, National programme for the health	8		1,2,
	care for the elderly, Social health programme;			3,4
	role of WHO in Indian national program.			
V	Community services in rural, urban and school		Understand and expressed the	9
	health: Functions of PHC, Improvement in rural		principles on the prevention and	1
	sanitation, national urban health mission, Health	7	control of communicable and	1,2,
	promotion and education in school.		non-communicable diseases	3,4

T1: Textbook of Preventive and Social Medicine, Prabhakara GN, 2nd Edition, 2010, ISBN:9789380704104, JAYPEE Publications.

T2: Textbook of Preventive and Social Medicine (Mahajan and Gupta), Edited by Roy Rabindra Nath, SahaIndranil, 4thEdition, 2013, ISBN: 9789350901878, JAYPEE Publications.

T3: Park Textbook of Preventive and Social Medicine, K Park, 21st Edition, 2011, ISBN-14: 9788190128285, BANARSIDAS BHANOT PUBLISHERS

#### **REFERENCE BOOKS:**

R1: Review of Preventive and Social Medicine (Including Biostatistics), Jain Vivek, 6th Edition, 2014, ISBN: 9789351522331, JAYPEE Publications.

R2: Research in Social and Administrative Pharmacy, Elsevier, Ireland.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Define the fundamental health concepts, including disease,	PO1,PO2,PO3,PO4,PO5,PO6,					
1	education, and hygiene practice.	PO8PO9,PO10, PO11					
2	Categorize disease symptoms and preventive measures	PO1,PO2,PO3,PO4,PO5,PO6,					
2	Categorize disease symptoms and preventive measures	PO8PO9,PO10, PO11					
3	Define objectives, functions, and outcomes of national	PO1,PO2,PO3,PO4,PO5,PO6,					
3	health programs	PO7,PO8,PO9,PO10, PO11					
4	Describe India's national health programmes concerning	PO1,PO2,PO3,PO4,PO5,PO6,					
4	WHO's role	PO7,PO8,PO9,PO10, PO11					
_	Emphasize the importance of health education in schools	PO1,PO2,PO3,PO4,PO5,PO6,					
3	and community service in rural and urban areas.	PO8PO9,PO10, PO11					

			SEMI	ESTE	ER – V	Ш						
Course	e Title		PHARMA	MAl	RKET	'IN	G M	ANA	GEM	ENT		
Course		BP	Total credits: 4	L	T	F		S	R			
		803ET	Total hours: 45T	3	1	0	)	0	0	0	4	
Pre- requisite Nil Co-requisite									Nil			
Progra					helor o			•				
	Semester Fall/VIII semester of first year of the programme											
Cou		1.	Acquire high conscious								o health a	nd
Objec	ctives	_	pharmaceutical problem				•					
		2.	Have a critical way of the		-						-	
3. Evaluate alternative ways of solving problems related to health and pharmaceutical issues.					nd							
GO1		TT 1 .	*		4 C	1			1 1		1 4	
CO1		l	and and enumerate the co	_	_	nar	mace	eutica	ıı mar	keting and	product	
CO2			ment in pharmaceutical i			4:		1		1	-4	
			per the various componen		_					icai produc	cis	
CO3		l	and the different pharmac per the role and responsib				_			agantati	and maining	~
CO4			es in India	mity (	oi proi	less	ional	sale	s repre	esemanve	anu pricin	g
CO5			es in India ne emerging concepts of r	nortra	ting c	nd +	he **	10	arlzat «	recentah		
Unit-	<u> </u>	Appry u	Content	Hai KC	Conta					ng Outcon	20	KL
No.			Content		Hou			L	eai iiii	ng Outcon	iie	KL
I I	Marke	ting		-	1100		Able	to a	rticula	te the defi	nition	
1	1	ition, general concepts and scope of eting; Distinction between marketing										
	1					general concepts, and the broad scope of marketing.				oroad		
	1	ng; Marketing environment; Industry					_			-	achievino	
	1	ompetitive analysis; Analyzing					This outcome focuses on achieving a solid understanding of					
		ner buyin	ving			foundational marketing principles.						
		or. <b>Phar</b> i								1		
		tative and qualitative aspects; size										
	1		on of the market;		10							1,2
	demog	raphic de	escriptions and socio-									
	psycho	logical c										
	consun	ner; mark	ting.									
	Consur	ner profi										
	Г	_	its of the physician; patie									
	1		cian and retail pharmacis									
	1	•	Market; Role of market									
-	researc						411		-		1	1.0
II		ct decisio								ate effectiv		1,2
	1	_	product line and product i	mıx			1	_		rategies, in	_	
			act life cycle,p roduct	Jour			1			d to produ roduct line		
	Г	-	is; product positioning; Nns; Product branding,	NEW	10		prod		_	roduct IIII	zs, allu	
	Г		abelling decisions, Produ	ıct	10		1			product	life	
	_	_	pharmaceutical industry.				cycle		na iliC	product	1110	
	inanage	-111011t III	Pharmaceurear moustry.	.			1 -		nrodi	act portfoli	io	
							_	-	_	nplement		
										luct positio	oning	
							strate		_	positi	6	
III	Promo	tion:		+				_		and imples	nent	
							1 10			p.101		l

promotional budget; personal selling, adve journals, sampling, re exhibition, public rel	ertising, direct mail, etailing, medical	10	integrated promotion strategies by understanding the determinants of the promotional mix.  This outcome emphasizes the ability to create cohesive and effective promotional campaigns that utilize multiple channels for	4,5
in channels, physical management: Strateg in physical distributi <b>Professional sales re</b> Duties of PSR, purposelection and training	channel members, riate channel, conflict distribution ic importance, tasks on management. epresentative (PSR): ose of detailing, g, supervising, norms otivating, evaluating,	10	maximum impact.  Able to design, analyze, and optimize pharmaceutical marketing channels.  This outcome emphasizes the ability to strategically navigate the complexities of pharmaceutical distribution to ensure efficient and effective product reach.	
V Pricing:  Meaning, importance determinants of price and strategies, issue	ce; pricing methods es in price armaceutical industry. DPCO (Drug Price NPPA (National	10	Able to learn to navigate and comply with regulatory frameworks while making pricing decisions.  Apply advanced marketing concepts, including vertical and horizontal marketing, rural marketing, consumerism, industrial marketing, and global marketing.	4,5

T1: Philip Kotler and Kevin Lane Keller: Marketing Management, Prentice Hall of India, New Delhi T2: Dhruv Grewal and Michael Levy: Marketing, Tata MC Graw Hill

T3: Arun Kumar and N Menakshi: Marketing Management, Vikas Publishing, India T4: Rajan Saxena: Marketing Management; Tata MC Graw-Hill (India Edition)

T5: Ramaswamy, U.S & Nanakamari, S: Marketing Managemnt: Global Perspective, Indian Context, Macmilan India, New Delhi.

T6: Shanker, Ravi: Service Marketing, Excell Books, New Delhi

#### **REFERENCE BOOKS:**

R1: Subba Rao Changanti, Pharmaceutical Marketing in India (GIFT – Excel series) Excel Publications R2: Walker, Boyd and Larreche: Marketing Strategy- Planning and Implementation, Tata MC GrawHill, New Delhi.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Understand and enumerate the concept of pharmaceutical marketing and product management in the pharmaceutical industry	PO1,PO2,PO3,PO4,PO5,PO6, ,PO8,PO9,PO11					
2	Use various components of promotion of pharmaceutical products.	PO1,PO2,PO3,PO4,PO5,PO6, ,PO8,PO9,PO11					
3	Understand the different pharmaceutical marketing channels	PO1,PO2,PO3,PO4,PO5,PO6, ,PO8,PO9,PO11					
4	Describe the role and responsibility of professional sales representatives and pricing authorities in India.	PO1,PO2,PO3,PO4,PO5,PO6, ,PO8,PO9,PO11					
5	Apply the emerging concepts of marketing and the role of market research.	PO1,PO2,PO3,PO4,PO5,PO6, ,PO8,PO9,PO11					

			SEMESTE	ZR – V	/III					
Course						CIENC				
Course	code	BP 809ET	Total credits: 4	L	T	P	S	R	O/F	C
<b>D</b>	• • .	3.101	Total hours: 45T	3	1	0	0	0	0	4
Pre-rec		Nil	Co-requisite	1	- C DL			Nil		
Progra						armac				
Seme Cou		1. Ac	Fall/ VIII semester							th and
Objec			armaceutical problems wi							lii aiiu
Objec	uves	_	ve a critical way of thinking			-				ıt.
			aluate alternative ways of	•					•	
			armaceutical issues.							
CC	1	Understand	how Indian and EU legis	lation	class	sifies co	osmeti	cs and	cosmeceuti	cals.
		Learn the ar	natomy of the skin, hair, an	nd ora	ıl cav	ities an	d aesth	etic the	erapy excip	ients
CC	2		d hair care product con	_		_		. Expla	in develop	oing
			npoo, toothpaste, and colo							
CC	3	1	sun protection cosmetic			s using	g your	new s	skills. Visi	t BIS for
~ -		_	kin cream, and shampoo			1 1 1		4	, 571 -1	
CC	94	_	thod for researching and	_			ın care	produ	cts. The the	eory
CC	<u> </u>		uments like sebumeters ar					-1 4-		1
	5	skin, and da	&D efforts focused on d	ieveio]	ping	novei a	ipproa	enes to	greasy nai	r, ary
Unit-		· ·	Content	Con	tact	Learn	ing Ωι	ıtcome		KL
No.			ontent	Ho		Learn	ing O	ittoine		
I	Classif	ication of cos	smetic and cosmeceutical			Under	stand t	he cosr	netic	4,5
	produc					produc	ts as p	er Indi	an & EU	
	Indian	and EU regul	lations, Evolution of			regulat	ion.			
		ceuticals from			Understand the key					
	•	nd OTC drug				_		uitable		
		_	ts: Surfactants, rheology					of vario	ous	
			nts, emollients,			cosme	•			
	_		ification and application	1	Δ				nents of	
			re and function of skin. are of hair. Hair growth		0	biolog	•	ction ar		
	cycle.	Dasic structi	ne of han. Han growth						i, hair and	
	•	Cavity: Com	mon problem associated					discus		
		eth and gums	•				•		lated to the	
		C				skin ar	_			
						Study	the s	ructure	and	
						functio	on of sl	kin		
II	_	_	ılation and building						ampoos	4,5
			products: Face wash,					-	air dye	
			Cold Cream, Vanishing						hanism of	
		and their adv				action		perspa	nts &	
		antages. Appl nulation of co	lication of these products			deodor	ants			
			eodorants- Actives &							
	_	nism of action								
			ılation and building							
	_	•	products: Conditioning	1	0					

	1 11: 1:4: 4: 1 1 00		T	
	shampoo, Hair conditioner, anti-dandruff			
	shampoo. Hair oils.			
	Chemistry and formulation of Para-			
	phylenediamine based hair dye.			
	Principles of formulation and building			
	blocks of oral care products: Toothpaste			
	for bleeding gums, sensitive teeth. Teeth			
	whitening, Mouthwash			
III	Sun protection, Classification of Sunscreens		Describe the evaluation of hair	4,5
	and SPF.		and skin preparations	
	Role of herbs in cosmetics:			
	Skin Care: Aloe and turmeric. Hair care:			
	Henna and amla. <b>Oral care:</b> Neem and clove	10		
	Analytical cosmetics: BIS specification and			
	analytical methods for shampoo, skin- cream			
	and toothpaste.			
IV	Principles of Cosmetic Evaluation:		Explain sebumeter, and	4,5
	Principles of sebumeter, corneometer.		corneometer.	
	Measurement of TEWL, Skin Color, Hair			
	tensile strength, Hair combing properties	8		
	Soaps, and syndet bars. Evaluation and			
	skin benefits.			
V	Oily and dry skin, causes leading to dry skin,		Explain the terms blemishes,	4,5
	skin moisturisation. Basic understanding of		wrinkles, acne etc.	
	the terms Comedogenic, dermatitis.			
	Cosmetic problems associated with Hair			
	and scalp: Dandruff, Hair fall causes			
	Cosmetic problems associated with skin:	7		
	blemishes, wrinkles, acne, prickly heat and			
	body odor.			
	pody odor.			
	Antiperspirants and Deodorants- Actives			
	•			

- T1: Cosmetics Formulation, Manufacture and quality control, PP.Sharma, 4thedition
- T2: Handbook of cosmetic science and Technology A.O.Barel, M.Paye and H.I. Maibach. 3rdedition.
- T3: Text book of cosmeticology by Sanju Nanda & Roop K. Khar, Tata Publishers.

#### **REFERENCE BOOKS:**

- R1: Harry's Cosmeticology, Wilkinson, Moore, Seventh Edition, George Godwin. R2: Poucher's perfume cosmetics and Soaps, 10thedition.
- R3: Cosmetic and Toiletries recent suppliers' catalogue.
- R4: CTFA (The Cosmetic, Toiletry & Fragrance Association) directory.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Understand how Indian and EU legislation classifies cosmetics and cosmeceuticals. Learn the anatomy of the skin, hair, and oral cavities and the excipients used in aesthetic therapy.	PO1,PO3,PO4,PO6,PO7,PO8, PO9,PO10,PO11					
2	Explain skin and hair care product components and processes and develop skills in cream, shampoo, toothpaste, and color formulation.	PO1,PO3,PO4,PO6,PO7,PO8, PO9,PO10,PO11					
3	Create many sun protection cosmetics formulas using your new skills.	PO1,PO3,PO4,PO6,PO7,PO8, PO9,PO10,PO11					
4	Design a method for researching and rating hair and skin care products.	PO1,PO3,PO4,PO6,PO7,PO8, PO9,PO10,PO11					
5	Apply cosmetics R&D efforts focused on developing novel approaches to greasy hair, dry skin, and dandruff.	PO1,PO3,PO4,PO6,PO7,PO8, PO9,PO10,PO11					

	SEMESTER – VIII								
Course Title		Project Work							
Course code	BP813PW	Total credits: 4	L	C					
		Total hours: 45T	3	1	2	0	0	0	6
Pre-requisite	Nil	Co-requisite					Nil		
Programme		Bac	helor	of Pha	rmacy	y			
Semester		Fall/VIII semest	er of fi	rst yea	ar of t	he pro	gramm	ie	
Course	1. Acqui	re research skills.							
Objectives	2. Develo	op scientific writing skil	ls.						
	3. Foster	critical thinking ability.							
	4. Apply	application-oriented lea	rning.						
	5. Impro	ve time management an	d orgai	nizatio	nal ski	lls.			
	6. Enhan	ce communication skills	S.						
CO1	Perform an In	nterdisciplinary work.							
CO2	Apply theore	tical knowledge from the	e litera	ture re	view. S	Start E	xperime	ent.	
CO3	Utilize the Li	terature Review and De	sign O	bjectiv	es and	Plan o	f work.		
CO4	Analyze the I	Results obtained and Ev	aluate	them u	sing St	tatistic	al meth	ods with a	
	Conclusion.								
CO5	Build Design	and Compose a Manuso	cript fo	r Publi	cation	•			

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Perform an Interdisciplinary work.	PO1,PO2,PO3,PO4,PO6,PO7,PO8				
1	r errorm an interdisciplinary work.	,PO9,PO10,PO11				
2	Apply theoretical knowledge from the literature review.	PO1,PO2,PO3,PO4,PO6,PO7,PO8				
	Start Experiment.	,PO9,PO10,PO11				
3	Utilize the Literature Review and Design Objectives and	PO1,PO2,PO3,PO4,PO6,PO7,PO8				
3	Plan of work.	,PO9,PO10,PO11				
4	Analyze the Results obtained and Evaluate them using	PO1,PO2,PO3,PO4,PO6,PO7,PO8				
4	Statistical methods with a Conclusion.	,PO9,PO10,PO11				
5	Build Design and Compose a Manuscript for Publication.	PO1,PO2,PO3,PO4,PO6,PO7,PO8				
3	Bund Design and Compose a Manuscript for Fublication.	,PO9,PO10,PO11				



### **ASSAM DOWN TOWN UNIVERSITY**

## Curriculum and Syllabus

## Master of Pharmacy (Pharmaceutics)

**OUTCOME BASED EDUCATION FRAMEWORK** 

**CHOICE BASED CREDIT SYSTEM** 

Version: 1.01

# FACULTY OF PHARMACEUTICAL SCIENCE

July, 2023

#### **PREAMBLE**

Assam down town University is a premier higher educational institution which offers Bachelor, Master, and Ph.D. degree programs across various faculties. These program, collectively embodies the vision and mission of the university. All the programs offered by the Faculty of Pharmaceutical Science of Assam down town University strictly follow the curriculum approved by the Pharmacy Council of India (PCI), the statutory body responsible for regulating the profession of pharmacy in India. This document contains outline of teaching and learning framework and complete detailing of the courses. This document is a guidebook for the students to choose desired courses for completing the program and to be eligible for the degree. This volume also includes the prescribed literature, study materials, texts, and reference books under different courses as guidance for the students to follow.

Recommended by the Board of Studies (BOS) meeting of the Faculty of Pharmaceutical Science held on dated 08/07/2023 and approved by the Emergent Academic Council(AC) meeting held on dated 28/07/2023

Chairperson, Board of Studies

Member Secretary, Academic Council

To become a Globally Recognized University from North Eastern Region of India, dedicated to the Holistic Development of Students and Making Society Better

#### **Missions**

- 1. Creation of curricula that address the local, regional, national, and international needs of graduates, providing them with diverse and well-rounded education.
- 2. Build a diverse student body from various socio-economic backgrounds, provide exceptional value-based education, and foster holistic personal development, strong academic careers, and confidence.
- 3. Achieve high placement success by offering students skill-based, innovative education and strong industry connections.
- 4. Become the premier destination of young people, desirous of becoming future professional leaders through multidisciplinary learning and serving society better.
- 5. Create a highly inspiring intellectual environment for exceptional learners, empowering them to aspire to join internationally acclaimed institutions and contribute to global efforts in addressing critical issues, such as sustainable development, Climate mitigation and fostering a conflict-free global society.
- 6. To be renowned for creating new knowledge through high quality interdisciplinary research for betterment of society.
- 7. Become a key hub for the growth and excellence of AdtU's stakeholders including educators, researchers and innovators.
- 8. Adapt to the evolving needs and changing realities of our students and community by incorporating national and global perspectives, while ensuring our actions are in harmony with our foundational values and objectives of serving the community.

#### **Programme Overview**

M.Pharm programme designed to enrich students' basic and advanced knowledge in the Pharmaceutical Science domain, the programme follows the courses mandated by Pharmacy Council of India (PCI) education regulations. The semester-wise course sequence and the entire M. Pharm curricula have been arranged to provide hands-on training and real-world exposure to traditional and modern practices, making graduates industry-ready. As pharmacists are true drug experts, M. Pharm students are exposed to allied science courses and core pharmaceutical courses, fostering their aptitude for research and advancements in new drug development technologies.

Rules & Syllabus for the Master of Pharmacy (M. Pharm) Course framed under Regulation of the 2014 as per by Pharmacy Council of India (PCI).

#### **Duration of the course**

The course of study for M.Pharm shall extend over a period of four semesters (two academic years).

#### **Specific Features of the Curriculum**

The M Pharm curriculum is designed to align with the evolving needs of the pharmacy field and society at large. It offers a comprehensive blend of theoretical knowledge and practical applications essential for a profound understanding of pharmaceuticals, fostering the development of a wide array of skills. This curriculum is thoughtfully designed to equip students with both theoretical acumen and hands-on proficiency, catering precisely to the requirements of the dynamic industry and the broader societal demands.

#### **ELIGIBILITY Criteria:**

A Pass in the following examinations:

B. Pharm Degree examination of an Indian university established by law in India from an institution approved by Pharmacy Council of India and has scored not less than 55% of the maximum marks (aggregate of 4 years of B.Pharm.)

Every student, selected for admission to post graduate pharmacy program in any PCI approved institution should have obtained registration with the State Pharmacy Council or should obtain the same within one month from the date of his/her admission, failing which the admission of the candidate shall be cancelled.

Note: It is mandatory to submit a migration certificate obtained from the respective university where the candidate had passed his/her qualifying degree (B.Pharm.)

#### **Program Educational Objectives (PEOs):**

- **PEO-1:** AdtU Pharmacy graduates will be well prepared for successful careers as Pharmaceutical Professionals across diverse sectors including the pharmaceutical industry, healthcare, corporate institutions and government organizations.
- **PEO-2:** Pharmacy graduates will be academically prepared to become Registered Pharmacists, poised to make significant contributions to the advancement of the healthcare sector.
- **PEO-3:** The graduates will engage in professional practices to elevate their stature with a sense of responsibility and be successful in higher education, if pursued.

#### **Programme Specific Outcomes (PSOs):**

- **PSO-1:** Professional Excellence: Translate the high-level of understanding of drug action into key stages in preclinical, clinical research studies and interpret data of pharmaceutical experiments in drug discovery and modifications as per the needs of pharmaceutical industries.
- **PSO-2:** Practice in Research: Apply pharmacy knowledge and competency in research, and collaborative projects thereby contributing to the continuous advancement of pharmaceutical science.

**PSO-3:** International Competency: Demonstrate global professional competencies by attaining interdisciplinary knowledge through specialized certifications offered on international learning platforms.

#### **Program Outcome (POs):**

- **PO.1:** Pharmacy Knowledge: Possess knowledge and comprehension of the core and basic knowledge associated with the profession of pharmacy, including biomedical sciences; pharmaceutical sciences; behavioural, social, and administrative pharmacy sciences; and manufacturing practices.
- **PO.2:** Planning abilities: Demonstrate effective planning abilities including time management, resource management, delegation skills and organizational skills. Develop and implement plans and organize work to meet deadlines.
- **PO.3:** Problem analysis: Utilize the principles of scientific enquiry, thinking analytically, clearly and critically, while solving problems and making decisions during daily practice. Find, analyse, evaluate and apply information systematically and shall make defensible decisions.
- **PO.4:** Modern tool usage: Learn, select, and apply appropriate methods and procedures, resources, and modern pharmacy-related computing tools with an understanding of the limitations.
- **PO.5:** Leadership skills: Understand and consider the human reaction to change, motivation issues, leadership and team-building when planning changes required for fulfilment of practice, professional and societal responsibilities. Assume participatory roles as responsible citizens or leadership roles when appropriate to facilitate improvement in health and well-being.
- **PO.6:** Professional identity: Understand, analyse and communicate the value of their professional roles in society (e.g. health care professionals, promoters of health, educators, managers, employers, employees).
- **PO.7:** Pharmaceutical ethics: Honour personal values and apply ethical principles in professional and social contexts. Demonstrate behaviour that recognizes cultural and personal variability in values, communication and lifestyles. Use ethical frameworks; apply ethical principles while making decisions and take responsibility for the outcomes associated with the decisions.
- **PO.8:** Communication: Communicate effectively with the pharmacy community and with society at large, such as, being able to comprehend and write effective reports, make effective presentations and documentation, and give and receive clear instructions.

#### **Career Prospects:**

M. Pharm graduates are equipped to assume diverse roles, such as Industrial Pharmacist (in the field of Production and Manufacturing, Formulation Development, Quality Assurance, Quality Control, Packaging, R&D etc.), Hospital and Community Pharmacist, Medical Representative, Sales Executive, Bulk Medicine Distributor, Lecturer, Entrepreneurship, Drug Inspector, Drug Analyst etc. After completion of M. Pharmacy, the students may go for higher studies in PhD programs.

#### **CHAPTER -I: REGULATIONS**

#### 1. Short Title and Commencement

These regulations shall be called as "The Revised Regulations for the Master of Pharmacy (M.Pharm) Degree Program - Credit Based Semester System (CBSS) of the Pharmacy Council of India, New Delhi". They shall come into effect from the Academic Year 2016-17. The regulations framed are subject to modifications from time to time by the authorities of the university.

#### 2. Minimum qualification for admission

A Pass in the following examinations:

B.Pharm Degree examination of an Indian university established by law in India from an institution approved by Pharmacy Council of India and has scored not less than 55% of the maximum marks (aggregate of 4 years of B.Pharm.)

Every student, selected for admission to post graduate pharmacy program in any PCI approved institution should have obtained registration with the State Pharmacy Council or should obtain the same within one month from the date of his/her admission, failing which the admission of the candidate shall be cancelled.

Note: It is mandatory to submit a migration certificate obtained from the respective university where the candidate had passed his/her qualifying degree (B.Pharm.)

#### 3. Duration of the program

The program of study for M.Pharm. Shall extend over a period of four semesters (two academic years). The curricula and syllabi for the program shall be prescribed from time to time by Pharmacy Council of India, New Delhi.

#### 4. Medium of instruction and examinations

Medium of instruction and examination shall be in English.

#### 5. Working days in each semester

Each semester shall consist of not less than 100 working days. The odd semesters shall be conducted from the month of June/July to November/December and the even semesters shall be conducted from the month of December/January to May/June in every calendar year.

#### 6. Attendance and progress

A candidate is required to put in at least 80% attendance in individual courses considering theory and practical separately. The candidate shall complete the prescribed course satisfactorily to be eligible to appear for the respective examinations.

#### 7. Program/Course credit structure

As per the philosophy of Credit Based Semester System, certain quantum of academic work viz. theory classes, practical classes, seminars, assignments, etc. are measured in terms of credits. On satisfactory completion of the courses, a candidate earns credits. The amount of credit associated with a course is dependent upon the number of hours of instruction per week in that course. Similarly, the credit associated with any of the other academic, co/extra- curricular activities is dependent upon the quantum of work expected to be put in for each of these activities per week/per activity.

#### 8. Credit assignment

#### 8.1 Theory and Laboratory courses

Courses are broadly classified as Theory and Practical. Theory courses consist of lecture (L) and Practical (P) courses consist of hours spent in the laboratory. Credits (C) for a course is dependent on the number of hours of instruction per week in that course, and is obtained by using a multiplier of one (1) for lecture and a multiplier of half (1/2) for practical (laboratory) hours. Thus, for example, a theory course having four lectures per week throughout the semester carries a credit of 4. Similarly, a practical having four laboratory hours per week throughout semester carries a credit of 2.

The contact hours of seminars, assignments and research work shall be treated as that of practical courses for the purpose of calculating credits. i.e., the contact hours shall be multiplied by 1/2. Similarly, the contact hours of journal club, research work presentations and discussions with the supervisor shall be considered as theory course and multiplied by 1.

#### 8.2 Minimum credit requirements

The minimum credit points required for the award of M. Pharm degree is 95. However based on the credit points earned by the students under the head of co-curricular activities, a student shall earn a maximum of 100 credit points. These credits are divided into Theory courses, Practical, Seminars, Assignments, Research work, Discussions with the supervisor, Journal club and Co-Curricular activities over the duration of four semesters. The credits are distributed semester-wise as shown in Table 12. Courses generally progress in sequence, building competencies and their positioning indicates certain academic maturity on the part of the learners. Learners are expected to follow the semester-wise schedule of courses given in the syllabus.

#### 9. Academic work

A regular record of attendance both in Theory, Practical, Seminar, Assignment, and Journal club, Discussion with the supervisor, Research work presentation and Dissertation shall be maintained by the department / teaching staff of respective courses.

#### 10. Course of study

The specializations in M.Pharm program is given in Table 1.

Table – 1: List of M.Pharm. Specializations and their Code

S. No.	Specialization	Code
1.	Pharmaceutics	MPH
2.	Industrial Pharmacy	MIP
3.	Pharmaceutical Chemistry	MPC
4.	Pharmaceutical Analysis	MPA
5.	Pharmaceutical Quality Assurance	MQA
6.	Pharmaceutical Regulatory Affairs	MRA
7.	Pharmaceutical Biotechnology	MPB
8.	Pharmacy Practice	MPP
9.	Pharmacology	MPL
10.	Pharmacognosy	MPG

The course of study for M.Pharm specializations shall include Semester wise Theory & Practical as given in Table -2 to 11. The number of hours to be devoted to each theory and practical course in any semester shall not be less than that shown in Table -2 and 3.

Table – 2: Course of study for M. Pharm (Pharmaceutics)

Course Code	Course	Credit Hours	Credit Points	Hrs./wk.	Marks				
	Semester I								
MPH101T	Modern Pharmaceutical	4	4 4	4	100				
	Analytical Techniques	4	4	4	100				
MPH102T	Drug Delivery System	4	4	4	100				
MPH103T	Modern Pharmaceutics	4	4	4	100				
MPH104T	Regulatory Affair	4	4	4	100				
MPH105P	Pharmaceutics Practical I	12	6	12	150				
MPH106NA	Seminar/Assignment	7	4	7	100				
	Total	35	26	35	650				

Semester II					
	Molecular Pharmaceutics	4	4	4	100
MPH201T	(Nano Tech and Targeted DDS)	4	4	4	100
	Advanced Biopharmaceutics &	4	4	4	100
MPH202T	Pharmacokinetics	4			100
MPH203T	Computer Aided Drug Delivery System	4	4	4	100
MPH204T	Cosmetic and Cosmeceuticals	4	4	4	100
MPH205P	Pharmaceutics Practical II	12	6	12	150
MPH206NA	Seminar/Assignment	7	4	7	100
	Total	35	26	35	650

Table – 3: Course of study for M. Pharm. III Semester

Course Code	Course	Credit Hours	Credit Points
MRM 301T	Research Methodology and Biostatistics*	4	4
MRM302NA	Journal club	1	1
MRM303NA	Discussion / Presentation (Proposal Presentation)	2	2
MRM304NA	Research Work	28	14
	35	21	

^{*} Non University Exam

Table – 4: Course of study for M. Pharm. IV Semester

Course Code	Course	Credit Hours	Credit Points
MRM401NA	Journal Club	1	1
MRM402NA	Discussion/Final Presentation	3	3
MRM403NA	Research Work and Colloquium	31	16
MRM404NA	Scholarly Activity		3
	Total	35	23

Table – 5: Semester wise credits distribution

Semester	Credit Points
I	26
II	26
III	23
IV	23
Co-curricular Activities/Scholarly Activities	
(Attending Conference, Scientific Presentations and Scholarly	Minimum=04
Activities) Credit Points will be included in IV semester and 4	Willillium—04
credit point will be allocated for other certificate courses	
Total Credit Points	100

Table – 6: Guidelines for Awarding Credit Points for Co-curricular Activities

Name of the Activity	Maximum Credit Points Eligible / Activity
Participation in National Level Seminar/ Conference/ Workshop/	
Symposium/ Training Programs (related to the specialization of the student)	01
Participation in international Level Seminar/ Conference/ Workshop/	
Symposium/ Training Programs (related to the specialization of the student)	02
Academic Award/Research Award from State Level/National Agencies	01
Academic Award/Research Award from International Agencies	02
Research / Review Publication in National Journals (Indexed in Scopus / Web of Science)	01
Research / Review Publication in International Journals (Indexed in Scopus / Web of Science)	02

Note: International Conference: Held outside India International Journal: The Editorial Board outside India

#### 11. Program Committee

The M. Pharm. programme shall have a Programme Committee constituted by the Head of the institution in consultation with all the Heads of the departments.

The composition of the Programme Committee shall be as follows:

A teacher at the cadre of Professor shall be the Chairperson; One Teacher from each M.Pharm specialization and four student representatives (two from each academic year), nominated by the Head of the institution.

#### **Duties of the Programme Committee:**

- Periodically reviewing the progress of the classes.
- Discussing the problems concerning curriculum, syllabus and the conduct of classes.
- Discussing with the course teachers on the nature and scope of assessment for the course and the same shall be announced to the students at the beginning of respective semesters.
- Communicating its recommendation to the Head of the institution on academic matters.
- The Programme Committee shall meet at least twice in a semester preferably at the end of each sessional exam and before the end semester exam.
- Examinations/Assessments
- The schemes for internal assessment and end semester examinations are given in Table -8.

#### 12. End semester examinations

12.1. The End Semester Examinations for each theory and practical course through semesters I to IV shall be conducted by the respective university except for the subject with asterix symbol (*) in table I and II for which examinations shall be conducted by the subject experts at college level and the marks/grades shall be submitted to the university

^{*}The credit points assigned for extracurricular and or co-curricular activities shall be given by the Principals of the colleges and the same shall be submitted to the University. The criteria to acquire this credit point shall be defined by the colleges from time to time.

 $Table-7: Schemes \ for \ internal \ assessments \ and \ end \ semester$ 

(Pharmaceutics- MPH)

	Internal Assessment				End Semester		1	
Course	Course	Continuous	Sessional	Evame		E	xams	Total Marks
Code	Course	Mode	Marks	Duration	Total	Marks	Duration	with KS
		CIE.						
		SE.	MESTER	l .				
MPH 101T	Modern Pharmaceutical Analytical Techniques	10	15	1 Hr	25	75	3 Hrs	100
MPH 102T	Drug Delivery System	10	15	1 Hr	25	75	3 Hrs	100
MPH 103T	Modern Pharmaceutics	10	15	1 Hr	25	75	3 Hrs	100
MPH104T	Regulatory Affair	10	15	1 Hr	25	75	3 Hrs	100
MPH105P	Pharmaceutics Practical I	20	30	6 Hrs	50	100	6 Hrs	150
MPH106NA	Seminar/Assignment	-	-	-	-	-	-	100
		Tot	al					650
		SE	MESTER I	I				
MPH 201T	Molecular Pharmaceutics (Nano Tech and Targeted DDS)	10	15	1 Hr	25	75	3 Hrs	100
MPH 202T	Advanced Biopharmaceutics & Pharmacokinetics	10	15	1 Hr	25	75	3 Hrs	100
MPH 203T	Computer Aided Drug Delivery System	10	15	1 Hr	25	75	3 Hrs	100
MPH204T	Cosmetic and Cosmeceuticals	10	15	1 Hr	25	75	3 Hrs	100
MPH 205P	Pharmaceutics Practical II	20	30	6 Hrs	50	100	6 Hrs	150
MPH206NA	Seminar/Assignment	-	-	-	-	-	-	100
Total					650			

Table – 8: Schemes for internal assessments and end semester examinations (Semester III& IV)

	Internal Assessment			End Semester Exams		Total		
Course Code	Course	Continuous Mode		al Exams	Total	Marks	Duration	Marks
			Marks	Duration				
		SEMEST	TER III					
MRM30 1T	Research Methodology							
	and	1.0	1.5	1 11.	25	75	2 11	100
	Biostatistics*	10	15	1 Hr	25	75	3 Hrs	100
MRM302NA	Journal club	-	-	-	25	-	-	25
MRM303NA	Discussion /							
	Presentation (Proposal				50			50
	Presentation)	-	-	-	50	-	-	50
MRM304NA	Research work*	-	-	-	-	350	1 Hr	350
		Total	•				1	525
		SEMEST	TER IV					
MRM401NA	Journal club	-	-	-	25	-	-	25
MRM402NA	Discussion / Final					7.5		7.5
	Presentation	_	_	-	-	75		75
NADNAAOONIA	Research work and					400		400
MRM403NA	Colloquium	-	_	-	-	400	-	400
MRM404NA	Scholarly Activity	-	-	-	-	175	-	175
Total					675			

^{*}Non University Examination

#### 12.2. Internal assessment: Continuous mode

The marks allocated for Continuous mode of Internal Assessment shall be awarded as per the scheme given below.

#### **Sessional Exams**

Two sessional exams shall be conducted for each theory / practical course as per the schedule fixed by the college(s). The scheme of question paper for theory and practical sessional examinations is given in the table. The average marks of two sectional exams shall be computed for internal assessment as per the requirements given in tables.

Table – 9: Scheme for awarding internal assessment: Continuous mode

Theory	
Criteria	Maximum Marks
Attendance (Refer Table – 28)	8
Student – Teacher interaction	2
Total	10
Practical	
Criteria	Maximum Marks
Attendance (Refer Table – 28	10
Based on Practical Records, Regular viva voce, etc.	10
Total	20

Table – 10: Guidelines for the allotment of marks for attendance

Percentage of Attendance	Theory	Practical
95 – 100	8	10
90 – 94	6	7.5
85 – 89	4	5
80 - 84	2	2.5
Less than 80	0	0

#### 13. Promotion and award of grades

A student shall be declared PASS and eligible for getting grade in a course of M.Pharm. Programme if he/she secures at least 50% marks in that particular course including internal assessment.

#### 13. Carry forward of marks

In case a student fails to secure the minimum 50% in any Theory or Practical course as specified in 12, then he/she shall reappear for the end semester examination of that course. However, his/her marks of the Internal Assessment shall be carried over and he/she shall be entitled for grade obtained by him/her on passing.

#### 14. Improvement of internal assessment

A student shall have the opportunity to improve his/her performance only once in the sectional exam component of the internal assessment. The re-conduct of the sectional exam shall be completed before the commencement of next end semester theory examinations.

#### 15. Re-examination of end semester examinations

Re-examination of end semester examination shall be conducted as per the schedule given in table 29. The exact dates of examinations shall be notified from time to time.

Table – 11: Tentative schedule of end semester examinations

Semester	For Regular Candidates	For Failed Candidates
I and III	November / December	May / June
II and IV	May / June	November / December

#### 16. Allowed to keep terms (ATKT):

No student shall be admitted to any examination unless he/she fulfils the norms given in 6. ATKT rules are applicable as follows:

A student shall be eligible to carry forward all the courses of I and II semesters till the III semester examinations. However, he/she shall not be eligible to attend the courses of IV semester until all the courses of I, II and III semesters are successfully completed.

A student shall be eligible to get his/her CGPA upon successful completion of the courses of I to IV semesters within the stipulated time period as per the norms.

Note: Grade AB should be considered as failed and treated as one head for deciding ATKT. Such rules are also applicable for those students who fail to register for examination(s) of any course in any semester.

#### 17. Grading of performances

Letter grades and grade points allocations:

Based on the performances, each student shall be awarded a final letter grade at the end of the semester for each course. The letter grades and their corresponding grade points are given in Table -12.

Table – 12: Letter grades and grade points equivalent to Percentage of marks and performances

Percentage of Marks Obtained	Letter Grade	Grade Point	Performance
90.00 - 100	О	10	Outstanding
80.00 – 89.99	A	9	Excellent
70.00 – 79.99	В	8	Good
60.00 - 69.99	С	7	Fair
50.00 - 59.99	D	6	Average
Less than 50	F	0	Fail
Absent	AB	0	Fail

A learner who remains absent for any end semester examination shall be assigned a letter grade of AB and a corresponding grade point of zero. He/she should reappear for the said evaluation/examination in due course.

The Semester grade point average (SGPA)

The performance of a student in a semester is indicated by a number called 'Semester Grade Point Average' (SGPA). The SGPA is the weighted average of the grade points obtained in all the courses by the student during the semester. For example, if a student takes five courses (Theory/Practical) in a semester with credits C1, C2, C3 and C4 and the student's grade points in these courses are G1, G2, G3 and G4, respectively, and then students' SGPA is equal to:

SGPA = 
$$\frac{\text{C1G1} + \text{C2G2} + \text{C3G3} + \text{C4G4}}{\text{C1} + \text{C2} + \text{C3} + \text{C4}}$$

The SGPA is calculated to two decimal points. It should be noted that, the SGPA for any semester shall take into consideration the F and ABS grade awarded in that semester. For example if a learner has a F or ABS grade in course 4, the SGPA shall then be computed as:

SGPA = 
$$\frac{\text{C1G1} + \text{C2G2} + \text{C3G3} + \text{C4* ZERO}}{\text{C1} + \text{C2} + \text{C3} + \text{C4}}$$

Cumulative Grade Point Average (CGPA)

The CGPA is calculated with the SGPA of all the IV semesters to two decimal points and is indicated in final grade report card/final transcript showing the grades of all IV semesters and their courses. The CGPA shall reflect the failed status in case of F grade(s), till the course(s) is/are passed. When the course(s) is/are passed by obtaining a pass grade on subsequent examination(s) the CGPA shall only reflect the new grade and not the fail grades earned earlier. The CGPA is calculated as:

$$CGPA = \frac{C1S1 + C2S2 + C3S3 + C4S4}{C1 + C2 + C3 + C4}$$

where C1, C2, C3,.... is the total number of credits for semester I, II,III,.... and S1, S2, S3,....is the SGPA of semester I,II,III,.....

#### Declaration of class

The class shall be awarded on the basis of CGPA as follows: First Class with Distinction = CGPA of. 7.50 and above

First Class =  $CGPA ext{ of } 6.00 ext{ to } 7.49$ 

Second Class =  $CGPA ext{ of } 5.00 ext{ to } 5.99$ 

Project work

All the students shall undertake a project under the supervision of a teacher in Semester III to IV and submit a report. 4 copies of the project report shall be submitted (typed & bound copy not less than 75 pages).

The internal and external examiner appointed by the University shall evaluate the project at the time of the Practical examinations of other semester(s). The projects shall be evaluated as per the criteria given below.

#### **Evaluation of Dissertation Book:**

Objective(s) of the work done	50 Marks
Methodology adopted	150 Marks
Results and Discussions	250 Marks
Conclusions and Outcomes	50 Marks

Total 500 Marks

#### **Evaluation of Presentation:**

Presentation of work 100 Marks
Communication skills 50 Marks
Question and answer skills 100 Marks
Total 250 Marks

#### **Award of Ranks**

Ranks and Medals shall be awarded on the basis of final CGPA. However, candidates who fail in one or more courses during the M.Pharm program shall not be eligible for award of ranks. Moreover, the candidates should have completed the M. Pharm program in minimum prescribed number of years, (two years) for the award of Ranks.

#### Award of degree

Candidates who fulfil the requirements mentioned above shall be eligible for award of degree during the ensuing convocation.

#### **Duration for completion of the program of study**

The duration for the completion of the program shall be fixed as double the actual duration of the program and the students have to pass within the said period, otherwise they have to get fresh Registration.

#### Revaluation I retotalling of answer papers

There is no provision for revaluation of the answer papers in any examination. However, the candidates can apply for retotalling by paying prescribed fee.

Re-admission after break of study

Candidate who seeks re-admission to the program after break of study has to get the approval from the university by paying a condonation fee.

			SEMESTE								
			cal Analytical Techniq	lues							
Course	code	MPH101T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 60T	Γ <b>4</b>	0	0	0	0	0	4	
Pre-req			Co-requisite					Nil			
Progr		Master of Pharmacy (									
	Semester I semester of first year of the program										
Cou			ourse student is able to	o know,							
Objec	tives	Chemicals and Excipients.									
			is drugs in single and o		tion	dosage	form	IS.			
Theoretical and practical skills of the instruments.  CO1 Compare and Utilize the Spectroscopy knowledge to Interpret various levels											
CO		_	the Spectroscopy kno	owledge	to	Interpre	et var	ious le	evels of m	olecular	
~~		spectra's.		15.37		<b>AD.</b> II	<b></b> 13		1100	. n. m	
CO			ion of N.M.R, compar	re 1D N	MR,	2DNM	1R, 1	HNMI	R and 13C.	NMR to	
~~		Propose structures.			-			41.0		11.00	
CO			and Importance, theo	•		•	•	•	ferentiate of	different	
~~		•	and Fragmentation rule								
CO		_	nromatographic techni	•			ınstrı	umenta	ation, choo	se each	
		_	mple and Develop a V				11	1	1 1 .	*,1	
CO		-	ectrophoretic technique				_			gn with	
<b>T</b> T •/		Importance of Electro	phoresis, X-Ray Cryst			and Bio	lumır	nescen	ce assays.		
Unit- No.		Conter	nt	Contac Hour	τ	L	earni	ng Ou	tcome	KL	
I I	HV.	Visible spectroscopy:	Introduction	Hour	9	tudents	xvi11 1	he able	e to know		
		ry, Laws, Instrumenta	· ·						behind		
		UV-Visible spectrosco				•			echnique.		
		ents and solvent effect				ion spe	CHOBC	opic t	comique.		
		V-Visible spectroscop	* *								
		pectroscopy: Theory,									
	_	cular vibrations, Sam									
		umentation of Dispers									
		sform IR Spectromete									
	affec	ting vibrational freque	encies and	11						2,	
	Appl	ications of IR spectros	scopy	11						3,4,5	
	Spec	troflourimetry: Theo	ry of								
	Fluo	rescence, Factors affect	eting								
	fluor	escence, Quenchers, In	nstrumentation								
	and A	Applications of fluores	scence								
	spect	rophotometer.									
		ne emission spectrosc									
		rption spectroscopy:	-								
		umentation, Interferen	ces and								
		cations.									
II		R spectroscopy: Quan							e to know		
		role in NMR, Principl				asic cor	_		•		
		ent requirement in NM						_	dict NMR		
	_	ess, NMR signals in va	•	11	_	ectra b		-		3,4,5	
		nical shift, Factors inf	•		O	rganic o	compo	ounds.			
		Spin-Spin coupling, (	1 0								
		ear magnetic double re									
	outli	ne of principles of FT-	INIMK and								

	13CNMR. Applications of NMR			
	spectroscopy.			
III	Mass Spectroscopy: Principle, Theory, Instrumentation of Mass Spectroscopy, Different types of ionization like electron impact, n chemical, field, FAB and MALDI, APCI, ESI, APPI Analysers of Quadrupole and Time of Flight, Mass fragmentation and its rules, Meta stable ions, Isotopic peaks and Applications of Mass spectroscopy	11	Students will be able to learn identify which Ionization technique is suitable for compounds and analyze the type of Analyzer to be used depending on the sample.	3,4,5
IV	Chromatography: Principle, apparatus, instrumentation, chromatographic parameters, factors affecting resolution and applications of the following:  a) Paper chromatography b) Thin Layer chromatography c) Ion exchange chromatography d)Column chromatography e) Gas chromatography f) High Performance Liquid chromatography g) Affinity chromatography	11	Students will be able to understand theory & application of different chromatographic techniques.	3,4,5
V	a. Electrophoresis: Principle, Instrumentation, working conditions, factors affecting separation and applications of the following: a) Paper electrophoresis b) Gel electrophoresis c) Capillary electrophoresis d) Zone electrophoresis e) Moving boundary electrophoresis f) Iso electric focusing b. X ray Crystallography: Production of X rays, Different X ray diffraction methods, Bragg's law, Rotating crystal technique, X ray powder technique, Types of crystals and applications of X ray diffraction.	11	Students will be able to Understand, Principle, Instrumentation, working conditions, factors affecting separation and applications of electrophoresis.	3,4,5
VI	Immunological assays: RIA (Radio immunoassay), ELISA, Bioluminescence assays.	5	Students will be able to know importance of RIA and ELISA techniques and their applications.	

T1: Elementary Organic Spectroscopy by Y. R Sharma.

T2: Chromatography by P.D Sethi.

#### **REFERENCE BOOKS:**

R1. Spectrometric Identification of Organic compounds- Robert M Silverstein, Sixth edition, John Wiley & Sons, 2004.

- R2. Principles of Instrumental Analysis- Doglas ASkoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
- R3. Instrumental methods of analysis-Willards, 7th edition, CBS publishers.
- R4. Practical Pharmaceutical Chemistry—Beckett and Stenlake, Vol II, 4th edition, CBS Publishers, New Delhi, 1997.
- R5. Organic Spectroscopy- William Kemp, 3rd edition, ELBS, 1991.
- R6. Quantitative Analysis of Drugs in Pharmaceutical formulation- P D Sethi, 3rd Edition, CBS Publishers, New Delhi, 1997.
- R7. Pharmaceutical Analysis- Modern methods- Part B- J W Munson, Volume 11, Marcel Dekker Series.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Compare and Utilize the Spectroscopy knowledge to Interpret various levels of molecular spectra.	PO1,PO2,PO3,PO4,PO5,PO8					
2	Analyze instrumentation of N.M.R, compare 1D NMR, 2DNMR, 1HNMR and 13CNMR to Propose structures.	PO1,PO2,PO3,PO4,PO5,PO8					
3	Assume the principle and Importance, theory of Mass, spectroscopy, differentiate different peaks and ionization and Fragmentation rules to Predict the structure.	PO1,PO2,PO3,PO4,PO5,PO8					
4	Compare different Chromatographic techniques and Evaluate instrumentation, choose each technique based on sample and Develop a Validated method.	PO1,PO2,PO3,PO4,PO5,PO8					
5	Justify different electrophoretic techniques, X-ray crystallography and design with Importance of Electrophoresis, X-Ray Crystallography and Bioluminescence assays.	PO1,PO2,PO3,PO4,PO5,PO8					

		SEN	MESTER –	I						
<b>Course Title</b>			Drug Deliv	ery Sys	tems					
Course code	MPH102T	Total credit		L	T	P	S	R	O/F	C
		Total hours:		4	0	0	0	0	0	4
Pre-requisite			equisite				N	[il		
Program		acy (Pharmaceutic								
Semester		t year of the progra								
Course		of the course, stu								
Objectives		roaches for develop		_				·		
		election of drugs a				_	or dei	ivering	g syste	m.
CO1		ne formulation and evaluation of Novel drug delivery systems.						and		
COI	_	nderstand principles and mechanisms of SR, CR, and Rate-Controlled Drug Systems teria for selection of drugs and polymers.						anu		
CO2		drug delivery appro	<u> </u>	ble for	specific	natie	nt cate	gories	and an	nlv
CO2	personalized med		Jacines suita	010 101	specific	pane	iii caic	gories	and ap	pry
CO3	•	te, and evaluate m	uco - adhes	sive Bu	ccal. O	cular.	and Ti	ransder	mal D	านฐ
	Delivery System				, ,	,				100
CO4		address the unique	challenges	in Pro	tein, Pe	eptide.	and V	Vaccin	e deliv	ery
	_	zing their formula	_			•				•
CO5	Analyze the cost	e-effectiveness, the	rapeutic effi	cacy, a	nd patie	ent-cer	tric be	nefits	of vari	ous
	novel drug delive	ery systems.								
Unit-No.	C	ontent	Contact		Lea	rning	Outco	me		KL
Omt-140.		Jittent	Hour							IXL
I	Sustained Rele	` '			ents wil					
	Controlled Rel			1	t drug d			ms and	1	
	formulations:				role in			.1		
	basic concepts,			1 ^	naceuti		•			
	disadvantages,	ysicochemical &		deliv	ct of va	rious i	actors	on aru	g	
	biological appro			denv	ery.					
		tion, Mechanism								
	of Drug Deliver									
	formulation. Po	•								
	introduction, de									
	classification, p		10							2,3,4
	application Dos	sage Forms for								
	Personalized M	edicine:								
	Introduction, D	efinition,								
	Pharmacogenet									
	Categories of P									
	Personalized M									
	Customized dru	-								
	systems, Bioele									
	Medicines, 3D	-								
	pharmaceutical	s, 1 ele								
II	pharmacy.  Rate Controlle	ed Drug Delivery		Stude	ents wil	l he ak	ole to C	Omnar	·e	
11	Systems: Prince				contrast			_		
	Fundamentals,	-	10		ery rout			_		2,3,4
	Activation; Mo	• •			bility of					, <del>-</del> , •
		ns; Mechanically			ery rout	-		-	gs.	

	activated, pH activated, Enzyme activated, and Osmotic activated Drug Delivery Systems Feedback regulated Drug Delivery Systems; Principles & Fundamentals.			
III	Gastro-Retentive Drug Delivery Systems: Principle, concepts advantages and disadvantages, Modulation of GI transit time approaches to extend GI transit. Buccal Drug Delivery Systems: Principle of muco adhesion, advantages and disadvantages, Mechanism of drug permeation, Methods of formulation and its evaluations.	10	Students will be able explain the mechanisms of controlled and sustained release drug delivery. And able to analyze different formulation approaches for controlled release.	3,4,5
IV	Ocular Drug Delivery Systems: Barriers of drug permeation, Methods to overcome barriers.	6	Students will be able to learn to Compare and contrast different drug delivery routes and evaluate the suitability of specific drug delivery routes for different drugs.	2,3,4
V	Transdermal Drug Delivery Systems: Structure of skin and barriers, Penetration enhancers, Transdermal Drug Delivery Systems, Formulation and evaluation.	10	Students will be able to evaluate the role of nanoparticles, liposomes, and micelles in targeted drug delivery.	2,3,4
VI	Protein and Peptide Delivery: Barriers for protein delivery. Formulation and Evaluation of delivery systems of proteins and other macromolecules.	8	Students will be able to know about the latest developments in drug delivery research. & critically evaluate the Potential impact of new technologies in drug delivery.	2,3,4,5
VII	Vaccine delivery systems: Vaccines, uptake of antigens, single shot vaccines, mucosal and transdermal delivery of vaccines.	6	Students will be able to Identify future directions and challenges for drug delivery system design.	2,3,4

- T1. Y W. Chien, Novel Drug Delivery Systems, 2nd edition, revised and expanded, Marcel Dekker, Inc., New York, 1992.
- T2. Robinson, J. R., Lee V. H. L, Controlled Drug Delivery Systems, Marcel Dekker, Inc., New York, 1992.
- T3. N.K. Jain, Controlled and Novel Drug Delivery, CBS Publishers & Distributors, New Delhi, First edition 1997 (reprint in 2001).
- T4. S.P.Vyas and R.K.Khar, Controlled Drug Delivery concepts and advances, VallabhPrakashan, New Delhi, First edition 2002)

#### **REFERENCE BOOKS:**

- R1. Y W. Chien, Novel Drug Delivery Systems, 2nd edition, revised and expanded, Marcel Dekker, Inc., New York, 1992.
- R2. Robinson, J. R., Lee V. H. L, Controlled Drug Delivery Systems, Marcel Dekker, Inc., New York, 1992.
- R3. Encyclopaedia of controlled delivery, Editor- Edith Mathiowitz, Published by WileyInterscience Publication, John Wiley and Sons, Inc, New York! Chichester/Wenham
- R4. N.K.Jain, Controlled and Novel Drug Delivery, CBS Publishers & Distributors, New Delhi, First edition 1997 (reprint in 2001).
- R5. S.P.Vyas and R.K.Khar, Controlled Drug Delivery- concepts and advances, VallabhPrakashan, New Delhi, First edition 2002.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Understand principles and mechanisms of SR, CR, and Rate- Controlled Drug Systems and criteria for selection of drugs and polymers.	PO1,PO3,PO4,PO5,PO6,PO8					
2	Design tailored drug delivery approaches suitable for specific patient categories and apply personalized medicine concepts.	PO1,PO3,PO4,PO5,PO6,PO8					
3	Design, formulate, and evaluate muco-adhesive Buccal, Ocular, and Transdermal Drug Delivery Systems.	PO1,PO3,PO4,PO5,PO6,PO8					
4	Recognize and address the unique challenges in Protein, Peptide, and Vaccine delivery systems emphasizing their formulations and evaluations	PO1,PO3,PO4,PO5,PO6,PO8					
5	Analyze the cost-effectiveness, therapeutic efficacy, and patient-centric benefits of various novel drug delivery systems.	PO1,PO3,PO4,PO5,PO6,PO8					

			SEME	STER	- I						
Course T	itle	Modern Pharmaceut	ics								
Course co	ode	MPH103T	Total credits:	4 1	L	T	P	S	R	O/F	C
			Total hours: 6	60T 4	4	0	0	0	0	0	4
Pre-requi	site	Nil	Co-requisite	Nil							
Program	n	Master of Pharmacy	(Pharmaceutics)								
Semeste	er	I semester of first ye	ear of the program								
Course	•	Upon completion of			e able	e to ui	ndersta	and			
Objectiv	es	The elements of refo									
		The Active Pharmac	~			drug	Produ	ct dev	elopm	ent	
		Industrial Managem									
		Optimization Techn	-		_	-					
601		Stability Testing, ste	•	_							1 *1*.
CO1		Recall and identify t		ncepts 1	elated	d to di	rug ex	cıpıen	t intera	actions, st	ability
CO1		testing, and theories		1.1:	£ 1	l	4 : .	.11	: 4 - 4 :		
CO2		Comprehend the sco									
CO3		Apply cGMP princip production planning	-	рпансе,	1114116	ige m	aiCI läl	s c116	cuvery	, and mip	CHICHI
CO4		Analyze the impact		rameter	s on to	ahlet 1	roner	ties ar	nd disse	olution	
CO5		Analyze the signific									lts in
		pharmaceutical research			, 500001				-P		111
Unit-No		Conte			ıtact		Lea	rning	Outc	ome	KL
				Н	our						
I	a. F	Preformation Concep	ts- Drug			Stuc	lents v	vill be	able to	o learn	
	Exc	cipient interactions d	ifferent				_	_		tions -	
		thods, kinetics of stal								etics of	
		ing. Theories of disp				stab	ility, S	Stabili	ty testi	ng.	
		rmaceutical Dispersi									
		Suspension, SMED									
		stability large and s									
	_	ental physiological									
		sideration, Manufacturation.	turing and	1	0						1245
		nuation. Optimization techniq	ues in	2	U						1,2,4,5
		armaceutical Formula									
		parameters of optim	-								
		timization techniques									
	_	rmaceutical formula									
	_	cessing. Statistical d									
	sur	face method, Contou	r designs,								
	Fac	torial designs and ap	plication in								
	for	nulation									
II		lidation: Introductio							able to		
		rmaceutical Validati	_					_	rmace		
		rits of Validation, Va								nerits of	
		bration of Master pla		_	0				dation		2245
		HO guidelines for cal		1	0				aster p		2,3,4,5
		idation of equipment						_	uidelin	es ior	
	_	cific dosage form, T	-			can	bration	1.			
		idation. Government nufacturing Process	~								
	ıvıa	nutacturing Process	wiouci, UKS,								

	DQ, IQ, OQ &P.Q. of facilities.			
III	cGMP& Industrial Management: Objectives and policies of current good manufacturing practices, layout of buildings, services, equipment's and their maintenance Production management: Production organization, materials management, handling and transportation, inventory management and control, production and planning control, Sales forecasting, budget and cost control, industrial and personal relationship. Concept of Total Quality Management.	10	Students will be able to learn about Production organization, materials management, handling and transportation, inventory management and control, production and planning control, Sales forecasting, budget and cost control, industrial and personal relationship.	2,3,4,5
IV	Compression and compaction: Physics of tablet compression, compression, consolidation, effect of friction, distribution of forces, compaction profiles. Solubility.	10	Students will be able to understand the Physics of tablet compression, compression, compression, consolidation, effect of friction, distribution of forces, compaction profiles. Solubility.	2,3,4,5
V	Study of consolidation parameters; Diffusion parameters, Dissolution parameters and Pharmacokinetic parameters, Heckel plots, Similarity factors—f2 and f1, Higuchi and Peppas plot, Linearity Concept of significance, Standard deviation, Chi square test, students T-test, ANOVA test.	10	Students will be able to learn Diffusion parameters, Dissolution parameters and Pharmacokinetic parameters, Heckel plots, Similarity factors – f2 and f1, Higuchi and Peppas plot, Linearity Concept of significance, Standard deviation, Chi square test, students T-test, ANOVA test.	2,3,4,5

T1. Dr. Ashok A. Hajare; Modern Pharmaceutics; Nirali Prakashan; 1st edition, 2023.

#### REFERENCE BOOKS:

- R1. Theory and Practice of Industrial Pharmacy ByLachmann and Liebermann
- R2. Pharmaceutical dosage forms: Tablets Vol. 1-3 by Leon Lachmann.
- R3. Pharmaceutical Dosage forms: Disperse systems, Vol, 1-2; By Leon Lachmann.
- R4. Pharmaceutical Dosage forms: Parenteral medications Vol. 1-2; By Leon Lachmann.
- R5. Modern Pharmaceutics; By Gilbert and S. Banker.
- R6. Remington's Pharmaceutical Sciences.
- R7. Advances in Pharmaceutical Sciences Vol. 1-5; By H.S. Bean & A.H. Beckett.
- R8. Physical Pharmacy; By Alfred Martin
- R9. Bentley's Textbook of Pharmaceutics—by Rawlins.
- R10.Goodmanufacturing practices for Pharmaceuticals: A plan for total quality control, Second edition; By Sidney H. Willing.
- R11. Quality Assurance Guide; By Organization of Pharmaceutical producers of India.
- R12. Drug formulation manual; By D.P.S. Kohl and D.H.Shah.Eastern publishers, New Delhi.

- R13. How to practice GMPs; By P.P.Sharma. Vandhana Publications, Agra.
- R14. Pharmaceutical Process Validation; By Fra. R. Berry and Robert A. Nash.
- R15. Pharmaceutical Preformulations; By J.J. Wells.
- R16. Applied production and operations management; By Evans, Anderson, Sweeneyand Williams.
- R17. Encyclopaedia of Pharmaceutical technology, Vole I– III.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Recall and identify the fundamental concepts related to drug excipient interactions, stability testing, and theories of dispersion	PO1,PO3,PO4,PO5,PO6,PO8
2	Comprehend the scope, merits, and guidelines for pharmaceutical validation and calibration	PO1,PO3,PO4,PO5,PO6,PO8
3	Apply cGMP principles to ensure compliance, manage materials effectively, and implement production planning and control	PO1,PO3,PO4,PO5,PO6,PO8
4	Analyze the impact of compression parameters on tablet properties and dissolution.	PO1,PO3,PO4,PO5,PO6,PO8
5	Analyze the significance of data interpretation, statistical tests, and experimental results in pharmaceutical research and development.	PO1,PO3,PO4,PO5,PO6,PO8

Course		Regulatory Affairs								
Course	e code	MPH104T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 60T		0	0	0	0	0	4
	quisite	Nil	Co-requisite	Nil						
Prog		Master of Pharmacy	,							
Semo		I semester of first ye	1 0							
Cou			the course, it is expec							stand
Objec	ctives	_	of innovator and gene	_		-	_	_		
			y guidance's and guid			•		-		
		•	Dossiers and their sub			_			n differer	it countri
			regulatory requiremen				ag pro	oducts		
			global documents in (				alini	ool triol	C.	
<ul><li>6. Clinical trials requirements for approvals for conducting clinical trial</li><li>7. Pharmacovigilance and process of monitoring in clinical trials</li></ul>							8			
CO	<b>71</b>		cepts of innovator and		-				een the d	ฑาด
	<i>9</i> 1		ses, and summarize th	-	_					rug
		• •	latory guidelines and d		• •			* *		ssions
CO	02	_	es for new drugs, gene	_	_					5510115
			ses, documentation, an	_		_			-	clinical
CO	<b>U3</b>	_	development to safety	_		1			8	
		-	d process related docum			tify crit	ical c	quality a	attributes	and
CO	<b>O4</b>	prepare master batch	h records, Drug Master	r Files, t	echni	cal rep	orts,	and oth	er CMC	
		regulatory documen	ts.							
CO	75	Apply ICH guidelin	es, CTD, and regional	regulato	ory re	quirem	ents t	o prepa	are a com	plete
		regulatory submission	on dossier for drug app			JS, EU	, and	other g	global ma	rkets.
Unit- No.		Conten	nt (	Contact	t	Lea	rnin	g Outco	ome	KL
NO.	a Doc	umentation in Pharma	acoutical industry:	Hour	Stu	dente v	vi11 h	e able to	o learn	
		formula record, DM	•			cument			o icarii	
		listribution records. C	` -					industi	rv &.	
	,		- I		1				-	
		product development Introduction, Hatch Regulatory requirement for								
1	Waxman act and amendments, CFR (CODE product approval						_	al		
		•	,				_	al		
	OF FE	an act and amendmer	ION), drug product				_	al		
T	OF FE	an act and amendmen DERAL REGULAT	ION), drug product OA regulatory	12			_	al		2245
I	OF FE perforr approv	an act and amendmer DERAL REGULATI nance, in-vitro, AND	ION), drug product DA regulatory proval process, BE	12			_	al		2,3,4,5
I	OF FE perform approversion	an act and amendmer DERAL REGULATI nance, in-vitro, AND al process, NDA app	ION), drug product OA regulatory Proval process, BE at, in–vivo, scale up	12			_	al		2,3,4,5
I	OF FE perform approve and druprocess surveil	an act and amendmer DERAL REGULATI mance, in-vitro, AND all process, NDA appug product assessments approval changes, plance, outsourcing B.	ION), drug product OA regulatory proval process, BE at, in–vivo, scale up post marketing A and BE to CRO.	12			_	al		2,3,4,5
I	OF FE perform approver and druprocess surveil b. Reg	an act and amendmen DERAL REGULATI mance, in-vitro, AND al process, NDA app ag product assessments approval changes, plance, outsourcing Bulatory requirement f	ION), drug product OA regulatory Proval process, BE Int, in–vivo, scale up Post marketing A and BE to CRO. For product	12			_	al		2,3,4,5
I	OF FE perform approve and druprocess surveil b. Regular approv	an act and amendment DERAL REGULATION ance, in-vitro, AND all process, NDA apputed product assessments approval changes, plance, outsourcing Bullatory requirement fal: API, biologics, no	ION), drug product OA regulatory Proval process, BE Int, in–vivo, scale up Post marketing A and BE to CRO. For product Evel, therapies	12			_	al		2,3,4,5
I	OF FE perform approve and dru process surveil b. Regulapprove obtaini	an act and amendment DERAL REGULATION of the nance, in-vitro, AND all process, NDA apputes product assessments approval changes, plance, outsourcing Bulatory requirement fal: API, biologics, noting NDA, ANDA for	ION), drug product OA regulatory Proval process, BE Int, in–vivo, scale up Prost marketing A and BE to CRO. For product Provel, therapies I generic drugs	12			_	al		2,3,4,5
I	OF FE perform approve and druprocess surveil b. Regular approve obtaining ways a	an act and amendment DERAL REGULATION ance, in-vitro, AND all process, NDA apputed product assessments approval changes, plance, outsourcing Bullatory requirement fal: API, biologics, no	ION), drug product OA regulatory Proval process, BE Int, in–vivo, scale up Prost marketing A and BE to CRO. For product Provel, therapies I generic drugs	12			_	al		2,3,4,5
I	OF FE perform approve and dru process surveil b. Regulapprove obtaining ways a drugs	an act and amendment DERAL REGULATION ance, in-vitro, AND all process, NDA apputes product assessments approval changes, plance, outsourcing Bulatory requirement fal: API, biologics, noting NDA, ANDA for and means of US register.	ION), drug product OA regulatory proval process, BE at, in–vivo, scale up post marketing A and BE to CRO. For product ovel, therapies generic drugs stration for foreign	12	pro	duct ap	prov			2,3,4,5
I	OF FE perform approve and dru process surveil b. Regular approve obtaining ways a drugs CMC,	an act and amendment DERAL REGULATION ance, in-vitro, AND all process, NDA apputed product assessments approval changes, plance, outsourcing Bulatory requirement fal: API, biologics, not not means of US registrost approval regulations.	ION), drug product OA regulatory proval process, BE at, in–vivo, scale up post marketing A and BE to CRO. Cor product povel, therapies generic drugs stration for foreign  tory affairs.	12	pro	duct ap	vill bo	e able to	0	2,3,4,5
I	OF FE perform approve and dru process surveil b. Regulary approve obtaining ways a drugs CMC, Regulary	an act and amendment DERAL REGULATION and process, NDA apputed product assessments approval changes, plance, outsourcing Bulatory requirement for al: API, biologics, not not means of US registration for combination	ION), drug product DA regulatory proval process, BE at, in–vivo, scale up post marketing A and BE to CRO. For product ovel, therapies generic drugs stration for foreign  tory affairs. products and	12	Stu-	duct ap	vill bo	e able to	o t	2,3,4,5
п	of FE perform approve and dru process surveil b. Regular approve obtaining ways a drugs CMC, Regular medicare	an act and amendment DERAL REGULATION and process, NDA apputed product assessments approval changes, plance, outsourcing Bulatory requirement fal: API, biologics, nor nd means of US registropost approval regulation for combinational devices' and ECTE	ION), drug product OA regulatory Proval process, BE Int, in—vivo, scale up Prost marketing OA and BE to CRO. For product Ovel, therapies I generic drugs I stration for foreign  I tory affairs. I products and O format, industry	12	Stu-	duct ap	vill bo	e able to	o t	2,3,4,5
	of FE perform approve and dru process surveil b. Regular approve obtains ways a drugs CMC, Regular medical and FE	an act and amendment DERAL REGULATION ance, in-vitro, AND all process, NDA apparent product assessments approval changes, plance, outsourcing Bulatory requirement for al: API, biologics, not not means of US registropy approval regulation for combinational devices' and ECTE DA liaison. ICH- Guident and DERAL REGULATION AND A GOVERNOUS AND A GOVERNOU	ION), drug product OA regulatory proval process, BE at, in–vivo, scale up post marketing A and BE to CRO. For product evel, therapies generic drugs stration for foreign  tory affairs. products and O format, industry delines of ICH-Q, S		Stu-	duct ap	vill bo	e able to	o t	
	of FE perform approve and drup process surveil b. Regular approve obtaining ways a drugs CMC, Regular medical and FE E, M. I	an act and amendment DERAL REGULATION and process, NDA apputed product assessments approval changes, plance, outsourcing Builtory requirement for all API, biologics, not not means of US registropy approval regulation for combinational devices' and ECTE DA liaison. ICH- Guickley and approved requirement of the second process approval regulation for combinational devices and ECTE DA liaison. ICH- Guickley and ECTE DA liaison.	ION), drug product OA regulatory proval process, BE at, in–vivo, scale up post marketing A and BE to CRO. For product evel, therapies generic drugs stration for foreign  tory affairs. products and O format, industry delines of ICH-Q, S		Stu-	duct ap	vill bo	e able to	o t	
II	OF FE perform approve and drup process surveil b. Regularing approve obtaining ways a drugs CMC, Regular medical and FE E, M. ITGA a	an act and amendment DERAL REGULATION and process, NDA applying product assessments approval changes, plance, outsourcing Bulatory requirement for al: API, biologics, not may NDA, ANDA for and means of US registropist approval regulation for combinational devices' and ECTE DA liaison. ICH- Guick Regulatory requirement and ROW countries.	ION), drug product OA regulatory proval process, BE at, in—vivo, scale up post marketing A and BE to CRO. For product ovel, therapies generic drugs stration for foreign  tory affairs. products and O format, industry delines of ICH-Q, S ents of EU, MHRA,	12	Stu- Und app	dents v derstan roval r	vill bod CM	e able to IC, pos tory afi	o t	2,3,4,5
	OF FE perform approve and dru process surveil b. Regular approve obtains ways a drugs CMC, Regular medical and FE E, M. ITGA a Non cl	an act and amendment DERAL REGULATION and process, NDA apputed product assessments approval changes, plance, outsourcing Builtory requirement for all API, biologics, not not means of US registropy approval regulation for combinational devices' and ECTE DA liaison. ICH- Guickley and approved requirement of the second process approval regulation for combinational devices and ECTE DA liaison. ICH- Guickley and ECTE DA liaison.	ION), drug product OA regulatory proval process, BE at, in—vivo, scale up post marketing A and BE to CRO. For product ovel, therapies generic drugs estration for foreign  tory affairs. products and O format, industry delines of ICH-Q, S ents of EU, MHRA, ment: Global		Stu- Und app	dents v derstan roval r	vill bod CM egula	e able to IC, pos tory aff	o t fairs.	

	Investigation of medicinal products dossier, dossier (IMPD) and investigator brochure (IB).		development	
	Clinical trials: Developing clinical trial		Students will be able to Learn	
	protocols. Institutional review board/		about requirement to clinical	
	independent ethics committee Formulation and		study process,	
IV	working procedures informed Consent process	12	pharmacovigilance safety	2,3,4,5
	and procedures. HIPAA- new, requirement to		monitoring in clinical trials.	
	clinical study process, pharmacovigilance			
	safety monitoring in clinical trials.			

1. Merchant S.H. and Dr.J.S.Quadry. A textbook of hospital pharmacy, 4th ed. Ahmadabad: B.S. Shah Generic Drug Product Development, Solid Oral Dosage forms, Leon Shargel and IsaderKaufer,Marcel Dekker series, Vol.143 2.

#### **REFERENCE BOOKS:**

- R1. Generic Drug Product Development, Solid Oral Dosage forms, Leon Shargel and IsaderKaufer, Marcel Dekker series, Vol.143
- R2. The Pharmaceutical Regulatory Process, Second Edition Edited by Ira R. Berry and Robert P.Martin, Drugs and the Pharmaceutical Sciences, Vol. 185, Informa Health care Publishers.
- R3. New Drug Approval Process: Accelerating Global Registrations By Richard A Guarino, MD,5th edition, Drugs and the Pharmaceutical Sciences, Vol.190.
- R4. Guidebook for drug regulatory submissions / Sandy Weinberg.By John Wiley &Sons.Inc.
- R5. FDA regulatory affairs: a guide for prescription drugs, medical devices, and biologics/edited By Douglas J. Pisano, David Mantus.
- R6. Clinical Trials and Human Research: A Practical Guide to Regulatory Compliance By Fay A.Rozovsky and Rodney K. Adams.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand the concepts of innovator and generic drugs, differentiate between the drug development processes, and summarize the regulatory pathways for approval.	PO1,PO2,PO3,PO6,PO8
2	Remember the regulatory guidelines and dossier preparation for international submissions to regulatory agencies for new drugs, generic drugs, and post-approval changes.	PO1,PO2,PO3,PO6,PO8
3	Evaluate the processes, documentation, and regulatory expectations for conducting clinical trials from protocol development to safety monitoring.	PO1,PO2,PO3,PO6,PO8
4	Analyze product and process related documents to identify critical quality attributes and prepare master batch records, Drug Master Files, technical reports, and other CMC regulatory documents.	PO1,PO2,PO3,PO6,PO8
5	Apply ICH guidelines, CTD, and regional regulatory requirements to prepare a complete regulatory submission dossier for drug approval in the US, EU, and other global markets.	PO1,PO2,PO3,PO6,PO8

SEMESTER – I	
--------------	--

Course Tit	le Pharmaceutics Practi	cal- I							
Course coo	le MPH105P	Total credits: 12	L	T	P	S	R	O/F	C
		Total hours: 12	0	0	12	0	0	0	6
Pre-requisi		Co-requisite					Nil		
Program	· ·	Master of Pharmacy (Pharmaceutics)							
Semester	1	I semester of first year of the program							
Course		n of the course, it is e						able to und	erstand
Objective	2. Formulation and evaluation of pharmaceutical formulations.								
	3. Estimate the drug by spectroscopy								
CO1						ounds			
		s through the utilizati							
CO2		The students will understand proficiency in executing preformulation studies for tablets,							
	_	showcasing their ability to apply knowledge and analysis skills.							
CO3		he students will reveal their understanding of High-Performance Liquid Chromatography IPLC) experiments through the application of analytical skills and knowledge, showcasing							
	, , , , , , , , , , , , , , , , , , ,			-		s and I	cnowle	edge, show	casing
CO4	_	te and interpret comp	_				. 1		. 1 41.
CO4		ablish their analytical							
	pharmaceutical comp	ed dosage forms, show	veasing i	neir a	ability	to anai	yze an	id categoriz	ze
CO5		etical and practical sk	ille acco	ciated	1 with 1	the inci	rumer	nt	
Unit-No.	Cont		Contact			earnin			KL
01111-110.	Cont		Hour		L	.a	g Out	come	KL
I	1. Analysis of pharmaco	peial compounds		St	udents	will be	able 1	to Learn	
	and their formulations b			Aı	nalysis	of pha	rmaco	peial	
	spectrophotometer				mpoun	_		•	
	2. Simultaneous estimat	ion of multi						roscopic	
	component containing for	ormulations by UV		teo	chnique	es			
	spectrophotometry								
	3. Experiments based or	HPLC							
	4. Experiments based or	n Gas							
	Chromatography								
	5. Estimation of riboflav	/in/quinine sulphate							
	by fluorimetry								
	6. Estimation of sodium	potassium by							
	flame photometry								
	7. To perform In-vitro d	-							4,5,6
	of CR/ SR marketed for								<i>y- y-</i>
	8. Formulation and eval	uation of sustained							
	release matrix tablets								
	9. Formulation and eval	uation osmotically							
	controlled DDS	matica of Election							
	10. Preparation and eval DDS- hydro dynamicall	~							
	11. Formulation and eva								
	adhesive tablets.	iluation of muco-							
	12. Formulation and eva	luation of trans							
	dermal patches.	naation of trails							
	13. To carry out preform	ulation studies of							
	tablets.	idiation studies of							
	14. Tostudy the effect of	f compressional							
	1 r. 1 ostady the chect 0.	Compressional							

force on tablets disintegration time.		
16. Tostudy the effect of particle size on		
dissolution of a tablet.		
17. Tostudy the effect of binders on		
dissolution of a tablet.		
18. To plot Heckel plot, Higuchi and		
Peppas plot and determine similarity		
factors.		

T1. A Practical Book of Pharmaceutics-I (M.Pharm), Khalil Wagh, Usman, Ahmed, S Vikas And Company (Pvt)

#### **REFERENCE BOOKS:**

R1. Pharmaceutics: Practical Manual 3Rd Edition by Abraham, Pharma Med Press

CO PO Mapping				
SN	Course Outcome (CO)	Mapped Program Outcome		
1	The students will demonstrate comprehension of the analysis of pharmacopeial compounds and their formulations through the utilization of UV Vis spectrophotometer.	PO1, PO2, PO3, PO4, PO5, PO6, PO8		
2	The students will understand proficiency in executing preformulating studies for tablets, showcasing their ability to apply knowledge and analysis skills.	PO1, PO2, PO3, PO4, PO5, PO6, PO8		
3	The students will reveal their understanding of High- Performance Liquid Chromatography (HPLC) experiments through the application of analytical skills and knowledge, showcasing their ability to evaluate and interpret complex procedures.	PO1, PO2, PO3, PO4, PO5, PO6, PO8		
4	The students will establish their analytical skills by examining various drugs present in both singular and combined dosage forms, showcasing their ability to analyse and categorize pharmaceutical compositions	PO1, PO2, PO3, PO4, PO5, PO6, PO8		
5	The mastery of theoretical and practical skills associated with the instrument.	PO1, PO2, PO3, PO4, PO5, PO6, PO8		

Course Title	Course Title   Molecular Pharmaceutics (Nano Technology & Targeted DDS) (NTDS)												
Course code	MPH201T	Total credits: 4	L	T	P	S	R	O/F	С				
		Total hours: 60T	4	0	0	0	0	0	4				
Pre-requisite	Nil	Co-requisite			Nil								
Program	Master of Pharmac	y (Pharmaceutics)	•										
Semester	II semester of first	year of the program											
Course	After completion o	f course student is ab	ole to kno	w,									
Objectives	1. The various ap	proaches for develop	oment of	nove	l drug	delive	ry syst	ems.					
	2. The criteria for	selection of drugs a	nd polym	ers f	for the	develo	opment	t of NTDS					
		n and evaluation of r					•						
CO1		ry systems for target						ain					
CO2		te nano particles and											
COZ		rmulating, preparing							.d				
CO3	systems	rmulating, preparing	g, and eva	ıuatı	ng aiv	erse n	апо ра	rticie-base	cu				
	<u> </u>	for improving nasal	absorption	n in	the day	ion o	f nocal	drug dali	7 <b>017</b> 37				
CO4		ze pulmonary delive	•			_		•	•				
004	and dry powder inh		ny by the	acsi	ign or s	unuoi	ic acros	3013, H <b>C</b> 041	IIZCIS				
		of antisense molecule	es and en	nhas	sizino o	zene tl	herany	annroach	2.5				
CO5		, and the future poter		_		-			23,				
			Contact										
Unit-No.	Cor	ntent	Hour		Lea	arning	g Outc	ome	KL				
I	Targeted Drug De	livery Systems:		S	Students will be able to learn								
	Concepts, Events	-			Design drug delivery systems								
	process involved i		12		_	_	_	tumours	3,4,5				
	Tumor targeting a	nd Brain specific			and to the brain								
	delivery.			ļ_									
II	Targeting Method						be able						
	preparation and ev Particles & Lipose		12		Prepare and evaluate nano								
	preparation and ev	• •			particles and liposomes as carriers for drug targeting								
III	Micro Capsules /							to learn					
111	Types, preparation	-					drugs a						
		oodies; preparation					urugs a						
	and application, p		12	_ ^	icrospl			311 01	2.4.5				
	and application, p	-	12		_		and for va	rious	3,4,5				
	Aquasomes, Phyto	·			oplicati	•	101 va	11005					
	Electrosomes.	osomes,		"	pnean	0113							
IV	Pulmonary Drug I	Delivery Systems:		S	tudents	will l	he able	to learn					
1,	Aerosols, propella	* *					ulmona						
	Types, preparation				_	_	e desig	-					
		Delivery systems;	12		•	•	_	oulizers					
	Types, preparation							lers and	3,4,5				
	71 /1 1					-	gies fo						
					_		-	delivery					
					stems		8	,					
V	Nucleic acid base	d therapeutic				will l	be able	to					
	delivery system: (	•	12				erapy i		3,4,5				
	introduction (ex-v	= -					ancer a						

gene therapy). Potential target diseases for gene therapy (inherited	inherited diseases	
disorder and cancer). Gene		
expression systems (viral and nonviral gene transfer). Liposomal		
gene delivery systems.		
Bio distribution and		
Pharmacokinetics. knowledge of therapeutic antisense molecules and		
aptamers as drugs of future.		

- T1. Y W. Chien, Novel Drug Delivery Systems, 2nd edition, revised and expanded, Marcel Dekker, Inc., New York, 1992.
- T2. S.P.Vyas and R.K.Khar, Controlled Drug Delivery concepts and advances, VallabhPrakashan, New Delhi, First edition 2002.

#### **REFERENCE BOOKS:**

- R1. Y W. Chien, Novel Drug Delivery Systems, 2nd edition, revised and expanded, Marcel Dekker, Inc., New York, 1992.
- R2. S.P. Vyas and R.K. Khar, Controlled Drug Delivery- concepts and advances, VallabhPrakashan, New Delhi, First edition 2002.
- R3. N.K. Jain, Controlled and Novel Drug Delivery, CBS Publishers & Distributors, NewDelhi, First edition 1997 (reprint in 2001).

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Design drug delivery systems for targeting drugs to tumours	PO1, PO2, PO3, PO4, PO5, PO6,					
1	and to the brain	PO8					
2	Prepare and evaluate nano particles and liposomes as carriers	PO1, PO2, PO3, PO4, PO5, PO6,					
	for drug targeting	PO8					
3	Acquire skills in formulating, preparing, and evaluating diverse	PO1, PO2, PO3, PO4, PO5, PO6,					
3	nanoparticle-based systems	PO8					
	Develop strategies for improving nasal absorption in the design						
4	of nasal drug delivery systems and optimize pulmonary	PO1, PO2, PO3, PO4, PO5, PO6,					
4	delivery by the design of suitable aerosols, nebulizers and dry	PO8					
	powder inhalers						
	Apply knowledge of antisense molecules and emphasizing	PO1, PO2, PO3, PO4, PO5, PO6,					
5	gene therapy approaches, expression systems, and the future	PO8					
	potential of antisense molecules and aptamers.	100					

SEMESTER –	П
------------	---

Course Ti	tle Advanced Biopharma	aceutics & Pharmacoki	netics	3						
Course co	de MPH202T	Total credits: 4	L	T	P	S	R	O/F	C	
		Total hours: 60T	4	0	0	0	0	0	4	
Pre-requis		Co-requisite				N	il			
Progran	-	· · · · · · · · · · · · · · · · · · ·								
Semeste	•									
		course student is able to								
		biopharmaceutics and	•							
		derive the pharmacoki				•		e best des	scribe	
Course	1 -	bsorption, distribution,								
Objective	NG	n of biopharmaceutic s			_			_		
3	i ne design and evalu	ation of dosage regime	ns of	the dr	ugs us	ing ph	armaco	kınetic ai	nd	
	biopharmaceutic para		1	1	1. ,.	CI		C		
	1 -	pharmacokinetic prob	lems a	and ap	plicati	on of t	oasics o	Ī		
CO1	pharmacokinetic.	1 T (C	IT)	1.D	A 1		N 1	•	1	
CO1		astrointestinal Tract (G			•	•			10	
	1	nat influence drug absortical principles to optin	•		_				.mayrad	
CO2	11 7 1			_			_		rovea	
COZ	r	performance and evaluate and interpret in vitro data to predict drug behavior and								
	1	performance in vivo Apply mathematical models to describe drug ADME processes and assess the impact of								
CO3	1		-	wie pi	ocesse	s and	assess i	не шрас	ι 01	
		drug-drug interactions on pharmacokinetics.  Comprehend the design and conduct of bioequivalence studies and apply pharmacokinetic								
CO4	_	~	•			s and a	ippry pr	iaiiiacok	inctic	
	r -	principles to evaluate and compare drug product performance  Apply pharmacokinetic principles to predict drug behavior in different physiological and								
CO5	1	pathological conditions and integrate pharmacokinetic principles into clinical decision-								
000	r	making.								
TI 4 NI			Co	ntact		r ,			171	
Unit-No.	Con	ntent	E	lour	l I	Learn	ing Ou	tcome	KL	
	Drug Gastrointestinal tra	act, Mechanism of			Stud	ents w	ill be a	ble to		
	drug absorption, Factors	affecting drug			1		asic co	-		
	absorption, pH-partition	theory of drug			in bi	opharı	naceuti	cs and		
	absorption. Formulation				phar	macok	inetics			
	physicochemical factors									
	Dissolution process, No	•								
	equation and drug dissol									
	affecting the dissolution									
	absorption: role of the de	-								
I	(elixir, syrup and solution	,		12					3,4,5	
	Suspension as a dosage	-								
	dosage form, Tablet as a	-								
		ssolution methods, Formulation and								
	processing factors, Corre									
	data with in vitro dissolu	-								
	model: Permeability-Sol	•								
	and the pH Partition Hyj	•								
	of the Costmaint - time 1 T	root (CIT) all								
	of the Gastrointestinal T	· · ·								
	Microclimate Intracellul	ar Complex. pH								
	Microclimate Intracellul Environment, Tight-June	ar Complex. pH ction			Ct. J	onto *-	ill <b>h</b> a a'	hla ta		
II	Microclimate Intracellul	ar Complex. pH ction erations in drug		12			rill be a	ble to	3,4,5	

	D.C. I. 1. 1. 1. 1.	1	CC .: 1	1
	Performance: Introduction, biopharmaceutic		affecting drug	
	factors affecting drug bioavailability, rate-		bioavailability,	
	limiting steps in drug absorption,		alternative methods of	
	physicochemical nature of the drug		dissolution testing,	
	formulation factors affecting drug product		meeting dissolution	
	performance, in vitro: dissolution and drug		requirements, problems	
	release testing, compendial methods of		of variable control in	
	dissolution, alternative methods of		dissolution testing	
	dissolution testing, meeting dissolution		performance of drug	
	requirements, problems of variable control in		products	
	dissolution testing performance of drug			
	products. In vitro-in vivo correlation,			
	dissolution profile comparisons, drug product			
	stability, considerations in the design of a			
	drug product.			
	Pharmacokinetics: Basic considerations,		Students will be able to	
	pharmacokinetic models, compartment		know pharmacokinetic	
	modelling: one compartment model- IV		models	
	bolus, IV infusion, extra-vascular. Multi			
	compartment model: two compartment-			
	model in brief, non-linear pharmacokinetics:			
III	cause of non-linearity, Michaelis- Menten	12		2,3,4
	equation, estimation of $K_{max}$ and $V_{max}$ . Drug			
	interactions: introduction, the effect of			
	protein binding interactions, the interactions,			
	cytochrome effect of tissue-binding p450-			
	based drug interactions, drug interactions			
	linked to transporters.			
	Drug Product Performance, in vivo:		Students will be able to	
	Bioavailability and Bioequivalence: drug		know Bioavailability and	
	product performance, purpose of		Bioequivalence	
	bioavailability studies, relative and absolute		•	
	availability methods for assessing			
	bioavailability, bioequivalence studies,			
	design and evaluation of bioequivalence			
	studies, study designs, crossover study			
IV	designs, evaluation of the data,	12		3,4,5
	bioequivalence example, study submission			- , ,-
	and drug review process. biopharmaceutics			
	classification system, methods. Permeability:			
	In-vitro, in-situ and In-vivo methods. Generic			
	biologics (bio similar drug products), clinical			
	significance of bioequivalence studies,			
	special concerns in bioavailability and			
	bioequivalence studies, generic substitution.			
	Application of Pharmacokinetics: Modified-		Students will be able to	
	Release Drug Products, Targeted Drug		know Pharmacokinetics	
	Delivery Systems and Biotechnological		and pharmacodynamic,	
V	Products. Introduction to Pharmacokinetics	12	drug interactions. &	3,4,5
	and pharmacodynamic, drug interactions.		Application of	
	Pharmacokinetics and pharmacodynamics of		Pharmacokinetics.	
•	т паннасокинсись ана рнагнасодупатись от	1	r marmacokinetics.	1

biotechnology drugs. Introduction, Protein	s
and peptides, Monoclonal antibodies,	
Oligonucleotides, Vaccines	
(immunotherapy), Gene therapies	

- R1. Bio pharmaceutics and Clinical Pharmacokinetics by Milo edition, Philadelphia, Lea and Febiger, 1991
- R2. Biopharmaceutics and Pharmacokinetics, A. Treatise, D.M. Brahmankar and Sunil B. Jaiswal., Vallab Prakashan, Pitampura, Delhi.

#### **REFERENCE BOOKS:**

- R1. Biopharmaceutics and Clinical Pharmacokinetics by Milo Gibaldi, 4th edition, Philadelphia, Lea and Febiger, 1991
- R2. Biopharmaceutics and Pharmacokinetics, A. Treatise, D.M. Brahmankar and Sunil B. Jaiswal., Vallab Prakashan, Pitampura, Delhi
- R3. Applied Biopharmaceutics and Pharmacokinetics by Shargel. Land YuABC, 2ndedition, Connecticut Appleton Century Crofts, 1985
- R4. Textbook of Biopharmaceutics and Pharmacokinetics, Dr.Shobha Rani R. Hiremath, Prism Book
- R5. Pharmacokinetics by Milo Gibaldi and D. Perrier, 2nd edition, Marcel Dekker Inc., New York, 1982
- R6. Current Concepts in Pharmaceutical Sciences: Biopharmaceutics, Swarbrick. J, LeaandFebiger, Philadelphia, 1970
- R7. Clinical Pharmacokinetics, Concepts and Applications 3rd edition by Malcolm Rowland and Thom~N. Tozer, Lea and Febiger, Philadelphia, 1995
- R8. Dissolution, Bioavailability and Bioequivalence, Abdou. H.M, Mack Publishing Company, Pennsylvania 1989
- R9. Biopharmaceutics and Clinical Pharmacokinetics, An Introduction, 4th edition, revised and expanded by Robert. E. Notari, Marcel Dekker Inc,New York and Basel,1987.
- R10. Bio pharmaceutics and Relevant Pharmacokinetics by John.G Wagner and M.Pemarowski, 1st edition, Drug Intelligence Publications, Hamilton, Illinois, 1971.
- R11. Encyclopaedia of Pharmaceutical Technology, Vol 13, James Swarbrick, James.G.Boylan, Marcel Dekker Inc, New York, 1996.
- R12. Basic Pharmacokinetics,1st edition, Sunil S Jambhekar and Philip J Breen, pharmaceutical press, RPS Publishing,2009.
- R13. Absorption and Drug Development- Solubility, Permeability, and Charge State, Alex Avdeef, John Wiley & Sons, Inc,2003.

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Understanding the Gastrointestinal Tract (GIT) and Drug Absorption Mechanisms and Identify the factors that influence drug absorption in the gastrointestinal tract	PO1, PO2, PO3, PO5, PO8				
2	Apply biopharmaceutical principles to optimize drug formulation and design for improved performance and evaluate and interpret in vitro data to predict drug behavior and performance in vivo	PO1, PO2, PO3, PO5, PO8				
3	Apply mathematical models to describe drug ADME processes and assess the impact of drug-drug interactions on pharmacokinetics.	PO1, PO2, PO3, PO5, PO8				
4	Comprehend the design and conduct of bioequivalence studies and apply pharmacokinetic principles to evaluate and compare drug product performance	PO1, PO2, PO3, PO5, PO8				
5	Apply pharmacokinetic principles to predict drug behavior in different physiological and pathological conditions and integrate pharmacokinetic principles into clinical decisionmaking.	PO1, PO2, PO3, PO5, PO8				

Course Ti	tle Computer Aided Drug	g Development							
Course co	MPH203T	Total credits: 4	L	T	P	S	R	O/F	C
		Total hours: 60T	4	0	0	0	0	0	4
Pre-requisite Nil		Co-requisite				l	Nil		
Progran	-	•							
Semeste									
	_	course student is able							
		outers in Pharmaceutic			and D	evelo	pment		
	_	Modeling of Drug Dis	•						
Course	_	eclinical Developmen			1				
Objective	NG   -	chniques in Pharmace	eutical F	ormı	llation				
	5. Computers in Ma 6. Computers in Cl	inical Development							
	_	gence (AI) and Roboti	ios						
		luid dynamics (CFD)							
CO1	Learn how to use com	• ` ′		'Anro	h and c	lovel	nmen	t	
CO2	Utilize the knowledge	<u> </u>					_	ι.	
CO2	Construct computer-a							reening	
	Computer-aided phar								vzing the
CO4	comprehensiveness.	macokinetics and pil	141111400	a y 1101.		mar at	.cl1Zal	anai, anai	, zing uic
	Invent and anticipat	te artificial intellige	ence. ro	botio	es. cu	rrent	challe	enges, ar	d future
CO5	developments.	e artificial interng	, 10	,0011	, ca		• Hall	enges, un	ia iavait
Unit-	Conte	nt	Contact		Lea	rning	g Outo	ome	KL
No.			Hour			•	•		
I	a. Computers in Pharma	aceutical Research		St	udents	will	be able	e to	
	and Development: A Go			U	ndersta	and th	e Con	nputers	
	History of Computers in	n Pharmaceutical			Pharm	naceu	tical R	esearch	
	Research and Developn	nent. Statistical			nd Dev	elopn	nent.		
	modelling in pharmaceu	itical research and							
	development: Descriptiv	ve versus							
	Mechanistic Modelling,	Statistical							
	Parameters, Estimation,		12						3,4,5
	Regions, Nonlinearity a	t the Optimum,	12						3,7,3
	Sensitivity Analysis, Op	otimal Design,							
	Population Modelling								
	b. Quality-by-Design in								
	Development: Introduct	-							
	guideline, Regulatory an	•							
	on QbD, scientifically b	-							
TT	examples of application			04	udents	,,, <u>,</u> ,11	ho al-1	a to	
II	Computational Modellin Disposition: Introduction	-			tilize tl				
	Techniques: Drug Abso	-						lelling of	
	Intestinal Permeation, I	-			rug Di			oning of	
	Drug Excretion, Active	-	12		. 45 1	Posi	.1011.		3,4,5
	BCRP, Nucleoside Tran								
	hPEPT1, ASBT, OCT,	_							
	Choline Transporter.	,							
III	Computer-aided formul	ation		St	udents	will	be able	e to	
	development: Concept of		12					nputer-	3,4,5
	Optimization parameter	-	-		ded for			•	, ,-
	r parameter	,							1

	design, Optimization technology & Screening design. Computers in Pharmaceutical Formulation: Development of pharmaceutical emulsions, micro-emulsion drug carriers Legal Protection of Innovative Uses of Computers in R&D, The Ethics of Computing in Pharmaceutical Research, Computers in Market analysis		development	
IV	a. Computer-aided biopharmaceutical characterization: Gastrointestinal absorption simulation. Introduction, Theoretical background, Model construction, Parameter sensitivity analysis, Virtual trial, Fed vs. fasted state, In vitro dissolution and in vitro in vivo correlation, Bio waiver considerations b. Computer Simulations in Pharmacokinetics and Pharmacodynamics: Introduction, Computer Simulation: Whole Organism, Isolated Tissues, Organs, Cell, Proteins and Genes. c. Computers in Clinical Development: Clinical Data Collection and Management, Regulation of Computer Systems	12	Students will be able to Understand the scope of Computer-aided Pharmacokinetics and Pharmaco-dynamics characterization	3,4,5
V	Artificial Intelligence (AI), Robotics and Computational fluid dynamics: General overview, Pharmaceutical Automation, Pharmaceutical applications, Advantages and Disadvantages. Current Challenges and Future Directions.	12	Students will be able to Understand and summarize the general Artificial Intelligence (AI), Robotics and Computational fluid dynamics	3,4,5

- T1. Computer Applications in Pharmaceutical Research and Development, Sean Ekins, 2006, John Wiley & Sons.
- T2. Computer-Aided Applications in Pharmaceutical Technology, 1st Edition, JelenaDjuris, Woodhead Publishing.

#### **REFERENCE BOOKS:**

- R1. Computer Applications in Pharmaceutical Research and Development, Sean Ekins, 2006, John Wiley & Sons.
- R2. Computer-Aided Applications in Pharmaceutical Technology, 1st Edition, JelenaDjuris, Woodhead Publishing
- R3. Encyclopedia of Pharmaceutical Technology, Vol 13, James Swarbrick, James.G.Boylan, Marcel Dekker Inc, New York, 1996.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Learn how to use computers for pharmaceutical research and development.	PO1, PO2, PO3, PO4, PO5, PO8					
2	Utilize the knowledge for computational modeling of drug disposition.	PO1, PO2, PO3, PO4, PO5, PO8					
3	Construct computer-aided formulation development, optimization, and screening.	PO1, PO2, PO3, PO4, PO5, PO8					
4	Computer-aided pharmacokinetics and pharmacodynamics characterization: analyzing the comprehensiveness.	PO1, PO2, PO3, PO4, PO5, PO8					
5	Invent and anticipate artificial intelligence, robotics, current challenges, and future developments.	PO1, PO2, PO3, PO4, PO5, PO8					

Course Tit	ele Cosmetics And Cosm	neceuticals								
Course co		Total credits: 4	L	T	P	S	R	O/F	С	
		Total hours: 60T	4	0	0	0	0	0	4	
Pre-requis	ite Nil		Nil	•	•	•				
Program		1	111							
Semester		` ,								
Course	-	course student is able to	o know.							
Objective	•	used in cosmetics and			als.					
Objective		ocks for various formul		Catic	ais.					
	1	ogies in the market	idilolis.							
	I	redients and basic scien	nce to d	evelo	n cost	netics	and co	smecenti	cals	
		ledge to develop cosme			_					
	stability, and eff	-	ores and			il Cuis	Willia ac	onea sar	<i>-</i> ,	
CO1		regulatory provisions re	elated to	imn	ort and	l man	ufactur	e of cosm	etics	
CO2		skin and hair related to				· IIIdii	aractar	<b>C</b> 01 <b>C</b> 0511		
CO3	_	s for different product		•		smeti	es and o	cosmeceu	ticals	
CO4	_	s for different product								
CO5	_	gredients used in hair of						COSITICCCU	iticais	
Unit-No.	Conte		Contact	II Cai			g Outc	ome	KL	
UIII-110.	Cont		Hour		Lea	41 IIIII	g Ouit	OHIC	KL	
I	Cosmetics- Regulatory:		IIUuI	Stu	dents v	will b	ahle t	0		
1	cosmetic products as per			Students will be able to Understand the cosmetic						
	Indian regulatory requir	-			products as per Indian					
	labelling of cosmetics R			_						
	provisions relating to im					regulation. Understand the regulatory provisions related				
	Misbranded and spuriou	-	12		2,3,4,5					
	Regulatory provisions re		12	to the import and manufacture of cosmetics					2,3,4,3	
	manufacture of cosmetic				osine	105				
	obtaining license, prohil									
	manufacture and sale of									
	loan license, offences ar	·								
II	Cosmetics- Biological a	-		Stu	dents v	will b	e able t	0		
	skin relating to problem	•					Regula			
	acne, pigmentation, pric	•			visions		_	J		
	and body odour. Structu	· ·		_			cosme	tics		
	growth cycle. Common		12						2, 3,4,5	
	associated with oral cav	-								
	care needs for face, eye	•								
	feet, nail, scalp, neck, bo	ody and under-arm.								
III	Formulation Building bl	locks: Building		Stu	dents v	will b	e able t	o learn		
	blocks for different prod	luct formulations		abo	ut Stru	ıcture	of skir	ı		
	of cosmetics/cosmeceut	icals. Surfactants		rela	iting to	prob	lems			
	Classification and applic	cation. Emollients,								
	rheological additives: cl	assification and								
	application. Antimicrob	ial used as	12						3,4,5	
	preservatives, their meri	its and demerits.	14						3,4,3	
	Factors affecting microb	pial preservative								
	efficacy. Building block	s for formulation								
	of a moisturizing cream	, vanishing cream,								
	cold cream, shampoo an	nd toothpaste.								
	Soaps and syndet bars. I	Perfumes;								
	soaps and syndet bars. I	renumes;								

	Classification of perfumes. Perfume			
	ingredients listed as allergens in EU			
	regulation. Controversial ingredients:			
	Parabens, formaldehyde liberators,			
	dioxane.			
IV	Design of cosmeceutical products: Sun protection, sunscreens classification and regulatory aspects. Addressing dry skin, acne, sun-protection, pigmentation, prickly heat, wrinkles, body odour., dandruff, dental cavities, bleeding gums, mouth	12	Students will be able to Understand the Structure of hair and hair growth cycle. Understand the problems associated with oral cavity	3,4,5
	Odor and sensitive teeth through cosmeceutical formulations.			
V	Herbal Cosmetics: Herbal ingredients used in Hair care, skin care and oral care. Review of guidelines for herbal cosmetics by private bodies like cosmos with respect to preservatives, emollients, foaming agents, emulsifiers and rheology modifiers. Challenges in formulating herbal cosmetics.	12	Students will be able to Recognize the role of Surfactants	2, 3,4,5

- T1. Cosmetics Formulation, Manufacture and quality control, PP. Sharma, 4th edition
- T2. Handbook of cosmetic science and Technology A.O.Barel, M.Paye and H.I. Maibach. 3rd edition

#### **REFERENCE BOOKS:**

- R1. Harry's Cosmeticology.8th edition.
- R2. Poucher'sperfumecosmeticsandSoaps,10th edition.
- R3. Cosmetics-Formulation, Manufacture and quality control, PP. Sharma,4th edition
- R4. Handbook of cosmetic science and Technology A.O.Barel, M.Paye and H.I. Maibach. 3rdedition
- R5. Cosmetic and Toiletries recent supplier's catalogue.
- R6. CTFA directory

#### RELATIONSHIP BETWEEN COURSE OUTCOMES (CO) AND PROGRAM OUTCOMES

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Describe the Indian regulatory provisions related to import and manufacture of cosmetics	PO1, PO2, PO5, PO6, PO8					
2	Explain structure of skin and hair related to various problems	PO1, PO2, PO5, PO6, PO8					
3	Select building blocks for different product formulations of cosmetics and cosmeceuticals	PO1, PO2, PO5, PO6, PO8					
4	Select building blocks for different product formulations of cosmetics and cosmeceuticals	PO1, PO2, PO5, PO6, PO8					
5	Discuss the herbal ingredients used in hair care, skin care and oral care	PO1, PO2, PO5, PO6, PO8					

#### SEMESTER - II

Course Titl	e Pharmaceutics Practi	cal's- II									
Course cod	e MPH205P	Total credits: 6	L	T	P	S	R	O/F	С		
		Total hours: 12	0	0	12	0	0	0	6		
Pre-requisit	te Nil	Nil Co-requisite Nil									
Program	Master of Pharmacy	(Pharmaceutics)									
Semester	, , , , , , , , , , , , , , , , , , ,										
Course	Course After completion of course student is able to know,										
Objectives	Objectives Develop a novel drug delivery system.  Perform pharmacokinetic and pharmacodynamic studies.										
	_	•	mic	studie	S.						
	Prepare and evaluate										
CO1	· ·					ne, nio	osome	etc.			
CO2 Compare dissolution of two different marketed products											
CO3		data using Design Expe									
CO4		products incorporating h				al acti	ives				
CO5	-	e cream, shampoo and t									
Unit-No	Con	itent		ontac	t I	Learn	ing O	utcome	KL		
			]	Hour							
	. To study the effect of t	-					vill be a				
	change,non-solvent addit							oncepts			
_	oolymer addition in micr				l l		ncapsul				
	2. Preparation and evalua	-						rs effect			
	3. Formulation and evalu						_	ulation			
	gelatine/albumin microsp						about v				
	Formulation and evaluation	ation of			_	_	_	ocedures			
	iposome's/niosomes				l l		n Prepa				
	5. Formulation and evalu	•						various			
	5. Improvement of dissol						_	delivery			
	lightly soluble drug by S	solid dispersion			dosaş	ge for	ms				
	echnique.	tion of town 1:00									
	7. Comparison of dissolu narketed products /brand										
	3. Protein binding studies										
	oound drug & poorly pro	<b>U V Y</b>									
	9. Bioavailability studies										
	nimals.	of functumor in		12					3,4,5,6		
	0. Pharmacokinetic and	IVIVC data analysis									
	by WinnolineR software	,									
	1. In vitro cell studies fo	or permeability and									
	netabolism										
1	2. DoE Using Design Ex	xpert® Software									
1	3. Formulation data ana	lysis Using Design									
E	Expert® Software										
1	4. Quality-by-Design in	Pharmaceutical									
Ι	Development										
	5. Computer Simulation	s in Pharmacokinetics									
	and Pharmacodynamics										
	6. Computational Mode	lling of Drug									
	Disposition										
	7. To develop Clinical I										
	8. To carry out Sensitivi	ty Analysis, and									
F	Population Modelling.										

19. Development and evaluation of Creams		
20. Development and evaluation of Shampoo		
and Toothpaste base		
21. To incorporate herbal and chemical actives		
to develop products		
22. To address Dry skin, acne, blemish,		
Wrinkles, bleeding gums and dandruff		

- T1. Bio pharmaceutics and Clinical Pharmacokinetics by Milo Gibaldi, 4_{th}edition, Philadelphia, Lea and Febiger, 1991 2
- T2. Bio pharmaceutics and Pharmacokinetics, A. Treatise, D.M. Brahmankar and Sunil B. Jaiswal., VallabPrakashan, Pitampura, Delhi
- T3. Textbook of Bio pharmaceutics and Pharmacokinetics, Dr.Shobha Rani R. Hiremath, Prism Book
- T4. Pharmacokinetics by Milo Gibaldi and D. Perrier, 2nd edition, Marcel Dekker Inc., New York, 1982

#### REFERENCE BOOKS:

- R1. Clinical Pharmacokinetics, Concepts and Applications 3rd edition by MalcolmRowland and Thom~ N. Toner, Lea and Febiger, Philadelphia,1995 2
- R2. Dissolution, Bioavailability and Bioequivalence, Abdou. H.M, Mack PublishingCompany, Pennsylvania 1989
- R3. Bio pharmaceutics and Clinical Pharmacokinetics, An Introduction, 4th edition, revised and expanded by Robert. E. Notari, Marcel Dekker Inc, New York and Basel, 1987.
- R4. Bio pharmaceutics and Relevant Pharmacokinetics by John. G Wagner and M.Pemarowski, 1st edition, Drug Intelligence Publications, Hamilton, Illinois, 1971 R5 Encyclopedia of Pharmaceutical Technology, Vol 13, James Swarbrick, James. G.Boylan, Marcel Dekker Inc, New York, 1996

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Formulate and examine Alginate beads, microspheres, liposome, niosome etc.	PO1, PO2, PO3, PO4, PO5, PO8						
2	Compare dissolution of two different marketed products	PO1, PO2, PO3, PO4, PO5, PO8						
3	Analyse formulation data using Design Expert® Software	PO1, PO2, PO3, PO4, PO5, PO8						
4	Design and develop products incorporating herbal and chemical actives	PO1, PO2, PO3, PO4, PO5, PO8						
5	Develop and evaluate cream, shampoo and toothpaste base	PO1, PO2, PO3, PO4, PO5, PO8						

<b>Course Tit</b>	tle Research Methodolo	gy and Biostatistics							
Course co	de MRM301T	Total credits: 4	L	T	P	S	R	O/F	C
Course coo	ue MRM5011	Total hours: 60T	4	0	0	0	0	0	4
Pre-requisi	ite Nil	Co-requisite		-		ľ	Vil		
Program Master of Pharmacy			•						
Semester	· III semester of Second	year of the program							
Course	Upon completion of th	e course, the student sl	hall be ab	ole to	)				
Objective	s 1. Know the operation	on of M.S. Excel, SPSS	S, R and I	MIN	ITAB	®, Do	E (Des	ign of	
	Experiment).								
	2. Know the various	statistical techniques t	to solve st	tatist	tical p	roblen	ns		
	3. Appreciate statistical techniques in solving the problems								
CO1	Analyse the value, sco			s of	resear	ch.			
CO2	Discuss the basic conc	- '							
CO3	Apply the basic principal								
CO4		ines for the maintenar	nce of lal	bora	tory a	nimals	and d	esign res	earch
	work.								
CO5	Create efficiency in so								
Unit-No.	Cont	ent	Contact		Le	arnin	g Outo	ome	KL
			Hour	<u> </u>					
	General Research Metho	.					earn ab	out	
	objective, requirements,	-			sics of		rch		
I	difficulties, review of lit		4.6	me	ethodo	logy			2 4 5
	design, types of studies,	-	12						3,4,5
	eliminate errors/bias, co								
	randomization, crossove	er design, placebo,							
	blinding techniques.	1: .: 1		G	1 4	'11 1	1		
	Biostatistics: Definition,	* *					earn ab tics and		
	size, importance of samplinfluencing sample size,	·			sies oi plicati		tics and	1 118	
	tests of significance, typ	-		ap	pncan	OII.			
	tests, parametric tests(st								
l II	ANOVA, Correlation co	·	12						3,4,5
11	regression), non-parame	· ·	12						3,7,3
	rank tests, analysis of va	,							
	chi square test), null hyp								
	degree of freedom, inter								
	values.	provincia er r							
	Medical Research: Histo	orv, values in		Sti	udents	will 1	earn ab	out	
	medical ethics, autonom				edical				
	maleficence, double effe	•							
	between autonomy and l								
	maleficence, euthanasia,								
	confidentiality, criticism								
III	medical ethics, importar	nce of	12						3,4,5
	communication, control	resolution,							
	guidelines, ethics comm	ittees, cultural							
	concerns, truth telling, o	online business							
	practices, conflicts of in	terest, referral,							
	vendor relationships, tre	-							
	members, sexual relation								
IV	CPCSEA guidelines for	laboratory animal	12	Stı	udents	will 1	earn ab	out	3,4,5

	facility: Goals, veterinary care, quarantine,		CPCSEA guidelines	
	surveillance, diagnosis, treatment and			
	control of disease, personal hygiene,			
	location of animal facilities to laboratories,			
	anaesthesia, euthanasia, physical facilities,			
	environment, animal husbandry, record			
	keeping, SOPs, personnel and training,			
	transport of lab animals.			
	Declaration of Helsinki: History,		Students will learn about	
	introduction, basic principles for all medical		medical ethics	
V	research, and additional principles for	12		3,4,5
	medical research combined with medical			
	care.			

#### **REFERENCE BOOKS:**

- R1. Pharmaceutical statistics- Practical and clinical applications, Sanford Bolton, publisher Marcel Dekker Inc. NewYork.
- R2. Fundamental of Statistics Himalaya Publishing House- S.C.Gupta
- R3. Design and Analysis of Experiments –PHI Learning Private Limited, R. Pannerselvam,
- R4. Design and Analysis of Experiments Wiley Students Edition, Douglas and C. Montgomery

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Analyse the value, scope, objectives, and requirements of research.	PO1,PO3,PO4,PO5,PO6,PO8						
2	Discuss the basic concepts of statistical analysis.	PO1,PO3,PO4,PO5,PO6,PO8						
3	Apply the basic principles of medical research and ethics.	PO1,PO3,PO4,PO5,PO6,PO8						
4	Understand the guidelines for the maintenance of laboratory animals and design research work.	PO1,PO3,PO4,PO5,PO6,PO8						
5	Create efficiency in solving practical difficulties.	PO1,PO3,PO4,PO5,PO6,PO8						

<b>Course Title</b>	Journal Club								
Course code	MRM302NA	Total credits: 1	L	T	P	S	R	O/F	C
		Total hours:	0	0	0	0	0	0	1
<b>Pre-requisite</b>	Nil	Co-requisite				1	Nil		
Program	Master of Pharmacy								
Semester	III semester of Secon	d year of the program							
	1. To teach and de	evelop critical appraisa	l skill	ls, inci	ease e	xposu	re to ra	pidly evol	ving
Course	literature, and h	nelp in informed clinica	ıl pra	ctice.					
Objectives	2. To provide a ur	nique opportunity to pr	omote	e inter	est in r	esearc	h whil	e learning	from
	experts about k	experts about knowledge gaps and future research questions.							
CO1	Retrieve and recall essential information from scientific literature, summarizing key								
	concepts and findings discussed in the assigned journal articles.								
CO2	Interpret and comprehend the chosen journal articles' methodologies, results, and								
	implications, demonstrating a clear understanding of the research content.								
(())	Apply critical analysis skills to assess the experimental design and methodologies								
	employed in the selected journal articles, evaluating their appropriateness and validity.								
	Analyze and synthesize information from multiple journal articles, comparing and								
	contrasting methodologies, results, and conclusions to identify patterns, trends, and								
	potential areas for further investigation.								
	Evaluate the overall significance and relevance of the journal articles in the broader								
	•	armaceutical research,	consi	dering	g ethica	al impl	licatior	ns, limitati	ons,
	and potential contributions to the field.								

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Retrieve and recall essential information from scientific literature, summarizing key concepts and findings discussed in the assigned journal articles.	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8
2	Interpret and comprehend the chosen journal articles' methodologies, results, and implications, demonstrating a clear understanding of the research content.	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8
3	Apply critical analysis skills to assess the experimental design and methodologies employed in the selected journal articles, evaluating their appropriateness and validity.	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8
4	Analyse and synthesize information from multiple journal articles, comparing and contrasting methodologies, results, and conclusions to identify patterns, trends, and potential areas for further investigation.	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8
5	Evaluate the overall significance and relevance of the journal articles in the broader context of current pharmaceutical research, considering ethical implications, limitations, and potential contributions to the field.	PO1,PO2,PO3,PO4,PO5,PO6,P O7,PO8

Semester III	
Course Title   DISCUSSION / PRESENTATION (PROPOSAL PRESENTATION)	

Course code	MRM303NA	Total credits: 2	L	T	P	S	R	O/F	C
		Total hours:	0	0	0	0	0	0	2
<b>Pre-requisite</b>	Nil	Co-requisite		•		N	Vil	•	
Program	Master of Pharmacy	·							
Semester	III semester of Second year of the program								
Course	1. Develop scientific writing skills								
Objectives	2. Enable critical thinking ability								
	3. Enhance con	nmunication skills							
	4. Follow ethic	al considerations							
CO1	Identify the research	problem							
CO2	Discuss research pro	blem with team and gr	uide fo	or solu	tion				
CO3	Develops protocol re	Develops protocol report with an aim and objectives							
CO4	Analyse research pro	Analyse research problem							
CO5	Develops plan of wo	ork for research project							

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Identify the research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
2	Discuss research problem with team and guide for solution	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
3	Develops protocol report with an aim and objectives	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
4	Analyse research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
5	Develops plan of work for research project	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						

<b>Course Title</b>	Research Work								
Course code	MRM304NA	Total credits: 14	L	T	P	S	R	O/F	C
		Total hours:	4	0	0	0	0	0	14
Pre-requisite	Nil	Co-requisite	Nil		·	•			
Program	Master of Pharm	acy							
Semester	III semester of S	econd year of the progr	am						
Course	1. Acquire resea	arch skills							
<b>Objectives</b>	2. Develop scient	tific writing skills							
	3. Enable critical	thinking ability							
	<ul><li>4. Adopt application-oriented learning</li><li>5. Appreciate time management and organizational skills:</li><li>6. Enhance communication skills</li></ul>								
	7. Follow ethical considerations								
CO1	· ·	naceutical concepts and	principle	es pert	inent to	the N	1. Phari	m project's	S
	research focus.								
CO2	_	chanism of action of the		_		cal age	ents, de	monstratir	ng a
		nderstanding of their m		_					
CO3		pharmaceutical research		ques to	analys	se drug	g formu	llations an	d
		acy in practical experin							
CO4		thesize experimental da					sions ab	out the	
		• •	nts of the formulated drugs.						
CO5	_	eutical research's ethical	_	•	consid	deratio	ns, ensi	uring	
	alignment with e	stablished guidelines ar	nd princip	oles.					

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Recall key pharmaceutical concepts and principles pertinent to the M. Pharm project's research focus.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
2	Interpret the mechanism of action of the selected pharmaceutical agents, demonstrating a comprehensive understanding of their molecular pathways.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
3	Utilize advanced pharmaceutical research techniques to analyse drug formulations and assess their efficacy in practical experiments.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
4	Examine and synthesize experimental data to draw informed conclusions about the effectiveness and potential improvements of the formulated drugs.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
5	Assess pharmaceutical research's ethical and regulatory considerations, ensuring alignment with established guidelines and principles.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					

	Semester IV
<b>Course Title</b>	Journal Club

Course code	MRM401NA	Total credits: 1	L	T	P	S	R	O/F	С
		Total hours:	0	0	0	0	0	0	1
Pre-requisite	Nil	Co-requisite				ľ	Nil		
Program	Master of Pharmacy								
Semester	IV semester of Second	l year of the program							
Course	1. To teach and deve	lop critical appraisal sl	xills, i	ncreas	se expo	sure t	o rapio	ily evolvin	g
Objectives	literature, and help	in informed clinical p	ractic	e.					
	2. To provide a unique	ue opportunity to prom	ote in	terest	in rese	earch v	while lo	earning fro	m
	experts about know	wledge gaps and future	e research questions.						
CO1	Retrieve and recall ess	ential information from	from scientific literature, summarizing key						
	concepts and findings	cepts and findings discussed in the assigned journal articles.							
CO2	Interpret and compreh	end the chosen journal	l articles' methodologies, results, and						
	implications, demonst	rating a clear understar	nding	of the	resear	ch con	itent.		
CO3	Apply critical analysis	skills to assess the exp	perime	ental d	lesign	and m	ethodo	logies	
	employed in the select	ed journal articles, eva	luatin	g their	r appro	priate	ness a	nd validity	
CO4	Analyse and synthesiz	e information from mu	ltiple	journa	al artic	les, co	mpari	ng and	
	contrasting methodolo	gies, results, and concl	and conclusions to identify patterns, trends, and						
	potential areas for furt	her investigation.							
CO5	Evaluate the overall si	e overall significance and relevance of the journal articles in the broader context							
	of current pharmaceut	ical research, consideri	ng eth	nical in	mplica	tions,	limitat	ions, and	
	potential contributions	to the field.							

	CO PO Mapping					
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Retrieve and recall essential information from scientific literature, summarizing key concepts and findings discussed in the assigned journal articles.	PO1,PO2,PO3,PO4,PO5,PO6,PO7, PO8				
2	Interpret and comprehend the chosen journal articles' methodologies, results, and implications, demonstrating a clear understanding of the research content.	PO1,PO2,PO3,PO4,PO5,PO6,PO7, PO8				
3	Apply critical analysis skills to assess the experimental design and methodologies employed in the selected journal articles, evaluating their appropriateness and validity.	PO1,PO2,PO3,PO4,PO5,PO6,PO7, PO8				
4	Analyse and synthesize information from multiple journal articles, comparing and contrasting methodologies, results, and conclusions to identify patterns, trends, and potential areas for further investigation.	PO1,PO2,PO3,PO4,PO5,PO6,PO7, PO8				
5	Evaluate the overall significance and relevance of the journal articles in the broader context of current pharmaceutical research, considering ethical implications, limitations, and potential contributions to the field.	PO1,PO2,PO3,PO4,PO5,PO6,PO7, PO8				

Semester IV	
Course Title DISCUSSION / FINAL PRESENTATION	

Course code	MRM402NA	<b>Total credits: 2</b>	L	T	P	S	R	O/F	С
		<b>Total hours:</b>	0	0	0	0	0	0	2
Pre-requisite	Nil	Co-requisite			'	N	Vil	1	
Program	Master of Pharmacy								
Semester	IV semester of Secon	nd year of the program							
Course	1. Develop scientif	ic writing skills							
Objectives	2. Enable critical thinking ability								
	3. Enhance commu	nication skills							
	4. Follow ethical co	onsiderations							
CO1	Identify the research	problem							
CO2	Discuss research problem with team and guide for solution								
CO3	Develops protocol report with an aim and objectives								
CO4	Analyse research pro	blem							
CO5	Develops plan of work for research project								

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Identify the research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
2	Discuss research problem with team and guide for solution	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
3	Develops protocol report with an aim and objectives	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
4	Analyse research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					
5	Develops plan of work for research project	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8					

Course code	MRM403NA	Total credits: 14	L	T	P	S	R	O/F	C
		Total hours:	0	0	0	0	0	0	14
Pre-requisite	Nil	Co-requisite	•	•	•	N	Vil		
Program	Master of Pharmacy								
Semester	IV semester of Secon	nd year of the program							
Course	1. Acquire research	ı skills							
<b>Objectives</b>	2. Develop scientif	ic writing skills							
	3. Enable critical thinking ability								
	4. Adopt application-oriented learning								
	5. Appreciate time management and organizational skills:								
	6. Enhance communication skills								
	7. Follow ethical considerations								
CO1	Recall key pharmaceutical concepts and principles pertinent to the M. Pharm project's								
	research focus.								
CO2	Interpret the mechanism of action of the selected pharmaceutical agents, demonstrating a								
	comprehensive understanding of their molecular pathways.								
CO3	Utilize advanced pharmaceutical research techniques to analyse drug formulations and								
		in practical experiments							
CO4	Examine and synthesize experimental data to draw informed conclusions about the								
	effectiveness and potential improvements of the formulated drugs.								
CO5	Assess pharmaceutical research's ethical and regulatory considerations, ensuring alignment								
	with established guidelines and principles.								

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Recall key pharmaceutical concepts and principles	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	pertinent to the M. Pharm project's research focus.	1 0 1,1 0 2,1 0 0,1 0 1,1 0 0,1 0 0,1 0 0
	Interpret the mechanism of action of the selected	
2	pharmaceutical agents, demonstrating a comprehensive	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	understanding of their molecular pathways.	
	Utilize advanced pharmaceutical research techniques to	
3	analyse drug formulations and assess their efficacy in	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	practical experiments.	
	Examine and synthesize experimental data to draw	
4	informed conclusions about the effectiveness and	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	potential improvements of the formulated drugs.	
	Assess pharmaceutical research's ethical and regulatory	
5	considerations, ensuring alignment with established	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	guidelines and principles.	



### ASSAM DOWN TOWN UNIVERSITY

## Curriculum and Syllabus

## **Master of Pharmacy**

(Pharmacology)

# OUTCOME BASED EDUCATION FRAMEWORK CHOICE BASED CREDIT SYSTEM

Version: 1.01

# FACULTY OF PHARMACEUTICAL SCIENCE

July, 2023

#### **PREAMBLE**

Assam down town University is a premier higher educational institution which offers Bachelor, Master, and Ph.D. degree programs across various faculties. These program, collectively embodies the vision and mission of the university. All the programs offered by the Faculty of Pharmaceutical Science of Assam down town University strictly follow the curriculum approved by the Pharmacy Council of India (PCI), the statutory body responsible for regulating the profession of pharmacy in India. This document contains outline of teaching and learning framework and complete detailing of the courses. This document is a guidebook for the students to choose desired courses for completing the program and to be eligible for the degree. This volume also includes the prescribed literature, study materials, texts, and reference books under different courses as guidance for the students to follow.

Recommended by the Board of Studies (BOS) meeting of the Faculty of Pharmaceutical Science held on dated 08/07/2023 and approved by the Emergent Academic Council(AC) meeting held on dated 28/07/2023

Chairperson, Board of Studies

Member Secretary, Academic Council

#### Vision

To become a Globally Recognized University from North Eastern Region of India, dedicated to the Holistic Development of Students and Making Society Better

#### Missions

- 1. Creation of curricula that address the local, regional, national, and international needs of graduates, providing them with diverse and well-rounded education.
- 2. Build a diverse student body from various socio-economic backgrounds, provide exceptional value-based education, and foster holistic personal development, strong academic careers, and confidence.
- 3. Achieve high placement success by offering students skill-based, innovative education and strong industry connections.
- 4. Become the premier destination of young people, desirous of becoming future professional leaders through multidisciplinary learning and serving society better.
- 5. Create a highly inspiring intellectual environment for exceptional learners, empowering them to aspire to join internationally acclaimed institutions and contribute to global efforts in addressing critical issues, such as sustainable development, Climate mitigation and fostering a conflict-free global society.
- 6. To be renowned for creating new knowledge through high quality interdisciplinary research for betterment of society.
- 7. Become a key hub for the growth and excellence of AdtU's stakeholders including educators, researchers and innovators
- 8. Adapt to the evolving needs and changing realities of our students and community by incorporating national and global perspectives, while ensuring our actions are in harmony with our foundational values and objectives of serving the community.

#### **Programme details**

M.Pharm (Pharmacology) programme designed to enrich students' basic and advanced knowledge in the Pharmaceutical Science domain, the programme follows the courses mandated by Pharmacy Council of India (PCI) education regulations. The semester-wise course sequence and the entire M.Pharm (Pharmacology) curricula have been arranged to provide hands-on training and real-world exposure to traditional and modern practices, making graduates industry-ready. As pharmacists are true drug experts, M.Pharm (Pharmacology) students are exposed to allied science courses and core pharmaceutical courses, fostering their aptitude for research and advancements in new drug development technologies.

Rules & Syllabus for the Master of Pharmacy (M. Pharm) Course framed under Regulation of the 2014 as per by Pharmacy Council of India (PCI).

#### **Duration of the course:**

The course of study for M.Pharm (Pharmacology) shall extend over a period of four semesters (two academic years).

#### **Specific Features of the Curriculum:**

The M. Pharm curriculum is designed to align with the evolving needs of the pharmacy field and society at large. It offers a comprehensive blend of theoretical knowledge and practical applications essential for a profound understanding of pharmaceuticals, fostering the development of a wide array of skills. This curriculum is thoughtfully designed to equip students with both theoretical acumen and hands-on proficiency, catering precisely to the requirements of the dynamic industry and the broader societal demands.

#### **ELIGIBILITY Criteria:**

A Pass in the following examinations:

B. Pharm Degree examination of an Indian university established by law in India from an institution approved by Pharmacy Council of India and has scored not less than 55 % of the maximum marks (aggregate of 4 years of B.Pharm.).

Every student, selected for admission to post graduate pharmacy program in any PCI approved institution should have obtained registration with the State Pharmacy Council or should obtain the same within one month from the date of his/her admission, failing which the admission of the candidate shall be cancelled.

Note: It is mandatory to submit a migration certificate obtained from the respective university where the candidate had passed his/her qualifying degree (B.Pharm.)

#### **Program Educational Objectives (PEOs):**

- **PEO-1:** Adtu Pharmacy graduates will be well prepared for successful careers as Pharmaceutical Professionals across diverse sectors including the pharmaceutical industry, healthcare, corporate institutions and government organizations.
- **PEO-2:**Pharmacy graduates will be academically prepared to become Registered Pharmacists, poised to make significant contributions to the advancement of the healthcare sector.
- **PEO-3:**The graduates will engage in professional practices to elevate their stature with a sense of responsibility and be successful in higher education, if pursued.

#### **Programme Specific Outcomes (PSOs):**

**PSO1:** Professional Excellence: Translate the high-level of understanding of drug action into key stages in preclinical, clinical research studies and interpret data of pharmaceutical experiments in drug discovery and modifications as per the needs of pharmaceutical industries.

- **PSO2:** Practice in Research: Apply pharmacy knowledge and competency in research, and collaborative projects thereby contributing to the continuous advancement of pharmaceutical science.
- **PSO3:** International Competency: Demonstrate global professional competencies by attaining interdisciplinary knowledge through specialized certifications offered on international learning platforms.

#### **Program Outcome (POs):**

- **PO1:** Pharmacy Knowledge: Possess knowledge and comprehension of the core and basic knowledge associated with the profession of pharmacy, including biomedical sciences; pharmaceutical sciences; behavioural, social, and administrative pharmacy sciences; and manufacturing practices.
- **PO2:** Planning abilities: Demonstrate effective planning abilities including time management, resource management, delegation skills and organizational skills. Develop and implement plans and organize work to meet deadlines.
- **PO3:** Problem analysis: Utilize the principles of scientific enquiry, thinking analytically, clearly and critically, while solving problems and making decisions during daily practice. Find, analyse, evaluate and apply information systematically and shall make defensible decisions.
- **PO4:** Modern tool usage: Learn, select, and apply appropriate methods and procedures, resources, and modern pharmacy-related computing tools with an understanding of the limitations.
- PO5: Leadership skills: Understand and consider the human reaction to change, motivation issues, leadership and team-building when planning changes required for fulfilment of practice, professional and societal responsibilities. Assume participatory roles as responsible citizens or leadership roles when appropriate to facilitate improvement in health and well-being.
- **PO6:** Professional identity: Understand, analyse and communicate the value of their professional roles in society (e.g. health care professionals, promoters of health, educators, managers, employers, employees).
- PO7: Pharmaceutical ethics: Honour personal values and apply ethical principles in professional and social contexts. Demonstrate behaviour that recognizes cultural and personal variability in values, communication and lifestyles. Use ethical frameworks; apply ethical principles while making decisions and take responsibility for the outcomes associated with the decisions.
- **PO8:** Communication: Communicate effectively with the pharmacy community and with society at large, such as, being able to comprehend and write effective reports, make effective presentations and documentation, and give and receive clear instructions.

#### **Career Prospects:**

M.Pharm (Pharmacology) graduates are equipped to assume diverse roles, such as Industrial Pharmacist (in the field of Production and Manufacturing, Formulation Development, Quality Assurance, Quality Control, Packaging, R&D etc.), Hospital and Community Pharmacist, Medical Representative, Sales Executive, Bulk Medicine Distributor, Lecturer, Entrepreneurship, Drug Inspector, Drug Analyst etc. After completion of M. Pharm the students may go for higher studies in PhD programs.

#### **CHAPTER -I: REGULATIONS**

#### 1. Short Title and Commencement

These regulations shall be called as "The Revised Regulations for the Master of Pharmacy (M.Pharm) Degree Program - Credit Based Semester System (CBSS) of the Pharmacy Council of India, New Delhi". They shall come into effect from the Academic Year 2016-17. The regulations framed are subject to modifications from time to time by the authorities of the university.

#### 2. Minimum qualification for admission

A Pass in the following examinations:

B.Pharm degree examination of an Indian university established by law in India from an institution approved by Pharmacy Council of India and has scored not less than 55 % of the maximum marks (aggregate of 4 years of B.Pharm.)

Every student, selected for admission to post graduate pharmacy program in any PCI approved institution should have obtained registration with the State Pharmacy Council or should obtain the same within one month from the date of his/her admission, failing which the admission of the candidate shall be cancelled.

Note: It is mandatory to submit a migration certificate obtained from the respective university where the candidate had passed his/her qualifying degree (B.Pharm.)

#### 3. Duration of the program

The program of study for M.Pharm shall extend over a period of four semesters (two academic years). The curricula and syllabi for the program shall be prescribed from time to time by Pharmacy Council of India, New Delhi.

#### 4. Medium of instruction and examinations

Medium of instruction and examination shall be in English.

#### 5. Working days in each semester

Each semester shall consist of not less than 100 working days. The odd semesters shall be conducted from the month of June/July to November/December and the even semesters shall be conducted from the month of December/January to May/June in every calendar year.

#### 6. Attendance and progress

A candidate is required to put in at least 80% attendance in individual courses considering theory and practical separately. The candidate shall complete the prescribed course satisfactorily to be eligible to appear for the respective examinations.

#### 7. Program/Course credit structure

As per the philosophy of Credit Based Semester System, certain quantum of academic work viz. theory classes, practical classes, seminars, assignments, etc. are measured in terms of credits. On satisfactory completion of the courses, a candidate earns credits. The amount of credit associated with a course is dependent upon the number of hours of instruction per week in that course. Similarly, the credit associated with any of the other academic, co/extra-curricular activities is dependent upon the quantum of work expected to be put in for each of these activities per week/per activity.

#### 8. Credit assignment

#### 8.1 Theory and Laboratory courses

Courses are broadly classified as Theory and Practical. Theory courses consist of lecture (L) and Practical (P) courses consist of hours spent in the laboratory. Credits (C) for a course is dependent on the number of hours of instruction per week in that course, and is obtained by using a multiplier of one (1) for lecture and a multiplier of half (1/2) for practical (laboratory) hours. Thus, for example, a theory course having four lectures per week throughout the semester carries a credit of 4. Similarly, a practical having four laboratory hours per week throughout semester carries a credit of 2.

The contact hours of seminars, assignments and research work shall be treated as that of practical courses for the purpose of calculating credits. i.e., the contact hours shall be multiplied by 1/2. Similarly, the contact hours of journal club, research work presentations and discussions with the supervisor shall be considered as theory course and multiplied by 1.

#### 8.2 Minimum credit requirements

The minimum credit points required for the award of M. Pharm degree is 95. However based on the credit points earned by the students under the head of co-curricular activities, a student shall earn a maximum of 100 credit points. These credits are divided into Theory courses, Practical, Seminars, Assignments, Research work, Discussions with the supervisor, Journal club and Co-Curricular activities over the duration of four semesters. The credits are distributed semester-wise as shown in Table 6. Courses generally progress in sequence, building competencies and their positioning indicates certain academic maturity on the part of the learners. Learners are expected to follow the semester-wise schedule of courses given in the syllabus.

#### 9. Academic work

A regular record of attendance both in Theory, Practical, Seminar, Assignment, and Journal club, Discussion with the supervisor, Research work presentation and Dissertation shall be maintained by the department / teaching staff of respective courses.

#### 10. Course of study

The specialization in M.Pharm (Pharmacology) program is given in Table 1.

Table – 1: List of M.Pharm Specializations and their Code

S. No.	Specialization	Code
1.	Pharmaceutics	MPH
2.	Industrial Pharmacy	MIP
3.	Pharmaceutical Chemistry	MPC
4.	Pharmaceutical Analysis	MPA
5.	Pharmaceutical Quality Assurance	MQA
6.	Pharmaceutical Regulatory Affairs	MRA
7.	Pharmaceutical Biotechnology	MPB
8.	Pharmacy Practice	MPP
9.	Pharmacology	MPL
10.	Pharmacognosy	MPG

The course of study for M.Pharm (Pharmacology) specializations shall include Semester wise Theory & Practical as given in Table -2 to 11. The number of hours to be devoted to each theory and practical course in any semester shall not be less than that shown in Table -2 and Table -3.

Table – 2: Course of study for M.Pharm (Pharmacology)

Course	Course	Credit	Credit	Hrs./	Marks
Code		Hours	<b>Points</b>	wk	
	Semester I				
MPL101T	Modern Pharmaceutical		4	4	100
	Analytical Techniques				
MPL102T	Advanced Pharmacology-I	4	4	4	100
MPL 103T	Pharmacological and Toxicological	4	4	4	100
	Screening Methods-I				
MPL104T	Cellular and Molecular Pharmacology	4	4	4	100
MPL105P	Pharmacology Practical I	12	6	12	150
MPL106NA	MPL106NA Seminar/Assignment		4	7	100
	Total	35	26	35	650
	Semester II				
MPL201T	Advanced Pharmacology II	4	4	4	100
MPL 202T	Pharmacological and	4	4	4	100
	Toxicological Screening Methods-II				
MPL203T	Principles of Drug Discovery	4	4	4	100
MPL204T	Experimental Pharmacology	4	4	4	100
	practical- II				
MPL205P	Pharmacology Practical II	12	6	12	150
MPL206NA	Seminar/Assignment	7	4	7	100
	Total	35	26	35	650

Table - 3: Course of study for M. Pharm. III Semester

<b>Course Code</b>	Course	Credit Hours	Credit
			<b>Points</b>
MRM 301T	Research Methodology and Biostatistics*	4	4
MRM302NA	Journal club	1	1
MRM303NA	RM303NA Discussion / Presentation		2
	(Proposal Presentation)		
MRM304NA	Research Work	28	14
	Total	35	21

^{*} Non University Exam

Table – 4: Course of study for M. Pharm. IV Semester

Course Code	Course	<b>Credit Hours</b>	Credit
			Points

MRM401NA	Journal Club	1	1
MRM402NA	Discussion/Final Presentation	3	3
MRM403NA	Research Work and Colloquium	31	16
MRM404NA	Scholarly Activity		3
	Total	35	23

Table – 5: Semester wise credits distribution

Semester	Credit Points
I	26
II	26
III	23
IV	23
Co-curricular Activities/Scholarly Activities (Attending Conference, Scientific Presentations and Scholarly Activities) Credit Points will be included in IV semester and 4 credit point will be allocated for other certificate courses	Minimum=04
Total Credit Points	100

^{*}Credit Points for Co-curricular Activities

Table - 6: Guidelines for Awarding Credit Points for Co-curricular Activities

Name of the Activity	Maximum Credit Points Eligible / Activity
Participation in National Level	<b>9</b> · · · · · · · · · · · · · · · · · · ·
Seminar/Conference/Workshop/Symposium/Training Programs	01
(related to the specialization of the student)	
Participation in international Level	
Seminar/Conference/Workshop/Symposium/ Training	02
Programs (related to the specialization of the student)	
Academic Award/Research Award from State	01
Level/National Agencies	
Academic Award/Research Award from International	02
Agencies	
Research / Review Publication in National Journals	01
(Indexed in Scopus / Web of Science)	
Research / Review Publication in International Journals	02
(Indexed in Scopus / Web of Science)	

Note: International Conference: Held outside India International Journal: The Editorial Board outside India

#### 11. Program Committee

The M. Pharm programme shall have a Programme Committee constituted by the Head of the institution in consultation with all the Heads of the departments.

^{*}The credit points assigned for extracurricular and or co-curricular activities shall be given by the Principals of the colleges and the same shall be submitted to the University. The criteria to acquire this credit point shall be defined by the colleges from time to time.

The composition of the Programme Committee shall be as follows:

A teacher at the cadre of Professor shall be the Chairperson; One Teacher from each M.Pharm (Pharmacology) specialization and four student representatives (two from each academic year), nominated by the Head of the institution.

#### **Duties of the Programme Committee:**

- Periodically reviewing the progress of the classes.
- Discussing the problems concerning curriculum, syllabus and the conduct of classes.
- Discussing with the course teachers on the nature and scope of assessment for the course and the same shall be announced to the students at the beginning of respective semesters.
- Communicating its recommendation to the Head of the institution on academic matters.
- The Programme Committee shall meet at least twice in a semester preferably at the end of each sessional exam and before the end semester exam.
- Examinations/Assessments
- The schemes for internal assessment and end semester examinations are given in Table –
   8.

#### 12. End semester examinations

12.1. The End Semester Examinations for each theory and practical course through semesters I to IV shall be conducted by the respective university except for the subject with asterix symbol (*) in table I and II for which examinations shall be conducted by the subject experts at college level and the marks/grades shall be submitted to the university.

Tables -7: Schemes for internal assessments and end semester examinations (Pharmacology-MPL)

		Inter	Internal Assessment				Semester xams	Total
Course	Course	Continuous	Sessiona	l Exams				Marks
Code		Mode	Marks	Duration	Total	Marks	Duration	
	1	SEMES	STER I	1				
	Modern Pharmaceutical	10	15	1 Hr	25	75	3 Hrs	100
MPL 101T	Analytical Techniques							
MPL102T	Advanced	10						
	Pharmacology-I		15	1 Hr	25	75	3 Hrs	100
	Pharmacological and							
MPL103T	Toxicological	10	15	1 Hr	25	75	3 Hrs	100
	Screening Methods-I							
MPL104T	Cellular and Molecular	10	15	1 Hr	25	75	3 Hrs	100
	Pharmacology							
MPL105P	Experimental	20	30	6 Hrs	50	100	6 Hrs	150
	Pharmacology - I							
MPL106NA	Seminar /Assignment	-	-	-	-	-	-	100
		Total						650
	SEMESTER II							
MPL201T	Advanced							
	Pharmacology II	10	15	1 Hr	25	75	3 Hrs	100
	Pharmacological and							

MPL102T	Toxicological	10	15	1 Hr	25	75	3 Hrs	100
	Screening Methods-II							
MPL203T	Principles of							
	Drug Discovery	10	15	1 Hr	25	75	3 Hrs	100
MPL204T	Clinical research and	10	15	1 Hr	25	75	3 Hrs	100
	Pharma covigilance							
MPL205P	Experimental	20	30	6 Hrs	50	100	6 Hrs	150
	Pharmacology -II							
MPL206NA	Seminar /Assignment	-	-	-	-	-	-	100
Total					650			

 $Tables-8: Schemes \ for \ internal \ assessments \ and \ end \ semester \ examinations \ (Semester \ III \& \ IV)$ 

		Internal Assessment			Semester kams	Total		
<b>Course Code</b>	Course	Continuous	Session	al Exams				Marks
		Mode	Marks	Duration	Total	Marks	Duration	
		SEMES	TER III					
MRM30 1T	Research Methodology							
	and Biostatistics*	10	15	1 Hr	25	75	3 Hrs	100
MRM302NA	Journal club	-	-	-	25	-	-	25
MRM303NA	Discussion /							
	Presentation (Proposal	-	-	-	50	-	-	50
	Presentation)							
MRM304NA	Research work*	-	-	-	-	350	1 Hr	350
		Total						525
		SEMEST	ER IV					
MRM401NA	Journal club	-	-	-	25	-	-	25
MRM402NA	Discussion / Final					75		75
	Presentation	-	-	-	-	73		/3
MRM403NA	Research work and					400		400
WIKWI4USINA	Colloquium	_	_	_	-	400	_	400
MRM404NA	Scholarly Activity	-	-	-	-	175	-	175
Total					675			

^{*}Non University Examination

#### 12.2. Internal assessment: Continuous mode

The marks allocated for Continuous mode of Internal Assessment shall be awarded as per the scheme given below.

#### **Sessional Exams**

Two sessional exams shall be conducted for each theory / practical course as per the schedule fixed by the college(s). The scheme of question paper for theory and practical sessional examinations is given in the table. The average marks of two sessional exams shall be computed for internal assessment as per the requirements given in tables.

Table – 9: Scheme for awarding internal assessment: Continuous mode

Theory			
Criteria Cri	Maximum Marks		
Attendance (Refer Table – 28)	8		
Student – Teacher interaction	2		
Total	10		
Practical			
Attendance (Refer Table – 28	10		
Based on Practical Records, Regular viva voce, etc.	10		
Total	20		

Table – 10: Guidelines for the allotment of marks for attendance

Percentage of Attendance	Theory	Practical
95 – 100	8	10
90 – 94	6	7.5
85 – 89	4	5
80 – 84	2	2.5
Less than 80	0	0

#### 13. Promotion and award of grades

A student shall be declared PASS and eligible for getting grade in a course of M.Pharm Programme if he/she secures at least 50% marks in that particular course including internal assessment.

#### 14. Carry forward of marks

In case a student fails to secure the minimum 50% in any Theory or Practical course as specified in 12, then he/she shall reappear for the end semester examination of that course. However, his/her marks of the Internal Assessment shall be carried over and he/she shall be entitled for grade obtained by him/her on passing.

#### 15. Improvement of internal assessment

A student shall have the opportunity to improve his/her performance only once in the sessional exam component of the internal assessment. The re-conduct of the sessional exam shall be completed before the commencement of next end semester theory examinations.

#### 16. Re-examination of end semester examinations

Re-examination of end semester examination shall be conducted as per the schedule given in table 29. The exact dates of examinations shall be notified from time to time.

Table – 11: Tentative schedule of end semester examinations

Semester	For Regular Candidates	For Failed Candidates
I and III	November / December	May / June
II and IV	May / June	November / December

#### 17. Allowed to keep terms (ATKT):

No student shall be admitted to any examination unless he/she fulfils the norms given in 6. ATKT rules are applicable as follows:

A student shall be eligible to carry forward all the courses of I and II semesters till the III semester examinations. However, he/she shall not be eligible to attend the courses of IV semester until all the courses of I, II and III semesters are successfully completed.

A student shall be eligible to get his/her CGPA upon successful completion of the courses of I to IV semesters within the stipulated time period as per the norms.

Note: Grade AB should be considered as failed and treated as one head for deciding ATKT. Such rules are also applicable for those students who fail to register for examination(s) of any course in any semester.

#### 18. Grading of performances

Letter grades and grade points allocations:

Based on the performances, each student shall be awarded a final letter grade at the end of the semester for each course. The letter grades and their corresponding grade points are given in Table -30.

Table – 12: Letter grades and grade points equivalent to Percentage of marks and performances

Percentage of	Letter Grade	Grade Point	Performance
Marks Obtained			
90.00 – 100	О	10	Outstanding
80.00 – 89.99	A	9	Excellent
70.00 – 79.99	В	8	Good
60.00 – 69.99	С	7	Fair
50.00 - 59.99	D	6	Average
Less than 50	F	0	Fail
Absent	AB	0	Fail

A learner who remains absent for any end semester examination shall be assigned a letter grade of AB and a corresponding grade point of zero. He/she should reappear for the said evaluation/examination in due course.

The Semester grade point average (SGPA)

The performance of a student in a semester is indicated by a number called 'Semester Grade Point Average' (SGPA). The SGPA is the weighted average of the grade points obtained in all the courses by the student during the semester. For example, if a student takes five courses (Theory/Practical) in a semester with credits C1, C2, C3 and C4 and the student's grade points in these courses are G1, G2, G3 and G4, respectively, and then students' SGPA is equal to:

$$C1G1 + C2G2 + C3G3 + C4G4$$
  
 $SGPA = C1 + C2 + C3 + C4$ 

The SGPA is calculated to two decimal points. It should be noted that, the SGPA for any semester shall take into consideration the F and ABS grade awarded in that semester. For example, if a learner has a F or ABS grade in course 4, the SGPA shall then be computed as:

$$C1G1 + C2G2 + C3G3 + C4* ZERO$$
  
 $SGPA = C1 + C2 + C3 + C4$ 

Cumulative Grade Point Average (CGPA)

The CGPA is calculated with the SGPA of all the IV semesters to two decimal points and is indicated in final grade report card/final transcript showing the grades of all IV semesters and their courses. The CGPA shall reflect the failed status in case of F grade(s), till the course(s) is/are passed. When the course(s) is/are passed by obtaining a pass grade on subsequent examination(s) the CGPA

Shall only reflect the new grade and not the fail grades earned earlier. The CGPA is calculated as:

$$C1S1 + C2S2 + C3S3 + C4S4$$
 $CGPA = C1 + C2 + C3 + C4$ 

where C1, C2, C3, is the total number of credits for semester I,II,III,.... and S1,S2, S3,....is the SGPA of semester I,II,III......

Declaration of class

The class shall be awarded on the basis of CGPA as follows: First Class with Distinction = CGPA of, 7.50 And above

First Class =  $CGPA ext{ of } 6.00 ext{ to } 7.49$ 

Second Class =  $CGPA ext{ of } 5.00 ext{ to } 5.99$ 

Project work

All the students shall undertake a project under the supervision of a teacher in Semester III to IV and submit a report. 4 copies of the project report shall be submitted (typed & bound copy not less than 75 pages).

The internal and external examiner appointed by the University shall evaluate the project at the time of the Practical examinations of other semester(s). The projects shall be evaluated as per the criteria given below.

#### **Evaluation of Dissertation Book:**

Objective(s) of the work done 50 Marks

Methodology adopted 150 Marks

Results and Discussions 250 Marks

Conclusions and Outcomes 50 Marks

Total 500 Marks

#### **Evaluation of Presentation:**

Presentation of work 100 Marks
Communication skills 50 Marks
Question and answer skills 100 Marks
Total 250 Marks

#### **Award of Ranks**

Ranks and Medals shall be awarded on the basis of final CGPA. However, candidates who fail in one or more courses during the M.Pharm (Pharmacology)program shall not be eligible for award of ranks. Moreover, the candidates should have completed the M.Pharm (Pharmacology)program in minimum prescribed number of years, (two years) for the award of Ranks.

#### Award of degree

Candidates who fulfil the requirements mentioned above shall be eligible for award of degree during the ensuing convocation.

#### **Duration for completion of the program of study**

The duration for the completion of the program shall be fixed as double the actual duration of the program and the students have to pass within the said period, otherwise they have to get fresh Registration.

#### **Revaluation I Retotalling of answer papers**

There is no provision for revaluation of the answer papers in any examination. However, the candidates can apply for retotalling by paying prescribed fee.

#### Re-admission after break of study

Candidate who seeks re-admission to the program after break of study has to get the approval from the university by paying a condonation fee.

			SEMESTEI							
Course '			eutical Analytical Te							1
Course	code	MPL101T	Total credits: 4	L	T	P	S	R	O/F	C
D.	• •,	NT*1	Total hours: 60T	4	0	0	0	0	0	4
Pre-requ		Nil	Co-requisite	Nil						
Progra			acy (Pharmacology)							
Semes			t year of the program							
Cour		^	of course student is a	ble to Kr	iow,					
Objecti	ives		s and Excipients vsis of various drugs i	n single	and a	amhi	notion	dosas	ra forms	
		1	al and practical skills	_			ianon	uosag	c ioiiis	
COI	1		tilize the Spectrosco				Inter	nret v	various le	vels of
	L	molecular spectra	•	эру кис	Wica	ge to	mici	pret	various ic	VC15 01
CO2	<u>.                                    </u>	•	entation of N.M.R,	compa	re 11	) NN	IR. 2	DNM	R. 1HNN	AR and
		13CNMR to prop		o o i i i p u			, -	21,1,1	,	
CO ₃	<b>,</b>		ciple and Importance	e, theo	ry of	Mass	s Sne	ctrosc	opy: diffe	erentiate
		-	d ionization and Frag		-		_			
CO4		-	t Chromatographic to				•			choose
		_	sed on sample and de	_					ĺ	
CO5	;	•	electrophoretic techr						and design	gn with
			ectrophoresis, X-Ray	_			_			_
Unit-		Conte	nt	Contac	t	Lea	arnin	g Out	come	KL
No.				Hour						
I	UV-V	visible spectrosco	py: Introduction,		S	tudent	s will	be ab	le to	
	Theo	ry, Laws, Instrume	entation associated		k	now tł	neory	& App	olication	
		UV-Visible spectro		- 1			pectro	scopic		
		nts and solvent eff		te	chniq	ue.				
		cations of UV-Vis								
		ectroscopy: Theo	•							
		ecular vibrations, Sample handling,								
		-	ersive and Fourier							
	l .	form IR Spectrom								
		ting vibrational frequencies and ications of IR spectroscopy		10						2,
		t <b>roflurimetry:</b> Th								3,4,5
	_	escence, Factors a	•							
	l .	escence, Quencher	-							
	l .	applications of fluo	·							
		rophotometer.								
	_	e emission spectr	oscopy and							
		nic absorption spe								
	l .	iple, Instrumentati								
		pplications.								
II		R spectroscopy: Q		10	S	tudent	s will	be ab	le to	
	and tl	neir role in NMR,	Principle,			now b		_		
		mentation, Solven	-			neory l				
		, Relaxation proce	-		1 ^	redict		_	-	3,4,5
		rious compounds,				pplica		o orga	nic	3,1,3
		rs influencing che	•		C	ompou	ınds.			
	_		g constant, Nuclear							
	magn	etic double resona	nce, Brief outline							

	of principles of FT-NMR and 13CNMR.			
	Applications of NMR spectroscopy.			
III	Mass Spectroscopy: Principle, Theory,	10	Students will be able to	
	Instrumentation of Mass		learn identify which	
	Spectroscopy, Different types of		Ionization technique is	
	ionization like electron impact, chemical,		suitable for compounds and	
	field, FAB and MALDI, APCI, ESI, APPI		analyze the type of Analyzer	3,4,5
	Analyzers of Quadruple and Time of		to be used depending on the	
	Flight, Mass fragmentation and its rules,		sample.	
	Meta stable ions, Isotopic peaks and			
	Applications of Mass spectroscopy			
IV	Chromatography: Principle, apparatus,	10	Students will be able to	
	instrumentation, chromatographic		understand theory &	
	parameters, factors affecting resolution		application of different	
	and applications of the following:		chromatographic	
	a) Paper chromatography		techniques.	
	b) Thin Layer chromatography			
	c) Ion exchange chromatography			3,4,5
	d)Column chromatography			
	e) Gas chromatography			
	f) High Performance Liquid			
	chromatography			
	g) Affinity chromatography			
	h) Gel Chromatography			
V	a. Electrophoresis: Principle,	10	Students will be able to	
	Instrumentation, working conditions,		Understand, Principle,	
	factors affecting separation and		Instrumentation, working	
	applications of the following:		conditions, factors affecting	
	a) Paper electrophoresis		separation and applications	
	b) Gel electrophoresis		of electrophoresis.	
	c) Capillary electrophoresis			
	d) Zone electrophoresis			3,4,5
	e) Moving boundary electrophoresis			3, 1,5
	f) Iso electric focusing			
	b. X ray Crystallography: Production of			
	X rays, Different X ray diffraction			
	methods, Bragg's law, Rotating crystal			
	technique, X ray powder technique, Types			
	of crystals and applications of X ray			
	diffraction.	1.0		
VI	Potentiometry: Principle, working, Ion	10	Students will be able to	
	selective Electrodes and Application of		know importance of RIA	
	potentiometry. Thermal Techniques:		and ELISA techniques and	
	Principle, thermal transitions and		their applications.	
	Instrumentation (Heat flux and power-			3,4,5
	compensation and designs), Modulated			
	DSC, Hyper DSC, experimental			
	parameters (sample			
	Preparation, experimental conditions,			
	calibration, heating and cooling rates,			

resolution, source of errors) and their	
influence, advantage and disadvantages,	
pharmaceutical applications. Differential	
Thermal Analysis (DTA): Principle,	
instrumentation and advantage and	
disadvantages, pharmaceutical	
applications, derivative differential	
thermal analysis (DDTA). TGA: Principle,	
Instrumentation, factors affecting results,	
advantage and disadvantages,	
pharmaceutical applications.	

#### **TEXT BOOKS:**

#### **REFERENCE BOOKS:**

- R1. Spectrometric Identification of Organic compounds Robert M Silverstein, Sixth edition, John Wiley & Sons, 2004.
- R2. Principles of Instrumental Analysis Douglas A Skoog, F. James Holler, Timothy A. Nieman, 5th edition, Eastern press, Bangalore, 1998.
- R3. Instrumental methods of analysis Willards, 7th edition, CBS publishers.
- R4. Practical Pharmaceutical Chemistry Beckett and Stenlake, Vol II, 4thedition, CBS Publishers, New Delhi, 1997.
- R5. Organic Spectroscopy William Kemp, 3rd edition, ELBS, 1991.
- R6. Quantitative Analysis of Drugs in Pharmaceutical formulation P D Sethi, 3rd Edition, CBS Publishers, New Delhi, 1997.
- R7. Pharmaceutical Analysis Modern Methods Part B J W Munson, Vol11, Marcel. Dekker Series
- R8. Spectroscopy of Organic Compounds, 2nd edn., P.S/Kalsi, Wiley easternLtd., Delhi.
- R9. Textbook of Pharmaceutical Analysis, KA. Connors, 3rd Edition, John Wiley & Sons, 1982.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Compare and Utilize the Spectroscopy knowledge to Interpret various levels of molecular spectra.	PO1,PO2,PO3,PO4,PO5,PO8
2	Analyse instrumentation of N.M.R, compare 1D NMR, 2DNMR, 1HNMR and 13CNMR to Propose structures.	PO1,PO2, PO3, PO4, PO5, PO8
3	Assume the principle and Importance, theory of Mass, spectroscopy; differentiate different peaks and ionization and Fragmentation rules to Predict the structure.	PO1,PO2, PO3, PO4, PO5, PO8
4	Compare different Chromatographic techniques and Evaluate instrumentation, choose each technique based on sample and Develop a Validated method.	PO1,PO2, PO3, PO4, PO5, PO8
5	Justify different electrophoretic techniques, X-ray crystallography and design with Importance of Electrophoresis, X-Ray Crystallography and Bioluminescence assays.	PO1,PO2, PO3, PO4, PO5, PO8

SEMESTER – I  Course Title   ADVANCED PHARMACOLOGY – I											
Course T	itle					1	1				
Course c	ode	MPL102T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 60T	4	0	0	0	0	0	4	
Pre-requ		Nil	Co-requisite	Ni	il						
Progra			cy (Pharmacology)								
Semest	er		year of the program								
			of the course, student shall b								
Cours	e		e path physiology and pharr								
Objectiv	ves	_	e mechanism of drug action								
			d the adverse effects, contra	indic	atıc	ns a	and c	linic	al uses	of di	ugs
601			atment of diseases	٠ 1		1			. ,	41 1	1
CO1			ic knowledge in the field of	phari	mac	colo	gy po	ertan	ning to	the d	lrugs
		and its therapeutic				1 C.	41	44		C	•
CO2		· ·	the recent advances in the d	rugs ı	use	1 IOI	r tne	treat	ment o	ı var	ious
CO2		diseases.	naonte of drug action and	aahaa	iar	20 :	17701-	704			
CO3			ncepts of drug action and medarlying mechanism of drug						d ma1-	01110	laval
CO4			derlying mechanism of drug effects, contraindications an								
CO5		of diseases	effects, contraindications an	ia ciii	lica	ıı us	es oi	aruş	gs usea	m u	eaument
Unit-No.			ontent	Cor	110	ot		Lac	rning		KL
UIIIt-140.		Co	ontent		our				come		KL
I	Gen	eral Pharmacolog	v	110	oui		Stud		will be		
1		narmacokinetics: Th	-						now ab		
			, biotransformation and						okineti		
		-	of linear and non-linear				&				
		-	ignificance of Protein				Phar	mac	odynan	nics	
	bind	•	8	12					J		2,3,4,5
		·	Mechanism of drug action								
	and	the relationship bet	ween drug concentration								
	and	effect. Receptors, s	tructural and functional								
	fami	lies of receptors, qu	uantitation of drug								
	rece	ptors interaction and	d elicited effects.								
II	Neu	rotransmission		12			Stud	lents	will be	!	
	a. G	eneral aspects and s	steps involved in				able	to k	now ab	out	
	neur	otransmission.					Neu	rohu	moral		
			nission in autonomic						sion in		
		ous system(Detaile	· · · · · · · · · · · · · · · · · · ·				auto	nom	ic nerv	ous	
		otransmitters- Adre	enaline and Acetyl				syste				
		line).							numora	l	
			nission in central nervous						sion in		
system (Detailedstudy		•							ervous		2,3,4,5
histamine, serotonin, do			•				syste	em			
	_	amate and glycine].									
		_	cholinergic transmission								
(NANC). Co-transmission  Systemic Pharmacology											
	_										
		etaned study on pati hanism ofaction, ph	h physiology of diseases,								
		-	s well as noveldrugs used								
		e following system									
	լու ա	c ronowing system	ט								

	Autonomic Pharmacology			
	Parasympathomimetics and lytic,			
	sympathomimetics and lytic, agents affecting			
	neuromuscular junction			
III	Central Nervous System Pharmacology	12	Students will be	
	General and local anesthetics Sedatives and		able to learn about	
	hypnotics, drugs used to treat anxiety.		General and local	
	Depression, psychosis, mania, epilepsy, neuro-		anesthetics	2,3,4,5
	degenerative diseases. Narcotic and non-narcotic		Sedatives and	2,3,4,3
	analgesics		hypnotics, Narcotic	
			and non-narcotic	
			analgesics	
IV	Cardiovascular Pharmacology	12	Students will be	
	Diuretics, antihypertensive, anti-ischemic, anti-		able to learn	
	arrhythmic, drugs for heart failure and		mechanism,	
	hyperlipidemia. Hematinic, coagulants,		classification &	
	anticoagulants, fibrinolytics and anti- platelet		Pharmacological	2,3,4,5
			action of Diuretics,	
			antihypertensive,	
			anti-ischemic, anti-	
			arrhythmic, drugs	
V	Autocoid Pharmacology	12	Students will be	
	The physiological and pathological role of		able to understand	
	Histamine, Serotonin, Kinins Prostaglandins		the physiological	2,3,4,5
	Opioid autocoids. Pharmacology of		and pathological	
	antihistamines, 5HT antagonist		role of Histamine	

### **TEXT BOOKS:**

#### **REFERENCE BOOKS:**

- R1. The Pharmacological Basis of Therapeutics, Goodman and Gillman's
- R2. Principles of Pharmacology. The Pathophysiologic basis of drug Therapyby David E Golan, Armen H, TashjianJr, EhrinJ, Armstrong, April W, Armstrong, Wolters, Kluwer-Lippincott Williams & Wilkins Publishers.
- R3. Basic and Clinical Pharmacology by B.G Katzung
- R4. Hand book of Clinical Pharmacokinetics by Gibaldi and Prescott.
- R5. Applied biopharmaceutics and Pharmacokinetics by Leon ShargelandAndrewB.C.Yu.
- R6. Graham Smith. Oxford textbook of Clinical Pharmacology.
- R7. Avery Drug Treatment
- R8. Dipiro Pharmacology, Pathophysiological approach.
- R9. Green Pathophysiology for Pharmacists.
- R10. Robbins &Cortan Pathologic Basis of Disease, 9th Ed. (Robbins Pathology)
- R11. A Complete Textbook of Medical Pharmacology by Dr. S.K Srivastavapublished by APC Avichal Publishing Company
- R12. KD. Tripathi. Essentials of Medical Pharmacology.
- R13. Modern Pharmacology with Clinical Applications, Craig Charles R. &StitzelRobert E., Lippincott Publishers.
- R14. Clinical Pharmacokinetics & Pharmacodynamics: Concepts and Applications Malcolm Rowland and Thomas N.Tozer, Wolters Kluwer, Lippincott Williams & Wilkins Publishers.
- R15. Applied biopharmaceutics and Pharmacokinetics, Pharmacodynamics andDrug metabolism for industrial scientists.
- R16. Modern Pharmacology, Craig CR. & Stitzel RE, Little Brown & Company

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Appreciate the basic knowledge in the field of pharmacology	PO1, PO3, PO4, PO5, PO7,
1	pertaining to the drugs and its therapeutic applications.	PO8
2	Elaborately learnt the recent advances in the drugs used for the	PO1, PO3, PO4, PO5,PO7,PO8
	treatment of various diseases.	101,103,104,103,107,108
3	Understood the concepts of drug action and mechanisms	PO1, PO3, PO4, PO5,PO7,PO8
3	involved.	101,103,104,103,107,108
4	Understood the underlying mechanism of drug actions at	PO1, PO3, PO4, PO5,PO7,PO8
<b>-</b>	cellular and molecular level.	101,103,104,103,107,108
5	Learn the adverse effects, contraindications and clinical uses of	PO1, PO3, PO4, PO5,PO7,PO8
3	drugs used in treatment of diseases	101,103,104,103,107,108

			SEMESTE							
Course	Title	Pharmacological and	l toxicological screeni	ng metl	hods -	- I				
Course	code	MPL103T	Total credits: 4	L	T	P	S	R	O/F	C
			Total hours: 60T	4	0	0	0	0	0	4
Pre-req	uisite	Nil	Co-requisite					Nil		
Progr	am	Master of Pharmacy	(Pharmacology)							
Semes	ster	I semester of first ye	ar of the program							
Cour	·se	Upon completion of t	he course, student sha	ll be ab	le to					
Object	ives	1. Appraise the reg	ulations and ethical re	quirem	ent fo	r the	usag	e of e	xperimen	tal animals.
			ous animals used in the drug discovery process and good laboratory							
		_	ntenance and handling	_						
			ious newer screening i		s invo	lved	in th	e drug	g discover	y process
			correlate the preclinica							
CO			e of laboratory animal						U	
		_ ·	ogical and toxicologic	al resea	arch, s	servi	ng as	inval	uable too	ls for
			new chemical entities.							
CO2			trate the comprehensi	_						ew
			central and autonomic		•				_	4-14:
		vitro methods.	mer's, and multiple sc	1610818	by un	IIZIII	g vai	ious a	illilliai illo	deis and m
CO3			trate the comprehensi	ve <b>nr</b> ec	linica	l scre	enin	a nroc	ess for ne	NX7
			-	_						, vv
		substances targeting central and autonomic nervous system diseases, including Parkinsonism, Alzheimer's, and multiple sclerosis through the utilization of various animal								
		models and in vitro methods, as well as exploring alternative models to ensure a thorough								
			al therapeutic interven	_						S
CO		Compare and demonstrate the preclinical screening methods for various new chemical								
		entities targeting cardiovascular, endocrine, immune, and digestive system diseases,								
		hypertension, arrhythmia, angina pectoris, atherosclerosis, diabetes, dyslipidemia, cancer,								
		and hepatotoxicity. E	valuation involves a combination of diverse animal models, in vitro							
		assays, and potential alternative models to comprehensively assess the safety and efficacy of								
		chemical entities before transitioning to clinical trials.								
CO			e preclinical screening	_						
			and immunostimula	-					-	
** • ·			ell as in in-vitro and o							
Unit-		Conte	nt		tact	L	_earn	ing O	outcome	KL
No.	I ak-	mataux A mim ala		H	our	CL	ıdarıtı		be able	
1		oratory Animals mon laboratory anima	1c. Description				know		de able	
		ling and applications	•						nd ethical	
		trains of animals.	or different species			_			or the	
Trai		sgenic animals:				1 1	-		rimental	
		luction, maintenance a	and applications				mals.	_		
		sthesia and euthanasia		1	2		-3.			2,3,4,5
anin			breeding of laboratory	7						
		als. CPCSEA guidelir								
	expe	riments on animals								
		l laboratory practice								
		ssay-Principle, scope	and limitations and							
	meth	ods								

П	Preclinical screening of new substances for the		Students will be able	
11	pharmacological activity using in vivo, in vitro,		to understand the	
	and other possible animal alternative models.		various newer	
	General principles of preclinical screening. CNS		screening methods	
	Pharmacology: behavioral and muscle co-		involved in the drug	
		12	discovery process	2,3,4,5
	ordination, CNS stimulants and depressants,		discovery process	
	anxiolytics, anti-psychotics, anti-epileptics and			
	no tropics. Drugs for neurodegenerative diseases			
	like Parkinsonism, Alzheimer's and multiple			
	sclerosis. Drugs acting on Autonomic Nervous		G: 1 : '111 11	
III	Preclinical screening of new substances for the		Students will be able	
	pharmacological activity using in vivo, in vitro,		to know the various	
	and other possible animal alternative models.		newer screening	
	Respiratory Pharmacology: anti-asthmatics,		methods involved in	
	drugs for COPD and antiallergics. Reproductive	12	the drug discovery	2,3,4,5
	Pharmacology: Aphrodisiacs andant fertility		process	
	agents Analgesics, anti-inflammatory and			
	antipyretic agents. Gastrointestinal drugs: anti-			
	ulcer, anti -emetic, anti- diarrheal and laxatives.			
IV	Preclinical screening of new substances for the		Students will be able	
	pharmacological activity using in vivo, in vitro,		to understand the	
	and other possible animal alternative models.		various newer	
	Cardiovascular Pharmacology: antihypertensive,		screening methods	
	antiarrhythmics, antianginal, ant atherosclerotic	12	involved in the drug	2,3,4,5
	agents and diuretics. Drugs for metabolic		discovery process	
	disorders like anti-diabetic, antidyslipidemic			
	agents. Anti-cancer agents. Hepatoprotective			
	screening			
V	Preclinical screening of new substances for the		Students will be able	
	pharmacological activity using in vivo, in vitro,		to learn the various	
	and other possible animal alternative models.		newer screening	
	Immunomodulators, Immunosuppressant's and		methods involved in	
	immunostimulants General principles of		the drug discovery	
	immunoassay: theoretical basis and optimization		process	
	of immunoassay, heterogeneous and			
	homogenous immunoassay systems.	12		2,3,4,5
	Immunoassay methods evaluation; protocol			
	outline, objectives and preparation.			
	Immunoassay for Dioxin and insulin Limitations			
	of animal experimentation and alternate animal			
	experiments. Extrapolation of in vitro data to			
	preclinical and preclinical to humans			
	precimical and precimical to humans			

#### **TEXT BOOKS:**

## **REFERENCE BOOKS:**

- R1. Biological standardization by J.H. Burn D.J. Finney and I.G. Goodwin
- R2. Screening methods in Pharmacology by Robert Turner. A
- R3. Evaluation of drugs activities by Laurence and Bachrach
- R4. Methods in Pharmacology by Arnold Schwartz.
- R5. Fundamentals of experimental Pharmacology by M.N.Ghosh
- R6. Pharmacological experiment on intact preparations by Churchill Livingstone

- R7. Drug discovery and Evaluation by Vogel H.G.
- R8.Experimental Pharmacology by R.K.Goyal.
- R9. Preclinical evaluation of new drugs by S.K. Guta
- R10. Handbook of Experimental Pharmacology, SK.Kulkarni
- R11. Practical Pharmacology and Clinical Pharmacy, SK.Kulkarni, 3rdEdition.
- R12. David R.Gross. Animal Models in Cardiovascular Research, 2ndEdition, Kluwer Academic Publishers, London, UK.
- R13. Screening Methods in Pharmacology, Robert A.Turner.
- R14. Rodents for Pharmacological Experiments, Dr. Tapan Kumar Chatterjee.
- R15. Practical Manual of Experimental and Clinical Pharmacology by BikashMedhi (Author), Ajay Prakash (Author)

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Recall the crucial role of laboratory animal models both conventional and transgenic in advancing pharmacological and toxicological research, serving as invaluable tools for screening drugs and new chemical entities.	PO1. PO2, PO3, PO4, PO6, PO7, PO8						
2	Illustrate and demonstrate the comprehensive preclinical screening process for new substances targeting central and autonomic nervous system diseases, including Parkinsonism, Alzheimer's, and multiple sclerosis by utilizing various animal models and in vitro methods.	PO1. PO2, PO3, PO4, PO6, PO7, PO8						
3	Illustrate and demonstrate the comprehensive preclinical screening process for new substances targeting central and autonomic nervous system diseases, including Parkinsonism, Alzheimer's, and multiple sclerosis through the utilization of various animal models and in vitro methods, as well as exploring alternative models to ensure a thorough evaluation of potential therapeutic interventions.	PO1. PO2, PO3, PO4, PO6, PO7, PO8						
4	Compare and demonstrate the preclinical screening methods for various new chemical entities targeting cardiovascular, endocrine, immune, and digestive system diseases, hypertension, arrhythmia, angina pectoris, atherosclerosis, diabetes, dyslipidemia, cancer, and hepatotoxicity. Evaluation involves a combination of diverse animal models, in vitro assays, and potential alternative models to comprehensively assess the safety and efficacy of chemical entities before transitioning to clinical trials.	PO1. PO2, PO3, PO4, PO6, PO7, PO8						
5	Explain and analyse preclinical screening of new substances like immunomodulators, immunosuppressants and immunostimulants by various immunoassays conducted in different animal as well as in invitro and other possible animal alternative models.	PO1. PO2, PO3, PO4, PO6, PO7, PO8						

		SEM	IESTER-	<u> </u>					
Course Ti	tle Cellular and	Molecular Pharmacolo	ogy						
Course co	de MPL104T	Total credits: 4	L T	P	P	S	R	O/F	C
		Total hours: 60T	3 1	0	0	0	0	0	4
Pre-requis	site Nil	Co-requisite	Nil	•			•		•
Progran	n Master of Ph	narmacy (Pharmacolog	y)						
Semeste	r I semester of	f first year of the progr	am						
Course	Upon comple	etion of the course, it is	expected	that th	he st	udents	will be a	ble to	
Objective	es 1. Expl	ain the receptor signal	transduction	on pro	ocess	ses.			
	2. Expl	ain the molecular pathy	ways affec	ted by	y dru	ıgs.			
	3. Appr	reciate the applicability	of molect	ılar pl	harm	nacolog	y and bio	omarkers i	n drug
	disco	overy process.							
		onstrate molecular biol		_					
CO1	1 -	ous cellular events, fun	ctions, pat	hways	s and	l transd	uction m	echanisms	s and how
	a gene is exp								
CO2		l signaling pathways ba		_				-	
CO3		bout principles and app	plications	of gen	nomi	c; prote	eomic too	ols, gene tl	nerapy and
	rDNA techno								
CO4		mmunotherapeutic and							
CO ₅	_	nciples and applications	s of variou	s assa	ays, ł	oiosimi	lars, cell	culture tec	chniques,
		f flow cytometry.							
Unit-	(	Content	Contac	t		Learni	ing Outc	ome	KL
No.			Hour						
I	Cell biology					be able			
		renome organization. Gene expression and its regulation,			understand the Structure and functions of cell and its				
	organelles								
	-				organelles				
	-					_	nization		
	_	RNA and micro	12		_		nd its reg		2,3,4,5
	RNA, gene map	ping and gene			•		f si RNA		,-,,-
	sequencing							oping and	
	Cell cycles and	_		ge	ene s	sequenc	ing		
	Cell death— ever								
		rinsic pathways of							
77		osis and autophagy		G :	4 1	11	11.1	h -	
II	Cell signaling	l intro o allesta a si a a a t					be able t		
		l intracellular signaling					learn Co	<b>711</b>	
	family and mole	ification of receptor		51	ignal	ınıg			
	•	channels; G-protein							
		rs, tyrosine kinas							
	receptors and nu	•							
	•	engers: cyclic AMP,							
	· ·	cium ion,inositol	12						2,3,4,5
	· ·								
		4,5-trisphosphate, (IP3), NO, and acylglycerol. <b>Detailed study of</b>							
	following intra	· · · · · · · · · · · · · · · · · · ·							
	signalingpathw								
	CyclicAM Psigr	•							
	•	ed protein kinase							
	(MAPK)signalir	•							

	(JAK)/signal transducer and activator of transcription (STAT) signaling			
III	Principles and applications of genomic and proteomic tools DNA electrophoresis, PCR (reverse transcription and real time), Gene sequencing, micro array technique, SDS page, ELISA and Western blotting, Recombinant DNA technology and gene therapy Basic principles of recombinant DNA technology-Restriction enzymes, various types of vectors. Applications of recombinant, DNA technology. Gene therapy- Various types of gene transfer techniques, clinical applications and recent advances in gene therapy.	12	Students will be able to understand the Principles and applications of genomic and proteomic tools DNA electrophoresis, PCR (reverse transcription and real time), Gene sequencing, micro array technique, SDS page, ELISA and western blotting, Recombinant DNA technology	2,3,4,5
IV	Pharmacogenomics Gene mapping and cloning of disease gene. Genetic variation and its role in health/ pharmacology, Polymorphisms affecting drug metabolism, Genetic variation in drug transporters, Genetic variation in G protein coupled receptors, Applications of proteomics science: Genomics, proteomics, metabolomics, functionomics, nutrigenomics Immunotherapeutic Types of immunotherapeutic, humanization antibody therapy, Immunotherapeutics in clinical practice	12	Students will be able to know about Pharmacogenomics & Immunotherapeutic	2,3,4,5
V	a. Cell culture techniques Basic equipment used in cell culture lab. Cell culture media, various types of cell culture, general procedure for cell cultures; isolation of cells, subculture, cryopreservation, characterization of cells and their application. Principles and applications of cell viability assays, glucose uptake assay, Calcium influx assays, Principles and applications of flow cytometry b. Biosimilars	12	Students will be able to learn about Cell culture techniques	2,3,4,5

- R1. The Cell, A Molecular Approach. Geoffrey M Cooper.
- R2. Pharmacogenomics: The Search for Individualized Therapies. Edited by J. Licinio and M -L. Wong.
- R3. Handbook of Cell Signaling (Second Edition) Edited by Ralph A. et.al.
- R4. Molecular Pharmacology: From DNA to Drug Discovery. John Dickensonet.al.
- R5. Basic Cell Culture protocols by CherilD.Helgason and Cindy L.Miller.
- R6. Basic Cell Culture (Practical Approach) by J. M. Davis (Editor).
- R7. Animal Cell Culture: A Practical Approach by John R. Masters (Editor).
- R8. Current protocols in molecular biology vol I to VI edited by FrederickM. Ausuvel et al.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Explain various cellular events, functions, pathways and transduction mechanisms and how a gene is expressed.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
2	Illustrate Cell signalling pathways based on receptors and second messengers in the cell.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
3	Knowledge about principles and applications of genomic; proteomic tools, gene therapy and rDNA technology.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
4	Understand Immunotherapeutic and application of omics in clinical practice.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
5	Describe principles and applications of various assays, biosimilars, cell culture techniques, application of flow cytometry.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8

		SEMESTE	R-I							
Course Title	Pharmacological 1	Practical –I								
Course code	MPL105P	Total credits: 6	L	T	P	S	R	O/F		C
		Total hours: 12	0	0	12	0	0	0		6
Pre-requisite	Nil	Co-requisite			1	-	Nil		-	
Program	Master of Pharma	cy (Pharmacology)								
Semester	I semester of first	year of the program								
Course	Upon completion of	of the course, it is exp	ected	that tl	ne stu	dents	will	be able t	o u	ınderstand
Objectives										
CO1	Demonstrate comp	rehension of the anal	ysis of	f Phar	maco	peial	comp	pounds a	nd	their
	formulations throu	gh the utilization of U	JV Vis	s spec	trosco	ору.				
CO2	Understand the im	portance and process	impor	tance	anim	al ex	perim	entation.		
CO3	The students will r	eveal their understand	ling o	f Higl	n-Perf	forma	nce I	Liquid		
	Chromatography (	HPLC) experiments to	nrougl	h the	applic	ation	of ar	nalytical	ski	lls and
	knowledge, showc	asing their ability to e	valua	te and	inter	pret	comp	lex proce	du	res.
CO4	Establish their ana	lytical skills by exam	ning	differ	ent dr	ug ar	alysi	s, estima	tio	n,
CO5	The mastery of the	oretical and practical	skills	assoc	iated	with	adva	nce analy	tic	al
	techniques and ins	truments								
Unit-No.	C	ontent	Co	ontac	t Lo	earni	ing O	utcome		KL
			I	Hour						
I	1. Analysis of pha	_			St	uden	ts wil	l be able		
	compounds and the	heir formulations by			to	Han	ds-on	with		
	UV Visspectroph				va	rious	spec	troscopio	2	
	2. Simultaneous estimation of multi							ographic		
	component containing formulations by						_	Handling		
	UV spectrophoto	· · · · · · · · · · · · · · · · · · ·					-	y animals		
	3. Experiments ba						_	erimental		
	4. Experiments ba	ased on Gas					acolog			
	Chromatography						onstr			
	5. Estimation of r	•						ening		
	sulphate by fluori						ls use			
		odium/potassium by			pr	eclin	ical r	esearch.		
	flame photometry									
	Handling of labor	•								
	1. Various routes	of drug								22456
	administration.	1.1 1 1.		12						2,3,4,5,6
	2. Techniques of anesthesia and eu									
	experimentalanin									
	(modified Irwin t	ervation battery tests								
	4. Evaluation of C									
		genics and anxiolytic								
	anticonvulsant ac	-	'							
	5. Evaluation of	•								
	inflammatory, loc	~								
	mydriaticandmio									
	6. Evaluation of o	· · · · · · · · · · · · · · · · · · ·								
		intiulcer activity by								
	pylorus ligation n	· · · · · · · · · · · · · · · · · · ·								
	8. Oral glucose to									
	o. Oral glucose it	nerance test.								

9. Isolation and identification of DNA		
from various sources (Bacteria,		
Cauliflower, onion, Goat liver).		
10. Isolation of RNA from yeast		
11. Estimation of proteins by		
Braford/Lowry's in biological samples.		
12. Estimation of RNA/DNA by UV		
Spectroscopy		
13. Gene amplification by PCR.		
14. Protein quantification Western		
Blotting.		
15. Enzyme based in-vitro assays		
(MPO, AChEs, α amylase, α		
glycosidase).		
16. Cell viability assays (MTT/Trypan		
blue/SRB).		
17. DNA fragmentation assay by		
agarose gel electrophoresis.		
18. DNA damage study by Comet		
assay.		
19. Apoptosis determination by		
fluorescent imaging studies.		
20. Pharmacokinetic studies and data		
analysis of drugs given by different		
routes of administration using		
software's.		
21. Enzyme inhibition and induction		
activity		
22. Extraction of drug from various		
biological samples and estimation of		
drugs in biological fluids using		
different analytical techniques (UV)		
23. Extraction of drug from various		
biological samples and estimation of		
1 . 1.1 . 10.1 .		ĺ

R1. CPCSEA, OECD, ICH, USFDA, Schedule Y, EPA guidelines,

different analytical techniques (HPLC)

R2. Fundamentals of experimental Pharmacology by M.N.Ghosh

drugs in biological fluids using

- R3. Handbook of Experimental Pharmacology by S.K. Kulkarni.
- R4. Drug discovery and Evaluation by Vogel H.G.
- R5. Spectrometric Identification of Organic compounds Robert M Silverstein.
- R6. Principles of Instrumental Analysis Doglas A Skoog, F. James Holler, Timothy A. Nieman.
- R7. Vogel's Text book of quantitative chemical analysis Jeffery, Basset, Mendham, Denney.
- R8. Basic Cell Culture protocols by Cheril D. Helgason and Cindy L.Mille.
- R9. Basic Cell Culture (Practical Approach ) by J. M. Davis (Editor).
- R10. Animal Cell Culture: A Practical Approach by John R. Masters (Editor).
- R11. Practical Manual of Experimental and Clinical Pharmacology by BikashMedhi(Author), Ajay Prakash (Author) Jaypee brothers' medical publishersPvt.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
	Demonstrate comprehension of the analysis of	
1	Pharmacopeial compounds and their formulations	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	through the utilization of UV Vis spectroscopy.	
2	Understand the importance and process importance	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	animal experimentation.	101,102,103,104,103,100,107,108
	The students will reveal their understanding of High-	
	Performance Liquid Chromatography (HPLC)	
3	experiments through the application of analytical skills	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	and knowledge, showcasing their ability to evaluate and	
	interpret complex procedures.	
4	Establish their analytical skills by examining different	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
4	drug analysis, estimation,	101,102,103,104,103,100,107,108
	The mastery of theoretical and practical skills	
5	associated with advance analytical techniques and	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
	instruments	

			SEMESTE	ER – II	[						
Course		Advanced pharmac									
Course	code	MPL201T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 60T	3	1	0	0	0	0	4	
Pre-ree		Nil	Co-requisite					N	il		
Prog		Master of Pharmac									
Seme			year of the program								
Cou		_	f course student is ab								
Objec	ctives	_	mechanism of drug								
			Path physiology and	_							1
			the adverse effects, t of diseases.	contra	inaica	ation	is and	ı ciin	icai uses	oi aru	igs used
CC	<b>71</b>		t of diseases.  It of diseases.	achanie	em of	Cocti	on of	Aiffe	rant har	mones	and
	<i>J</i> 1		the cellular and mole				011 01	uiiic	Tent non	mones	anu
CC	)2		ism of action and ad				ions	25500	riated wit	th drug	rs used in
	<i>)</i> <u></u>		s, helminthiasis, and								
			B drugs, on a cellula								
		development of dru	-							J	
CC	)3		o elucidate the path p	hysiol	logy,	thera	ару, а	and a	dverse dr	ug rea	ctions of
		drugs used in cance	r, inflammation, alle	rgy, as	sthma	, and	CO	PD as	s well as	unders	stand the
		principles behind in	nmune suppressants	and im	nmune	e stir	nulaı	nts.			
CC	)4		nysiology, therapy, a			_					_
		r	etic, anti-diarrheal, a		-		_				wel
			aluate the application								
CC	)5		f free radical generat								
		_	ortance of the protec		-					-	
		knowledge on receidiabetes.	nt advances in the tre	atmen	t oi n	euro	dege	nerati	ive diseas	ses, ca	ncer, and
Unit-		Content	C	ontact	t I	L	arni	ing (	utcome		KL
No.		Content		Hour				ing C	utcome		
I	Endo	crine Pharmacolog			Stu	dent	s wil	l be a	ble to		
		ular and cellular me	·						anism of	drug	
	action	of hormones such a	s growth		acti	ons	at ce	llular	and	C	
	hormo	one, prolactin, thyroi	d, insulin and		mol	lecul	ar le	vel			
	sex ho	rmones		12							2,3,4,5,
		hyroid drugs, Oral h									
	•	, Oral contraceptive									
		osteroids. Drugs aff	ecting calcium								
	regula				G:	1 .	• • •	1.1	1.1 -		
II		otherapy	ahaniam of						ble to	dm:-	
		ar and molecular me s and resistance of a			_			necna Ilular	anism of	urug	
			12			ar le		anu		2,3,4,5,	
	_	such as ß-lactams, sides, quinolones,	wiiiiii0	14	11101	ccul	a1 10	, 01			∠,∍,¬,∍,
		olideantibiotics. Anti	fungal, antiviral.								
		nti-TB drugs									
III		otherapy			Stu	dent	s wil	l be a	ble to kn	ow	
			fections: Drugs		the	Dru	gs us	ed in	the treat	ment	
1		used in 1 lotozodi ii	ed in Protozoal Infections; Drugs the Drugs used in the treatment								
	_	n the treatment of H	_	12			-	asis,			2,3,4,5,
	used in		elminthiasis,	12	of I	Helm emot	inthi herap	asis,	cancer,		2,3,4,5,

IV	biochemical mediators of inflammation and immune response. Allergic or Hypersensitivity reactions. Pharmacotherapy of asthma and COPD. Immunosuppressants and Immunostimulants.  GIT Pharmacology Antiulcer drugs, Prokinetics, anti-emetics,		Students will be able to know the Path physiology and	
	anti-diarrheal and drugs for constipation and irritable bowel syndrome. Chronopharmacology, Biological and circadian rhythms, applications of chronotherapyin various diseases like cardiovascular disease, diabetes, asthma and peptic ulcer.	12	pharmacotherapy of certain diseases Understand the adverse effects, contraindications and clinical uses of drugs used in treatment of diseases	2,3,4,5,
V	Free radicals Pharmacology Generation of free radicals, role of free radicals in etiopathology of various diseases such as diabetes, neurodegenerative diseases and cancer. Protective activity of certain important antioxidant  Recent Advances in Treatment: Alzheimer's disease, Parkinson's disease, Cancer, Diabetes mellitus.	12	Students will be able to understand generation of free radicals, role of free radicals in etiopathology of various diseases such as diabetes, neurodegenerative diseases and cancer.	2,3,4,5,

- R1. The Pharmacological basis of therapeutics- Goodman and Gill man's.
- R2. Principles of Pharmacology. The Path physiologic basis of drug therapy by David E Golan et al.
- R3. Basic and Clinical Pharmacology by B.G -Katzung.
- R4. Pharmacology by H.P. Rang and M.M. Dale.
- R5. Hand book of Clinical Pharmacokinetics by Gibaldi and Prescott.
- R6. Text book of Therapeutics, drug and disease management by E T. Herfindal and Gourley.
- R7. Applied biopharmaceutics and Pharmacokinetics by Leon ShargelandAndrewB.C.Yu.
- R8. Handbook of Essential Pharmacokinetics, Pharmacodynamics and DrugMetabolism for Industrial Scientists.
- R9. Robbins &Cortan Pathologic Basis of Disease, 9th Ed. (RobbinsPathology).
- R10. A Complete Textbook of Medical Pharmacology by Dr. S.K Srivastavapublished by APC Avichal Publishing Company.
- R11. KD. Tripathi. Essentials of Medical Pharmacology.
- R12. Principles of Pharmacology. The Pathophysiologic basis of drug Therapyby David E Golan, Armen H, TashjianJr, EhrinJ,Armstrong, April W, Armstrong, Wolters, Kluwer-Lippincott Williams & Wilkins Publishers.

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Demonstrate an understanding of the mechanism of action of different hormones and associated drugs at the cellular and molecular levels.	PO1,PO3,PO5,PO7,PO8						
2	Explain the mechanism of action and adverse drug reactions associated with drugs used in protozoal infections, helminthiasis, and various antimicrobial agents, including antifungal, antiviral, and anti-TB drugs, on a cellular and molecular level as well as analyse the development of drug resistance.	PO1,PO3,PO5,PO7,PO8						
3	Apply knowledge to elucidate the path physiology, therapy, and adverse drug reactions of drugs used in cancer, inflammation, allergy, asthma, and COPD as well as understand the principles behind immunosuppressants and immunostimulants.	PO1,PO3,PO5,PO7,PO8						
4	Analyse the path physiology, therapy, and adverse drug reactions of antiulcer drugs, prokinetics, antiemetic, anti-diarrheal, and drugs for constipation and irritable bowel syndrome and to evaluate the applications of chronotherapy in various diseases.	PO1,PO3,PO5,PO7,PO8						
5	Correlate the role of free radical generation with the aetiology of various diseases and understand the importance of the protective activity of antioxidants as well as synthesize knowledge on recent advances in the treatment of neurodegenerative diseases, cancer, and diabetes.	PO1,PO3,PO5,PO7,PO8						

			SEMESTE	R – II						
Course T	itle	Pharmacological an	d Toxicological Scre	eening N	/leth	ods-II				
Course c	ode	MPL202T	Total credits: 4	L	T	P	S	R	O/F	С
			Total hours: 60T	4	0	0	0	0	0	4
Pre-requi	isite	Nil	Co-requisite					Nil		
Progra		Master of Pharmacy	(Pharmacology)							
Semeste		II semester of first y	ear of the program							
Course	e	After completion of		e to,						
Objectiv	ves	1. Explain the	various types of toxi	city stud	lies.					
		2. Appreciate t	he importance of eth	nical and	l reg	ulatory	requ	iremer	its for tox	icity
		studies.								
		3. Demonstrate	e the practical skills	required	to c	conduct	the p	reclin	ical toxici	ty studies
CO1		Recall the basics of	toxicology as well as	s good la	abor	atory p	ractic	e and	discuss th	e role of
		various toxicology r	egulatory agencies.							
CO2		Apply various guide	lines by regulatory a	gencies	to c	onduct	toxic	ologic	al assays	in oral,
		dermal, and inhalation	on routes.							
CO3		Discuss and apply va	arious guidelines by	regulato	ory a	gencies	s to co	onduct	reproduc	tive and
		genotoxicity assays	in in-vivo and in-vita	ro setups	S.					
CO4		Evaluate various asp								
CO5		Analyze toxic kineti	cs and alternative me	ethods to	o an	imal to	xicity	testin	g and ana	lyze their
		role in drug develop	ment.							
Unit-		Conte	nt	Conta	ct	Le	earni	ng Ou	tcome	KL
No.				Hour	•					
I	Bas	sic definition and typ	es of toxicology			Studen				
		neral, mechanistic, re			Explai	•				
		criptive) Regulatory	12		toxicit	y stuc	lies.			
		ducting toxicity stud							3,4,5	
		A and Schedule Y O							3,1,3	
		od Laboratory Practic	· · · · · · · · · · · · · · · · · · ·							
		cept and its importar	nce in drug							
		elopment								
II		ute, sub-acute and ch	ŕ			Studen				3,4,5
		l inhalational studies	-			knowtl		-		
	-	delines. Acute eye ir				ethical		-	•	
	1	sitization, dermal irri		12		require		s for t	oxicity	
	1	icity studies. Test ite				studies	S.			
	1	racterization- import								
TTT	_	egulatory toxicology				G: 1	11	1 1	1 ,	2.4.5
III	1 ^	productive toxicology				Studen			le to	3,4,5
		roductive toxicity stu				unders			1	
	1 -	roductive studies (segment III), teratogenic	-			Reprod		e toxi	cology	
-		, -	*	12		studies	<b>5.</b>			
	1 '	gment II) Genotoxici st, in vitro and in vivo	· '							
		t, in vitro and in vivo l Chromosomal aberr								
		o carcinogenicity.	anons studies) in							
IV		O enabling studies (II	VD studies)			Studen	4 337:11	he ob	le to	
1 1		finition of IND, impo	· ·						ie to and in	
		ustry perspective, lis		12					that help	3,4,5
		IND submission. Sat							cological	
for		TAD SUUIIIISSIUII. Sal	icty pharmacology			acrine	ше р	11411111	cological	

	studies- origin, concepts and importance of safety pharmacology. Tier1- CVS, CNS and respiratory safety pharmacology, HERAssay. Tier2- GI, renal and other studies.		and toxicological properties of a drug.	
V	Toxic kinetics- Toxic kinetic evaluation in preclinical studies, saturation kinetics Importance and applications of toxic kineticstudies. Alternative methods to animal toxicity testing.	12	Student will be able to know Toxic kinetic evaluation in preclinical studies, saturation kinetics Importance and applications of toxicokinetic studies	3,4,5

- R1. Hand book on GLP, Quality practices for regulated non-clinical research and development (<a href="http://www.who.int/tdr/publications/documents/glp-handbook.pdf">http://www.who.int/tdr/publications/documents/glp-handbook.pdf</a>).
- R2. Schedule Y Guideline: drugs and cosmetics (second amendment) rules, 2005, ministry of health and family welfare (department of health) New Delhi
- R3. Drugs from discovery to approval by Rick NG.
- R4. Animal Models in Toxicology, 3rdEdition, Lower and Bryan
- R5. OECD test guidelines.
- R6. Principles of toxicology by Karen E. Stine, Thomas M. Brown.
- R7. Guidance for Industry M3(R2) Nonclinical Safety Studies for the Conduct of Human Clinical Trials and MarketingAuthorization for Pharmaceuticals (<a href="http://www.fda.gov/downloads/drugs/guidancecomplianceregulatoryinform">http://www.fda.gov/downloads/drugs/guidancecomplianceregulatoryinform</a> ation/guidances/ucm073246.pdf).

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Recall the basics of toxicology as well as good laboratory practice and discuss the role of various toxicology regulatory agencies.	PO1,PO2,PO3,PO4,PO5,PO6,PO8
2	Apply various guidelines by regulatory agencies to conduct toxicological assays in oral, dermal, and inhalation routes.	PO1,PO2,PO3,PO4,PO5,PO6,PO8
3	Discuss and apply various guidelines by regulatory agencies to conduct reproductive and genotoxicity assays in in-vivo and in-vitro setups.	PO1,PO2,PO3,PO4,PO5,PO6,PO8
4	Evaluate various aspects of investigational new drugs and safety pharmacology studies.	PO1,PO2,PO3,PO4,PO5,PO6,PO8
5	Analyse toxicokinetics and alternative methods to animal toxicity testing and analyse their role in drug development.	PO1,PO2,PO3,PO4,PO5,PO6,PO8

			SEMESTE	R – II							
Cours	e Title	Principles of Drug	Discovery								
Cours	e code	MPL203T	Total credits: 4	L	T	P	S	R	O/F	C	
			Total hours: 60T	4	0	0	0	0	0	4	
Pre-re	quisite	Nil	Co-requisite	Nil							
Prog	gram	Master of Pharmac									
Semo	Semester II semester of first year of the program										
Cou			course student is able								
Objec	ctives	_	rious stages of drug di		•			_			
			importance of the rol	_		_		es and	bioinform	atics in	
			Explain various targe		_		-				
		_	s lead seeking method importance of the rol			_		dogia	m in dmia		
		discovery.	importance of the for	e 01 co	при	ter aru	ea arug	, desig	gii iii drug		
CO	<u></u>		scovery process and s	stages in	n the	e drug	discove	erv nro	oram		
CC		_	of Targets, its identific	_		_			_		
	14		g discovery approach.		v all U	iaiIUII,	Protein	SHUC	iures & IIS		
CO	)3	,	leads and protocol to		fy le	ads. its	ontim	izatio	n procedure	es and its	
	. •	_	differentiate lead and		- J 10		. opuin		procedure		
CC	)4		rug designing protoco		licat	ion of	QSAR	in dru	ıg discover	y and	
		lead developments,	its statistical methodo	ology to	val	idate (	QSAR ε	quatio	ons.	•	
CC	)5	Understand rational	drug discovery, phar	macopl	nore	identif	fication	, in si	lico drug sy	nthesis	
		using software'spro	gram and significance	e of pro	drug	g conce	epts.				
Unit-		Conten	t	Conta	ct	L	earnin	g Out	tcome	KL	
No.				Hou	r						
I		erview of modern d	•			Students will be able to					
	_	ss: Target identificat				_	in the b				
		tion, lead identificati				Targe					
	_	ization. Economics		validation, protein structure							
	_	Discovery and valid		12	& its modification in drug discovery approach 4,						
		mics, Proteomics and Bioinformatics.  12 of Nucleic acid microarrays, Protein				uisco	4,5,6				
		rrays, Antisense tec	•								
		nseoligonucleotides,	-								
		ns. Role of transgeni									
	validat	tion	-								
II	Lead l	Identification- com	binatorial chemistry			Stude	nts will	be ab	ole to		
	& high	throughput screening	ng, in silico lead			Under	stand o	levelo	p leads		
		ery techniques, Assa	ay development for			•	rotocol		•		
							_				
			D : .:.	4.6		-				1.5.6	
		-	, Domains, motifs,	12				ifferei	ntiate lead	4,5,6	
		-	-madiation of mustain			and h	its.				
		~									
	crystallography in protein structure prediction										
III			presidential			Stude	nt will	be abl	e to learn		
		ional vs. rational dru	g design, Methods	12					the role of	4,5,6	
		ed in traditional drug	-				-		ug design		
III	hit ider Protei Levels and for Structus structus method crystal Ration Traditi	ntification.  n structure of protein structure lds in protein are. Computational pare: Threading and hads. Application of N lography in protein al Drug Design ional vs. rational dru	, Domains, motifs, prediction of protein omology modeling MR and X-ray structure prediction ag design, Methods	12		leads, proceed and all and his	its optidures a ble to dits.	mizat nd its ifferen be abl	e to learn	4,5,6	

	throughput screening, Concepts of Rational Drug Design, Rational Drug Design Methods: Structure and Pharmacophore based approaches, Virtual Screening techniques: Drug likeness screening, Concept of pharmacophore mapping and pharmacophore-		in drug discovery	
	based Screening			
IV	Molecular docking: Rigid docking, flexible docking, manual docking; Docking based screening. De novo drug design.  Quantitative analysis of Structure Activity Relationship  History and development of QSAR, SAR versus QSAR, Physicochemical parameters, Hansch analysis, Fee Wilson analysis and relationship between them	12	Student will be able to learn Molecular docking & QSAR	4,5,6
V	QSAR Statistical methods – regression analysis, partial least square analysis (PLS) and other multivariate statistical methods. 3D-QSAR approaches like COMFA and COMSIA Pro drug design-Basic concept, Pro drugs to improve patient acceptability, Drug solubility, Drug absorption and distribution, site specific drug delivery and sustained drug action. Rationale of pro drug design and practical consideration of pro drug design	12	Student will be able to understand the importance of the QSAR Statistical methods in drug design	4,5,6

- 1. MouldySioud. Target Discovery and Validation Reviews and Protocols: Volume 2 Emerging Molecular Targetsand Treatment Options. 2007 Humana Press Inc.
- 2. Darryl León. Scott MarkelIn. Silico Technologies in Drug Target Identification and Validation.2006 by Taylor and Francis Group, LLC.
- 3. Johanna K. DiStefano. Disease Gene Identification. Methods and Protocols. Springer New York Dordrecht Heidelberg London.
- R4.Hugo Kubiny. QSAR: Hansch Analysis and Related Approaches. Methods and Principles in Medicinal Chemistry.Publisher Wiley-VCH.
- R5. Klaus Gubernator, Hans-Joachim Böhm. Structure-Based Ligand Design.Methods and Principles in Medicinal Chemistry.Publisher Wiley-VCH.
- R6. Abby L .Parrill.M .Rami Reddy.Rational Drug Design. Novel Methodology and Practical Applications. ACS Symposium Series; American Chemical Society: Washington, DC, 1999.
- R7. J. Rick Turner. New drug development design, methodology and, analysis. John Wiley & Sons, Inc., New Jersey.

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Understand Drug discovery process and stages in the drug	PO1, PO2, PO3, PO4, PO5, PO8
	discovery programme.	
2	Explain the basics of Targets, its identification, validation,	PO1, PO2, PO3, PO4, PO5, PO8
	protein structures & its modification in drug discovery	
	approach.	
3	Understand develop leads and protocol to identify leads, its	PO1, PO2, PO3, PO4, PO5, PO8
	optimization procedures and its sources and able to	
	differentiate lead and hits.	
4	Knowledge about drug designing protocols, application of	PO1, PO2, PO3, PO4, PO5, PO8
	QSAR in drug discovery and lead developments, its statistical	
	methodology to validate QSAR equations.	
5	Understand rational drug discovery, pharmacophore	PO1, PO2, PO3, PO4, PO5, PO8
	identification, in silico drug synthesis using software	
	programmes and significance of pro drug concepts.	

		SEME	STE	R – II					
<b>Course Title</b>	Clinical Resea	rch and Pharmacovigi	lanc	e					
Course code	MPL 204T	Total credits: 4	L	T	P	S	R	O/F	C
		Total hours: 60T	4	0	0	0	0	0	4
<b>Pre-requisite</b>	Nil	Co-requisite				ľ	Nil		
Program	Master of Phan	macy (Pharmacology	)						
Semester	II semester of first year of the program								
Course	1 ^	on of course student is							
Objectives	1								
		51							
	_	the responsibilities of	-	_				ials.	
		e safety monitoring, r	_	_		ut activ	ities.		
	_	the principles of Pha		_					
		new adverse drug rea						. ,	
		n the adverse drug rea	ictioi	ı reportii	ig syste	ms and	commi	inication in	
CO1		acovigilance.	<b>C</b>	4 4:	1;;.	-1 4ui-1			
CO1 CO2		ulatory requirements e types of clinical tria			g clinic	ai irial.			
CO2	1	e types of clinical tria		_	d in ali	nical tr	iola		
CO3		nciple of Pharmacovis	-					<u> </u>	
CO4	_	dverse drug reactions							
C <b>O</b> 3	Pharmacoepide	_	ana	inch asso	SSIIICIII	and un	iderstan	u	
Unit-No.	•	Content	C	ontact	Learning		arning Outcome		KL
				Hour		Leur II	ing out	Come	
I	Regulatory Pe	rspectives of			Stude	nts will	be able	to	
	Clinical Trials	-			Expla	in the r	egulator	ſy	
	Principles of I	-			_		for con		
	Conference or	Harmonization -			clinic	al trial			
	Good Clinical	Practice (ICH-							
	GCP) guidelin	ies							
		ittee: Institutional							
		, Ethical Guidelines		12					4,5,6
		l Research and		12					1,5,0
		pant-Schedule Y,							
	ICMR								
	Informed Con								
	Structure and	content of onsent Process							
	Ethical princip								
	informed cons								
II		: Types and Design			Stude	nts will	be able	e to	
**		Study- RCT and					the type		
	-	servation Study:					designs		
	Cohort, Case	•					J		
	sectional	•							
	Clinical Trial	Study Team		12					4,5,6
		onsibilities of							
	Clinical Trial	Personnel:							
	Investigator, S	Study Coordinator,							
	Sponsor, Cont	ract Research							
	Organization a	and							

	its management			
III	Clinical Trial Documentation-		Students will be able to learn	
	Guidelines to the preparation of		Clinical Trial Documentation	
	documents, Preparation of			
	protocol, Investigator Brochure,			
	Case Report Forms, Clinical			
	Study Report Clinical Trial			
	Monitoring: Safety Monitoring in			
	CTA diverse Drug Reactions:	12		4,5,6
	Definition and types. Detection	12		1,5,0
	and reporting methods. Severity			
	and seriousness assessment.			
	Predictability and preventability			
	assessment, Management of			
	adverse drug reactions;			
	Terminologies of ADR			
IV			Students will be able to learn	
IV	Basic aspects, terminologies and establishment of		Basic aspects, terminologies	
			and establishment of	
	pharmacovigilance History and progress of			
	1		pharmacovigilance	
	pharmacovigilance, Significance			
	of safety monitoring,			
	Pharmacovigilance in India and			
	international aspects, WHO			
	international drug monitoring	12		4,5,6
	program, WHO and Regulatory			
	terminologies of ADR, evaluation			
	of medication safety, Establishing pharmacovigilance centers in			
	Hospitals, Industry and			
	National program related to			
	pharmacovigilance. Roles and			
	responsibilities in			
	1 -			
<b>X</b> 7	Pharmacovigilance		Students will be able to learn	
V	Methods, ADR reporting and			
	tools used in Pharmacovigilance International classification of		Methods, ADR reporting and tools used in	
	diseases, International Non-		Pharmacovigilance	
	proprietary names for drugs,			
	Passive and Active surveillance,			
	Comparative observational			
	studies, Targeted clinical	12		156
	investigations and Vaccine safety surveillance. Spontaneous	12		4,5,6
	_			
	reporting systemand Reporting to			
	regulatory authorities, Guidelines			
	for ADRs reporting. Argus, Aris			
	G Pharmacovigilance, VigiFlow, Statistical			
	Methods for evaluating			
	medication safety data			

VI	Pharmacoepidemiology,		Students will be able to learn	
	Pharmacoeconomics, safety	12	Pharmacoepidemiology,	150
	pharmacology	12	Pharmacoeconomics, safety	4,5,6
			pharmacology	

- R1. Central Drugs Standard Control Organization- Good Clinical Practices, Guidelines for Clinical Trials on Pharmaceutical Products in India. New Delhi: Ministry of Health; 2001.
- R2. International Conference on Harmonization of Technical requirements for registration of Pharmaceuticals for human use. ICH Harmonized Tripartite Guideline. Guideline for Good Clinical Practice.E6; May 1996.
- R3. Ethical Guidelines for Biomedical Research on Human Subjects 2000.Indian Council of Medical Research, New Delhi.
- R4. Textbook of Clinical Trials edited by David Machine, Simon Day and SylvanGreen, March 2005, John Wiley and Sons.
- R5. Clinical Data Management edited by R K Rondels, S A Varley, C F Webbs. Second Edition, Jan 2000, Wiley Publications.
- R6. Handbook of clinical Research. Julia Lloyd and Ann Raven Ed. Churchill Livingstone.
- R7. Principles of Clinical Research edited by Giovanna di Ignacio, Di Giovanna and Haynes.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Explain the regulatory requirements for conducting clinical trial.	PO1. PO2, PO3, PO4, PO5, PO8					
2	Demonstrate the types of clinical trial designs.	PO1. PO2, PO3, PO4, PO5, PO8					
3	Understand responsibilities of key players involved in clinical trials.	PO1. PO2, PO3, PO4, PO5, PO8					
4	Understand principle of Pharmacovigilance and safety monitoring system.	PO1. PO2, PO3, PO4, PO5, PO8					
5	Analyse new adverse drug reactions and their assessment and understand Pharmacoepidemiology.	PO1. PO2, PO3, PO4, PO5, PO8					

		SEMESTER	R – II						
Course Tit	2,						_	T	T
Course cod	le MPL205P	Total credits: 6 Total hours: 12	L	T	P 12	S	R	O/F	C
Duo magnisi	te Nil		0	0	12	0	0 Nil	0	6
Pre-requisi		Co-requisite					NII		
Program Semester									
Course	After completion of c		to know	7					
Objective	_	potency of test sample			ious bi	oassa	v proc	redures	
Objectives		echnique of recording	-						ı rats.
	•	toxicity studies as pre	-						
	development.	•					8	J	
CO1	Examine the effect of		ent in v	tro s	study n	netho	ds.		
CO2	Estimate the effect of	drugs through anima	ıl model	S.					
CO3	Analyze the effect of				nethod	ls.			
CO4	Design and develop A								
CO5	Develop and evaluate			stud	у.				
Unit-No.	Conte		Contac			arnin	g Out	come	KL
	Conte		Hour				g ~ <b></b> •		
I	1. To record the DRC o	f agonist using		5	Studen	ts wil	l be al	ole to	
	suitable isolated tissues			]	Estimate the potency of			cy of	
	2. To study the effects of	of		t	est sar	nples	using	various	
	antagonist/potentiating	ntagonist/potentiating agents on DRC of			oioassa	ıy pro	cedur	es.	
	agonist using suitable is	gonist using suitable isolated tissue			Acquir	e the	techni	que of	
	preparation.	reparation.				-		ressure,	
		. To determine to the strength of			Heart r				
	unknown sample by ma							toxicity	
	by using suitable tissue				studies	_		cal	
	4. To determine to the s	-			evaluat		_	_	
	unknown sample by int	-		(	discove	ery an	id deve	elopment	
	bioassay by using suital	ole tissue							
	preparation	4							
	5. To determine to the s	-							
	unknown sample by bra by using suitable tissue		12						3,4,5,6
	6. To determine to the s		12						3,4,3,0
	unknown sample by mu								
	bioassay by using suital								
	preparation.								
	7. Estimation of PA2 va	lues of various							
	antagonists using suitab								
	preparations.								
8. To study the effe		of various drugs on							
	isolated heart preparation								
	9. Recording of rat BP,	heart rate and							
	ECG.								
	10. Recording of rat EC	G							
	11. Drug absorption stu	dies by averted rat							
	ileum preparation.								
	12. Acute oral toxicity s	studies as per							

-	
	OECD guidelines.
	13. Acute dermal toxicity studies as per
	OECD guidelines.
	14. Repeated dose toxicity studies- Serum
	biochemical, hematological, urine
	analysis, functional observation tests and
	histological studies.
	15. Drug mutagen city study using mice
	bone-marrow chromosomal aberration
	test.
	16. Protocol design for clinical trial.(3
	Nos.)
	17. Design of ADR monitoring protocol.
	18. In-silico docking studies. (2 Nos.)
	19. In-silico pharmacophore based
	screening.
	20. In-silico QSAR studies.
	21. ADR

- R1. Fundamentals of experimental Pharmacology-by M.N.Ghosh.
- R2. Hand book of Experimental Pharmacology-S.K.Kulkarni.
- R3. Text book of in-vitro practical Pharmacology by Ian Kitchen.
- R4. Bioassay Techniques for Drug Development by Atta-ur-Rahman, IqbalChoudhary and William Thomsen.
- R5. Applied biopharmaceutics and Pharmacokinetics by Leon ShargelandAndrewB.C.Yu.
- R6. Handbook of Essential Pharmacokinetics, Pharmacodynamics and DrugMetabolism for Industrial Scientists.

	CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome					
1	Examine the effect of drugs through different in vitro study	PO1. PO2, PO3, PO4, PO5,					
1	methods.	PO6, PO7, PO8					
2	Estimate the effect of drugs through animal models.	PO1. PO2, PO3, PO4, PO5,					
2	Estimate the effect of drugs through animal models.	PO6, PO7, PO8					
3	Analyze the effect of drugs using advance instrument	PO1. PO2, PO3, PO4, PO5,					
3	methods.	PO6, PO7, PO8					
4	Design and develop ADR, clinical trial protocol	PO1. PO2, PO3, PO4, PO5,					
4	Design and develop ADK, chinical trial protocol	PO6, PO7, PO8					
5	Develop and evaluate drug mutagen city, toxicity study.	PO1. PO2, PO3, PO4, PO5,					
3	Develop and evaluate drug initiagen city, toxicity study.	PO6, PO7, PO8					

		SEMESTER -	- III						
Course Title	Research Method	ology and Biostatistic	cs						
Course code	MRM301T	Total credits: 4	L	T	P	S	R	O/F	C
		Total hours: 60T	4	0					4
Pre-requisite	Nil	Co-requisite				]	Nil		
Program	Master of Pharmac	•							
Semester		ond year of the progra							
Course		f the course, the studen							
Objectives		pperation of M.S. Exce	el, SPSS	S, R a	and MI	NITA	AB®, I	DoE (Des	ign of
	1. Experimen								
		various statistical techn	_				_	ems	
CO1		nte statistical technique							
CO1	*	scope, objectives, and			ts of re	searc	h.		
CO2		oncepts of statistical ar							
CO3		nciples of medical rese							
CO4	_	delines for the mainten	nance of	flab	oratory	anim	als an	d design	research
205	work.								
CO5	_	solving practical diffi							
Unit-No.	Cor	ntent	Contac	t	Lea	ırnin	g Out	come	KL
	G 15 1	26.1.1.1	Hour			***	•		
I	General Research	•••			Student			about	
	Research, objectiv	-			asics o				
	practical difficulti			r	methodology				
	literature, study de studies, strategies	12						3,4,5	
	_	ls, randomization,							
	crossover design,								
	techniques.	piaceoo, officing							
II	_	nition, application,		5	Student	s will	learn	about	
	sample size, impo				asics o				
		encing sample size,			pplicat				
	dropouts, statistica				• •				
	significance, type	of significance							
	tests, parametric to	ests(students "t"							
	test, ANOVA, Co	rrelation	12						3,4,5
	coefficient, regres	sion), non-							
	-	vilcoxan rank tests,							
	1 *	ce, correlation, chi							
	square test), null h								
	values, degree of								
***	interpretation of P				Y. 1 .	*11	1	1 .	
III		: History, values in			Student			about	
	medical ethics, au	•		l n	nedical	resea	irch		
	beneficence, non-								
	autonomy and ber		12						3,4,5
	malfeasance, euth		14						3,7,3
		riality, criticisms of							
		ethics, importance							
	of communication	-							
		-,							

	resolution, guidelines, ethics			
	committees, cultural concerns, truth			
	telling, online business practices,			
	conflicts of interest, referral, vendor			
	relationships, treatment of family			
	members, sexual relationships,			
	fatality.			
IV	CPCSEA guidelines for laboratory		Students will learn about	
	animal facility: Goals, veterinary		CPCSEA guidelines	
	care, quarantine, surveillance,			
	diagnosis, treatment and control of			
	disease, personal hygiene, location			
	of animal facilities to laboratories,	12		3,4,5
	anesthesia, euthanasia, physical			
	facilities, environment, animal			
	husbandry, record keeping, SOPs,			
	personnel and training, transport of			
	lab animals.			
V	Declaration of Helsinki: History,		Students will learn about	
	introduction, basic principles for all		medical ethics	
	medical research, and additional	12		3,4,5
	principles for medical research			
	combined with medical care.			

- R1. Pharmaceutical statistics- Practical and clinical applications, Sanford Bolton, publisher Marcel Dekker Inc. New York.
- R2. Fundamental of Statistics Himalaya Publishing House- S.C.Gupta.
- R3. Design and Analysis of Experiments –PHI Learning Private Limited, R. Pannerselvam.
- R4. Design and Analysis of Experiments Wiley Students Edition, Douglas and C. Montgomery.

CO PO Mapping						
SN	Course Outcome (CO)	Mapped Program Outcome				
1	Analyse the value, scope, objectives, and requirements of	PO1,PO3,PO4,PO5,PO6,PO8				
2	research.  Discuss the basic concepts of statistical analysis.	PO1,PO3,PO4,PO5,PO6,PO8				
3	Apply the basic principles of medical research and ethics.	PO1,PO3,PO4,PO5,PO6,PO8				
4	Understand the guidelines for the maintenance of laboratory animals and design research work.	PO1,PO3,PO4,PO5,PO6,PO8				
5	Create efficiency in solving practical difficulties.	PO1,PO3,PO4,PO5,PO6,PO8				

		Semester II	I						
Course Title	Journal Club								
Course code	MRM302NA	Total credits: 1	L	T	P	S	R	O/F	C
		Total hours:	0	0	0	0	0	0	1
Pre-requisite	Nil	Co-requisite				N	Vil		
Program	Master of Pharmacy	<i>I</i>							
Semester		ond year of the program							
Course	1. To teach and	d develop critical appi	aisal s	kills,	increas	se exp	osure to	o rapidly	
Objectives	evolving lite	erature, and help in int	ormed	l clinic	cal pra	ctice.			
	2. To provide	a unique opportunity t	o pron	note ir	nterest	in rese	earch w	hile learn	ing
		s about knowledge gap							
CO1		essential information f					summa	rizing key	/
	• •	gs discussed in the ass							
CO2		ehend the chosen jour				_			
		strating a clear unders							
CO3	1 ** *	sis skills to assess the	•		_	•		_	
		ected journal articles,							ity.
CO4	1 -	size information from	•	•			•	•	
		logies, results, and co	nclusi	ons to	identi	fy patt	erns, tr	ends, and	
	potential areas for fu								
CO5		significance and relev		•					
	•	narmaceutical research	i, cons	iderin	g ethic	cal imp	olicatio	ns, limitat	ions,
	and potential contrib	outions to the field.							

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Retrieve and recall essential information from scientific literature, summarizing key concepts and findings discussed in the assigned journal articles.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
2	Interpret and comprehend the chosen journal articles' methodologies, results, and implications, demonstrating a clear understanding of the research content.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
3	Apply critical analysis skills to assess the experimental design and methodologies employed in the selected journal articles, evaluating their appropriateness and validity.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
4	Analyze and synthesize information from multiple journal articles, comparing and contrasting methodologies, results, and conclusions to identify patterns, trends, and potential areas for further investigation.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
5	Evaluate the overall significance and relevance of the journal articles in the broader context of current pharmaceutical research, considering ethical implications, limitations, and potential contributions to the field.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8

	Semester III								
<b>Course Title</b>	DISCUSSION / PRESENTATION (PROPOSAL PRESENTATION)								
Course code	MRM303NA	Total credits: 2	L	T	P	S	R	O/F	C
		Total hours:	0	0	0	0	0	0	2
Pre-requisite	Nil	Co-requisite		•		N	Vil		•
Program	Master of Pharmacy		•						
Semester	III semester of Second year of the program								
Course	Develop scientific writing skills								
Objectives	2. Enable critic	al thinking ability							
	3. Enhance con	nmunication skills							
	4. Follow ethic	al considerations							
CO1	Identify the research	problem							
CO2	Discuss research pro	blem with team and gu	ide fo	r solut	ion				
CO3	Develops protocol report with an aim and objectives								
CO4	Analyse research problem								
CO5	Develops plan of wo	rk for research project							

CO PO Mapping								
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Identify the research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
2	Discuss research problem with team and guide for solution	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
3	Develops protocol report with an aim and objectives	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
4	Analyse research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
5	Develops plan of work for research project	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						

	Semester III								
<b>Course Title</b>	Research Work	Research Work							
Course code	MRM304NA	<b>Total credits: 14</b>	L	T	P	S	R	O/F	C
		Total hours:	4	0	0	0	0	0	14
<b>Pre-requisite</b>	Nil	Co-requisite				N	Vil		
Program	Master of Pharmacy								
Semester	III semester of Second	d year of the program							
Course	Acquire resear	ch skills							
Objectives	2. Develop scien	tific writing skills							
	3. Enable critical thinking ability								
	4. Adopt application-oriented learning								
	5. Appreciate time management and organizational skills:								
	6. Enhance com	nunication skills							
	7. Follow ethical	considerations							
CO1	Recall key pharmaceu	tical concepts and prin	ciples	pertir	ent to	the M	. Pharn	n project's	
	research focus.								
CO2	Interpret the mechanis	m of action of the sele	cted p	harma	ceutic	al age	nts, der	nonstratin	ıg a
	comprehensive unders	_	_	-					
CO3	Utilize advanced phara	naceutical research tec	hniqu	ies to a	analyz	e drug	formu	lations and	d assess
	their efficacy in practi	cal experiments.							
CO4	Examine and synthesiz	ze experimental data to	draw	infor	med co	onclus	ions ab	out the	
	effectiveness and potential improvements of the formulated drugs.								
CO5	Assess pharmaceutical	research's ethical and	regul	atory o	conside	eration	ıs, ensu	ıring align	ment
	with established guide	lines and principles.							

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Recall key pharmaceutical concepts and principles pertinent to the M. Pharm project's research focus.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
2	Interpret the mechanism of action of the selected pharmaceutical agents, demonstrating a comprehensive understanding of their molecular pathways.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
3	Utilize advanced pharmaceutical research techniques to analyse drug formulations and assess their efficacy in practical experiments.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
4	Examine and synthesize experimental data to draw informed conclusions about the effectiveness and potential improvements of the formulated drugs.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
5	Assess pharmaceutical research's ethical and regulatory considerations, ensuring alignment with established guidelines and principles.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8

		Semester I	V						
<b>Course Title</b>	Journal Club								
Course code	MRM401NA	Total credits: 1	L	T	P	S	R	O/F	С
		Total hours:	0	0	0	0	0	0	1
Pre-requisite	Nil	Co-requisite			•	N	Vil		
Program	Master of Pharmacy		•						
Semester	IV semester of Secon	d year of the program							
Course	1. To teach and	levelop critical apprai	sal ski	lls, in	crease	expos	ure to r	apidly evo	olving
Objectives	literature, and	help in informed clin	ical pra	actice.					
	2. To provide a u	inique opportunity to	promo	te inte	rest in	resear	rch whi	ile learnin	g from
	•	experts about knowledge gaps and future research questions.							
CO1	Retrieve and recall ess	sential information fro	m scie	entific	literati	ure, su	mmariz	zing key c	oncepts
	and findings discussed								
CO2	Interpret and compreh	•				-		s, and	
	implications, demonst								
CO3	Apply critical analysis		•		_			•	ployed
	in the selected journal			_					
CO4	Analyze and synthesiz		_	-			_	-	
contrasting methodologies, results, and conclusions to identify patterns, trend					ids, and po	otential			
	areas for further investigation.								
CO5	Evaluate the overall si	•							ontext
	of current pharmaceut		ring etl	hical i	mplica	tions,	limitati	ions, and	
	potential contributions to the field.								

	CO PO Mapping							
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Retrieve and recall essential information from scientific literature, summarizing key concepts and findings discussed in the assigned journal articles.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
2	Interpret and comprehend the chosen journal articles' methodologies, results, and implications, demonstrating a clear understanding of the research content.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
3	Apply critical analysis skills to assess the experimental design and methodologies employed in the selected journal articles, evaluating their appropriateness and validity.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
4	Analyse and synthesize information from multiple journal articles, comparing and contrasting methodologies, results, and conclusions to identify patterns, trends, and potential areas for further investigation.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
5	Evaluate the overall significance and relevance of the journal articles in the broader context of current pharmaceutical research, considering ethical implications, limitations, and potential contributions to the field.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						

	Semester IV								
<b>Course Title</b>	DISCUSSION / FINAL PRESENTATION								
Course code	se code MRM402NA Total credits: 2 L T P S R O/I					O/F	C		
		Total hours:	0	0	0	0	0	0	2
Pre-requisite	Nil	Co-requisite				N	Vil		
Program	Master of Pharmacy	,							
Semester	IV semester of Second year of the program								
Course	Develop scientific writing skills.								
Objectives	2. Enable critic	al thinking ability.							
	3. Enhance con	nmunication skills.							
	4. Follow ethic	al considerations.							
CO1	Identify the research	problem							
CO2	Discuss research pro	blem with team and go	uide fo	or solu	tion				
CO3	Develops protocol report with an aim and objectives								
CO4	Analyze research problem								
CO5	Develops plan of wo	rk for research project							

CO PO Mapping								
SN	Course Outcome (CO)	Mapped Program Outcome						
1	Identify the research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
2	Discuss research problem with team and guide for	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
	solution							
3	Develops protocol report with an aim and objectives	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
4	Analyse research problem	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						
5	Develops plan of work for research project	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8						

	Semester IV								
<b>Course Title</b>	Research Work and Colloquium								
Course code	MRM403NA	Total credits: 14	L	T	P	S	R	O/F	C
		Total hours:	0	0	0	0	0	0	14
<b>Pre-requisite</b>	Nil	Co-requisite				N	Nil		
Program	Master of Pharmacy								
Semester	IV semester of Secon	d year of the program							
Course	Acquire resear	ch skills							
Objectives	2. Develop scien	tific writing skills							
	3. Enable critical	thinking ability							
	4. Adopt application-oriented learning								
	<ol><li>Appreciate tin</li></ol>	ne management and org	ganiza	ational	skills				
	6. Enhance com	nunication skills							
	7. Follow ethical	considerations							
CO1	Recall key pharmace	utical concepts and p	rincij	ples p	ertine	nt to	the M	. Pharm լ	project's
	research focus.								
CO2	Interpret the mechani	sm of action of the so	electe	d pha	rmace	utical	agents	, demonst	rating a
	comprehensive unders								
CO3	Utilize advanced phara	maceutical research tec	hniqu	ies to	analyz	e drug	g formu	lations an	d assess
	their efficacy in practi-	•							
CO4	Examine and synthesiz	ze experimental data to	draw	infor	med co	onclus	ions ab	out the	
	effectiveness and pote								
CO5	Assess pharmaceutical		regul	atory	consid	eration	ns, ensi	ıring align	ment
	with established guide	lines and principles.							

	CO PO Mapping	
SN	Course Outcome (CO)	Mapped Program Outcome
1	Recall key pharmaceutical concepts and principles pertinent to the M. Pharm project's research focus.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
2	Interpret the mechanism of action of the selected pharmaceutical agents, demonstrating a comprehensive understanding of their molecular pathways.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
3	Utilize advanced pharmaceutical research techniques to analyse drug formulations and assess their efficacy in practical experiments.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
4	Examine and synthesize experimental data to draw informed conclusions about the effectiveness and potential improvements of the formulated drugs.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8
5	Assess pharmaceutical research's ethical and regulatory considerations, ensuring alignment with established guidelines and principles.	PO1,PO2,PO3,PO4,PO5,PO6,PO7,PO8